## Supporting Information for

# Rotamerism-Driven Large Magnitude Host-Guest Binding Change in a Crown Ether Derivatized Pyridinium-Phenolate Series 

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## 1. Materials and General Characterization Methods.

All reagents for synthesis, $\mathrm{N}, \mathrm{N}, \mathrm{N}^{\prime}, \mathrm{N}^{\prime}$-tetramethylguanidine (TMG), potassium thiocyanate (KSCN), tetrabutylammonium tetrafluoroborate $\left(\mathrm{n}-\mathrm{Bu}_{4} \mathrm{NBF}_{4}\right)$ were purchased from Aldrich. The solvents for synthesis were dried and purified by standard procedures. THF were freshly distilled from sodium/benzophenone and DMF was dried over $3 \AA$ molecular sieves. The acetonitrile employed for UV-Vis absorption and electrochemistry was Fluka spectroscopic grade.

All melting points were taken on a Kofler bench. IR spectra $\left(\mathrm{cm}^{-1}\right)$ were recorded on a Bruker Vertex 70 spectrometer. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ) and ${ }^{13} \mathrm{C}$ NMR (100.6 MHz) spectra were measured with a Bruker Avance (Serie 400) spectrometer at 295 K. High resolution MS were measured with an Agilent Technologies 6510 (Q-TOF) Spectrometer using a dual ESI source.

The cyclic voltammetry experiments (using a computer-controlled Radiometer Voltalab 6 potentiostat with a three-electrode single-compartment cell; the working electrode was a platinum disk; a saturated calomel electrode (SCE) used as a reference was placed in a separate compartment) were performed at 300 K , in $\mathrm{N}_{2}$-degassed acetonitrile with a constant concentration ( 0.1 M ) of $\mathrm{n}-\mathrm{Bu}_{4} \mathrm{BF}_{4}$. Ferrocene was used as an internal reference.

The absorption measurements were carried out with a Perkin Elmer Lambda 2 double beam UV-Vis spectrophotometers. The linearity of absorbance at $\lambda_{\text {MAX }}$ as a function of concentration was checked to assert that all chromophores do not aggregate for concentration below 0.1 mM in acetonitrile.

The $\mathrm{pK}_{a}$ values of the chromophores have been determined by UV-Visible absorption methods. The measurements are performed in aqueous medium (acetonitrile / water: 1v / 1v). A large excess of tetramethyl guanidine base $(0.02 \mathrm{M})$ is initially added in the solution containing the chromophore. Aliquots of a concentrated solution of hydrochloric acid ( 0.2 M ) are gradually added to decrease the pH . A typical evolution of the absorption spectrum of T-C as function of the pH is illustrated in the Figure 1.


Figure 1. pH dependence of the absorption spectrum of T-C.
(solvent : ACN/Water 1v/lv)

## 2. Synthesis of the chromophores.

The detailed synthesis and characterization of model chromophores $\mathbf{P}-\mathbf{M}$ and $\mathbf{T}-\mathbf{M}$ have been previously reported in ref. ${ }^{1}$. The synthetic route for the synthesis of crown ether derivatives is depicted in the following scheme:



P-C, T-C
(a) $\mathrm{I}_{2}, \mathrm{Ag}_{2} \mathrm{CO}_{2} \mathrm{CF}_{3}, \mathrm{CHCl}_{3}$, r.t., overnight, $90 \%$. (b) $\mathrm{PdP}\left(\mathrm{Ph}_{3}\right)_{4}, \mathrm{Cs}_{2} \mathrm{CO}_{3}, \mathrm{THF}$, reflux, overnight, $70.5 \%(\mathbf{P}-3)$ and $55.5 \%$ (T-3) . (c) EtSH, NaH, DMF, $100^{\circ} \mathrm{C}$, overnight, $78 \%$ (P-4) and $64 \%$ (T-4). (d) MeI, acetone, reflux, overnight, $95 \%$ (P-5) and $63.5 \%$ (T-5). (e) $\mathrm{Ag}_{2} \mathrm{O}, \mathrm{MeOH}$, r.t., $72 \%(\mathbf{P}-\mathrm{C})$ and $66 \%(\mathbf{T}-\mathbf{C})$.

The intermediate compounds, namely $2^{\prime}$ 'methoxy- 1 ', 3 '-xylyl-18-crown- $5,{ }^{2}$ tetramethyl-1,3,2-dioxaborolan-2-yl)pyridine ${ }^{3}$ and 3,5-dimethyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)pyridine ${ }^{4}$ were prepared using previously reported procedures ${ }^{2-4} . \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ was prepared according to reference ${ }^{5}$ and used directly or within three months at longest while stored under $\mathrm{N}_{2}$ at $-30^{\circ} \mathrm{C}$.

The known 2'-methoxy-1', 3'-xylyl-18-crown-5 was first para iodinated by treatment with iodine in dichloromethane in presence of silver trifluoroacetate (step a in scheme). Then, the following four step, namely the cross-coupling reactions, the oxygen deprotections of biaryls, the quaternizations of pyridine and the deprotonations were performed under the previously described conditions ${ }^{4}$.

## 5-Iodo-2-methoxy-1,3-xylyl-18-crown-5 (1).

To a solution of $100 \mathrm{mg}(0.3 \mathrm{mmol})$ of 2'-methoxy-1',3'-xylyl-18-crown-5 in chloroform ( 3.6 mL ) were added, under Ar , first $\mathrm{AgCO}_{2} \mathrm{CF}_{3}\left(81 \mathrm{mg}, 1.2\right.$ equiv), then, a solution of $\mathrm{I}_{2}$ ( 93.3 $\mathrm{mg}, 1.2$ equiv) in chloroform ( 3.6 mL ). The reaction mixture was stirred at room temperature overnight. The suspended silver salts were then filtered off. The filtrate was successively washed with aqueous saturated solution of $\mathrm{Na}_{2} \mathrm{SO}_{3}$ and water. The organic layer was dried over $\mathrm{MgSO}_{4}$ and evaporated under reduced pressure Purification by column chromatography (cyclohexane/AcOEt 1:9 then AcOEt/MeOH 9:1) afforded pure $\mathbf{1}$ as a colourless solid (125 $\mathrm{mg}, 90 \%$ ). M.P. $96{ }^{\circ} \mathrm{C}^{1}{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=3.38-3.52(\mathrm{~m}, 8 \mathrm{H}), 3.54-3.70(\mathrm{~m}$, 8 H ), $4.11(\mathrm{~s} \mathrm{3H}), 4.50(\mathrm{~s}, 4 \mathrm{H}), 7.58(\mathrm{~s} 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=65.2$ (1C), 68.4 (2C), 68.7 (2C), 70.2 (4C), 70.8 (2C), 86.4 (1C), 134.5 (2C), 140.6 (2C), 159.6 (1C) ppm. IR (KBr): $v=781,848,868,946,990,1024,1098,1223,1241,1268,1349,1429,1469,2867$ $\mathrm{cm}^{-1}$. HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{17} \mathrm{H}_{29} \mathrm{NNO}_{6}\left[\mathrm{M}+\mathrm{NH}_{4}\right]^{+} 470.1034$; found 470.1033 .

## Suzuki-Miyaura cross-coupling reaction.

Under an atmosphere of Ar, to a solution of $\mathbf{1}$ ( 1 equiv.) in anhydrous THF ( 50 mL per mmol of $\mathbf{1}$ ) were successively added, $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ (1.2 equiv.), boronic ester P-2 or T-2 (1.2 equiv.), and $\operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ ( 0.08 equiv.) were. The reaction mixture was refluxed overnight. The suspension was then filtered through Celite with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The solution was concentrated under reduced pressure and the residue was purified by chromatography on silicagel to afford pure $\mathbf{P - 3}$ and T-3.

## 5-(Pyridin-4'-yl)-2-methoxyl-1,3-xylyl-18-crown-5 (P-3).

After purification by chromatography on silicagel (cyclohexane/AcOEt, 1:9 then $\mathrm{AcOEt} / \mathrm{MeOH}, 9: 1$ ), compound $\mathbf{P - 3}(125 \mathrm{mg}, 70 \%)$ was obtained as colourless crystals from 1 ( $200 \mathrm{mg}, 0.4 \mathrm{mmol}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=3.43-3.54(\mathrm{~m}, 8 \mathrm{H})$, 3.57-3.70 (m, $8 \mathrm{H}), 4.20(\mathrm{~s}, 3 \mathrm{H}), 4.65(\mathrm{~s}, 4 \mathrm{H}), 7.59(\mathrm{~s}, 2 \mathrm{H}), 7.64\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=6.0 \mathrm{~Hz}, 2 \mathrm{H}\right), 8.65\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{H}, \mathrm{H}}=6.0\right.$ $\mathrm{Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR (100.6 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=65.3(1 \mathrm{C}), 68.8(2 \mathrm{C}), 69.2(2 \mathrm{C}), 70.2(2 \mathrm{C})$, 70.25 (2C), 70.8 (2C), 121.5 (2C), 130.6 (2C), 132.7(1C), 132.8 (2C), 147.8 (1C), 150.0 (2C), 160.7 (1C) ppm. HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{22} \mathrm{H}_{33} \mathrm{~N}_{2} \mathrm{O}_{6}\left[\mathrm{M}+\mathrm{NH}_{4}\right]^{+} 421.2333$; found 421.2330 .

## 5-[2',6'-Dimethyl-(pyridin-4'-yl)]-2-methoxyl-1,3-xylyl-18-crown-5 (T-3).

After purification by chromatography on silicagel (cyclohexane/AcOEt, 1:9 then AcOEt/MeOH, 9:1), compound T-3 ( $352 \mathrm{mg}, 55.5 \%$ ) was obtained as colourless crystals from $1(667 \mathrm{mg}, 1.5 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.07(\mathrm{~s}, 6 \mathrm{H}), 3.45-3.60(\mathrm{~m}, 8 \mathrm{H})$, 3.60-3.70 (m, 8H), $4.19(\mathrm{~s}, 3 \mathrm{H}), 4.60(\mathrm{~s}, 4 \mathrm{H}), 7.04(\mathrm{~s}, 2 \mathrm{H}), 8.33(\mathrm{sl}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=17.6$ (2C), 65.1 (1C), 68.6 (2C), 69.2 (2C), 70.3 (4C), 70.8 (2C), 131.1 (2C), 131.7 (2C), 132.4 (2C), 132.6 (1C), 148.3 (2C), 148.6 (1C), 158.9 (1C) ppm.HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{24} \mathrm{H}_{37} \mathrm{~N}_{2} \mathrm{O}_{6}\left[\mathrm{M}+\mathrm{NH}_{4}\right]^{+} 449.2646$; found 449.2657.

## Phenols deprotection.

Under an atmosphere of Ar, EtSH (7 equiv.) was added dropwise to NaH (8 equiv.) suspended in DMF ( 14 mL per mmol of $\mathbf{P}-\mathbf{3}$ or T-3). On completion of the $\mathrm{H}_{2}$ emission, $\mathbf{P}-\mathbf{3}$ or T-3 (1 equiv.) was introduced in the reaction mixture. After an overnight stirring at $100{ }^{\circ} \mathrm{C}$, $\mathrm{H}_{2} \mathrm{O}(2.5 \mathrm{~mL}$ per mmol of $\mathbf{P}-3$ or T-3), aqueous $\mathrm{HCl}(1 \mathrm{M}, 8 \mathrm{~mL}$ per mmol of $\mathbf{P}-\mathbf{3}$ or $\mathbf{T}-3)$, and a phosphate buffer $(0.5 \mathrm{M}, \mathrm{pH}=2)$ were successively added. The aqueous layer was extracted three times with $\mathrm{Et}_{2} \mathrm{O}$. The combined organic layers ( $\mathrm{pH}=6.5$ ) were extracted twice with $\mathrm{H}_{2} \mathrm{O}$ and dried over $\mathrm{MgSO}_{4}$. The solution was concentrated under reduced pressure and the residue was purified by chromatography on silicagel (cyclohexane/AcOEt) to afford pure P-4 or T-4.

## 5-(Pyridin-4'-yl)-2-hydroxy-1,3-xylyl-18-crown-5 (P-4).

After purification by chromatography on silicagel (cyclohexane/AcOEt, 1:9), compound P-4 ( $24 \mathrm{mg}, 78 \%$ ) was obtained as colourless crystals from $\mathbf{P}-3(32 \mathrm{mg}, 0.08 \mathrm{mmol}) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=3.63-3.81(\mathrm{~m}, 16 \mathrm{H}), 4.74(\mathrm{~s}, 4 \mathrm{H}), 7.45(\mathrm{~s}, 2 \mathrm{H}), 7.48\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=7.0 \mathrm{~Hz}\right.$, $1 \mathrm{H}), 7.49\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=7.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 8.39(\mathrm{~s}, 1 \mathrm{H}), 8.59\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=7.0 \mathrm{~Hz}, 2 \mathrm{H}\right) \quad$ ppm. ${ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=69.4$ (2C), 70.1 (2C), 70.3 (2C), 70.38 (2C), 70.7 (2C), 121.0 (2C), 125.6 (2C), 128.2 (2C), 129.0 (1C), 147.8 (1C), 150.0 (2C), 157.0 (1C) ppm. HRMS (ESI-QTof) : calcd for $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NNaO}_{6}[\mathrm{M}+\mathrm{Na}]^{+} 412.1731$; found 412.1736 .

## 5-[2',6'-Dimethyl-(pyridin-4'-yl)]-2-hydroxy-1,3-xylyl-18-crown-5 (T-4).

After purification by chromatography on silicagel (cyclohexane/AcOEt, 1:9), compound T-4 (68 mg, $64 \%$ ) was obtained as colourless crystals from T-3 ( $109 \mathrm{mg}, 0.25 \mathrm{mmol}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.90-20.5(\mathrm{sl}, 1 \mathrm{H}), 2.03(\mathrm{~s}, 6 \mathrm{H}), 3.60-3.80(\mathrm{~m}, 16 \mathrm{H}), 4.69(\mathrm{~s}, 4 \mathrm{H})$, $6.87(\mathrm{~s}, 2 \mathrm{H}), 8.25(\mathrm{sl}, 1 \mathrm{H}), 8.31(\mathrm{sl}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $\left.100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=17.5(2 \mathrm{C})$, 69.3 (2C), 70.1 (2C), 70.4 (2C), 70.5 (2C), 70.7 (2C), 125.2 (2C), 128.7 (1C), 129.3 (2C), 129.8 (1C), 131.4 (1C), 148.2 (2C), 149.1 (1C), 155.3 (1C) ppm. IR (KBr): $v=879,1023$, 1160, 1247, 1263, 1354, 1469, 2865, $3363 \mathrm{~cm}^{-1}$. HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{23} \mathrm{H}_{31} \mathrm{NaO}_{6}$ $[\mathrm{M}+\mathrm{Na}]^{+} 440.2044$; found 440.2046 .

## Alkylation of pyridine.

Under an atmosphere of Ar, to a suspension of the biaryl compounds P-4 or T-4 (1 equiv.) in acetone ( 16 mL per mmol of P-4 or T-4), iodomethane (4 equiv.) was added dropwise. The reaction mixture was refluxed overnight. The solvent was then removed under reduced pressure. The residue was washed first with $\mathrm{Et}_{2} \mathrm{O}$ then with AcOEt . The crude iodides $\mathbf{P}-5$ or T-5 were not further purified.

4-[17'-Hydroxy-2', $5^{\prime}, 8^{\prime}, 11$ ', 14 '-pentaoxa<15>metacyclophanyl]pyridinium iodide (P-5).
Compound P-5 (32 mg, $98 \%$ ) was obtained as colourless crystals from P-4 ( $24 \mathrm{mg}, 0.06$ mmol). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=3.69(\mathrm{~s}, 8 \mathrm{H}), 3.73-3.82(\mathrm{~m}, 4 \mathrm{H}), 3.88-3.96(\mathrm{~m}, 4 \mathrm{H})$, $4.29(\mathrm{~s}, 3 \mathrm{H}), 4.77(\mathrm{~s}, 4 \mathrm{H}), 7.86(\mathrm{~s}, 2 \mathrm{H}), 8.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=7.0 \mathrm{~Hz}, 2 \mathrm{H}\right), 8.51\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=7.0 \mathrm{~Hz}\right.$, 2H) ppm. ${ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=47.8$ (1C), 69.8 (4C), 69.9 (2C), 70.4 (2C),
70.5 (2C), 123.5 (2C), 123.7 (2C), 126.3 (1C), 130.2 (2C), 144.7 (2C), 153.8 (1C), 160.4 (1C) ppm. HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{NO}_{6}\left[\mathrm{M}^{*}\right]^{+} 404.2068$; found 404.2067.

## 3,5-Dimethyl-4-[17'-hydroxy-2',5', $8^{\prime}, 11$ ',14'-pentaoxa<15>metacyclophanyl]pyridinium iodide (T-5).

Compound T-5 ( $17 \mathrm{mg}, 63.5 \%$ ) was obtained as colourless hygroscopic crystals from T-4 ( $20 \mathrm{mg}, 0.05 \mathrm{mmol}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=2.29(\mathrm{~s}, 6 \mathrm{H}), 3.55-3.82(\mathrm{~m}, 16 \mathrm{H}), 4.58$ $(\mathrm{s}, 3 \mathrm{H}), 4.72(\mathrm{~s}, 4 \mathrm{H}), 6.91(\mathrm{~s}, 2 \mathrm{H}), 8.41(\mathrm{sl}, 1 \mathrm{H}), 8.84(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100.6 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=18.3$ (2C), 48.4 (1C), 69.5 (2C), $70.0(2 \mathrm{C}), 70.1$ (2C), 70.4 (2C), 70.7 (2C), 124.4 (1C), 126.2 (2C), 128.1 (2C), 138.0 (2C), 142.4 (2C), 157.0 (1C), 159.0 (1C) ppm.

## Deprotonation.

$\mathrm{Ag}_{2} \mathrm{O}$ (2 equiv.) was added to a solution of $\mathbf{P}-\mathbf{5}$ or $\mathbf{T} \mathbf{- 5}$ in MeOH ( 11 mL per mmol of $\mathbf{5 a - b}$ ). After 10 min stirring, excesses of both $\mathrm{Ag}_{2} \mathrm{O}$ and AgI precipitated. The suspension was then centrifugated. The supernatant organic phase was centrifugated once more and was then evaporated under reduced pressure to afford the pure pyridinium phenolates $\mathbf{P}-\mathbf{C}$ or $\mathbf{T} \mathbf{- C}$.

## 4-[20'-(17'-Oxydo)-2', $\left.5^{\prime}, 8^{\prime}, 11^{\prime}, 14^{\prime}-p e n t a o x a<15>m e t a c y c l o p h a n y l\right] p y r i d i n i u m ~(P-C)$.

Compound P-C ( $17.5 \mathrm{mg}, 72 \%$ ) was obtained as red-brown crystals from $\mathbf{P - 5}(32 \mathrm{mg}, 0.06$ $\mathrm{mmol})$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta=3.38-3.52(\mathrm{~m}, 8 \mathrm{H}), 3.57-3.64(\mathrm{~m}, 4 \mathrm{H}), 3.70-3.76$ $(\mathrm{s}, 4 \mathrm{H}), 4.04(\mathrm{~s}, 3 \mathrm{H}), 4.58(\mathrm{~s}, 4 \mathrm{H}), 7.75(\mathrm{~s}, 2 \mathrm{H}), 7.83\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=7.0 \mathrm{~Hz}, 2 \mathrm{H}\right), 8.16\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}, \mathrm{H}}=\right.$ $7.0 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta=46.3$ (1C), 70.3 (4C), 71.2 (2C), 71.5 (2C), 71.6 (2C), 114.5 (1C), 120.0 (2C), 130.4 (2C), 131.9 (2C), 144.2 (2C), 156.1 (1C), 176.6 (1C) ppm. IR (KBr): $v=495,529,827,1017,1074,1090,1103,1186,1312,1381$, 1454, 1482, 1590, 2862, 3272, 3272, $3499 \mathrm{~cm}^{-1}$. HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{NNaO}_{6}[\mathrm{M}+\mathrm{Na}]+426.1887$; found 426.1885 .

## 3,5-Dimethyl-4-[20'-(17'-oxydoy)-2',5', $8^{\prime}, 11$ ',14'-pentaoxa<15>metacyclophanyllpyridinium (T-C).

Compound T-C ( $10 \mathrm{mg}, 65 \%$ ) was obtained as red crystals from T-5 ( $20 \mathrm{mg}, 0.03 \mathrm{mmol}$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta=2.33$ (s, 6H), 3.49 (s, 8H), 3.53-3.63 (m, 4H), 3.64-3.78 (m, $4 \mathrm{H}), 4.25(\mathrm{~s}, 3 \mathrm{H}), 4.63(\mathrm{~s}, 4 \mathrm{H}), 6.97(\mathrm{~s}, 2 \mathrm{H}), 8.51(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 100.6 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right): \delta=18.8$ (2C), 47.7 (1C), $70.0(2 \mathrm{C}), 70.4$ (2C), 71.5 (4C), 71.6 (2C), 118.9 (1C),
128.6 (2C), 131.3 (2C), 138.6 (2C), 143.7 (2C), 161.5 (1C), 167.2 (1C) ppm. IR (KBr): $v \square=$ 1015, 1106, 1178, 1307, 1335, 1354, 1415, 1468, 1596, 1641, 2868, $3487 \mathrm{~cm}^{-1}$. HRMS (ESI-Q-Tof) : calcd for $\mathrm{C}_{24} \mathrm{H}_{34} \mathrm{NO}_{6}[\mathrm{M}+\mathrm{H}]^{+} 432.2380$; found 432.2382 .

## 3. Cyclic voltamogramms of the chromophores.



Figure S1. Cyclic voltamogramms of chromophores in acetonitrile $+\left(\mathrm{nBu}_{4} \mathrm{NBF}_{4}(0.1 \mathrm{M})\right.$ on platinum electrode at $100 \mathrm{mV} \mathrm{s}^{-1}$ (concentration of chromophores: $10^{-3} \mathrm{M}$ ).
4. Effects of the complexation with $\mathrm{K}^{+}$on the CVs of the ligands.


Figure S2. Experimental CVs for acetonitrile solutions of ligands in presence of KSCN. ( $\mathrm{v}=100 \mathrm{mV} \mathrm{s}^{-1}$ and concentration of ligands : $10^{-3} \mathrm{M}$ ).

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