

Supporting Information

Small molecule-mediated duplex formation of nucleic acids with 'incompatible' backbones

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Materials and Methods

Materials HPLC purified DNA and RNA oligonucleotides were purchased from Integrated DNA Technologies (Coralville, IA). 2',5'-RNA phosphoramidites were purchased from ChemGenes (Wilmington, MA). UNA phosphoramidites were purchased from Link Technologies (Bellshill, Scotland). 2',5'-RNA and UNA were synthesized on an automated Expedite synthesizer using standard phosphoramidite chemistry with modified coupling times (30 min for 2',5'-RNA and 5 min for UNA). Post synthesis, 2',5'-RNA and UNA oligonucleotides were deprotected using standard protocols, purified by IE-HPLC (Dionex DNAPac PA-100), and desalted on a Sep-Pak Plus C18 cartridge. isoGNA phosphoramidite and oligonucleotide synthesis was performed as previously reported by Karri et. al. (ref. 10). Proflavine hemisulfate (Sigma) was used as received and dissolved in dH₂O. The extinction coefficient used to determine stock solution concentration was $\epsilon_{444} = 38\,900\text{ M}^{-1}\text{ cm}^{-1}$.

Thermal denaturation studies UV-vis spectra were acquired on a Cary 50 UV-Vis spectrophotometer equipped with a LC 600 Quantum Northwest Peltier temperature controller. Solutions contained 5 μM of each oligonucleotide for non-self-complementary sequences or 10 μM of oligonucleotide for self-complementary sequences (90 μM bp), and 45 μM proflavine when noted, all solutions contained 1 M NaCl (or 100 mM NaCl when noted), 0.1 mM EDTA, and 10 mM Na₂HPO₄ (pH 7.0). Spectra were measured using a 2 mm quartz cuvettes unless otherwise noted. For each experiment, two heating and two cooling temperature ramps (of 0.4 $^{\circ}\text{C min}^{-1}$) were performed from 0 $^{\circ}\text{C}$ to 65 $^{\circ}\text{C}$, or from 0 $^{\circ}\text{C}$ to 80 $^{\circ}\text{C}$, and the absorbance intensity from 190-600 nm was recorded. Thermal melting profiles were produced from the resulting spectra by plotting A_{260 nm} (baseline corrected) as a function of temperature. The heating and cooling T_m for each duplex was determined by taking the first derivative of the UV melting profiles. The amount of change in UV absorbance observed upon duplex melting (i.e., hyperchromicity) for the different duplexes of this study is, as expected, variable given that both the degree of base stacking in the duplex state and the degree of intra-strand base stacking in the single stranded state, which will be particular to each pairing system, contribute to the hyperchromicity associated with duplex melting. Thermal melting profiles were also collected for each individual nucleic acid strand of this study in the presence of proflavine. In some cases a cooperative transition was observed, indicating a self-structure of the nucleic acid with proflavine. However, when the complementary strand was present in the same sample the cooperative melting transition of the duplex was always observed at a higher temperature than the self-structure of the individual strand. Additionally, differences in the longest wavelength absorption bands of proflavine in the presence of the individual strands versus the complementary strands also indicated duplex formation.

Dilution studies Dilution studies were performed on an Agilent 8453 UV-vis spectrophotometer equipped with an Agilent 89090A Peltier temperature controller. Typically, samples initially contained 5 μM of each oligonucleotide (90 μM bp), 45 μM proflavine, and buffer solution (1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0). Dilutions were performed at 4 $^{\circ}\text{C}$ by adding the buffer solution incrementally to each sample. The UV-vis absorption profile was acquired after each addition of buffer. In order to maintain an absorbance between 1.2 and 0.1 AUs, quartz cuvettes from 1 to 10 cm were employed. Spectra were normalized and the change in proflavine absorbance, which shifted upon dissociation of proflavine from the oligonucleotide duplex, was used to determine proflavine association constants (K_a).

Circular Dichroism CD spectra were acquired on a JASCO J-810 CD spectropolarimeter equipped with a NESLAB temperature controller. CD spectra (220 to 600 nm) were measured at either 4 $^{\circ}\text{C}$ or 20 $^{\circ}\text{C}$, as indicated, using a 1 cm path-length quartz cuvette.

Supplementary Figures

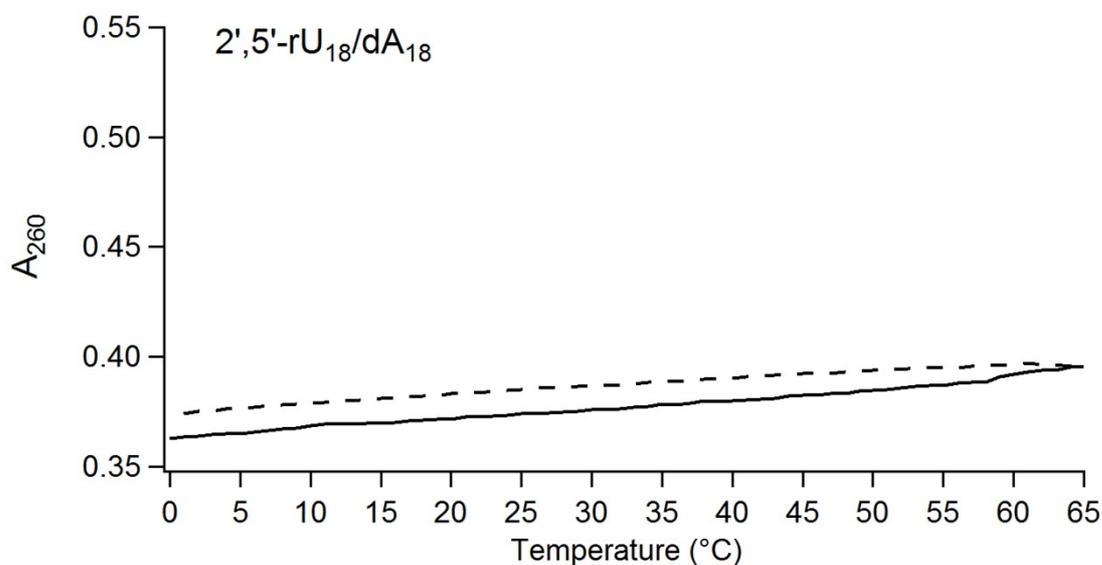


Fig. S1 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ and dA₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dotted line. The solution contained 0.1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

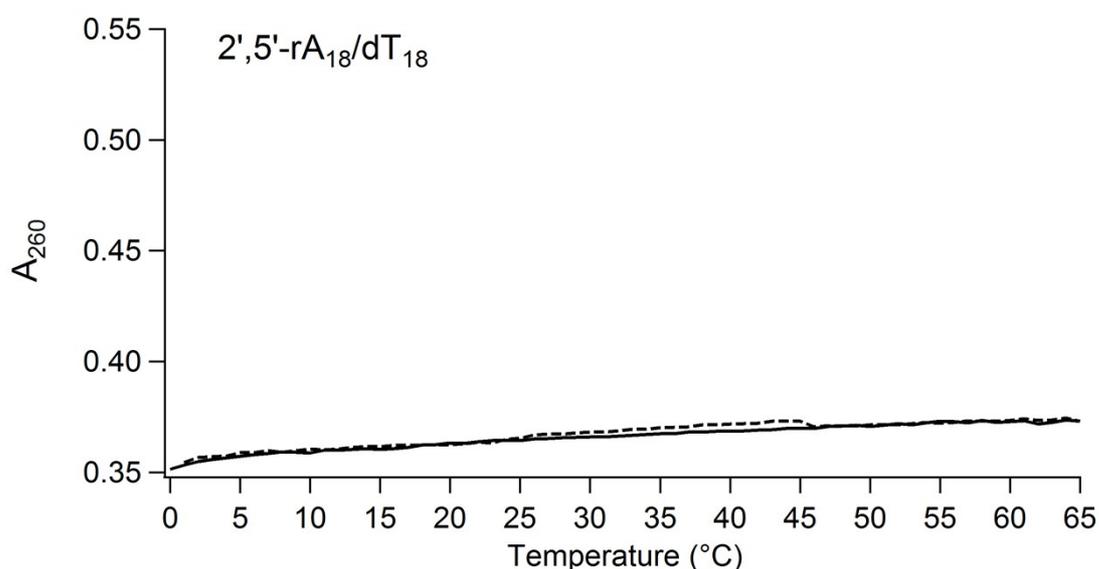


Fig. S2 UV-monitored thermal profiles of a solution containing 2',5'-rA₁₈ and dT₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dotted line. Solution contained 0.1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

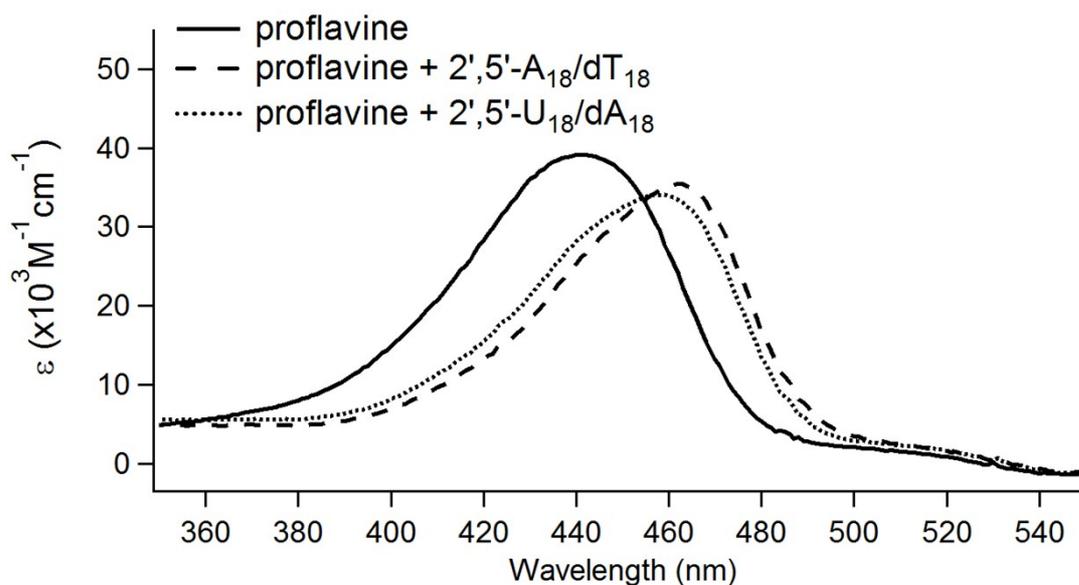


Fig. S3 UV spectra of proflavine in the absence (solid line) and in the presence of 2',5'-rU₁₈ and dA₁₈ (dotted line), or 2',5'-rA₁₈ and dT₁₈ (dashed line) at 4°C. Each solution was 45 μM in proflavine and 5 μM in each oligonucleotide (90 μM in base pair) for the solution containing 2',5'-rU₁₈ and dA₁₈. Solutions contained 0.1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

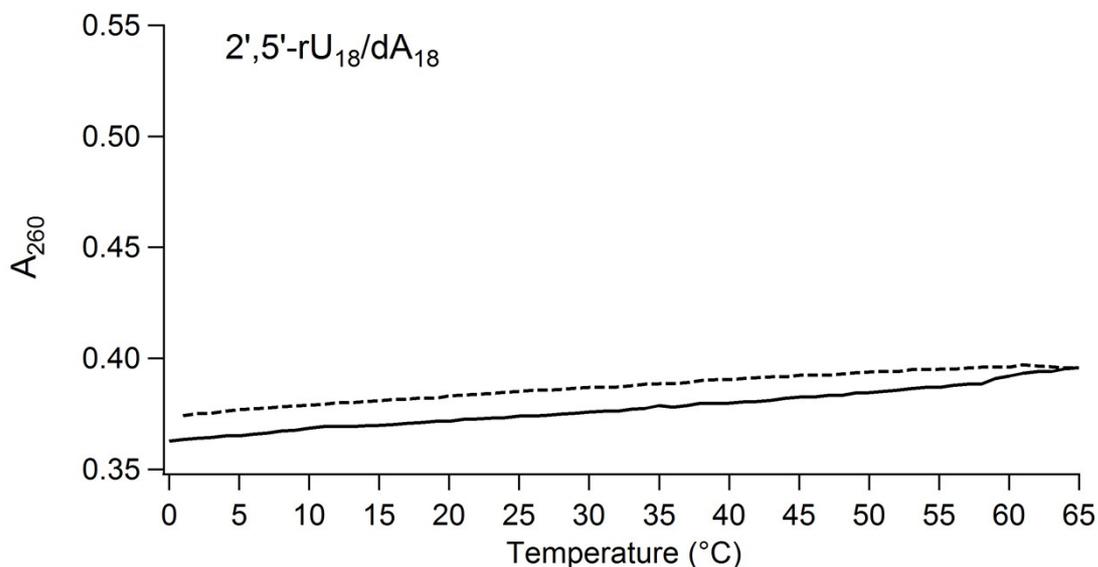


Fig. S4 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ and dA₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. The solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

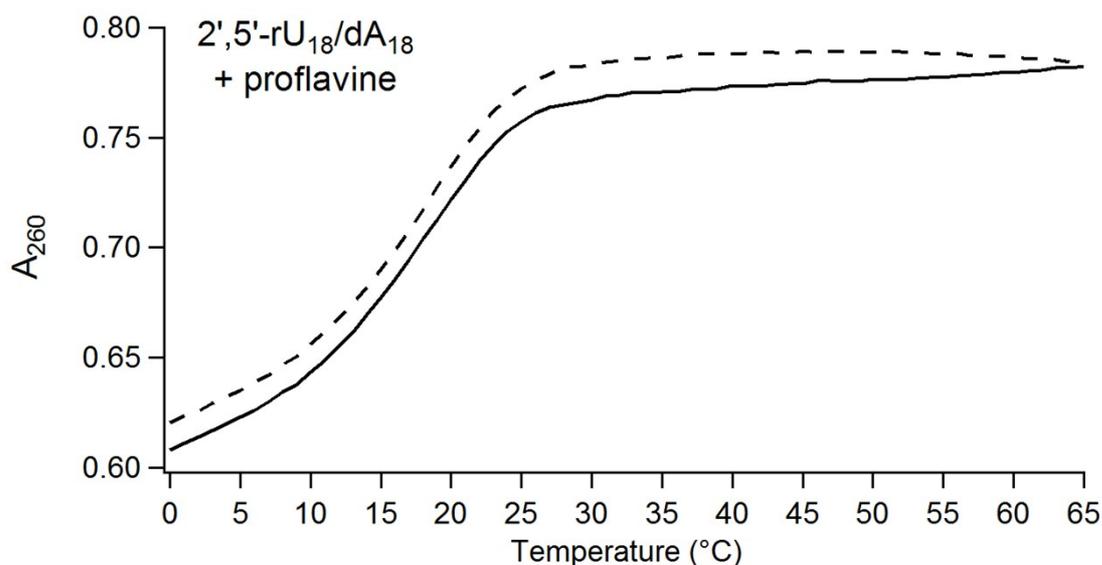


Fig. S5 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ and dA₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

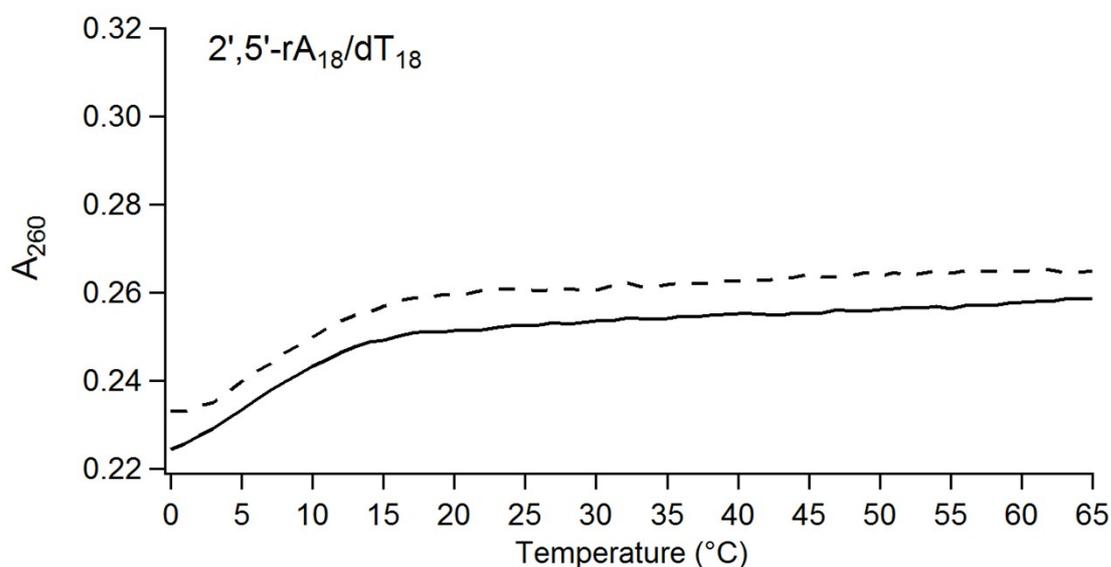


Fig. S6 UV-monitored thermal profiles of a solution containing 2',5'-rA₁₈ and dT₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0. Melt was performed in a 1 mm path length cell.

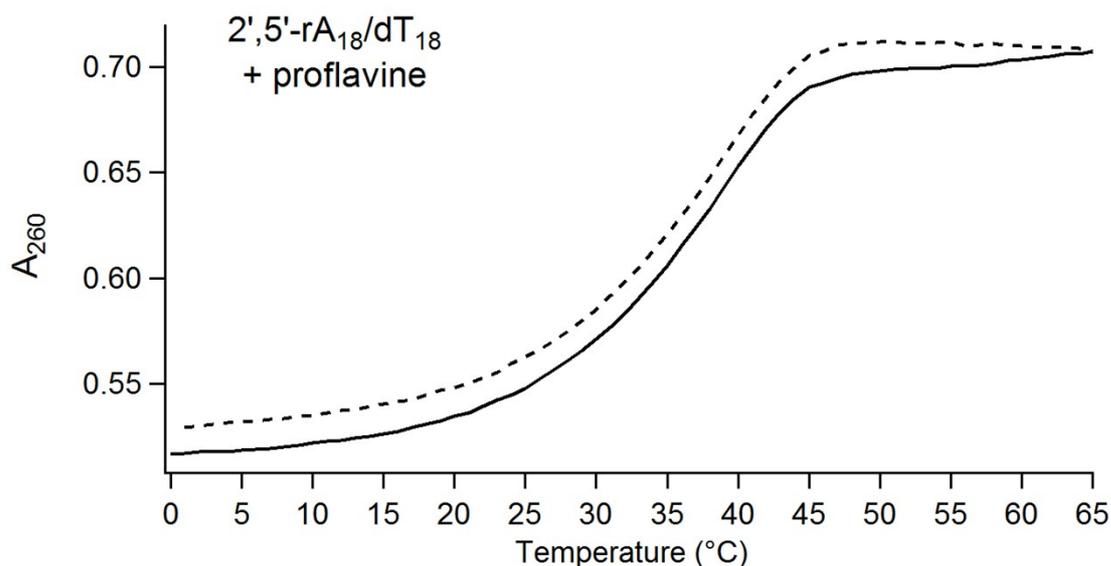


Fig. S7 UV-monitored thermal profiles of a solution containing 2',5'-rA₁₈ and dT₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

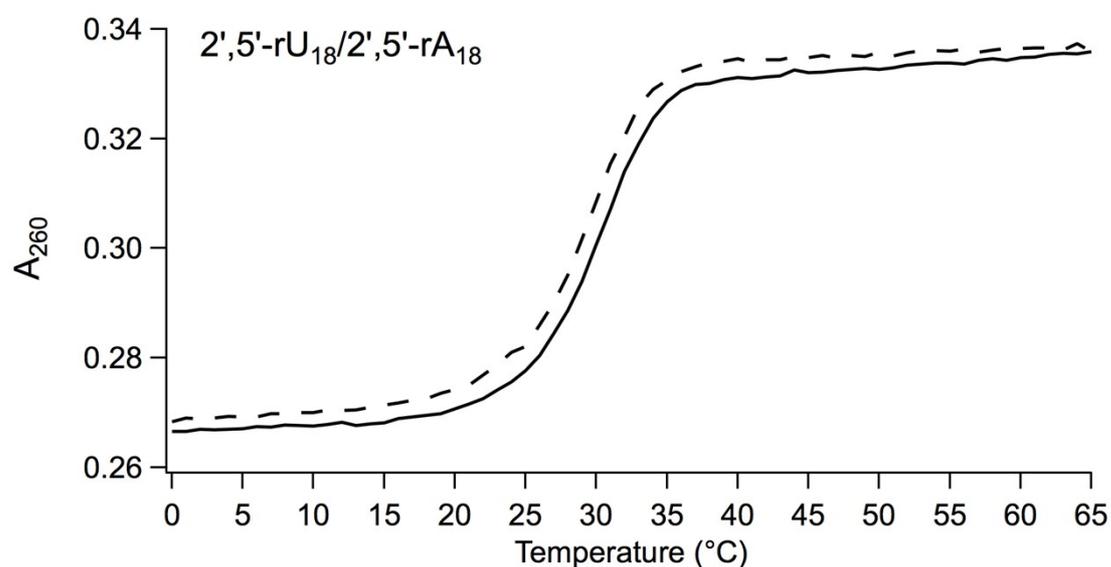


Fig. S8 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ and 2',5'-rA₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

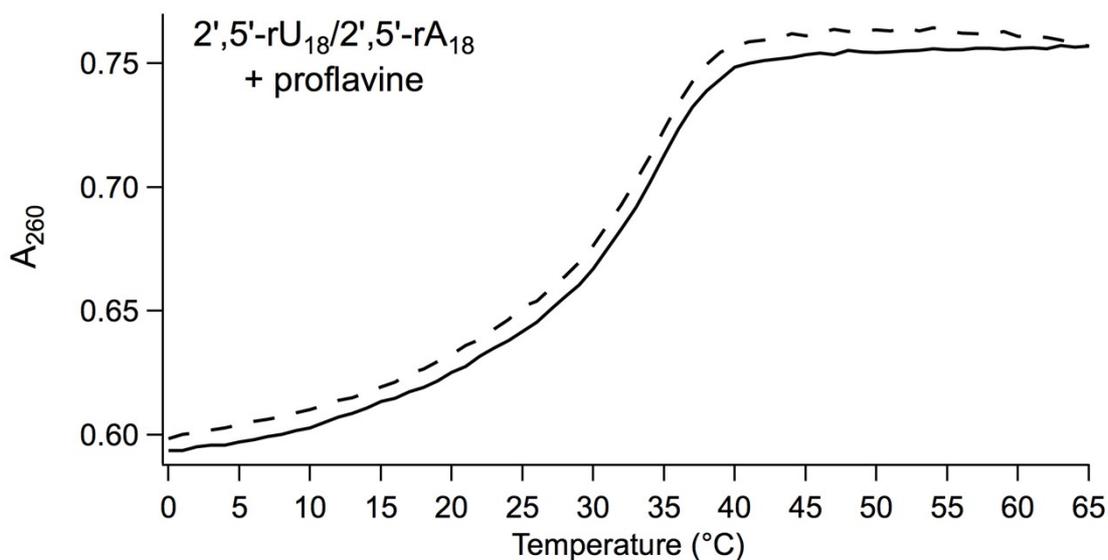


Fig. S9 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈, 2',5'-rA₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

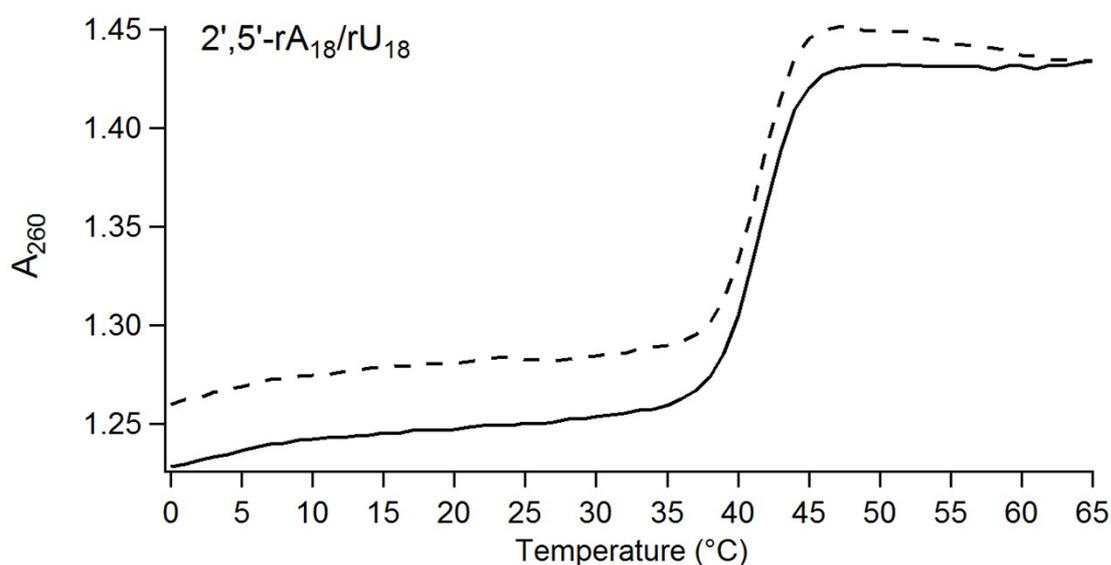


Fig. S10 UV-monitored thermal profiles of a solution containing 2',5'-rA₁₈ and rU₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

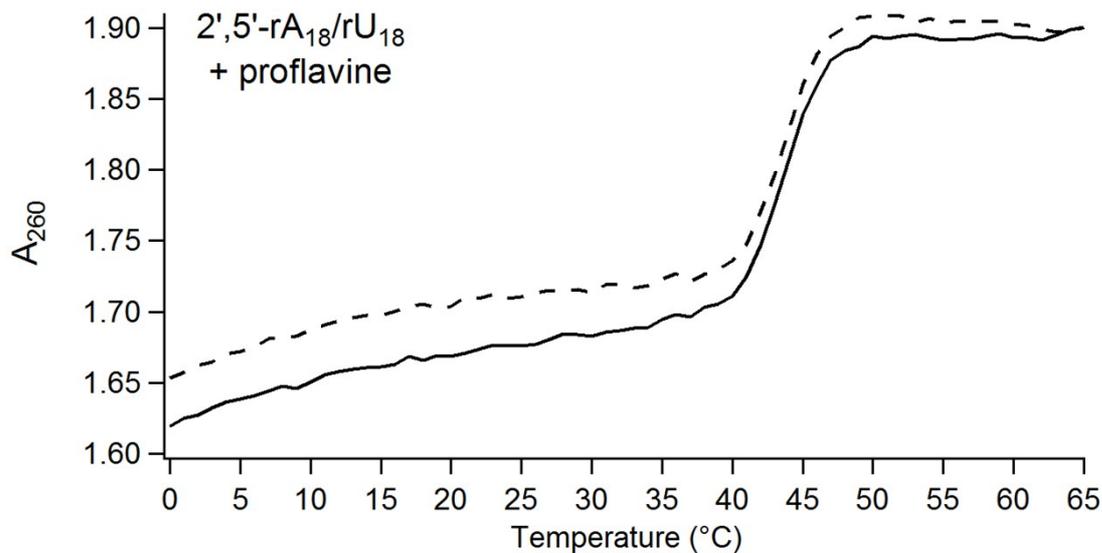


Fig. S11 UV-monitored thermal profiles of a solution containing 2',5'-rA₁₈ and rU₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

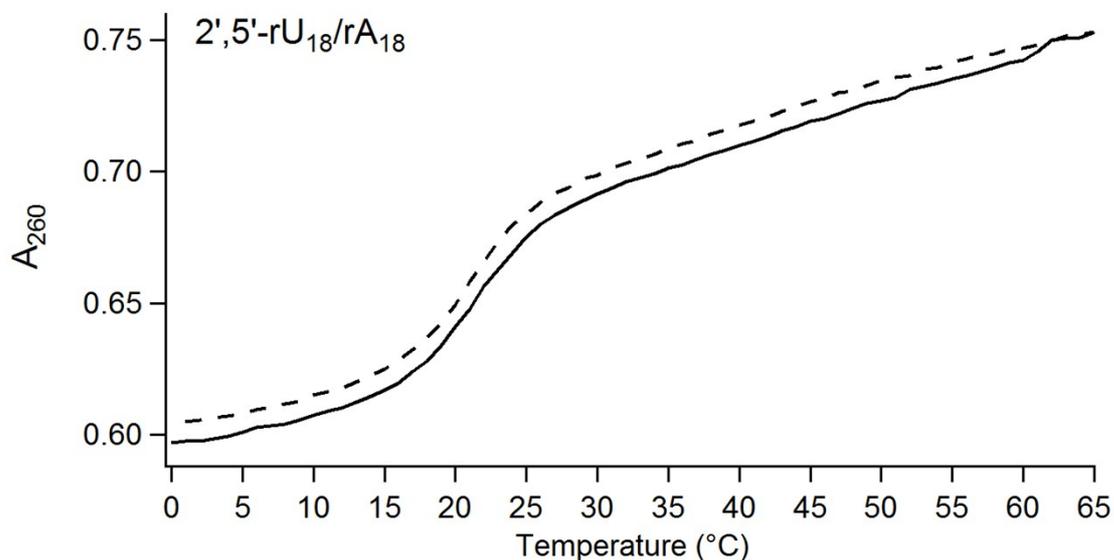


Fig. S12 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ and rA₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

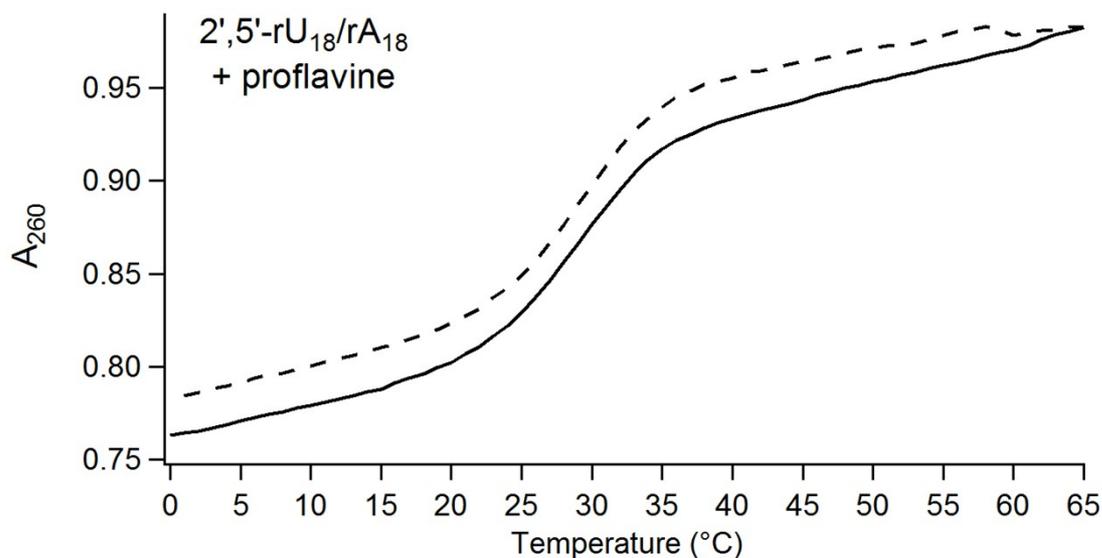


Fig. S13 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ and rA₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

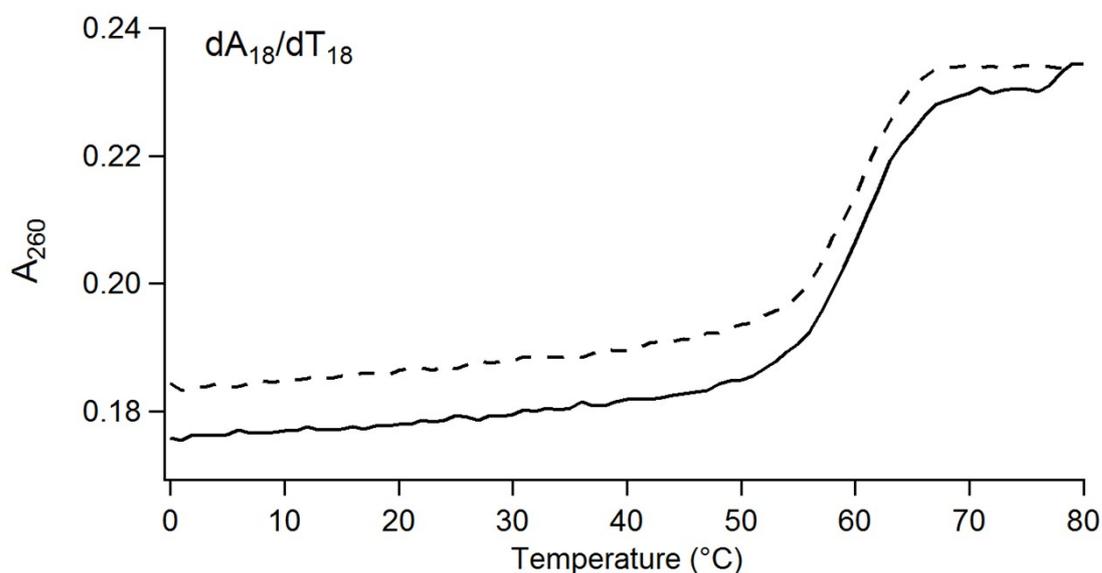


Fig. S14 UV-monitored thermal profiles of a solution containing dA₁₈ and dT₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0. Melt was performed in a 1 mm path length cell.

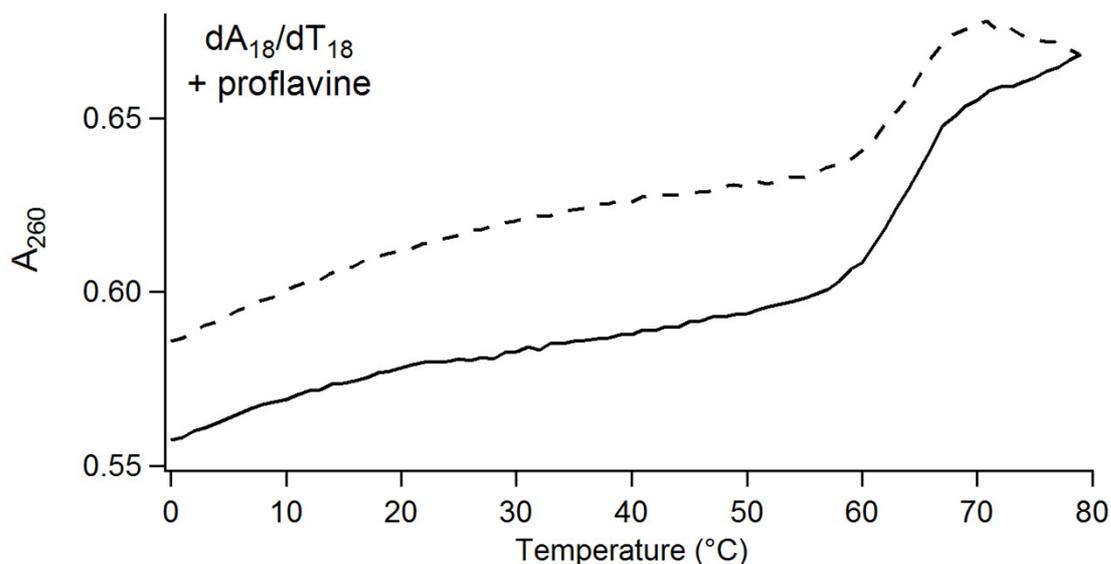


Fig. S15 UV-monitored thermal profiles of a solution containing dA₁₈ and dT₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0. Melt was performed in a 1 mm path length cell.

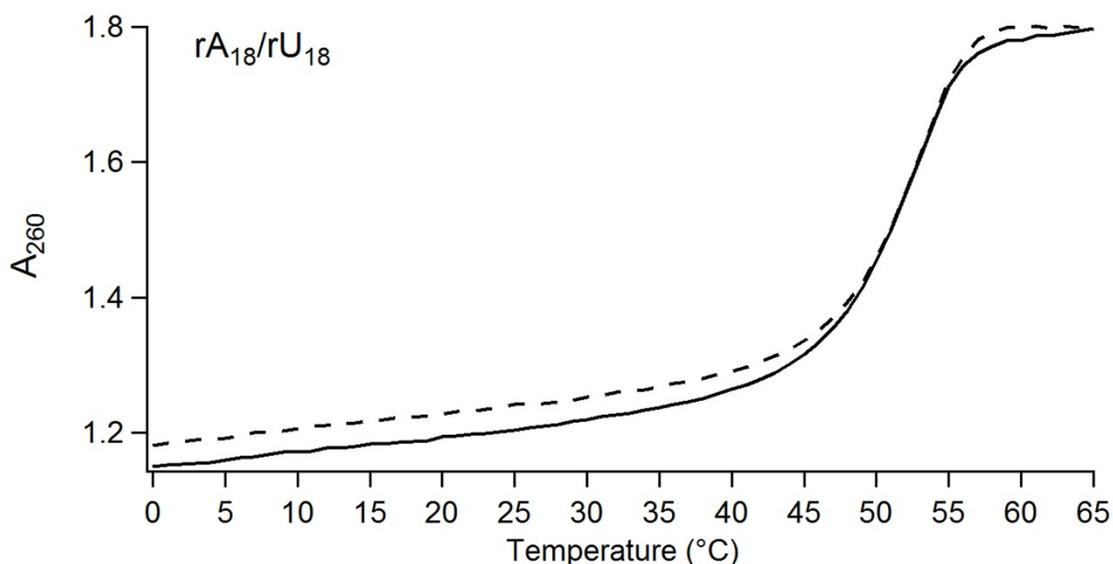


Fig. S16 UV-monitored thermal profiles of a solution containing rA₁₈ and rU₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

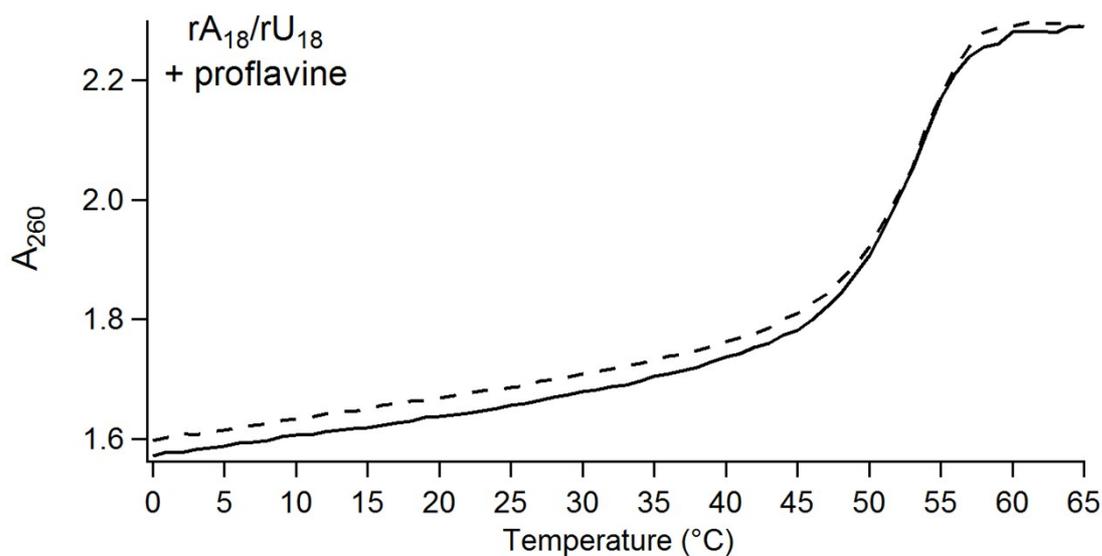


Fig. S17 UV-monitored thermal profiles of a solution containing rA₁₈ and rU₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

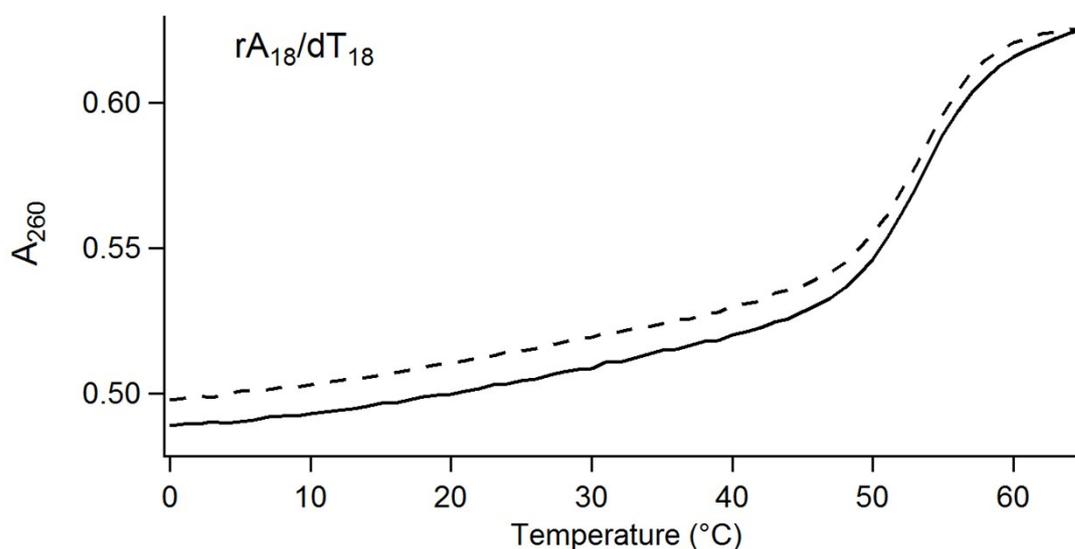


Fig. S18 UV-monitored thermal profiles of a solution containing rA₁₈ and dT₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

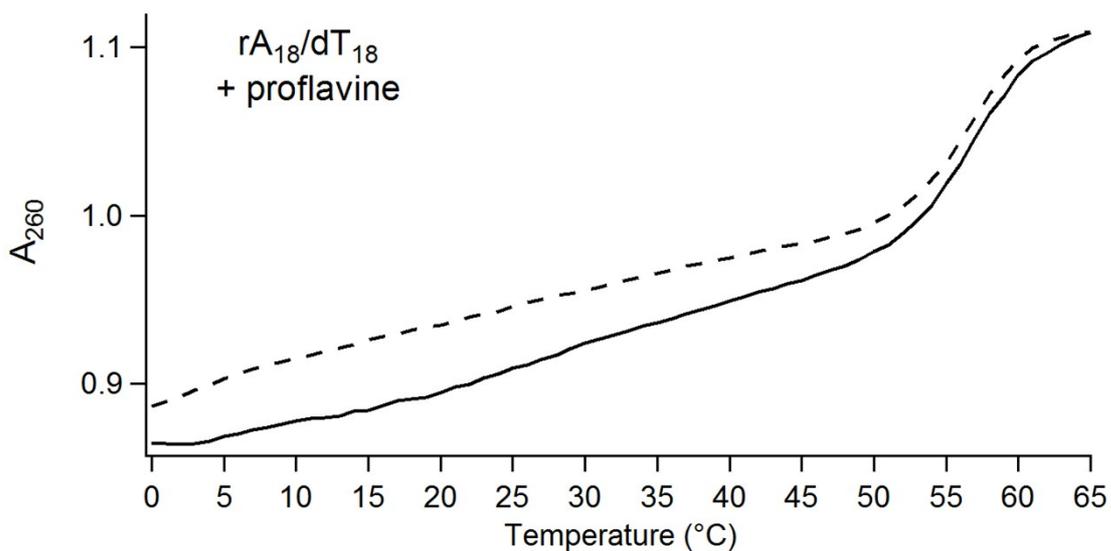


Fig. S19 UV-monitored thermal profiles of a solution containing rA₁₈ and dT₁₈ (5 μM in each oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

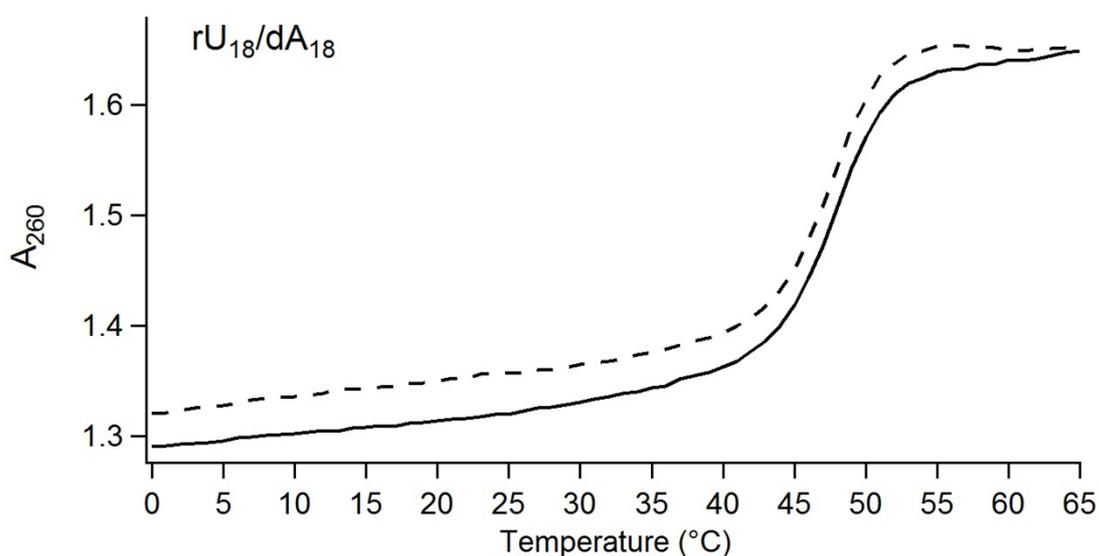


Fig. S20 UV-monitored thermal profiles of a solution containing rU₁₈ and dA₁₈ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

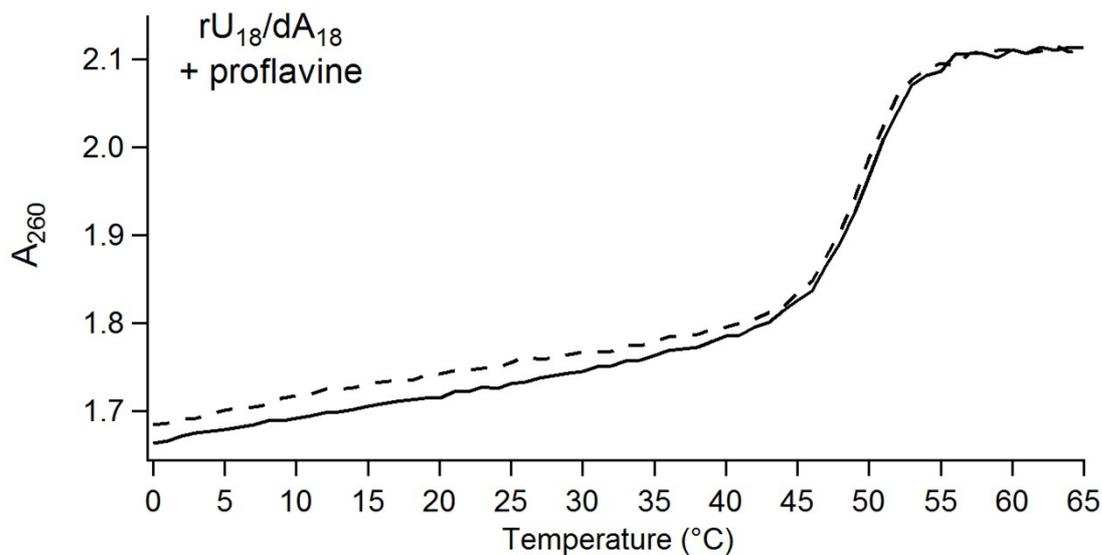


Fig. S21 UV-monitored thermal profiles of a solution containing rU₁₈ and dA₁₈ (5 μ M in each oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

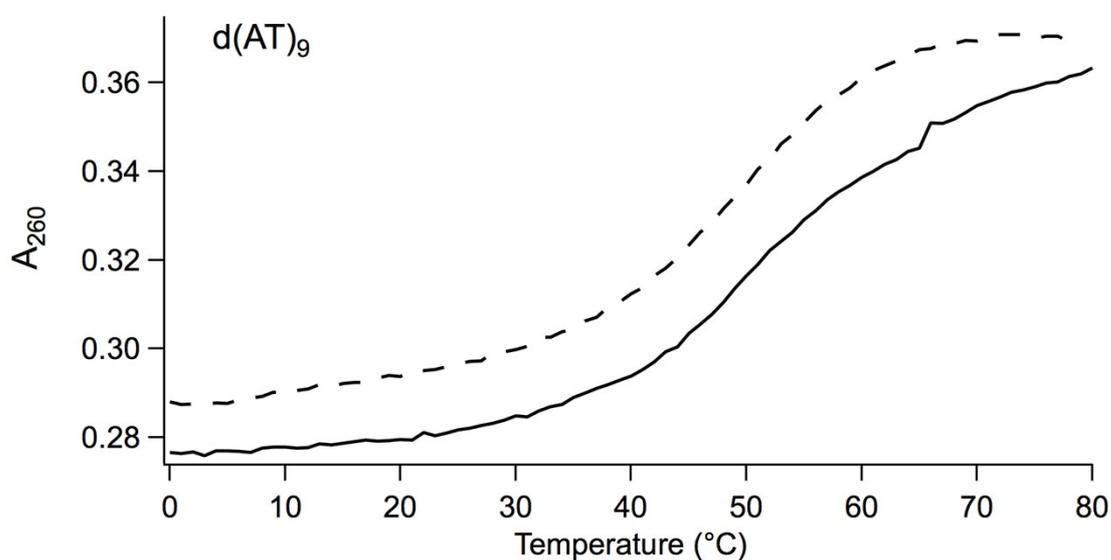


Fig. S22 UV-monitored thermal profiles of a solution containing d(AT)₉ (10 μ M in oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0. Melt was performed in a 1 mm path length cell.

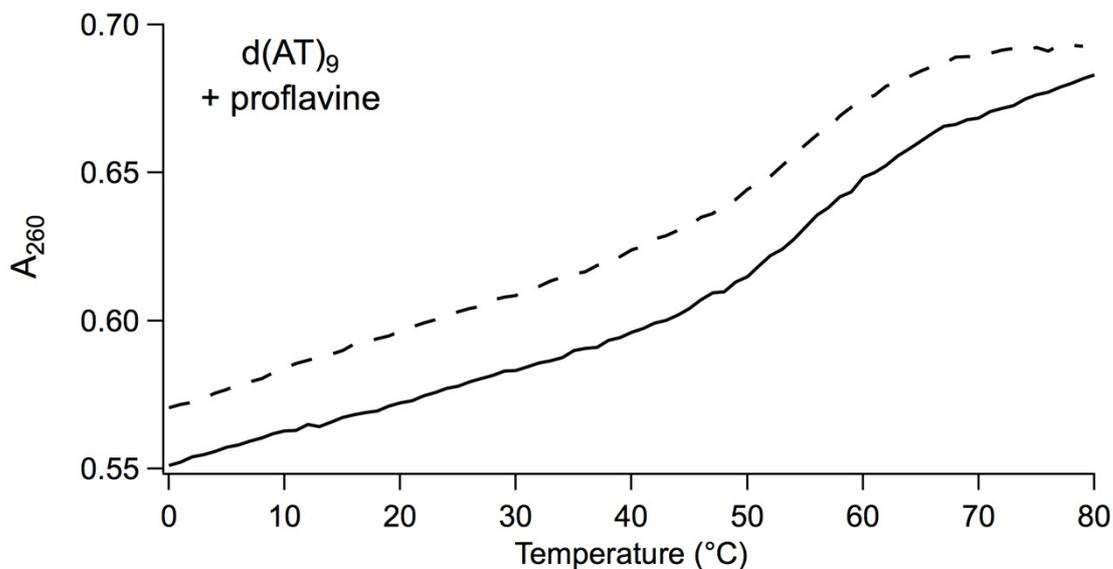


Fig. S23 UV-monitored thermal profiles of a solution containing d(AT)₉ (10 μM in oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0. Melt was performed in a 1 mm path length cell.

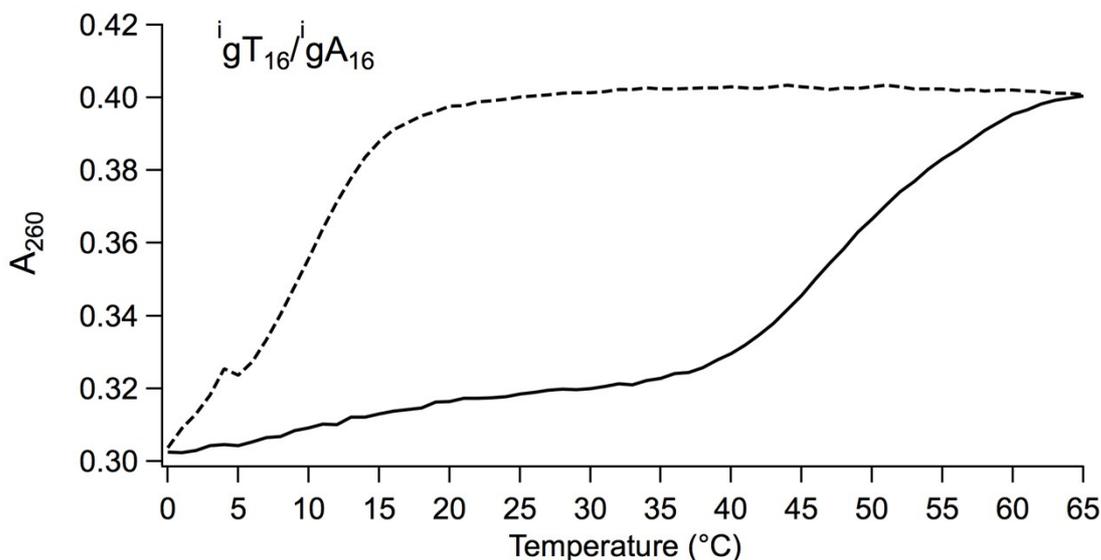


Fig. S24 UV-monitored thermal profiles of a solution containing *i*₉T₁₆ and *i*₉A₁₆ (5 μM in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

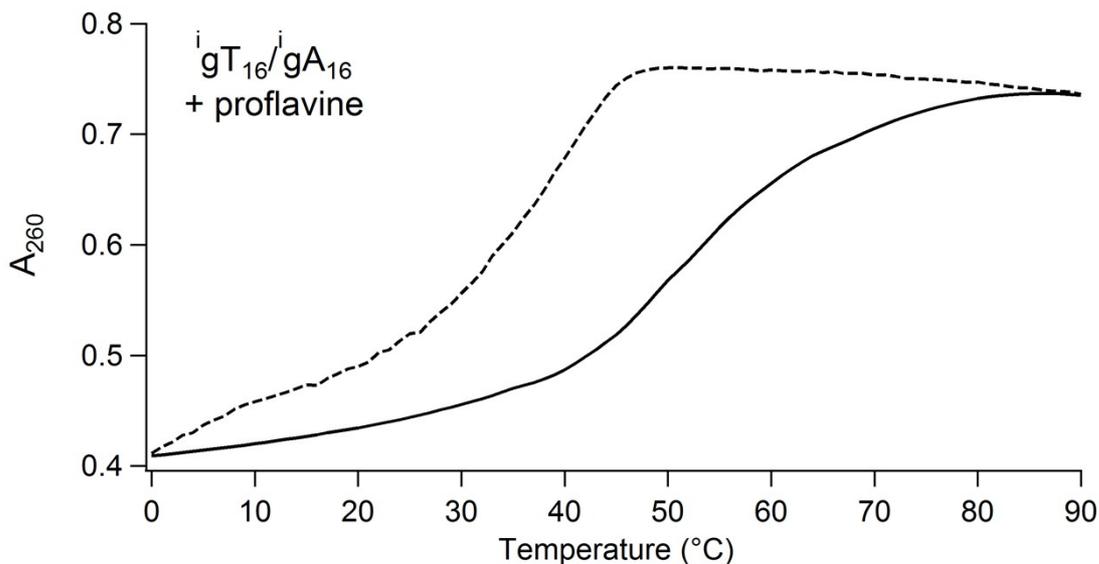


Fig. S25 UV-monitored thermal profiles of a solution containing i gT₁₆, i gA₁₆ (5 μ M in each oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

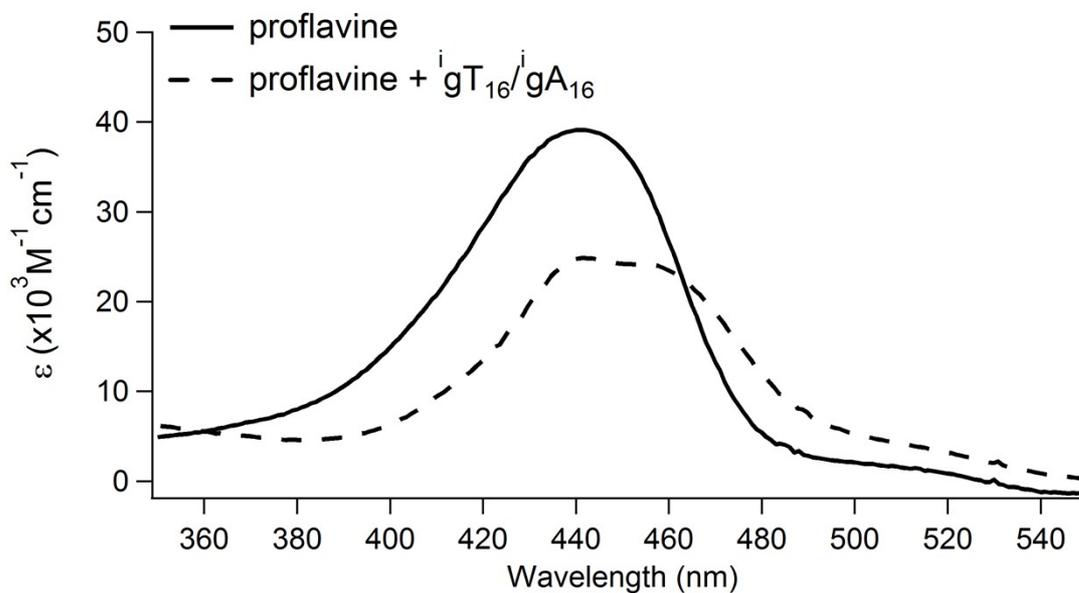


Fig. S26 UV spectra of proflavine in the absence (solid line) and in the presence of i gT₁₆ and i gA₁₆ (dashed line) at 4 °C. Each solution was 45 μ M in proflavine and 5 μ M in each oligonucleotide (90 μ M in base pair) for the solution containing 2',5'-rU₁₈ and dA₁₈. Solutions contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

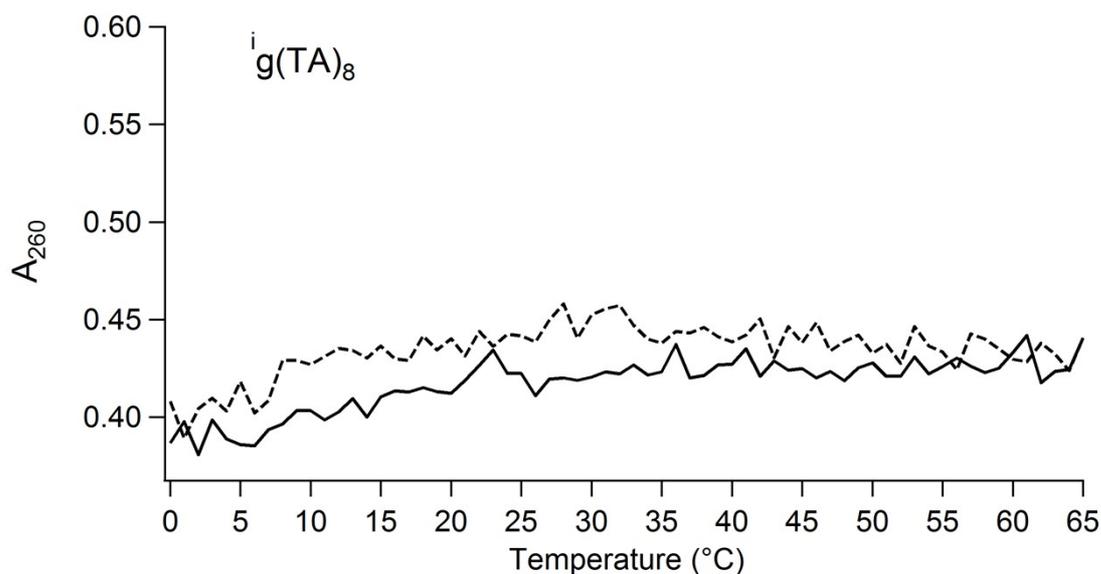


Fig. S27 UV-monitored thermal profiles of a solution containing ${}^i\text{g}(\text{TA})_8$ ($10\ \mu\text{M}$ in oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. The solution contained $1\ \text{M}\ \text{NaCl}$, $10\ \text{mM}\ \text{Na}_2\text{HPO}_4$, $0.1\ \text{mM}\ \text{EDTA}$, $\text{pH}\ 7.0$.

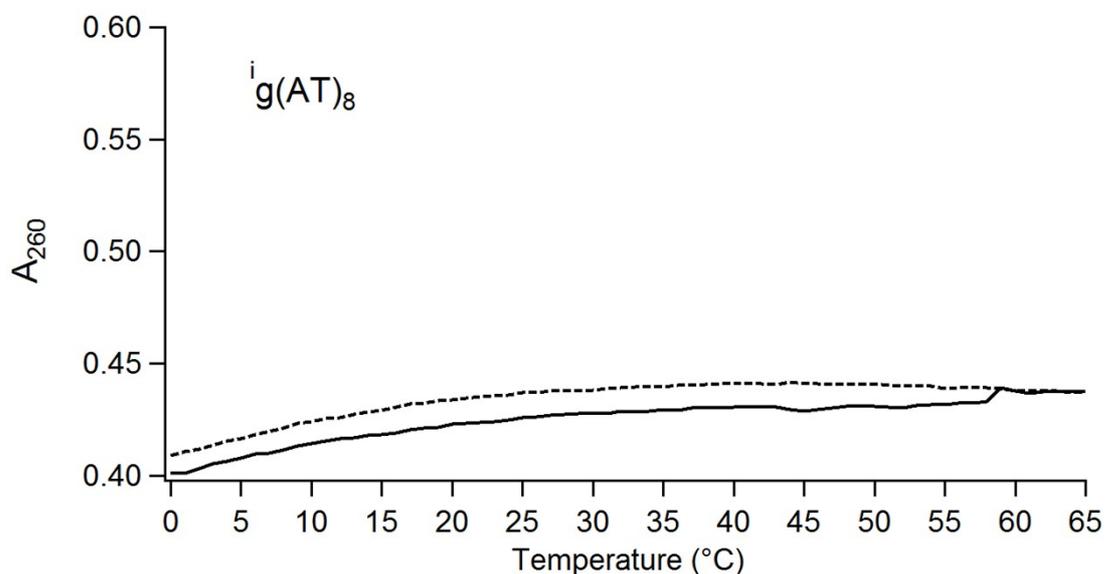


Fig. S28 UV-monitored thermal profiles of a solution containing ${}^i\text{g}(\text{AT})_8$ ($10\ \mu\text{M}$ in oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. The solution contained $1\ \text{M}\ \text{NaCl}$, $10\ \text{mM}\ \text{Na}_2\text{HPO}_4$, $0.1\ \text{mM}\ \text{EDTA}$, $\text{pH}\ 7.0$.

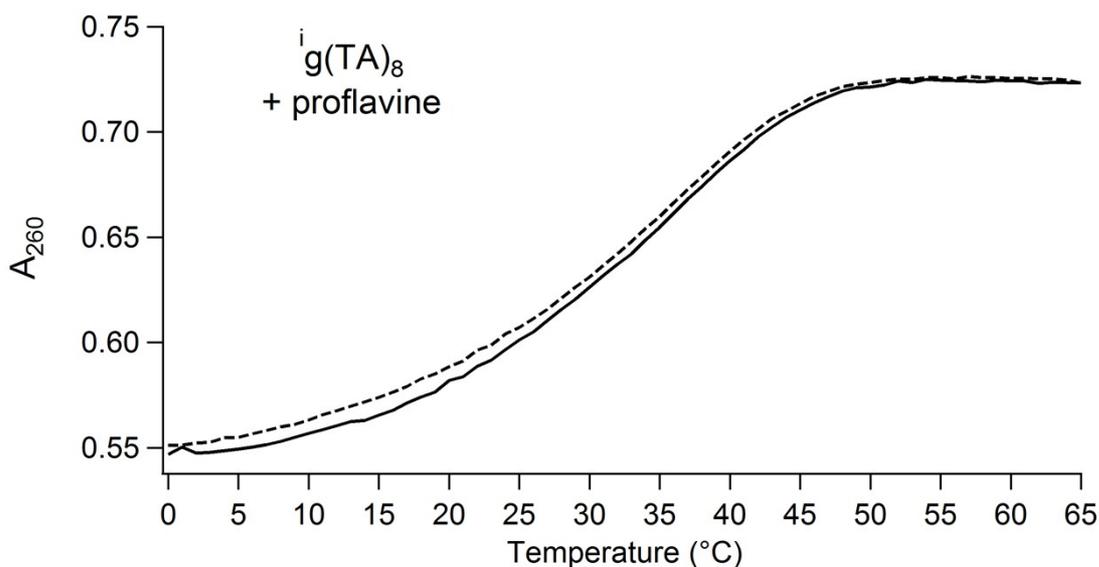


Fig. S29 UV-monitored thermal profiles of a solution containing i g(TA)₈ (10 μ M in oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

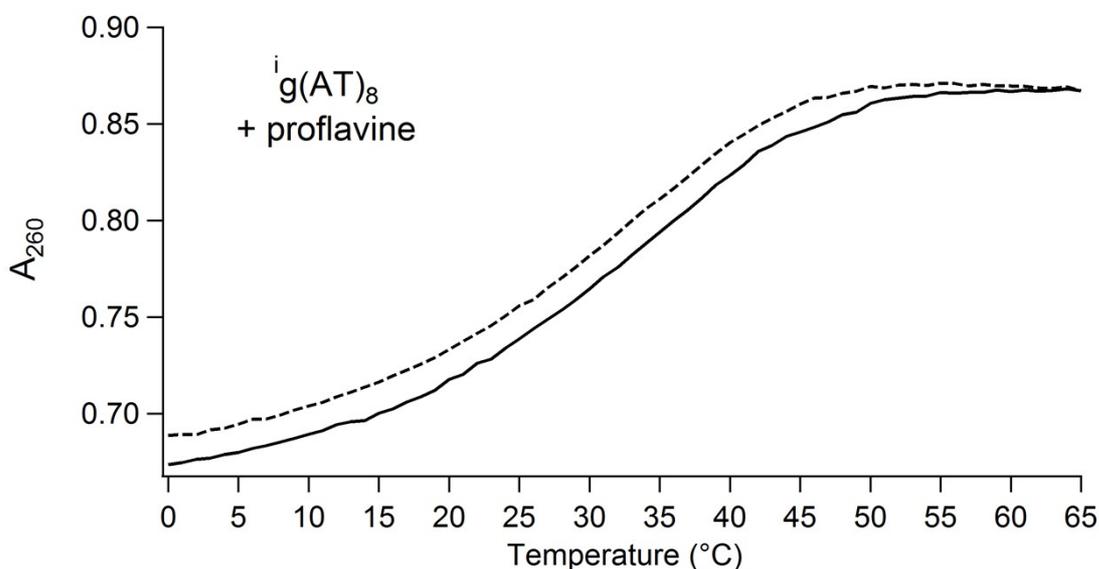


Fig. S30 UV-monitored thermal profiles of a solution containing i g(AT)₈ (10 μ M in oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

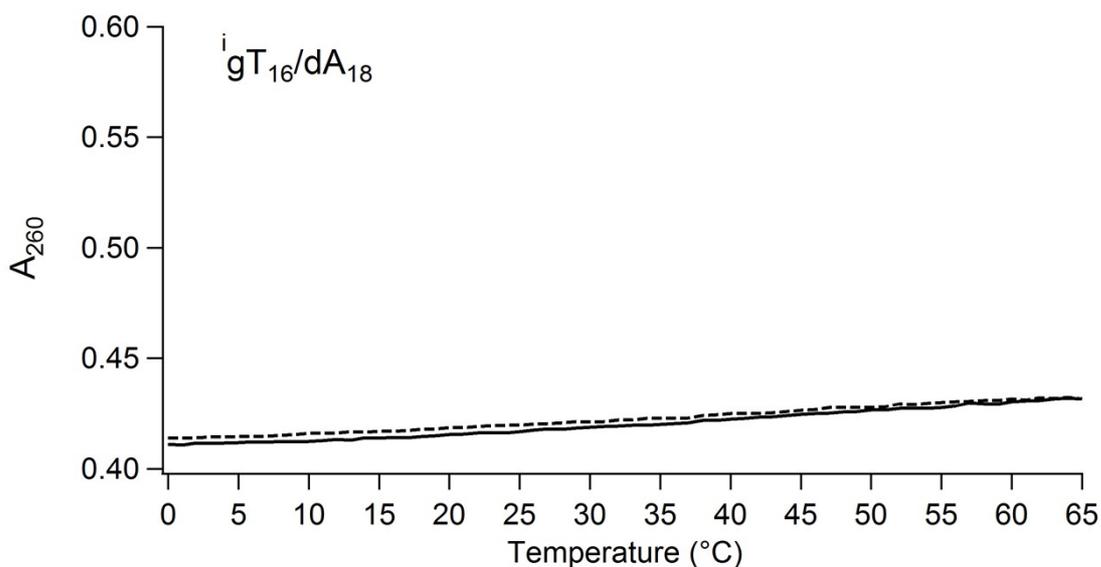


Fig. S31 UV-monitored thermal profiles of a solution containing i gT₁₆ and dA₁₈ (5 μ M in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. The solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

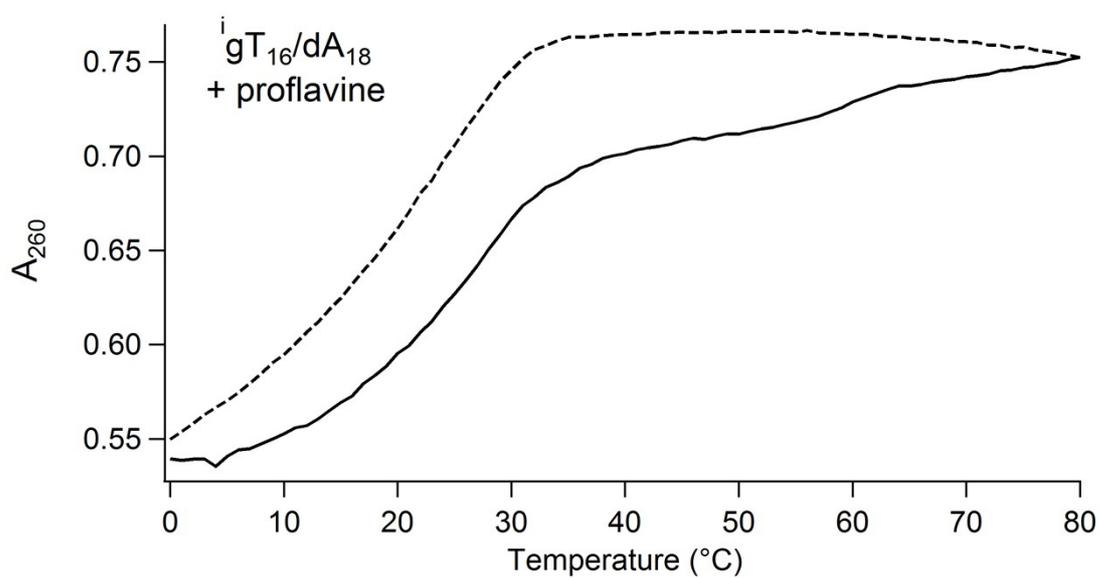


Fig. S32 UV-monitored thermal profiles of a solution containing i gT₁₆, dA₁₈ (5 μ M each in oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

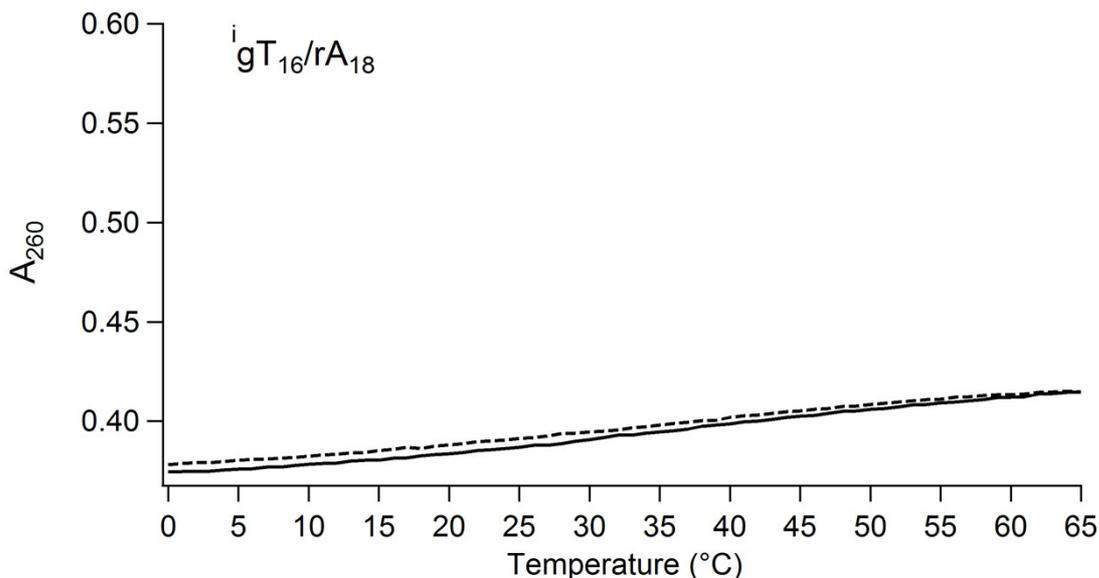


Fig. S33 UV-monitored thermal profiles of a solution containing $i\text{gT}_{16}$ and rA_{18} ($5\ \mu\text{M}$ in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. The solution contained 1 M NaCl, 10 mM Na_2HPO_4 , 0.1 mM EDTA, pH 7.0.

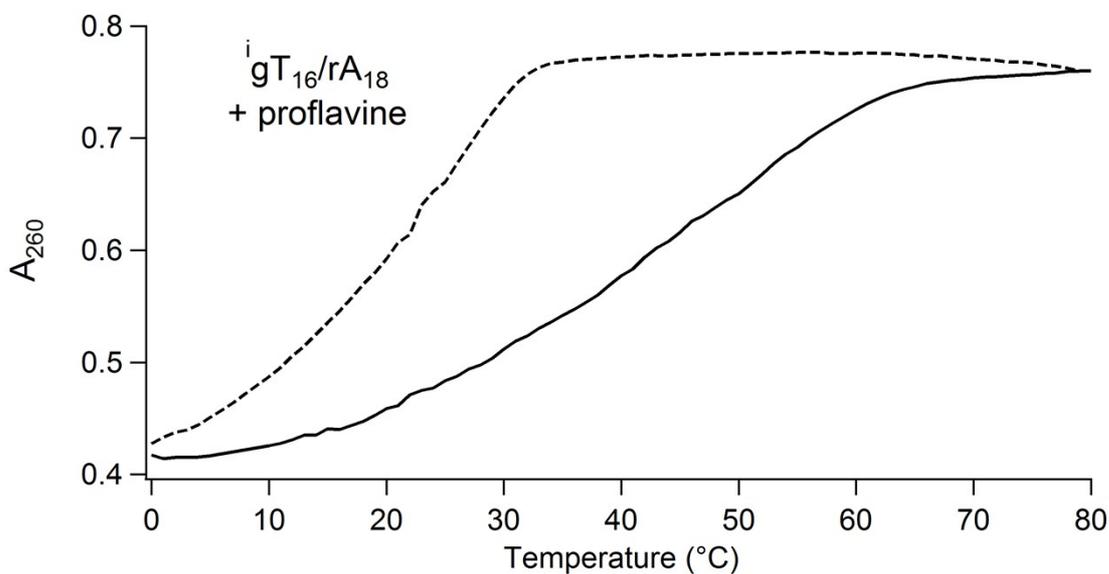


Fig. S34 UV-monitored thermal profiles of a solution containing $i\text{gT}_{16}$, rA_{18} ($5\ \mu\text{M}$ in each oligonucleotide) and proflavine ($45\ \mu\text{M}$). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na_2HPO_4 , 0.1 mM EDTA, pH 7.0.

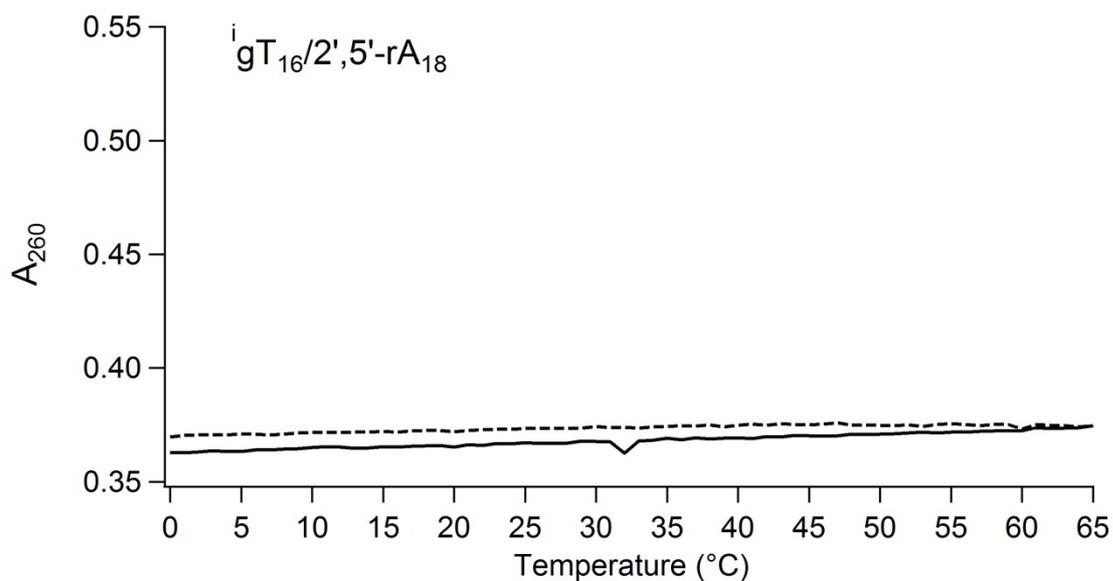


Fig. S35 UV-monitored thermal profiles of a solution containing i_gT_{16} and $2',5'\text{-rA}_{18}$ ($5\ \mu\text{M}$ in each oligonucleotide). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. The solution contained $1\ \text{M NaCl}$, $10\ \text{mM Na}_2\text{HPO}_4$, $0.1\ \text{mM EDTA}$, $\text{pH } 7.0$.

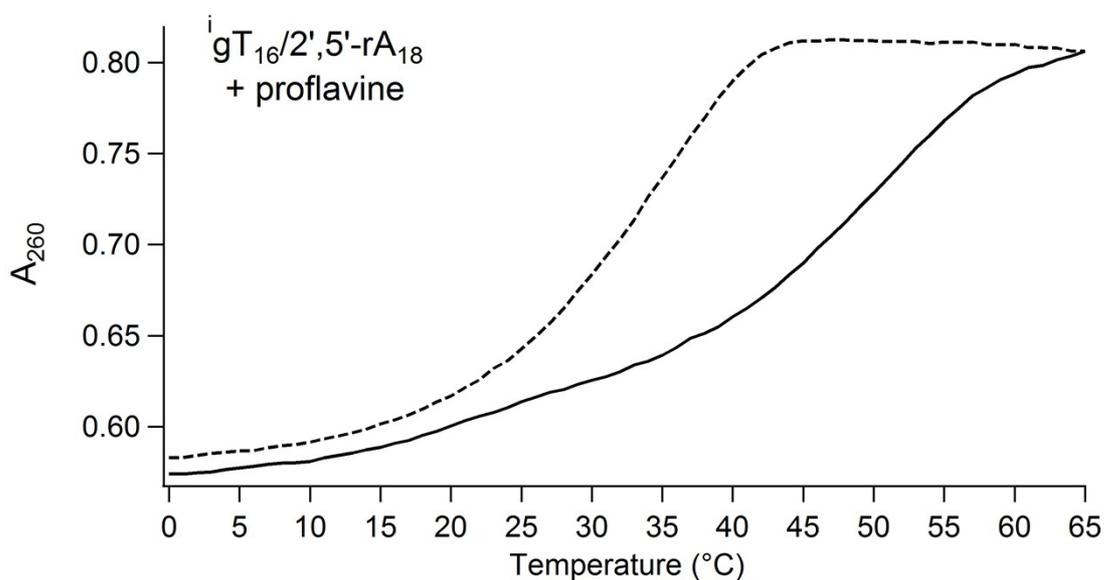


Fig. S36 UV-monitored thermal profiles of a solution containing $i_g(T)_{16}$, $2',5'\text{-rA}_{18}$ ($5\ \mu\text{M}$ in each oligonucleotide) and proflavine ($45\ \mu\text{M}$). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained $1\ \text{M NaCl}$, $10\ \text{mM Na}_2\text{HPO}_4$, $0.1\ \text{mM EDTA}$, $\text{pH } 7.0$. This particular duplex appears to have a pre-melting transition around $20\ ^\circ\text{C}$.

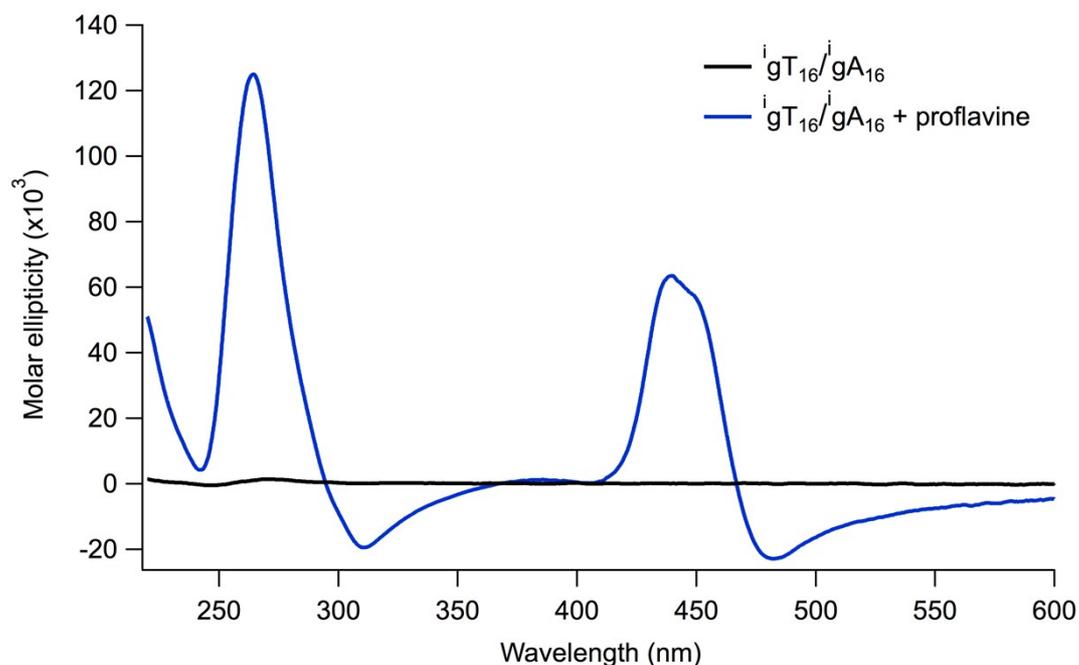


Fig. S37 CD spectra of the *i*_gT₁₆/*i*_gA₁₆ duplex in the absence (black line) and in the presence (blue line) of proflavine. Both solutions were 10 μM in each oligonucleotide. Sample containing proflavine was 90 μM in proflavine. Solutions contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

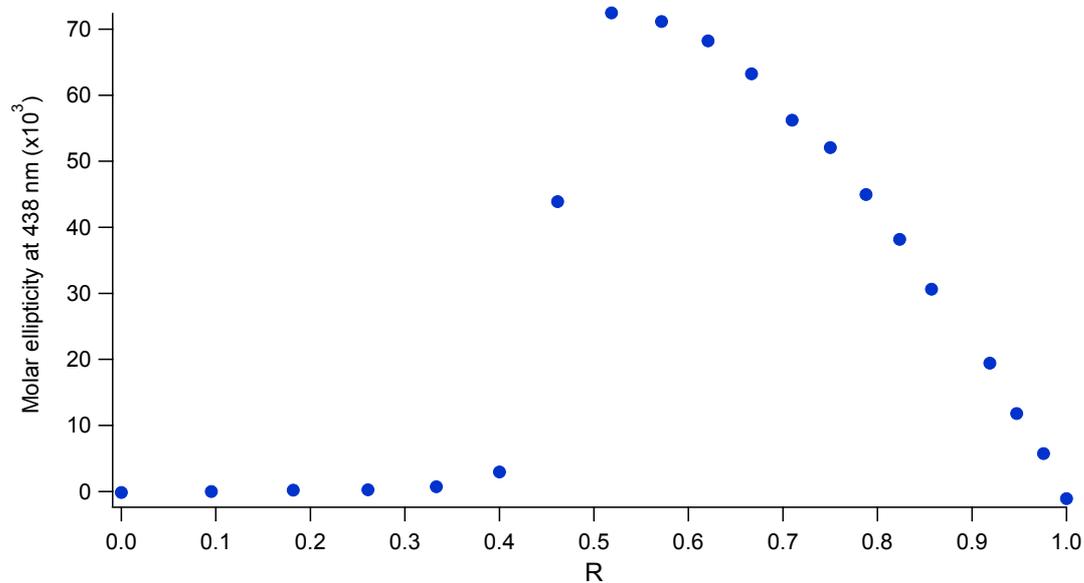


Fig. S38 Job plot analysis of the binding of proflavine to the *i*_gT₁₆/*i*_gA₁₆ duplex. $R = \frac{[\text{proflavine}]}{([\text{proflavine}] + [\text{base pair}]/2)}$. At $R = 0.5$, concentration of each oligonucleotide is 5 μM (90 μM in base pair) and proflavine is 45 μM. All solutions contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0. Measurements were performed at 4 °C.

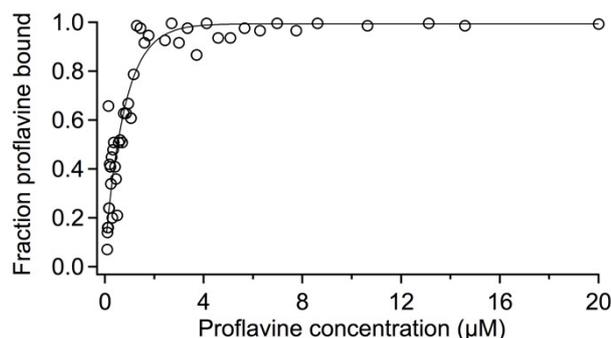


Fig S39 Data used for determination of the association constant (K_a) of proflavine with the duplex $i'gT_{16}/i'gA_{16}$ at 4 °C. Serial dilutions were performed using a solution originally 5 μM in each oligonucleotide and 45 μM in proflavine (one proflavine per two base pairs). The change in proflavine absorbance upon dissociation from the isoGNA duplex was used to monitor the fraction of proflavine bound to the duplex (normalized between 0 and 1). The K_a was determined to be $2 \times 10^6 \text{ M}^{-1}$. All solutions contained 1 M NaCl, 10 mM Na_2HPO_4 , 0.1 mM EDTA, pH 7.0.

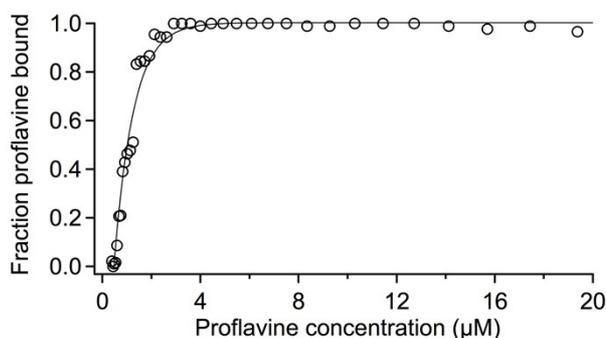


Fig. S40 Data used for determination of the association constant (K_a) of proflavine with the duplex $i'g(\text{AT})_8$ at 4 °C. Serial dilutions were performed using a solution originally 10 μM in oligonucleotide and 45 μM in proflavine (one proflavine per two base pairs). The change in proflavine UV absorbance upon dissociation from the isoGNA duplex was used to monitor the fraction of proflavine bound to the duplex (normalized here between 0 and 1). The K_a was determined to be $1 \times 10^6 \text{ M}^{-1}$. All solutions contained 1 M NaCl, 10 mM Na_2HPO_4 , 0.1 mM EDTA, pH 7.0.

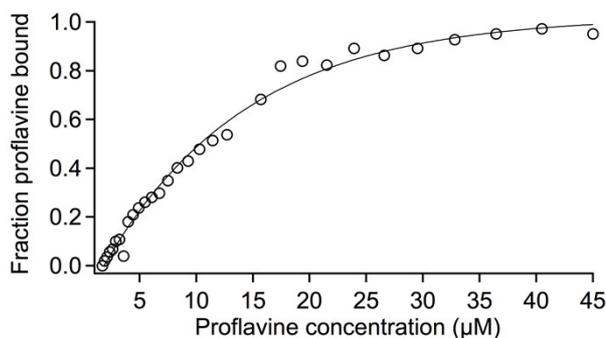


Fig. S41 Data used for determination of the association constant (K_a) of proflavine with the duplex $d(\text{AT})_9$ at 4 °C. Serial dilutions were performed using a solution originally 10 μM in oligonucleotide and 45 μM in proflavine (one proflavine per two base pairs). The change in proflavine UV absorbance upon dissociation from the DNA duplex was used to monitor the fraction of proflavine bound to the duplex (normalized here between 0 and 1). The K_a was determined to be $8.3 \times 10^4 \text{ M}^{-1}$. All solutions contained 1 M NaCl, 10 mM Na_2HPO_4 , 0.1 mM EDTA, pH 7.0.

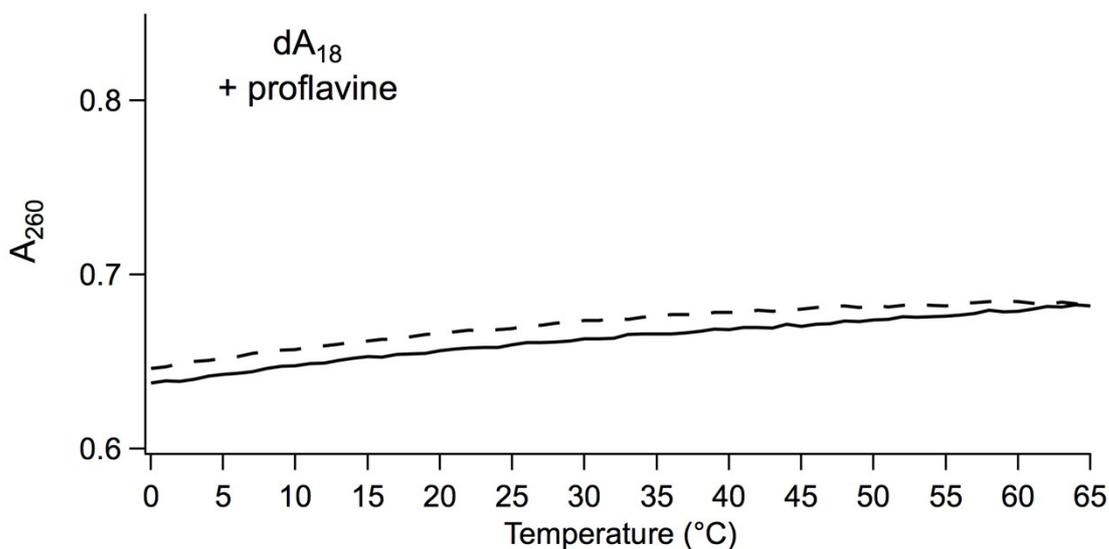


Fig. S42 UV-monitored thermal profiles of a solution containing dA₁₈ (5 μM in oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

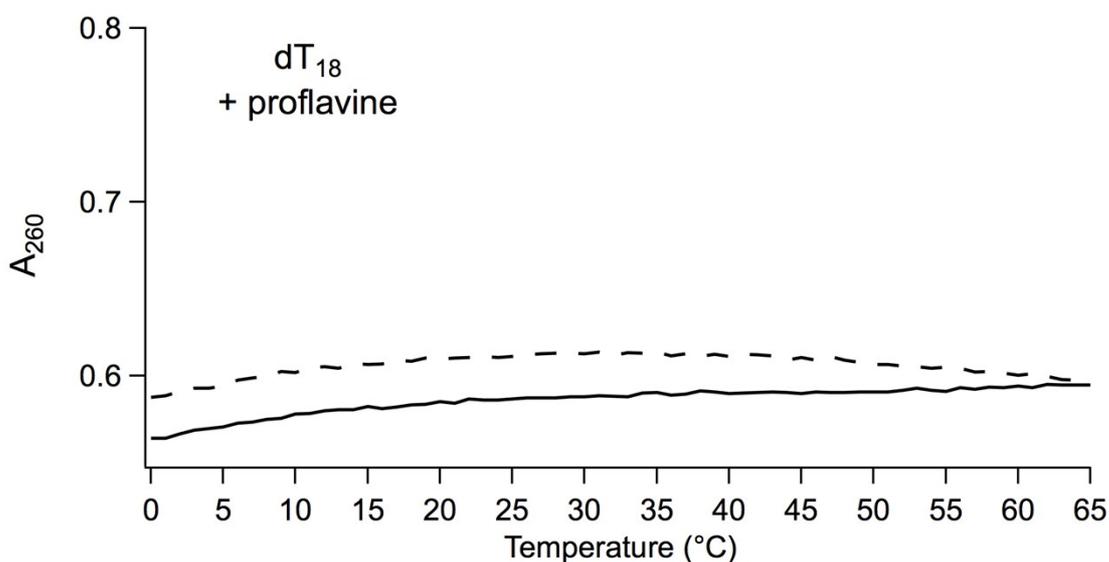


Fig. S43 UV-monitored thermal profiles of a solution containing dT₁₈ (5 μM in oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

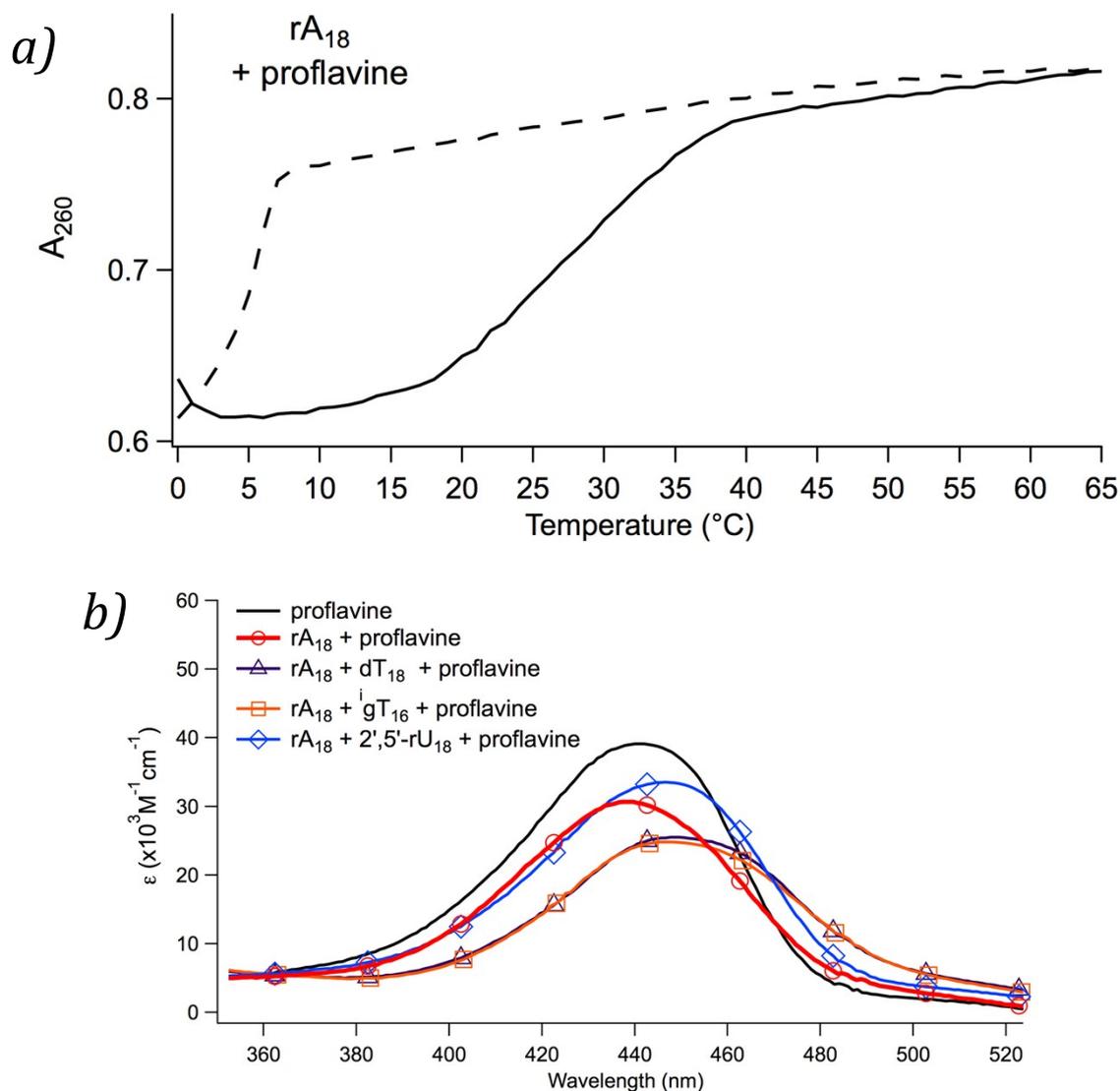


Fig. S44 *a)* UV-monitored thermal profiles of a solution containing rA_{18} ($5 \mu\text{M}$ in oligonucleotide) and proflavine ($45 \mu\text{M}$). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. rA_{18} forms a self-structure with proflavine with T_m of 28°C and 5°C for heating and cooling, respectively. We note that the self-structure formed by rA_{18} in the presence of proflavine is less stable than the duplexes formed with rA_{18} as one strand in the presence of proflavine. *b)* UV/vis spectra of longest wavelength proflavine absorption bands for solutions containing proflavine in the absence of oligonucleotide (solid black line), proflavine in the presence of rA_{18} only (red line-open circles), proflavine in the presence of rA_{18} and dT_{18} (purple line-open triangles), proflavine in the presence of rA_{18} and ${}^i gT_{16}$ (orange line-open squares), and proflavine in the presence of rA_{18} and $2',5'\text{-}rU_{18}$ (blue line-open diamonds). Note that the maximum of the proflavine absorption band is shifted to lower wavelengths (blue shift) for solutions containing proflavine and only rA_{18} relative to a solution with proflavine in the absence of oligonucleotide. The maxima of the proflavine absorption band is shifted to higher wavelengths (red shift) for solutions containing proflavine, rA_{18} and a complementary nucleic acid (i.e., dT_{18} , ${}^i gT_{16}$, $2',5'\text{-}rU_{18}$). A red shift in the absorption profile supports intercalative binding of proflavine to nucleic acid duplexes. Solutions contained 1 M NaCl , $10 \text{ mM Na}_2\text{HPO}_4$, 0.1 mM EDTA , $\text{pH } 7.0$.

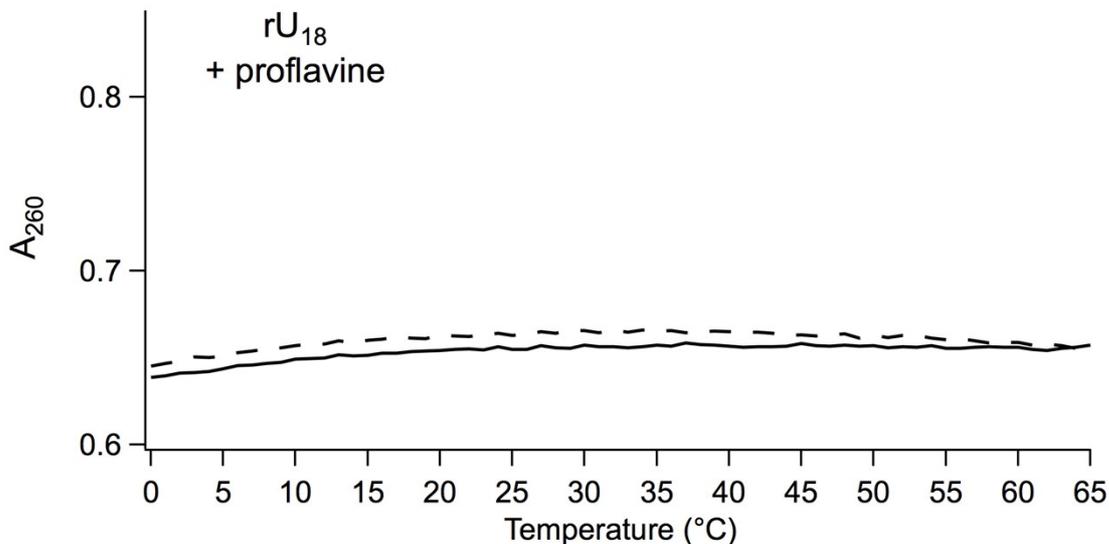


Fig. S45 UV-monitored thermal profiles of a solution containing rU₁₈ (5 μ M in oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

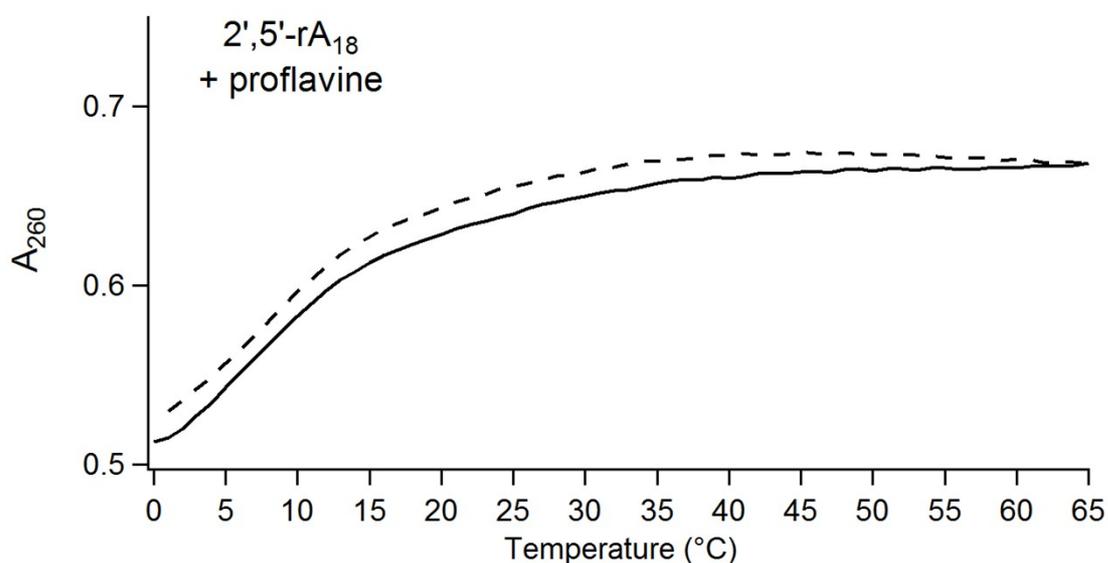


Fig. S46 UV-monitored thermal profiles of a solution containing 2',5'-rA₁₈ (5 μ M in oligonucleotide) and proflavine (45 μ M). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

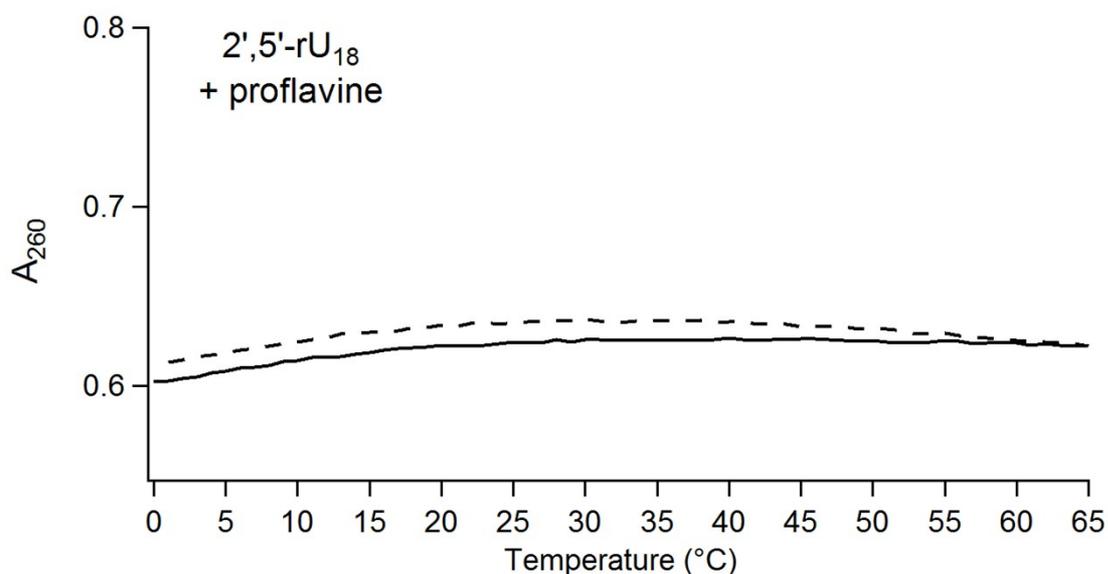


Fig. S47 UV-monitored thermal profiles of a solution containing 2',5'-rU₁₈ (5 μM in oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

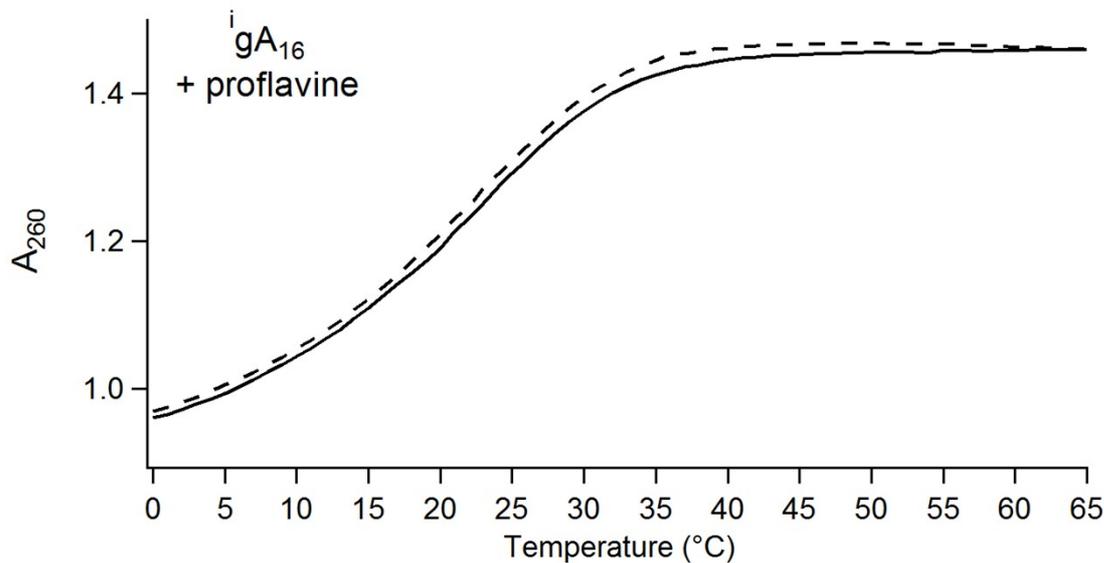


Fig. S48 UV-monitored thermal profiles of a solution containing *i*'gA₁₆ (5 μM in oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Note that *i*'gA₁₆ forms a self-structure with proflavine with a T_m of 20 °C, however, as indicated by the data in Figure S29, the duplex formed by *i*'gA₁₆ and *i*'gT₁₆ with proflavine is more stable (> 25 °C) than the *i*'gA₁₆ self-structure. Solution contained 1 M NaCl, 10 mM Na₂HPO₄, 0.1 mM EDTA, pH 7.0.

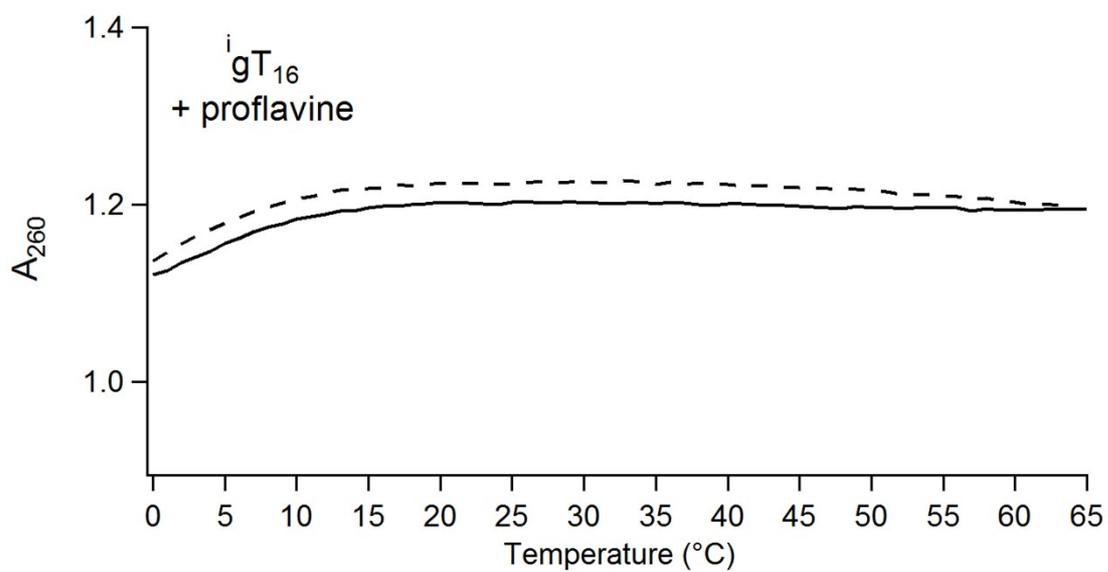


Fig. S49 UV-monitored thermal profiles of a solution containing ${}^i\text{gT}_{16}$ (5 μM in oligonucleotide) and proflavine (45 μM). A representative heating trace is shown as a solid line and a representative cooling trace is shown as a dashed line. Solution contained 1 M NaCl, 10 mM Na_2HPO_4 , 0.1 mM EDTA, pH 7.0.