

## Electronic Supplementary Information

# A single-stranded {Gd<sub>18</sub>} nanowheel with symmetric polydentate diacylhydrazone ligand

Kai Wang,<sup>a</sup> Zi-Lu Chen,<sup>b</sup> Hua-Hong Zou,<sup>b</sup> Kun Hu,<sup>b</sup> Hai-Ye Li,<sup>b</sup> Zhong Zhang,<sup>b</sup>

Wei-Yin Sun,<sup>c</sup> Fu-Pei Liang<sup>\*a,b</sup>

### Table of the content

1. General Materials and Methods
2. X-ray Structure Determination
3. Table S1. Crystal data and structure refinement parameters for **1**.
4. Synthesis of H<sub>4</sub>ovpho.
5. Scheme S1. Synthetic route of H<sub>4</sub>ovpho.
6. Synthesis of **1**.
7. Fig. S1. The asymmetry unit of **1**.
8. Table S2. The coordination environments, geometries and donors of Gd<sup>3+</sup> ions in **1**.
9. Fig. S2. The coordination modes of ovpho<sup>4-</sup> (a) and OAc<sup>-</sup> ions (b-d) in **1**.
10. Fig. S3. The building units of **1**.
11. Fig. S4. The arrangement of six ligands in **1**.
12. Fig. S5. The *D*<sub>3d</sub> symmetry structure of **1** in a side-view.
13. Fig. S6. Crystal packing diagram of the Gd-O cores of **1**.
14. Fig. S7. The  $\chi_m T$ -*T* plot for **1**.
15. Fig. S8. The  $\chi_m^{-1}$ -*T* plot for **1**.
16. Fig. S9. The plot of *M*-*H* for **1** in the field range of 0-50 kOe at 2-8 K.
17. Table S3. Selected bond lengths (Å) and angles (°) for **1**.
18. References

---

<sup>a</sup> Guangxi Key Laboratory of Electrochemical and Magnetochemical Functional Materials, College of Chemistry and Bioengineering, Guilin University of Technology, Guilin, 541004, China. E-mail address: [fliangoffice@yahoo.com](mailto:fliangoffice@yahoo.com) (F. Liang)

<sup>b</sup> State Key Laboratory for the Chemistry and Molecular Engineering of Medicinal Resources, School of Chemistry and Pharmacy, Guangxi Normal University, Guilin, 541004, China.

<sup>c</sup> Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, School of Chemistry and Chemical Engineering, Nanjing National Laboratory of Microstructures, Nanjing University, Nanjing, 210093, China.

## General Materials and Methods

All reagents were used as received without further purification. IR spectra were recorded in the range of 4000-400  $\text{cm}^{-1}$  on a Perkin-Elmer Spectrum One FT/IR spectrometer using a KBr pellet. Elemental microanalyses for C, H, and N were carried out on a Model 2400 II, Perkin-Elmer elemental analyzer. Magnetic susceptibility measurements were performed in the temperature range of 300-2 K using a Quantum Design MPMS SQUIDXL-5 magnetometer equipped with a 5 T magnetic field. The diamagnetic correction for the complex was estimated using Pascal's constants, and magnetic data were corrected for diamagnetic contributions of the sample holder.

## X-ray Structure Determination

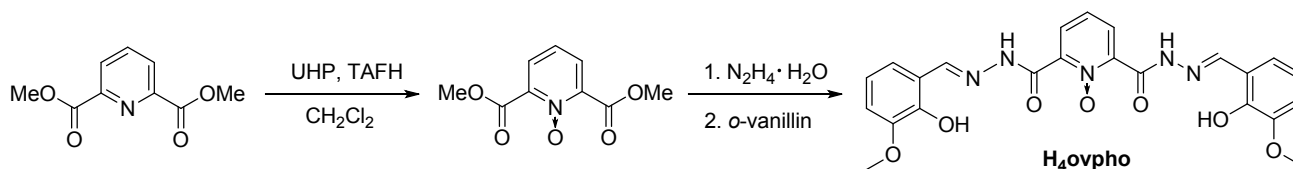
All the data for **1** were collected with a Bruker SMART CCD instrument by using graphite monochromatic Mo- $K\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) at 120(2) K. Absorption effect was corrected by semi-empirical methods. The structure was solved by direct methods and was refined by full-matrix least-squares methods with SHELXL-97 crystallographic software package.<sup>S1</sup> The non-hydrogen atoms were refined anisotropically. The hydrogen atoms were placed in calculated positions and refined by using a riding model. The final cycle of full-matrix least-squares refinement was based on observed reflections and variable parameters. A summary of crystal data and relevant refinement parameters is given in Table S1. Selected bond lengths and bond angles are given in Table S3.

**Table S1.** Crystal data and structure refinement parameters for **1**.

Identification	<b>1</b>	Identification	<b>1</b>
Formula	$\text{C}_{219}\text{H}_{324}\text{N}_{30}\text{O}_{147}\text{Gd}_{18}$	$Z$	6
Fw	8559.58	$D_c/\text{g cm}^{-3}$	1.504
Temp/K	120(2)	$m\mu/\text{mm}^{-1}$	3.190
Crystal system	Trigonal	$F(000)$	25056
Space group	$R\bar{3}c$	Reflns collected	95695
$a/\text{\AA}$	33.1977(3)	Unique reflns	12353
$b/\text{\AA}$	33.1977(3)	$R_{\text{int}}$	0.0963
$c/\text{\AA}$	59.4184(8)	GOF	1.160
$\alpha/^\circ$	90.00	$R_1, wR_2 (I > 2\sigma(I))$	$R_1 = 0.0568$
$\beta/^\circ$	90.00		$wR_2 = 0.1326$
$\gamma/^\circ$	120.00	$R_1, wR_2 (\text{all data})$	$R_1 = 0.0786$
$V/\text{\AA}^3$	56711.0(11)		$wR_2 = 0.1545$

## Synthesis of H<sub>4</sub>ovpho

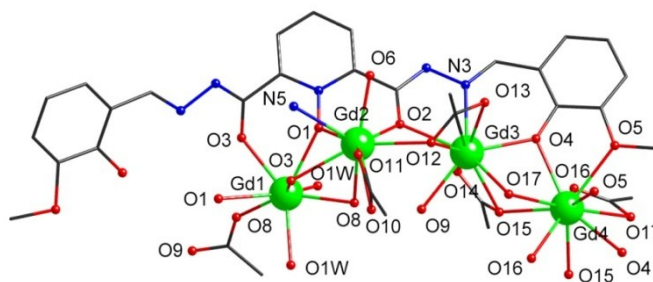
The synthetic route for H<sub>4</sub>ovpho is presented in Scheme S1. Starting from dimethyl 2,6-pyridinedicarboxylate, the dimethyl 2,6-pyridinedicarboxylate *N*-oxide was synthesized according to the literature.<sup>S2</sup> Subsequently, the pyridine-2,6-dicarbohydrazide *N*-oxide (2.1 g, 1 mmol) and N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O (10 ml) was refluxed in MeOH (100 ml) at 80 °C for 12 h. The *o*-vanillin (3.0 g, 2 mmol) was then added slowly and the reaction was kept at 80 °C for another 12 h. Upon cooling and filtering, yellow solid of *N,N'*-bis(*o*-vanillidene)pyridine-2,6-dicarbohydrazide *N*-oxide (H<sub>4</sub>ovpho) was obtained with a yield of 84%. ESI-MS *m/z*: 478.14 [M-H<sup>+</sup>]. Elemental analysis Calcd. (%): C, 57.62; H, 4.41; N, 14.61%. Found: C, 57.75; H, 4.33; N, 14.54%.



**Scheme S1.** Synthetic route of H<sub>4</sub>ovpho.

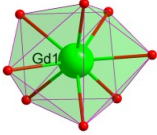
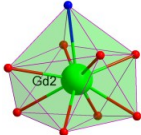
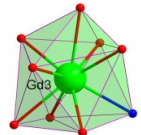
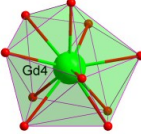
## Synthesis of 1

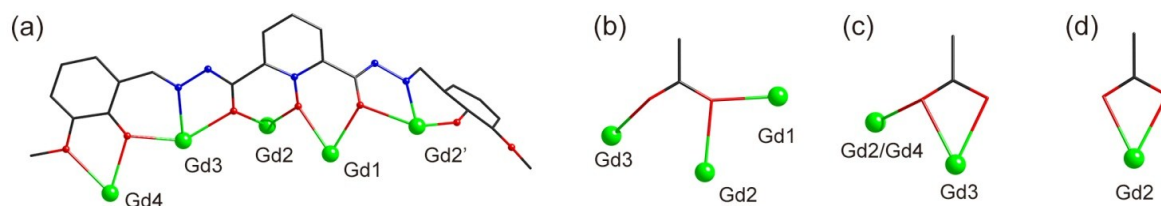
A mixture of Gd(OAc)<sub>3</sub>·6H<sub>2</sub>O (191 mg, 0.5 mmol), H<sub>4</sub>ovpho (24 mg, 0.05 mmol) and Et<sub>3</sub>N (0.1 mmol) in 1.5 mL CH<sub>3</sub>OH-CH<sub>3</sub>CN (*V/V* = 1/2) mixed solvent was sealed in a Pyrex tube and heated to 80 °C for 72 h. After cooled to room temperature at a rate of 0.5 °C/min, orange block-shaped crystals of **1** were obtained with a yield of 54% (based on ligand). Elemental analysis (%): calcd: C, 30.73; H, 3.82; N, 4.91; Found: C, 30.88; H, 3.93; N, 4.78. IR (KBr disk, cm<sup>-1</sup>): 3432 (vs), 2975 (w), 1615 (s), 1563 (m), 1456 (m), 1385 (m), 1216 (m), 1085 (m), 1050 (m), 946 (m), 741 (m), 669 (w), 437 (w).



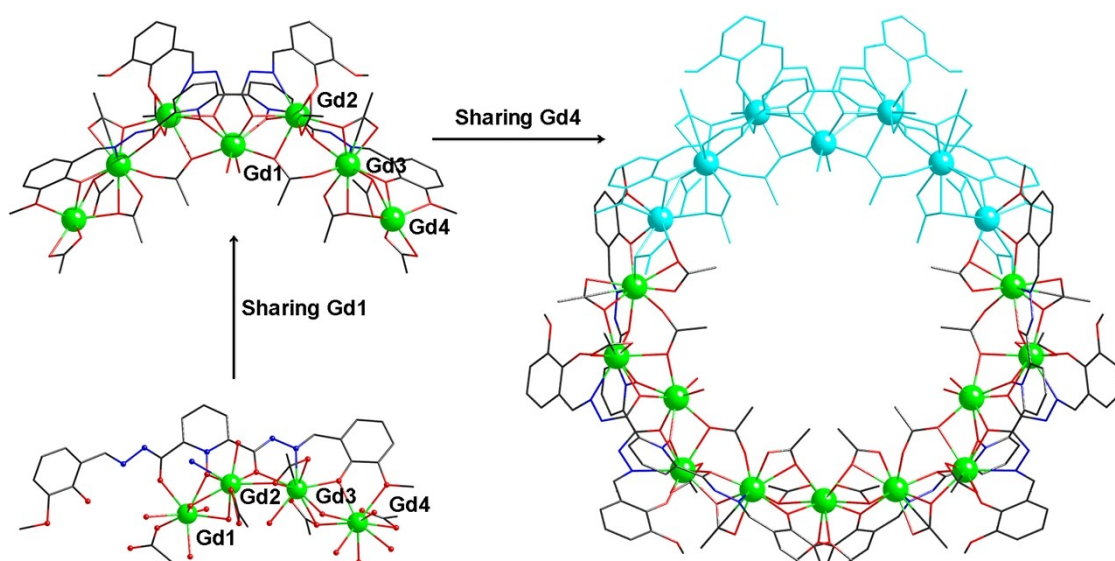
**Fig. S1.** The asymmetry unit of **1**.

**Table S2.** The coordination environments, geometries and donors of the Gd<sup>3+</sup> ions in **1**.

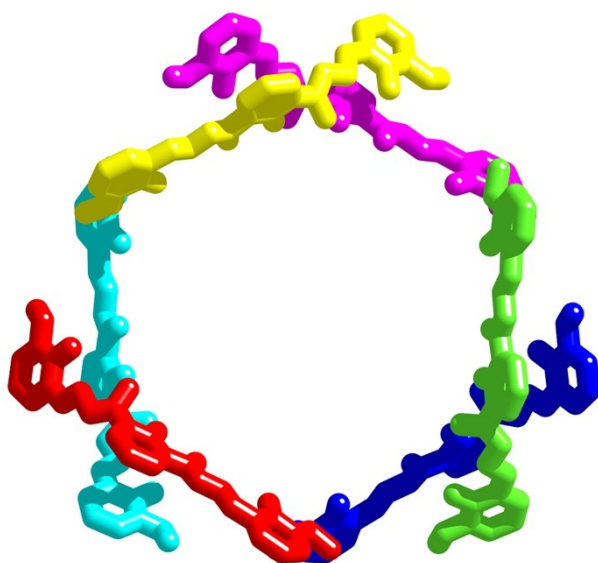
No.	Coordination Environments	Coordination Gometries	Coordination Donors
Gd1		Eight-coordinated Bicapped trigonal prism	O <sub>oxynitride</sub> , O <sub>enol</sub> , O <sub>acetate</sub> , O <sub>aqua</sub>
Gd2		Nine-coordinated Mono-capped square antiprism	O <sub>oxynitride</sub> , O <sub>phenol</sub> , O <sub>enol</sub> , O <sub>acetate</sub> , N <sub>acylhydrazone</sub>
Gd3		Nine-coordinated Mono-capped square antiprism	O <sub>acetate</sub> , O <sub>phenol</sub> , O <sub>enol</sub> , N <sub>acylhydrazone</sub>
Gd4		Ten-coordinated Bicapped dodecahedron	O <sub>acetate</sub> , O <sub>phenol</sub> , O <sub>methoxyl</sub>



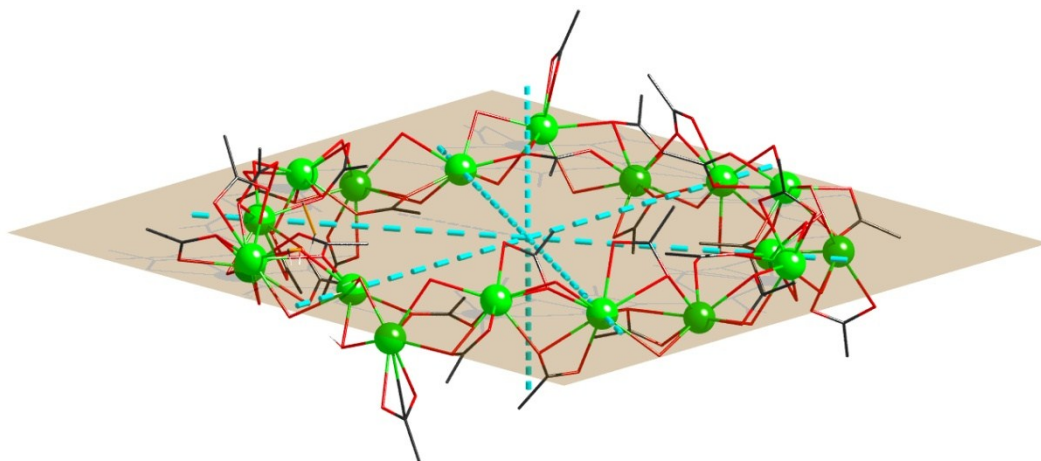
**Fig. S2.** The coordination modes of ovpho<sup>4-</sup> (a) and OAc<sup>-</sup> ions (b-d) in **1**.



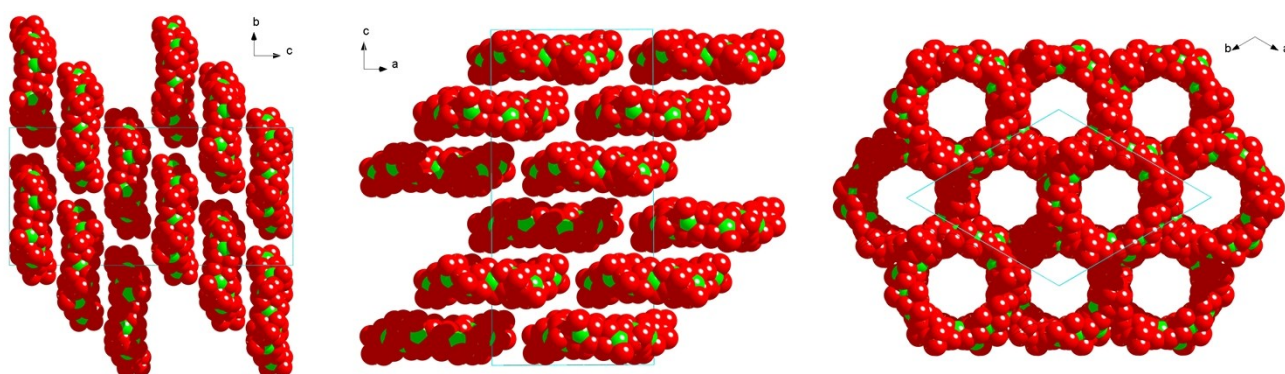
**Fig. S3.** The building units of **1**.



**Fig. S4.** The arrangement of six ligands in **1**.



**Fig. S5.** The  $D_{3d}$  symmetry structure of **1** in a side-view.



**Fig. S6.** Crystal packing diagram of the Gd–O cores of **1**.

Color code: Gd, green; O, red; cell edge, blue.

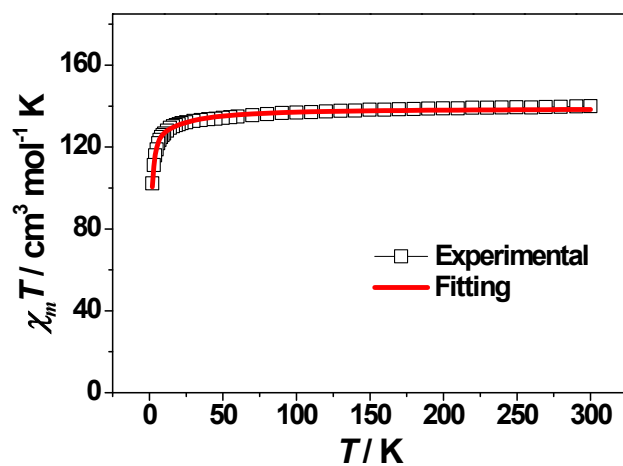


Fig. S7. The  $\chi_m T$ - $T$  plot for **1**. The red line represents the best fitting.

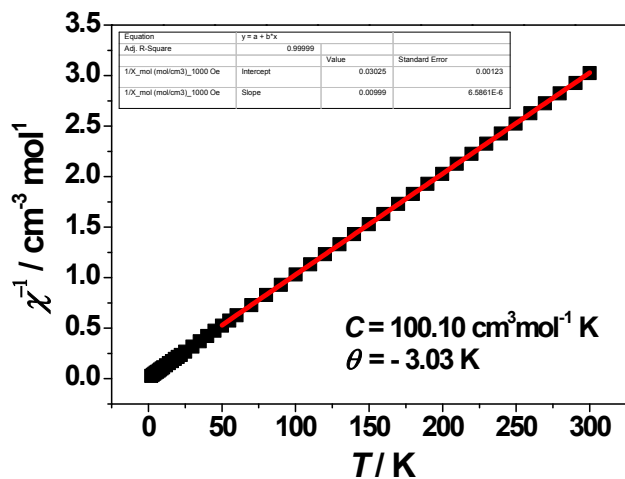


Fig. S8. The  $\chi_m^{-1}$ - $T$  plot for **1**.

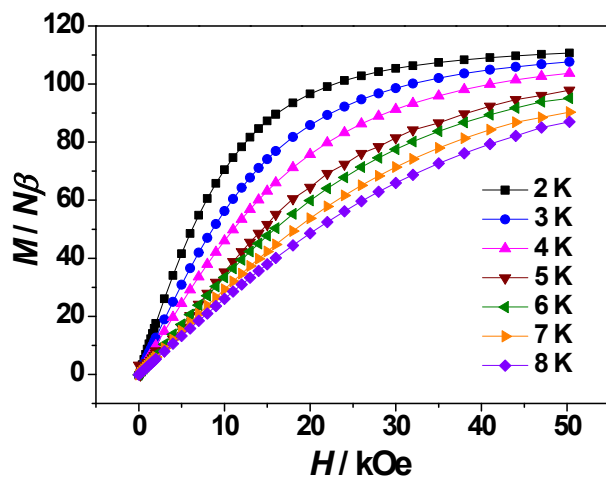


Fig. S9. The plot of  $M$ - $H$  for **1** in the field range of 0-50 kOe at 2-8 K.

**Table S3.** Selected bond lengths (Å) and angles (°) for **1**.

Bond	Length (Å)	Bond	Length (Å)	Bond	Length (Å)
Gd1-O8	2.334 (6)	Gd2-O10	2.467 (7)	Gd4-O15	2.379 (6)
Gd1-O8 <sup>i</sup>	2.335 (6)	Gd2-O8 <sup>i</sup>	2.550 (6)	Gd4-O4	2.439 (6)
Gd1-O1W <sup>i</sup>	2.355 (7)	Gd2-N5 <sup>ii</sup>	2.552 (9)	Gd4-O4 <sup>ii</sup>	2.439 (6)
Gd1-O1W	2.355 (7)	Gd2-N5 <sup>i</sup>	2.552 (9)	Gd4-O16 <sup>ii</sup>	2.482 (7)
Gd1-O3 <sup>i</sup>	2.387 (6)	Gd3-O4	2.325 (6)	Gd4-O16	2.482 (7)
Gd1-O3	2.387 (6)	Gd3-O9 <sup>i</sup>	2.342 (6)	Gd4-O5	2.523 (6)
Gd1-O1 <sup>i</sup>	2.402 (6)	Gd3-O17 <sup>ii</sup>	2.343 (6)	Gd4-O5 <sup>ii</sup>	2.523 (6)
Gd1-O1	2.402 (6)	Gd3-O2	2.357 (6)	Gd4-O17 <sup>ii</sup>	2.589 (7)
Gd2-O6 <sup>i</sup>	2.236 (7)	Gd3-O13	2.462 (6)	Gd4-O17	2.589 (7)
Gd2-O1	2.416 (6)	Gd3-N3	2.471 (7)	O3-Gd2 <sup>i</sup>	2.454 (6)
Gd2-O12	2.424 (6)	Gd3-O12	2.485 (6)	O8-Gd2 <sup>i</sup>	2.550 (6)
Gd2-O2	2.449 (6)	Gd3-O14	2.506 (6)	O9-Gd3 <sup>i</sup>	2.342 (6)
Gd2-O3 <sup>i</sup>	2.454 (6)	Gd3-O15	2.534 (6)	O17-Gd3 <sup>ii</sup>	2.343 (6)
Gd2-O11	2.464 (8)	Gd4-O15 <sup>ii</sup>	2.379 (6)	O6-Gd2 <sup>i</sup>	2.236 (7)
N5-Gd2 <sup>i</sup>	2.552 (9)				
Bond Angle	Angle(°)	Bond Angle	Angle(°)	Bond Angle	Angle(°)
Gd1-O1-Gd2	101.6 (2)	Gd3-O4-Gd4	101.9 (2)	Gd4-O15-Gd3	97.7 (2)
Gd3-O2-Gd2	107.2 (2)	Gd1-O8-Gd2 <sup>i</sup>	99.6 (2)	Gd3 <sup>ii</sup> -O17-Gd4	97.1 (2)
Gd1-O3-Gd2 <sup>i</sup>	100.9 (2)	Gd2-O12-Gd3	104.0 (2)		

Symmetry codes: (i)  $y-1/3, x+1/3, -z+5/6$ ; (ii)  $-x-4/3, -x+y-2/3, -z+5/6$ .

## References

- S1. (a) G. M. Sheldrick, *SHELXS-97, Program for Crystal Structure Solution and Refinement*. University of Göttingen, **1997**; (b) G. M. Sheldrick, *SHELXL 97, Program for the Refinement of Crystal Structure*. University of Göttingen, **1997**.
- S2. S. Caron, N. M. Do and J. E. Sieser, *Tetrahedron Lett.*, 2000, **41**, 2299.