Electronic Supplementary Information for

A Swelling-Suppressed Si/SiO_x Nanosphere Lithium Storage Material Fabricated by Graphene Envelopment

Hyundong Yoo,^a Eunjun Park,^a Hyekyoung Kim,^{bc} Juhye Bae,^a Hankwon Chang,^{bc}

Hee Dong Jang,*bc Hansu Kim*a

^a Department of Energy Engineering, Hanyang University, Seoul, 133-791, Republic of Korea.

^b Rare Metals Research Center, Korea Institute of Geoscience & Mineral Resources, Daejeon,
305-350, Republic of Korea.

^c Department of Nanomaterials Science and Engineering, University of Science and Technology, Daejeon 305-350, Republic of Korea.

* Corresponding authors: khansu@hanyang.ac.kr; hdjang@kigam.re.kr

Methods

Material preparation.

To prepare the SiO_x nanospheres, 10 mL of triethoxysilane ((C_2H_5O)₃SiH, Aldrich, 99.8%) was slowly dropped into a 0.1 M HCl solution (500 mL) under continuous stirring at 800 rpm. The solution was filtered and washed with deionized water repeatedly. The precipitates were collected and dried in a convection oven at 80 °C to eliminate residual moisture. After fine grinding, the obtained powder was heated at 1000 °C under a 4% H₂/Ar atmosphere with a flow rate of 0.5 L min⁻¹. The heating rate and time were fixed at 20 °C min⁻¹ and 1 h, respectively.

For the graphene-enveloped SiO_x, graphene oxide (GO) was prepared by a modified Hummers' Method and purified according to methods that have been reported previously.^{1,2} 1 mg mL⁻¹ of GO and 1 mg mL⁻¹ of the as-recovered SiO_x nanospheres were mixed in water and nebulized by a commercial mini spray dryer (CMSD, B290, Buchi). The CMSD had a two-fluid nozzle in which the precursor solution and dispersion air were introduced into the inner and outer tubes, respectively.³ Precursor droplets were formed from a continuous flow of the precursor solution at the nozzle tip. For the CMSD, the solution flow rate and the temperature were fixed at 6 L min⁻¹ and 200 °C, respectively. The product was collected by a cyclone and a bag filter, connected in series at the exhaust, and then further annealed at 900 °C under an Ar atmosphere for 30 min.

Material characterization.

The morphology and microstructure of the as-prepared graphene-enveloped SiO_x nanospheres were characterized using field-emission scanning electronic microscopy (FESEM; JEOL JSM- 7000F), high-resolution transmission electron microscopy (HRTEM; JEOL 2100F), and scanning transmission electron microscopy (STEM; JEOL 2100F). The microstructure was further identified by conducting X-ray diffraction (XRD) with an X-ray diffractometer (RTP 300 RC, Rigaku) and a Raman spectrometer (Dimension-P1 Raman, Lamda Solutions Inc.) with an Ar-ion continuous-wave laser at a wavelength of 514 nm. The amount of C in the graphene-enveloped SiO_x nanospheres was confirmed by thermogravimetric analysis (TGA; DTG-60H, Shimadzu).

Electrochemical characterization.

The electrodes were prepared by coating slurries consisting of the graphene-enveloped SiO_x nanospheres (70 wt%, as the active materials), a conducting agent (Super-P, 15 wt%), and a polyamide-imide binder (PAI, 15 wt%) dissolved in n-methyl-2-pyrrolidone (NMP, Sigma Aldrich) onto Cu foil with a mass loading of 1.0 mg cm⁻². The electrodes were dried at 200 °C for 2 h in a vacuum oven and were pressed under a pressure of 200 kg cm⁻². The electrochemical performance was examined by assembling CR2032 coin-type half cells in an Ar-filled glove box. These cells were assembled with a polyethylene (PE) membrane as the separator and Li metal as the counter and reference electrodes. 1 M LiPF₆ was dissolved in a mixed solvent of ethylene carbonate (EC) and ethyl methyl carbonate (EMC) (1:2 v/v, Panax Etec Co. Ltd.) with 2.0 wt% fluoroethylene carbonate (FEC); this was employed as the electrolyte. The cells were tested in constant current (CC) mode in a voltage window of 0.01–2.0 V vs. Li/Li⁺ at room temperature.



Fig. S1 Schematic illustrations of graphene-enveloped SiO_x nanospheres fabricated by a onestep, capillary force-driven aerosol process.



Fig. S2 HRTEM images of graphene-enveloped SiO_x nanosphere (pristine, before cycling): (a) low-magnification image and (b) high-magnification image with the corresponding selected area diffraction pattern (SADP, inset).



Fig. S3 (a) XRD pattern and (b) Raman spectrum collected from graphene-enveloped SiO_x nanospheres.



Fig. S4 TGA curve of graphene-enveloped SiO_x nanospheres.



Fig. S5 HRTEM images of graphene-enveloped SiO_x nanosphere after the first charge to 10 mV vs. Li/Li⁺: (a) low-magnification image and (b) high-magnification image with the corresponding selected area diffraction pattern (SADP, inset).



Fig. S6 HRTEM images of graphene-enveloped SiO_x nanosphere after the first discharge to 2.0 V vs. Li/Li⁺: (a) low-magnification image and (b) high-magnification image with the corresponding selected area diffraction pattern (SADP, inset).

Reference

- 1. W. S. Hummers Jr and R. E. Offeman, J. Am. Chem. Soc., 1958, 80, 1339-1339.
- 2. F. Kim, J. Luo, R. Cruz-Silva, L. J. Cote, K. Sohn and J. Huang, *Adv. Funct. Mater.*, 2010, **20**, 2867-2873.
- 3. P. J. Bruinsma, A. Y. Kim, J. Liu and S. Baskaran, *Chem. Mater.*, 1997, 9, 2507-2512.