# Electronic supplementary Information (ESI) 

# Enhancing the Magnetic Blocking Temperature and Magnetic Coercivityof $\left\{\mathrm{Cr}^{\prime \prime \prime}{ }_{2}\right.$ Ln" ${ }^{1 \mathrm{I}}{ }_{2}$ \} Single-Molecule Magnets via Bridging Ligand Modification 

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## Experimental details

General Information. All reactions were carried out under aerobic conditions. Chemicals and solvents were obtained from commercial sources and used without further purification. Elemental analyses (CHN) were carried out by Campbell Microanalytical Laboratory, University of Otago, Dunedin, New Zealand.

Synthesis of $\left[\mathrm{Cr}^{\mathrm{III}}{ }_{2} \mathrm{~Tb}^{\mathrm{III}}{ }_{2}(\mathrm{OMe})_{0.8}(\mathrm{OH})_{1.2}(2-\mathrm{Cl}-4,5-\mathrm{F}-\mathrm{benz})_{4}(\mathrm{mdea})_{2}\left(\mathrm{NO}_{3}\right)_{2}\right] \cdot 1.2 \mathrm{MeOH}$ (4). $\mathrm{CrCl}_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.13 \mathrm{~g}, 0.5 \mathrm{mmol})$ and $\mathrm{Tb}\left(\mathrm{NO}_{3}\right)_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.22 \mathrm{~g}, 0.5 \mathrm{mmol})$ were dissolved in $\mathrm{MeCN}(20 \mathrm{~mL})$, followed by the addition of N -methyldiethanolamine ( $0.06 \mathrm{~mL}, 0.5 \mathrm{mmol}$ ), 2-chloro-4,5-fluorobenzoic acid $(0.19 \mathrm{~g}, 1.0 \mathrm{mmol})$ and triethylamine ( $0.28 \mathrm{~mL}, 2.0 \mathrm{mmol}$ ) resulting in a green/blue solution. The solution quickly turned deep purple and was stirred for 2 hours after which the solvent was removed to give a purple oil. This was re-dissolved in MeOH and layered with diethylether $\left(\mathrm{Et}_{2} \mathrm{O}\right)$. Within $5-7$ days magenta crystals of 4 had appeared, in approximate yield of $42 \%$ Anal. Calculated (found) for 4 : $\mathrm{Cr}_{2} \mathrm{~Tb}_{2} \mathrm{C}_{40} \mathrm{H}_{38.4} \mathrm{O}_{21.2} \mathrm{~N}_{4} \mathrm{Cl}_{4} \mathrm{~F}_{8}$ : C, 29.54 (29.49); H, 4.36 (4.37); N, 3.44 (3.09).
Synthesis of $\left[\mathrm{Cr}^{I I I}{ }_{2} \mathrm{Dy}^{I I I}{ }_{2}(\mathrm{OMe})_{1.4}(\mathrm{OH})_{0.6}(2-\mathrm{Cl}-4,5-\mathrm{F}-\mathrm{benz})_{4}(\mathrm{mdea})_{2}\left(\mathrm{NO}_{3}\right)_{2}\right] \cdot 0.6 \mathrm{MeOH}$ (5).
The synthesis for 4 was followed but $\mathrm{Dy}\left(\mathrm{NO}_{3}\right)_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.22 \mathrm{~g}, 0.5 \mathrm{mmol})$ was used in place of $\mathrm{Tb}\left(\mathrm{NO}_{3}\right)_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}$. Magenta crystals of 5 appeared within5-7 days, in approximate yield of

49 \%. Anal. Calculated (found) for 5: $\mathrm{Cr}_{2} \mathrm{Dy}_{2} \mathrm{C}_{40} \mathrm{H}_{37.20} \mathrm{O}_{20.6} \mathrm{~N}_{4} \mathrm{Cl}_{4} \mathrm{~F}_{8}: \mathrm{C}, 29.42$ (29.21); $\mathrm{H}, 2.36$ (2.42); N, 3.44 (3.22).

Synthesis of $\left[\mathrm{Cr}^{I \mathrm{II}}{ }_{2} \mathrm{Ho}^{\mathrm{III}}{ }_{2}(\mathrm{OMe})_{1.16}(\mathrm{OH})_{0.84}(2-\mathrm{Cl}-4,5-\mathrm{F}-\mathrm{benz})_{4}(\text { mdea })_{2}\left(\mathrm{NO}_{3}\right)_{2}\right] \cdot 0.84 \mathrm{MeOH}$ (6).

The synthesis for 4 was followed but $\mathrm{Ho}\left(\mathrm{NO}_{3}\right)_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.22 \mathrm{~g}, 0.5 \mathrm{mmol})$ was used in place of $\mathrm{Tb}\left(\mathrm{NO}_{3}\right)_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}$. Magenta crystals of $\mathbf{6}$ appeared within5-7 days, in approximate yield of 39 \%. Anal. Calculated (found) for 6: $\mathrm{Cr}_{2} \mathrm{Ho}_{2} \mathrm{C}_{40} \mathrm{H}_{37.67} \mathrm{O}_{20.83} \mathrm{~N}_{4} \mathrm{Cl}_{4} \mathrm{~F}_{8}$ : C, 29.32 (29.40); H, 2.34 (2.41); N, 3.42 (3.32).

## Synthesis of $\left[\mathrm{Cr}^{\mathrm{III}}{ }_{2} \mathrm{Dy}^{I I I}{ }_{2}(\mathrm{OMM})(\mathrm{OH})\left(4-{ }^{-} \text {Bubenz }\right)_{4}\left(\right.\right.$ medea $_{2}\left(\mathrm{NO}_{3}\right)_{2} / \cdot \mathrm{MeOH} \cdot 2 \mathrm{Et}_{2} \mathrm{O}$ (7).

$\mathrm{CrCl}_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.13 \mathrm{~g}, 0.5 \mathrm{mmol})$ and $\mathrm{Dy}\left(\mathrm{NO}_{3}\right)_{3} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.22 \mathrm{~g}, 0.5 \mathrm{mmol})$ were dissolved in $\mathrm{MeCN}(20 \mathrm{~mL})$, followed by the addition of tert-butyldiethanolamine $(0.08 \mathrm{~g}, 0.5 \mathrm{mmol}), 4$ -tert-butylbenzoic acid $(0.18 \mathrm{~g}, 1.0 \mathrm{mmol})$ and triethylamine $(0.28 \mathrm{~mL}, 2.0 \mathrm{mmol})$ resulting in a green/blue solution. The solution quickly turned deep purple and was stirred for 2 hours after which the solvent was removed to give a purple oil. This was re-dissolved in MeOH and layered with diethylether $\left(\mathrm{Et}_{2} \mathrm{O}\right)$. Within $5-7$ days red/pink crystals of 7 had appeared, in approximate yield of $32 \%$.Anal. Calculated (found) for 7: $\mathrm{Cr}_{2} \mathrm{Dy}_{2} \mathrm{C}_{70} \mathrm{H}_{114} \mathrm{O}_{23} \mathrm{~N}_{4}$ : C, 46.48 (46.21); H, 6.35 (6.39); N, 3.10 (3.29).

X-ray Crystallography.X-ray crystallographic measurements for $\mathbf{4}$ were performed using a Bruker Smart Apex X8 diffractometer with $\mathrm{Mo} \mathrm{K} \alpha$ radiation. The data collection and integration were performed within SMART and SAINT+ software programs, and corrected for absorption using the Bruker SADABS program.X-ray diffraction measurements for 5 7were performed at 100(2) K at the Australian synchrotron MX1 beam-line. The data collection and integration were performed within Blu-Ice ${ }^{1}$ and XDS ${ }^{2}$ software programs. Compounds 4 - 7 were all solved by direct methods (SHELXS-97), and refined (SHELXL97) by full least matrix least-squares on all $F^{2}$ data. ${ }^{3}$ Crystallographic data and refinement parameters for $\mathbf{4 - 7}$ are summarized in Table S1. Crystallographic details are available in the Supporting Information (ESI) in CIF format. CCDC numbers 1495740 (4), 1495741 (5), 1495742 (6), 1495743 (7). These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

Table S1. Crystallographic data for complexes 4-7.

|  | 4 | 5 | 6 | 7 |
| :---: | :---: | :---: | :---: | :---: |
| Formula ${ }^{\text {a }}$ | $\begin{aligned} & \mathrm{Cr}_{2} \mathrm{~Tb}_{2} \mathrm{C}_{40} \mathrm{H}_{38.4} \\ & \mathrm{O}_{21.2} \mathrm{~N}_{4} \mathrm{Cl}_{4} \mathrm{~F}_{8} \end{aligned}$ | $\begin{aligned} & \mathrm{Cr}_{2} \mathrm{Dy}_{2} \mathrm{C}_{40} \mathrm{H}_{37.20} \\ & \mathrm{O}_{20.6} \mathrm{~N}_{4} \mathrm{Cl}_{4} \mathrm{~F}_{8} \end{aligned}$ | $\begin{aligned} & \mathrm{Cr}_{2} \mathrm{Ho}_{2} \mathrm{C}_{40} \mathrm{H}_{37.67} \\ & \mathrm{O}_{20.83} \mathrm{~N}_{4} \mathrm{Cl}_{4} \mathrm{~F}_{8} \end{aligned}$ | $\begin{aligned} & \mathrm{Cr}_{2} \mathrm{Dy}_{2} \mathrm{C}_{70} \mathrm{H}_{114} \\ & \mathrm{O}_{23} \mathrm{~N}_{4} \end{aligned}$ |
| $\mathrm{M}, \mathrm{gmol}^{-1}$ | 1629.98 | 1626.33 | 1635.34 | 1808.64 |
| Crystal system | Triclinic | Triclinic | Triclinic | Triclinic |
| Space group | $P-1$ | $P-1$ | $P-1$ | $P-1$ |
| $\mathrm{a} /[\AA]$ | 9.6481(6) | 9.5430(19) | 9.6670(19) | 11.164(2) |
| b/[ $\AA$ ] | 11.3861(7) | 11.414(2) | 11.331(2) | 11.907(2) |
| c/[ $\AA$ ] | 13.3598(8) | 13.306(3) | 13.316(3) | 16.534(3) |
| $\alpha /\left[{ }^{\circ}\right]$ | 67.617(5) | 67.56(3) | 67.54(3) | 70.77(3) |
| $\beta /\left[{ }^{\circ}\right]$ | 78.339(5) | 77.95 (3) | 78.27(3) | 73.76(3) |
| $\gamma /\left[{ }^{\circ}\right]$ | 87.410(5) | 87.25(3) | 87.69(3) | 82.43(3) |
| $\mathrm{V} /\left[\AA^{3}\right]$ | 1328.23(14) | 1309.2(5) | 1318.7(5) | 1990.5(7) |
| T/K | 123(2) | 100(2) | 100(2) | 100(2) |
| Z | 1 | 1 | 1 | 1 |
| $\rho$, calc $\left[\mathrm{g} \mathrm{cm}^{-3}\right]$ | 2.036 | 2.062 | 2.058 | 1.508 |
| $\lambda /[\AA]$ | 0.71073 | 0.71079 | 0.71079 | 0.71079 |
| Data <br> Measured | 18565 | 16519 | 27401 | 32265 |
| Ind. Reflns | 6101 | 4234 | 5907 | 8352 |
| $\mathrm{R}_{\text {int }}$ | 0.0427 | 0.0318 | 0.0391 | 0.0344 |
| Reflns with $\mathrm{I}>2 \sigma(\mathrm{I})$ | 5374 | 4040 | 5734 | 8051 |
| Parameters | 384 | 383 | 384 | 512 |
| Restraints | 0 | 0 | 0 | 66 |
| $\mathrm{R}_{1}{ }^{\mathrm{b}}$ (obs), $w R_{2}{ }^{\mathrm{b}}$ (all) goodness of fit | $0.0284,0.0645$ 1.091 | $0.0310,0.0826$ 1.068 | $0.0315,0.0819$ 1.059 | $0.0432,0.1177$ 1.059 |
| Largest residuals/ [ $\mathrm{e} \AA^{-3}$ ] | 0.991, -0.899 | 0.919, -1.435 | 1.106, -1.734 | 2.369, -1.463 |

${ }^{\text {a }}$ Includes solvate molecules.
${ }^{\mathrm{b}} \mathrm{R}_{1}=\Sigma| | \mathrm{Fo}|-|\mathrm{Fc}|| \Sigma|\mathrm{Fo}|, \mathrm{wR}_{2}=\left\{\Sigma\left[\mathrm{w}\left(\mathrm{Fo}^{2}-\mathrm{Fc}^{2}\right)^{2}\right] / \Sigma\left[\mathrm{w}\left(\mathrm{Fo}^{2}\right)^{2}\right]\right\}^{1 / 2}$.

Magnetic Measurements. The magnetic susceptibility measurements were carried out on a Quantum Design SQUID magnetometer MPMS-XL 7 operating between 1.8 and 300 K for dc-applied fields ranging from $0-5 \mathrm{~T}$. Microcrystalline samples were dispersed in Vaseline in order to avoid torquing of the crystallites. The sample mulls were contained in a calibrated gelatine capsule held at the centre of a drinking straw that was fixed at the end of the sample rod. Alternating current (ac) susceptibility measurements were carried out under an oscillating ac field of 3.5 Oe and frequencies ranging from 0.1 to 1500 Hz .



Fig. S1. Molecular structure of 4 (left) and 6 (right), the H -atoms are omitted for clarity. Colour scheme; Cr ${ }^{\text {III }}$, yellow; Ln ${ }^{\text {III }}$, purple; O, red; N, blue; C, light grey.

Table S2. Selected bond lengths $(\AA)$ and angles $\left(^{\circ}\right)$ for $\mathbf{4 - 7}$, using the labelling scheme shown below. The first coordination sphere for $\mathbf{7}$ is identical to that shown below.


| Atom | $\mathbf{4}\left\{\mathbf{C r}_{2} \mathbf{T b}_{\mathbf{2}}\right\}$ | $\mathbf{5}\left\{\mathbf{C r}_{2} \mathbf{D y}_{\mathbf{2}}\right\}$ | $\mathbf{6}\left\{\mathrm{Cr}_{2} \mathbf{H o}_{\mathbf{2}}\right\}$ | $\mathbf{7}\left\{\mathbf{C r}_{2} \mathbf{D y}_{\mathbf{2}}\right\}$ |
| :--- | :--- | :--- | :--- | :--- |
| Cr1-O4 | $1.952(2)$ | $1.953(3)$ | $1.953(3)$ | $1.955(3)$ |
| Cr1-O5 | $1.954(2)$ | $1.954(3)$ | $1.949(3)$ | $1.951(3)$ |
| Cr1-O1 | $1.967(2)$ | $1.968(3)$ | $1.972(3)$ | $1.970(3)$ |
| Cr1-O3 | $1.983(2)$ | $1.972(3)$ | $1.977(3)$ | $1.979(3)$ |
| Cr1-O6 | $1.993(2)$ | $1.989(3)$ | $1.993(3)$ | $1.982(3)$ |
| Cr1-N1 | $2.074(3)$ | $2.082(4)$ | $2.083(3)$ | $2.176(4)$ |
| Ln1-O5 | $2.260(2)$ | $2.241(3)$ | $2.236(3)$ | $2.243(3)$ |
| Ln1-O4' | $2.265(2)$ | $2.242(3)$ | $2.238(3)$ | $2.249(3)$ |
| Ln1-O2' | $2.388(2)$ | $2.387(3)$ | $2.365(3)$ | $2.365(3)$ |
| Ln1-07 | $2.391(2)$ | $2.384(3)$ | $2.365(3)$ | $2.356(3)$ |
| Ln1-01 | $2.429(2)$ | $2.432(3)$ | $2.418(3)$ | $2.423(3)$ |
| Ln1-O1' | $2.449(2)$ | $2.449(3)$ | $2.428(3)$ | $2.425(3)$ |
| Ln-O8 | $2.460(2)$ | $2.442(3)$ | $2.428(3)$ | $2.437(4)$ |
| Ln-O9 | $2.463(2)$ | $2.439(3)$ | $2.424(3)$ | $2.445(3)$ |
| Ln1-O1-Ln1' | $95.65(9)$ | $95.00(11)$ | $95.44(10)$ | $97.23(11)$ |
| Cr1-O1-Ln1 | $96.73(8)$ | $95.88(12)$ | $95.81(11)$ | $97.26(13)$ |
| Cr1-O4-Ln1' | $102.23(9)$ | $102.34(12)$ | $102.36(11)$ | $103.72(13)$ |


| Cr1-O5-Ln1 | $102.84(9)$ | $102.78(13)$ | $102.61(11)$ | $104.06(14)$ |
| :--- | :--- | :--- | :--- | :--- |
| Ln1-O1-Ln1 | $113.79(9)$ | $113.88(12)$ | $114.17(10)$ | $113.59(12)$ |



Fig. S2. Plots of $\chi_{M} T$ versus $T$ for $\mathbf{4 - 6}$ (top) and 7 (bottom), using an applied magnetic field of 0.1 T .


Fig. S3. Plots of $\chi_{M} T$ versus $T$ for $\mathbf{1}$ and $\mathbf{4}$ (top left), $\mathbf{2}$ and $\mathbf{5}$ (top right), $\mathbf{3}$ and $\mathbf{6}$ (bottom left) and $\mathbf{2}$ and 7 (bottom right) using an applied magnetic field of 0.1 T .


Fig. S4. Alternating current magnetic susceptibility measurements. In-phase ( $\chi_{\mathrm{m}}$ ) and out of phase ( $\chi_{\mathrm{m}}$ ") signals for $\mathbf{4}$ (top left), $\mathbf{5}$ (top right), $\mathbf{6}$ (bottom left) and $\mathbf{7}$ (bottom right).


Fig. S5. Relaxation data for $\mathbf{1}$ and $\mathbf{3}$ plotted as $\ln (\tau)$ versus $T^{-1}$. At the lowest temperatures the data begins to deviate from linearity.

## References

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