Electronic Supplementary Information (ESI) for:

Selective Separation of Americium from Europium Using 2,9-Bis(triazine)-1,10-phenanthrolines in Ionic Liquids: A New Twist on an Old Story

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Section S1. Materials and Synthetic Methods.

Materials

All reagents were obtained from Aldrich in their purest form and used without further purification. The solvent trifluoromethylphenyl sulfone (FS-13) was obtained from Marshallton Research Laboratories and used as received. The 1-alkyl-3-methylimidazolium bis[(trifluoromethyl)-sulfonyl]amide ([C4mim][NTf2]) ionic liquid was obtained from Sigma-Aldrich. Europium-152/154 was obtained from Isotope Products (presently owned by Eckert & Ziegler) and americium-241was produced at Oak Ridge National Laboratory.

Synthetic Methods

¹H and ¹³C NMR spectra were recorded using a Varian VNMRS 500 MHz spectrometer. The chemical shifts at 25°C (given in parts per million) were referenced to the residual protonated solvent. The mass peaks for protonated molecules were determined using DART (direct analysis in real time) at the Mass Spectrometry Center located in the Department of Chemistry at the University of Tennessee. The DART analyses were performed using a JEOL AccuTOF-D time-of-flight mass spectrometer with a DART ionization source from JEOL USA, Inc. (Peabody, MA).

Me-BTPhen was prepared according to a literature procedure¹ with a slight modification for the last step (Scheme 1). Neocuproine **1** was oxidized using selenium dioxide. The resulting dialdehyde **2** was treated with hydroxylamine hydrochloride. The dioxime **3** was then converted to the dinitrile 4. The latter was reacted with hydrazine hydrate to give the dicarbohydrazonamide **5**. Me-BTPhen was obtained by reaction of **5** with diacetyl in presence of triethylamine. Synthetic details for the last step and NMR spectra of Me-BTPhen are given below.



Scheme 1: synthetic route for the preparation of Me.BTPhen

To a suspension of 1,10-phenanthroline-2,9-dicarbohydrazonamide **5** (800 mg, 2.72 mmol) in THF (50 mL) was added diacetyl (0.5 mL, 5.75 mmol, 2.1 eq) and Et₃N (5 mL). The mixture was refluxed for 24h. After cooling, the solvent was removed under vacuum and the residue was washed with Et₂O. The crude product was purified by silica gel column chromatography (CH₂Cl₂/MeOH, 100:0 to 70:30 v/v) to give the desired compound as a yellow powder (660 mg, 62%). ¹H NMR (500 MHz, CDCl₃): δ 2.81 (s, 6H), 2.84 (s, 6H), 7.94 (s, 2H), 8.47 (dd, 2H, *J* = 8.4 Hz, 0.5 Hz), 8.99 (dd, 2H, *J* = 8.4 Hz, 0.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 19.7, 21.6, 123.0, 127.7, 130.1, 137.5, 146.4, 152.9, 157.5, 160.0, 161.3. HRMS: *m/z*: 395.1741 [M + H⁺]. C₂₂H₁₉N₈⁺ requires 395.1732.



Figure S1¹H NMR of Me-BTPhen in CDCl₃



Figure S2¹³C NMR of Me-BTPhen in CDCl₃

Section S2. Determination of distribution ratios.

The distribution ratio (D_M) for extraction of trivalent metal ions (M^{3+}) is defined by

$$D_M = \frac{C_{M,org}}{C_{M,aq}}$$

In this equation, $C_{m,aq}$ and $C_{m,org}$ represent the counts per a minute for the isotopes of ^{152/154}Eu or ²⁴¹Am in either the organic or aqueous phases, respectively. A volume ratio is needed in the calculation of distribution ratios to account for the difference in volume between the two immiscible phases. In all of our experiments, the volume ratio was close to 1:1 v/v. Separation factors SF for the metal ions *M* and *M*' are defined from:

$$SF_{M/M'} = D_M / D_M$$

Eu/Am measurements.

The distribution ratios for extraction using radiotracer techniques were calculated by measuring the amount of radioactivity of both aqueous and organic phases at equilibrium. Counting efficiency (241 Am or 152,154 Eu gamma ray absorption in solid scintillators) is identical for both phases; hence, the distribution ratio is defined by the ratio of specific radioactivity *S* (Bq/mL) of element *M* in the IL vs. aqueous phases.

$$D_M = \frac{C_{IL,f}}{C_{IL,f}} \propto \frac{S_{org,f}}{S_{aq,f}}$$

An equal volume of both IL (containing 4 mM complexant) and aqueous phases, 0.4 mL of each were used, respectively. Each sample was individually spiked with a 10 µL solution containing 1.85×10^6 Bq/mL (50 μ Ci/mL) of each radiotracer respectively. The solutions were mixed using a rotating wheel set at 60 rpm for 3 h at 25±0.2 °C. (Initial studies were performed at 1, 3, and 24 h, with samples contacted for 3 and 24 h observed to have the same D values. Due to this observation it was assumed that the samples achieved equilibrium by the 3 h time point. Samples contacted/mixed for 1 h had lower D values than those at 3 h. For the sake of brevity and consistency, all samples were contacted for 3 h). After 3 h the samples were centrifuged at 3.000 rpm for 5 min at 25 °C to ensure the phases separated from each other. Then a 100 µL aliquots were subsampled from each phase and placed into polypropylene tubes that were sealed with a cap. These tubes were then placed in a Canberra Gamma Analyst germanium spectrometer to determine the amounts of ²⁴¹Am and ^{152,154}Eu present in each sample. Prior to testing these samples, a quality assurance calibration was performed. The organic and aqueous samples were counted for a period of 30 min to ensure an accurate measurement. Additional blank samples (no isotopes present in solution) were run to ensure no background subtraction was necessary. Once the data was collected the total counts for each isotope in the samples was normalized to give the average counts per minute.

Slope Analysis

To determine the stoichiometry of the metal–ligand complex a slope analysis experiment was conducted. This experiment made use of the $^{152/154}$ Eu radiotracer employing similar techniques to those discussed in the above section discussing use of Eu and Am radiotracers. In this experiment the concentration of the Me-BTPhen was varied in [C₄mim][NTf₂]; the concentrations were as follows: 1.0, 3.0, 10.0, and 30.0 mM. The IL solutions were then contacted with 0.1 M HNO₃ which had been spiked with a 10 µL solution containing 50 µCi/mL of the Eu radiotracer.



Figure S3. Slope Analysis of Me-BTPhen with Eu extracted from 0.1 M HNO₃. The y-axis represents the $\log D_{Eu}$ and the x-axis is the $\log [Me-BTPhen]$.

From Figure S3 above, the trend line associated with the D_{Eu} for Me-BTPhen has a slope of two indicating that the ligand-metal complex is likely 2-to-1 during the extraction process. These results correlate with the solutions structural data presented and discussed in the manuscript. Additionally the 2-to-1 structure has been shown to exist in crystal structures in the previous reported examples in the literature.²

Quantification of Extracted Nitrate

The amount of nitrate extracted was determined by Ion Chromatography (IC) using conductivity detection. The IC used was a Dionex ICS-5000+ Reagent-Free High-Pressure IC

system, the detector was a conductivity detector, the columns used were a Thermo Scientific Dionex IonPac AS11-HC Hydroxide-Selective Anion-Exchange Column (4 x 250 mm) with a IonPac AG11-HC Guard Column (4 x 50 mm). The program used to run the instrument and process the data is Chromeleon version 6.2. The eluent used in the IC for the samples was 30 mM KOH, using a flow rate of 1.5 mL/min with the conductivity set at 112 mA. A calibration curve was made by dilution a purchased 1000 ppm nitrate standard purchased from Inorganic Ventures.

Section S3. Computational Methods

DFT calculations were performed on a series of Eu-Me-BTPhen and Am-Me-BTPhen complexes using the Gaussian 09,³ Revision D01, software package at the B3LYP^{4, 5} level of theory. A standard 6-311+G** basis sets were used for all light atoms. F-elements were modelled using the large-core (LC) relativistic effective core potential (RECP) and the associated (7s6p5d2f) /[5s4p3d2f]^{6, 7} basis sets. Since LC RECP calculations include the 4f electrons in the core, they were performed on a pseudo singlet state configuration. Solvent corrections we included as single-point energies using a generic ionic liquid implicit solvation model, SMD-GIL,⁸ with generic values of solvent descriptors. Cartesian coordinates of all the ligand-metal ion complexes accompanied their electronic energies obtained by at the B3LYP/LC(7s6p5d2f)/[5s4p3d2f]/6-311+G** level are given in Section S5.

Comparison of the relative stability of 2:1 ligand-metal ion complexes with monodentate and chelate coordination of NTf_2^- indicated that the chelate form with Eu^{3+} and Am^{3+} was 0.13–2.3 kcal/mol more stable than the monodentate form in the gas phase, but it was 5.4–5.7 kcal/mol less stable in the presence of implicit solvent. With the NO₃⁻ anion, we could locate only the Eu(III) and Am(III) complexes in the chelate mode, whereas the complex in the monodentate mode were not a stationary point on the potential energy surface, which converged to the chelating species during the geometry optimization. Eu/Am-ligand bond distances for the DFT optimized 2:1 ligand-metal ion complexes are listed in Table S1

	No Anion	OH-	H ₂ O	NTf ₂ -	NO ₃ -	NTf ₂ -
				(mono)	(chelate)	(chelate)
O _{Anion-1}		2.126/2.194	2.570/2.692	2.436/2.492	2.537/2.584	2.551/2.602
O _{Anion-2}					2.539/2.587	2.568/2.614
N _{Phen-1}	2.563/2.634	2.637/2.696	2.571/2.635	2.582/2.645	2.633/2.698	2.630/2.667
N _{Phen-2}	2.564/2.634	2.638/2.697	2.594/2.636	2.591/2.655	2.637/2.699	2.667/2.691
N _{Phen-3}	2.564/2.634	2.640/2.706	2.595/2.661	2.593/2.659	2.637/2.718	2.605/2.714
N _{Phen-4}	2.564/2.635	2.646/2.711	2.611/2.662	2.599/2.664	2.663/2.719	2.659/2.721
NAzide-1	2.587/2.635	2.703/2.738	2.611/2.656	2.636/2.680	2.652/2.695	2.658/2.700
NAzide-2	2.587/2.636	2.706/2.748	2.646/2.657	2.640/2.686	2.652/2.695	2.684/2.726
N _{Azide-3}	2.588/2.636	2.712/2.752	2.657/2.702	2.656/2.700	2.684/2.729	2.705/2.737
N _{Azide-4}	2.588/2.637	2.714/2.758	2.657/2.703	2.656/2.701	2.685/2.730	2.713/2.746
Average	2.576/2.635	2.614/2.667	2.599/2.667	2.599/2.654	2.635/2.685	2.644/2.692
Coord. #	8	9	9	9	10	10

Table S1. Eu/Am-Ligand Bond Distances in Å for Geometrically Optimized Molecules as Determined by DFT Calculations

Section S4. X-ray Absorption Fine Structure Spectroscopy

Data Collection and Processing

XAFS samples were prepared by mixing 2 mL of a 15 mM Me-BTPhen dissolved in $[C_4mim][NTf_2]$ with an equal volume of an aqueous solution of 15 mM Eu(NO₃)₃. The samples were mixed, then centrifuged to separate the two phases, then the top aqueous layer was removed by pipette and a fresh solution of Eu(NO₃)₃ was added to the IL solution and the process was repeated once more. The sample was contacted twice to ensure complete loading of the Me-BTPhen. After the second contact, 1.5 mL of Eu-loaded ionic liquid solution was pipetted into a screw-cap sample vial.

X-ray absorption data were collected at beamline 11-2 at Stanford Synchrotron Radiation Lightsource. Samples were inserted into a plexiglass sample box which was purged with He throughout the duration of the experiment. Spectra were collected at ambient temperature and pressures at the europium L₃-edge (6.977 keV) with fluorescence detection afforded by a Canberra 100-pixel Ge solid-state monolith detector. Sollar slits were used to decrease noise from x-ray scattering. Signal intensity was investigated both with and without use of a Cr-filter, with superior signal afforded when the filter was not used. A Eu(NO₃)₃ reference sample was measured simultaneously for energy calibration and data alignment. Samples were positioned on the beam to maximize fluorescence signal. The beam dimensions were 2×5 mm for all scans. Data were collected over four regions: -230 to -30 eV (10 eV step size, dwell time of 0.25 seconds), -30 to -5 eV (5 eV step size, dwell time of 0.5 seconds), -5 to 30 eV (1 eV step size), 3 Å⁻¹ to 13 Å⁻¹ (0.05 Å⁻¹ step size), with dwell time increasing as a function of k from 2 seconds at 3 Å⁻¹ to 16 seconds at 13 Å⁻¹.

Data were not collected further in k-space due to the occurrence of the Eu L_{II}-edge (7.617 keV, 12.95 Å⁻¹). Nine scans were collected.

The data were reformatted using SixPack⁹ then processed and analyzed using the Athena and Artemis programs of the IFEFFIT package based on FEFF 6.^{10, 11} Reference foil data were aligned to the first zero-crossing of the second derivative of the normalized $\mu(E)$ data, which was subsequently calibrated to the literature E₀ for the europium L_{III}-edge (6.977 keV). Any contributions from the Eu L_{II}-pre-edge were removed by truncating all data sets at 7.578 keV. Spectra were averaged in $\mu(E)$ prior to normalization. The background was removed and the data were assigned an Rbkg value of 1.1, less than one-half the value of the half-path length for the nearest scattering element, prior to normalizing to obtain a unit edge step.





Figure S4. The first six principal components derived from nine Eu L_{III}-edge k^2 -weighted EXAFS spectra of a Eu solution with 7.5 mM Me-BTPhen in an ionic liquid solvent. Components 7-9 (not shown) are similar to components 2 – 6. These data reveal only one mathematical component in the EXAFS spectra, indicating an adequate fit should be achievable from a single appropriate structure model.

EXAFS Analysis

The processed data set was initially qualitatively compared against scattering paths generated from DFT-derived potential structure models, as displayed in Figure S5. Inspection of the simulated direct scattering paths reveals all models display reasonable scattering paths for the majority of features, but that with the exception of $[Eu(Me-BTPhen)_2(OH)]^{2+}$, none are immediately capable of adequately fitting the shoulder observed at 1.5 Å. Of the remaining structure models, Eu-bound H₂O and monodentate-bound NTf₂⁻ both possess a single scattering



Figure S5. Single scattering paths for potential structure models compared against experimental EXAFS data, ordered by increasing coordination number and first-shell bond length. Representative DFT-calculated structures are provided in the upper corner of their corresponding spectrum. Top row, left to right: No anion, OH⁻; second row, left to right: H₂O, NTf₂ (monodentate); third row, left to right: NO₃ (chelate), NTf₂ (chelate);bottom row: Eu2(Phen)2(OH)2 dimer.

path capable of shifting to fit this shoulder without inducing physically unreasonable distortions in the Me-BTPhen coordinating ligands. Accordingly, structure models generated from chelating NO₃⁻ and NTf₂⁻ were discarded. Notably, both of these discarded models also would require a first coordination shell of 10 atoms. The final model was intended to interrogate whether the large feature at 4 Å could be attributed to formation of dimeric Eu species. A search of the CCDC database reveald that while no 4:2 Me-BTPhen:M complexes had been reported, several related 4:2 phenanthroline (Phen):Eu complexes were known. However, inspection of the Eu-Eu scattering path (3.73 Å) for this complex (plotted in red in Figure S5) was not located in a position that would beneficially contribute to a fit of the experimental spectrum. Additionally, as BTPhen possesses more *N*-donor atoms than Phen it would completely or (near completely) fill the Eu first coordination sphere, discouraging the prospects of forming a dimeric species.

The EXAFS data was initially fit with k-weighting of 1,2, and 3, then finalized with k^2 weighting in R-space. Structural parameters that were determined by the fits were the change in Ref (ΔR_i), the relative mean square displacement of the scattering element (σ^2_i), the passive electron reduction factor (S_0^2), and the energy shift of the photoelectron, (ΔE_0). The data range used for fitting was 1.15 - 3.4 Å in R-space and 3 - 10.3 Å⁻¹ in k-space, affording 10 independent points, with the range identified to minimize truncation effects in k-space and avoid spectral contributions from the artifact of the Fourier transform at approximately 0.8 Å in Rspace. The number of variables was not permitted to exceed 2/3 the number of independent points in keeping with the Nyquist criterion.^{12, 13} Preliminary models were constructed from all first and second shell single scattering paths, as well as all multiple scattering paths providing at least 15% of the contribution of the most intense single scattering peak and with half-path length les s than 3.7 Å. We asserted any changes in half path length would be attributable to translation of the Me-BTPhen with respect to Eu, rather than distortion of bond lengths within the rigid and electronically delocalized ligand. Therefore, scattering path lengths were fitted with only two parameters (one for translation of the phenanthroline portion, the second for translation of the triazine), while different mean square relative deviation parameters (σ^2) were afforded for each of the first two shells of coordinating atoms (Figure S6).



Figure S6. Assignment of ΔR and σ^2 parameters for the Me-BTPhen ligands.



Figure S7. (Left) Rejected XAFS fit from structure model $[Eu(Me-BTPhen)_2]^{3+}$. (Right) DFT-optimized structure model used for preparation of the structure model.



Figure S8. (Left) Rejected XAFS fit from structure model [Eu(Me-BTPhen)₂NO₃]²⁺. (Right) DFT-optimized structure model used for preparation of the structure model.



Figure S9. (Left) Rejected XAFS fit from structure model [Eu(Me-BTPhen)₂OH]²⁺. (Right) DFT-optimized structure model used for preparation of the structure model.



Figure S10. (Left) Rejected XAFS fit from structure model [Eu(Me-BTPhen)₂NTf₂]²⁺. (Right) DFT-optimized structure model used for preparation of the structure model.



Figure S11. k^2 -weighted $\chi(k)$ EXAFS data and fit for [Eu(Me-BTPhen)₂(H₂O)]³⁺.

Section S5. Cartesian coordinates and electronic energies of ligand-metal ion complexes obtained at the B3LYP/LC(7s6p5d2f)/[5s4p3d2f]/6-311+G** level.

 $[Eu(Me-BTPhen)_2]^{3+}; E(RB3LYP) = -2609.82409307$ A.U.

7.9366 12.8739 1.0053 Eu 11.1662 13.4254 -0.0923 Ν 9.9680 14.0239 -0.1119 Ν 10.8378 15.8637 -1.3138 Ν 7.5136 15.1480 -0.1005 Ν Ν 5.5238 13.7303 1.1453 Ν 6.2309 11.3628 2.2308 Ν 6.6310 10.1961 2.7532 Ν 4.0132 11.0015 2.9636 6.3052 12.4989 -2.0155 Ν Ν 7.2223 11.8693 -1.2696 7.4409 10.1511 -2.8775 Ν Ν 9.0638 10.6801 0.3039 Ν 9.6501 11.9411 2.6676 8.3223 14.2376 3.1706 Ν Ν 7.6387 15.3712 3.3752 Ν 9.4534 14.4746 5.2322 С 6.5360 10.7532 -3.6352 С 5.9456 11.9797 -3.1812 С 7.7574 10.7301 -1.7139 С 8.7714 10.0721 -0.8507 С 9.3805 8.8704 -1.2501 Η 9.1032 8.4247 -2.1950 С 10.3183 8.2946 -0.4201 Η 10.8048 7.3676 -0.7027 С 10.6489 8.9118 0.8034 С 9.9829 10.1222 1.1258

С 11.6160 8.3638 1.7089 Η 12.1120 7.4382 1.4411 11.9072 8.9893 2.8801 С Η 12.6392 8.5703 3.5607 10.2922 10.7871 2.3716 С С 11.2550 10.2137 3.2418 С 11.5280 10.8929 4.4464 Η 12.2573 10.4869 5.1386 С 10.8685 12.0672 4.7388 11.0532 12.6136 5.6530 Η С 9.9303 12.5633 3.8178 С 9.1974 13.8260 4.0907 С 7.8436 16.0637 4.4871 С 8.7911 15.5993 5.4596 С 12.0336 15.2935 -1.3100 С 12.2008 14.0199 -0.6702 С 9.8383 15.2088 -0.7125 С 8.4901 15.8317 -0.7068 С 8.2759 17.0800 -1.3154 Η 9.1039 17.5864 -1.7912 7.0096 17.6234 -1.2865 С 6.8155 18.5854 -1.7476 Η С 5.9592 16.9271 -0.6551 С 6.2665 15.6727 -0.0677 С 4.6213 17.4384 -0.5917 Η 4.4103 18.3990 -1.0470 3.6350 16.7356 0.0254 С 2.6253 17.1271 0.0698 Η С 5.2177 14.9255 0.5887 С 3.9059 15.4640 0.6298 С 2.9115 14.7036 1.2779 Η 1.8962 15.0805 1.3301 С 3.2331 13.4874 1.8409 2.4959 12.8785 2.3452 Η С 4.5596 13.0317 1.7540 С 4.9527 11.7286 2.3486 С 5.7506 9.4259 3.3773 С 4.3818 9.8425 3.4888 С 13.5193 13.3153 -0.6248 13.4231 12.3733 -0.0881 Η 13.8889 13.1122 -1.6344 Η Η 14.2757 13.9310 -0.1291 С 13.1667 15.9999 -1.9741 12.8280 16.9433 -2.3970 Η Η 13.9742 16.1940 -1.2606 Η 13.5963 15.3818 -2.7693

С	6.2417	8.1326 3.9458
Η	7.3049	8.0145 3.7450
Η	6.0783	8.0934 5.0270
Η	5.7022	7.2847 3.5134
С	3.3552	9.0093 4.1785
Η	2.3898	9.5109 4.1648
Η	3.2597	8.0325 3.6925
Η	3.6460	8.8132 5.2156
С	4.9193	12.7180 -3.9806
Η	4.6037	13.6138 -3.4490
Η	5.3180	13.0073 -4.9576
Η	4.0434	12.0902 -4.1698
С	6.1631	10.1306 -4.9379
Η	6.7265	9.2130 -5.0936
Η	5.0920	9.9049 -4.9668
Η	6.3582	10.8189 -5.7668
С	7.0577	17.3226 4.6742
Η	6.4068	17.4891 3.8179
Η	6.4461	17.2711 5.5799
Η	7.7211	18.1849 4.7891
С	9.0573	16.3445 6.7234
Η	9.8013	15.8219 7.3208
Η	9.4147	17.3569 6.5082
Η	8.1388	16.4571 7.3088

 $[Am(Me-BTPhen)_2]^{3+}; E(RB3LYP) = -2608.30226125$ A.U.

Am 7.9375 12.8758 1.0084 11.1607 13.5193 -0.1624 Ν 9.9554 14.1048 -0.1620 Ν 10.7873 15.9647 -1.3565 Ν 7.4752 15.2116 -0.1175 Ν Ν 5.4676 13.7828 1.1360 6.1366 11.3899 2.2311 Ν 6.5131 10.2164 2.7569 Ν Ν 3.9075 11.0663 2.9413 Ν 6.3448 12.3933 -2.0550 7.2618 11.7828 -1.2925 Ν Ν 7.5131 10.0456 -2.8725 9.1144 10.6223 0.3175 Ν 9.7048 11.8973 2.6983 Ν Ν 8.3832 14.2079 3.2390 7.7073 15.3430 3.4637 Ν Ν 9.5363 14.4129 5.2900

С 6.6093 10.6270 -3.6466 С 6.0013 11.8541 -3.2161 7.8134 10.6443 -1.7138 С С 8.8280 10.0052 -0.8344 С 9.4433 8.8039 -1.2260 9.1698 8.3518 -2.1689 Η С 10.3811 8.2377 -0.3910 Η 10.8729 7.3113 -0.6662 10.7059 8.8635 0.8293 С С 10.0347 10.0741 1.1465 11.6744 8.3173 1.7334 С 12.1711 7.3921 1.4650 Η С 11.9652 8.9443 2.9028 12.6982 8.5285 3.5844 Η 10.3445 10.7427 2.3942 С С 11.3105 10.1674 3.2619 С 11.5888 10.8420 4.4677 12.3204 10.4321 5.1550 Η 10.9318 12.0150 4.7666 С Η 11.1201 12.5582 5.6819 С 9.9906 12.5146 3.8505 С 9.2649 13.7792 4.1428 С 7.9274 16.0203 4.5815 С 8.8830 15.5378 5.5381 С 11.9894 15.4095 -1.3723 С 12.1805 14.1310 -0.7478 С 9.8022 15.2920 -0.7495 С 8.4464 15.9020 -0.7259 С 8.2266 17.1530 -1.3273 9.0511 17.6637 -1.8044 Η С 6.9598 17.6922 -1.2896 Η 6.7598 18.6558 -1.7447 С 5.9146 16.9896 -0.6573 С 6.2251 15.7317 -0.0757 С 4.5788 17.5047 -0.5916 Η 4.3709 18.4676 -1.0435 3.5921 16.8021 0.0231 С 2.5828 17.1941 0.0703 Η С 5.1727 14.9826 0.5807 С 3.8623 15.5282 0.6217 С 2.8609 14.7735 1.2649 Η 1.8484 15.1584 1.3150 С 3.1718 13.5540 1.8246 Η 2.4291 12.9486 2.3249 С 4.4955 13.0897 1.7398 С 4.8654 11.7786 2.3360

С 5.6142 9.4612 3.3718 С 4.2511 9.9012 3.4690 С 13.5077 13.4418 -0.7274 Η 13.4283 12.4899 -0.2055 13.8694 13.2605 -1.7439 Η 14.2614 14.0586 -0.2288 Η С 13.1060 16.1360 -2.0427 Η 12.7522 17.0814 -2.4486 13.9218 16.3288 -1.3381 Η Η 13.5302 15.5325 -2.8520 6.0777 8.1600 3.9456 С 7.1418 8.0261 3.7594 Η 5.8983 8.1220 5.0242 Η 5.5316 7.3209 3.5043 Η С 3.2038 9.0844 4.1472 2.2464 9.6008 4.1223 Η Η 3.0989 8.1088 3.6608 3.4798 8.8849 5.1877 Η С 4.9750 12.5701 -4.0354 Η 4.6453 13.4701 -3.5195 5.3800 12.8496 -5.0127 Η 4.1074 11.9305 -4.2235 Η С 6.2550 9.9820 -4.9437 6.8307 9.0692 -5.0818 Η 5.1870 9.7429 -4.9784 Η Η 6.4489 10.6611 -5.7804 7.1509 17.2811 4.7927 С H 6.4910 17.4613 3.9459 6.5496 17.2219 5.7049 Η Η 7.8205 18.1383 4.9102 С 9.1662 16.2661 6.8081 H 9.9144 15.7329 7.3910 Η 9.5256 17.2796 6.6016 Н 8.2546 16.3751 7.4048

 $[Eu(Me-BTPhen)_2(NO_3)]^{2+}$ chelate; E(RB3LYP) = -2890.57968384 A.U.

Eu 8.1641 12.8991 0.4133 8.2836 13.7758 -1.9644 0 0 9.6976 12.2153 -1.4909 9.6731 13.0764 -3.4996 0 11.4990 13.9379 0.4684 Ν 10.2906 14.4803 0.3105 Ν Ν 11.2394 16.5082 -0.4496 7.8095 15.5119 0.0378 Ν Ν 5.7440 13.8859 0.7629

6.4133 11.5291 1.9177 Ν 6.7879 10.4355 2.5832 Ν 4.1410 11.1248 2.4392 Ν Ν 5.6915 11.9543 -1.8660 Ν 6.7109 11.3925 -1.2145 6.5940 9.4657 -2.5799 Ν 8.7612 10.3148 0.1729 Ν 9.7484 11.8182 2.2235 Ν 8.4039 14.0969 2.8042 Ν 7.6463 15.1466 3.1261 Ν 9.7299 14.4011 4.7402 Ν 9.2396 13.0250 -2.3737 Ν С 5.6036 10.0188 -3.2609 С 5.1295 11.3120 -2.8809 С 7.1260 10.1782 -1.5769 С 8.2593 9.5876 -0.8222 8.7722 8.3255 -1.1771 С 8.3228 7.7833 -1.9969 Η 9.8446 7.8255 -0.4741 С Η 10.2684 6.8604 -0.7280 10.3976 8.5745 0.5839 С С 9.8083 9.8328 0.8682 С 11.5026 8.1099 1.3695 11.9464 7.1518 1.1247 Η С 11.9787 8.8460 2.4083 Η 12.8056 8.4828 3.0076 С 10.3275 10.6269 1.9608 С 11.3949 10.1117 2.7417 С 11.8251 10.8823 3.8394 12.6409 10.5260 4.4586 Η С 11.1897 12.0674 4.1336 Η 11.4678 12.6678 4.9880 С 10.1447 12.5015 3.2976 С 9.3926 13.7382 3.6245 С 7.8986 15.8242 4.2388 С 9.0033 15.4578 5.0653 С 12.4460 15.9790 -0.3293 С 12.5785 14.6411 0.1547 С 10.1920 15.7338 -0.1336 С 8.8292 16.2986 -0.2971 С 8.6525 17.5995 -0.8055 Η 9.5208 18.1901 -1.0609 7.3723 18.0755 -0.9754 С 7.2022 19.0695 -1.3733 Η С 6.2732 17.2636 -0.6304 С 6.5538 15.9702 -0.1210

С 4.9142 17.6987 -0.7638 Η 4.7229 18.6852 -1.1700 3.8845 16.8995 -0.3779 С Η 2.8594 17.2399 -0.4679 5.4578 15.1103 0.2705 С С 4.1284 15.5956 0.1646 С 3.0941 14.7570 0.6239 Η 2.0639 15.0888 0.5567 С 3.4005 13.5354 1.1795 Η 2.6380 12.8806 1.5764 С 4.7482 13.1353 1.2348 С 5.1207 11.8563 1.8888 С 5.8736 9.6860 3.1862 С 4.4906 10.0258 3.0873 С 13.9079 13.9740 0.3257 13.7680 12.9649 0.7096 Η Η 14.4383 13.9154 -0.6295 14.5474 14.5343 1.0141 Η 13.6321 16.8066 -0.7064 С Η 13.3131 17.8014 -1.0111 14.3337 16.8906 0.1290 Η 14.1783 16.3389 -1.5320 Η С 6.3573 8.4887 3.9454 Η 7.4425 8.4257 3.8855 6.0640 8.5465 4.9978 Η Η 5.9258 7.5666 3.5447 С 3.4184 9.1805 3.6964 Η 2.4407 9.6158 3.4989 3.4476 8.1634 3.2931 Η 3.5601 9.0939 4.7782 Η С 4.0054 11.9988 -3.5925 Η 3.8033 12.9600 -3.1232 Η 4.2580 12.1684 -4.6436 3.0954 11.3920 -3.5750 Η С 5.0138 9.2635 -4.4078 Η 5.5103 8.3013 -4.5173 Η 3.9415 9.1019 -4.2605 5.1228 9.8305 -5.3378 Η С 6.9943 16.9726 4.5657 6.2211 17.0641 3.8046 Η 6.5170 16.8313 5.5399 Η Η 7.5530 17.9119 4.6181 С 9.3675 16.2280 6.2938 10.2503 15.7945 6.7598 Η Η 9.5650 17.2771 6.0529 Η 8.5433 16.2207 7.0142

 $[Am(Me-BTPhen)_2(NO_3)]^{2+}$ chelate; E(RB3LYP) = -2889.06067191 A.U.

8.1607 12.8999 0.4219 Am 8.2915 13.7857 -2.0023 0 9.7036 12.2139 -1.5378 0 9.6708 13.0806 -3.5431 0 11.5040 14.0210 0.4544 Ν 10.2884 14.5491 0.3007 Ν 11.2069 16.5792 -0.4856 Ν 7.7823 15.5638 0.0340 Ν Ν 5.6961 13.9376 0.7776 6.3279 11.5509 1.9286 Ν Ν 6.6786 10.4404 2.5795 4.0459 11.1815 2.4266 Ν 5.6947 11.8608 -1.8753 Ν Ν 6.7215 11.3184 -1.2187 6.6507 9.3909 -2.5835 Ν 8.7943 10.2665 0.1857 Ν Ν 9.7794 11.7710 2.2637 8.4573 14.0747 2.8679 Ν 7.7212 15.1377 3.1972 Ν 9.8013 14.3470 4.7946 Ν 9.2425 13.0291 -2.4149 Ν 5.6533 9.9243 -3.2701 С С 5.1511 11.2079 -2.8936 С 7.1636 10.1134 -1.5774 С 8.3037 9.5395 -0.8161 С 8.8294 8.2839 -1.1750 Η 8.3899 7.7424 -2.0004 9.9005 7.7892 -0.4675 С Η 10.3342 6.8290 -0.7233 10.4403 8.5361 0.5983 С С 9.8415 9.7900 0.8872 С 11.5442 8.0690 1.3832 Η 11.9911 7.1138 1.1323 С 12.0155 8.7998 2.4269 12.8426 8.4367 3.0261 Η С 10.3564 10.5801 1.9889 С 11.4268 10.0612 2.7652 С 11.8611 10.8247 3.8660 Η 12.6785 10.4630 4.4799 С 11.2308 12.0098 4.1683 11.5152 12.6060 5.0234 Η С 10.1836 12.4503 3.3383 С 9.4464 13.6953 3.6772

С 7.9939 15.8070 4.3100 9.0954 15.4146 5.1290 С С 12.4204 16.0650 -0.3698 С 12.5729 14.7331 0.1251 С 10.1702 15.7962 -0.1546 С 8.7999 16.3490 -0.3135 С 8.6196 17.6461 -0.8301 Η 9.4857 18.2350 -1.0961 С 7.3385 18.1202 -0.9934 7.1648 19.1114 -1.3970 Η 6.2420 17.3113 -0.6348 С С 6.5236 16.0196 -0.1185 С 4.8854 17.7527 -0.7661 4.6982 18.7382 -1.1766 Η С 3.8539 16.9598 -0.3746 2.8296 17.3026 -0.4637 Η С 5.4226 15.1646 0.2812 С 4.0951 15.6576 0.1717 С 3.0536 14.8273 0.6292 Η 2.0264 15.1677 0.5587 3.3479 13.6030 1.1838 С 2.5787 12.9533 1.5758 Η С 4.6923 13.1926 1.2431 С 5.0425 11.9019 1.8911 С 5.7462 9.6986 3.1637 С 4.3708 10.0679 3.0626 С 13.9113 14.0829 0.2906 13.7864 13.0756 0.6842 Η 14.4342 14.0224 -0.6686 Η 14.5499 14.6565 0.9688 Η С 13.5938 16.9028 -0.7639 13.2610 17.8917 -1.0731 Η Η 14.3021 17.0008 0.0642 14.1374 16.4349 -1.5912 Η С 6.2019 8.4793 3.9049 7.2862 8.3971 3.8506 Η 5.9028 8.5243 4.9563 Η 5.7559 7.5727 3.4852 Η С 3.2791 9.2369 3.6567 2.3111 9.6932 3.4590 Η 3.2909 8.2238 3.2428 Η Η 3.4133 9.1362 4.7384 4.0178 11.8727 -3.6114 С 3.7967 12.8311 -3.1449 Η Η 4.2714 12.0443 -4.6618 Η 3.1186 11.2499 -3.5958

С	5.0852	9.1574 -4.4204
Η	5.5991	8.2038 -4.5248
Η	4.0148	8.9776 -4.2810
Η	5.1917	9.7249 -5.3503
С	7.1154	16.9729 4.6452
Η	6.3389	17.0810 3.8897
Η	6.6423	16.8397 5.6226
Η	7.6933	17.9007 4.6948
С	9.4782	16.1705 6.3608
Η	10.3572	15.7208 6.8186
Η	9.6891	17.2184 6.1264
Η	8.6581	16.1694 7.0860

 $[Eu(Me-BTPhen)_2(NTf_2)]^{2+}$ chelate; E(RB3LYP) = -4437.69912506 A.U.

Eu 8.1891 12.9165 0.3557 0 8.0396 13.7438 -2.0524 10.1765 12.2150 -1.1108 0 11.5167 14.1364 0.8340 Ν Ν 10.3142 14.5985 0.4867 11.2684 16.5968 -0.3480 Ν Ν 7.8440 15.5367 -0.0041 Ν 5.7924 13.9208 0.7586 6.4726 11.5716 1.9210 Ν 6.8528 10.4803 2.5863 Ν Ν 4.2138 11.2045 2.5232 5.4508 12.0284 -1.7406 Ν Ν 6.5173 11.4495 -1.1832 6.3141 9.5788 -2.6192 Ν Ν 8.6227 10.3114 0.0455 Ν 9.7027 11.7258 2.1095 Ν 8.4286 14.0152 2.7646 Ν 7.7065 15.0800 3.1130 Ν 9.7710 14.2298 4.7005 С 5.2983 10.1664 -3.2276 С 4.8330 11.4295 -2.7474 С 6.8972 10.2476 -1.6153 С 8.0480 9.6108 -0.9285 С 8.4878 8.3266 -1.3045 Η 7.9837 7.8109 -2.1096 С 9.5564 7.7723 -0.6385 Η 9.9244 6.7889 -0.9091 С 10.1750 8.4854 0.4094 С 9.6596 9.7696 0.7152 С 11.2720 7.9588 1.1664 11.6646 6.9838 0.9014 Η

С 11.8012 8.6570 2.2061 Η 12.6200 8.2463 2.7856 10.2312 10.5193 1.8108 С С 11.2794 9.9407 2.5714 С 11.7451 10.6637 3.6877 12.5491 10.2582 4.2919 Η С 11.1542 11.8610 4.0217 Η 11.4555 12.4236 4.8938 С 10.1247 12.3597 3.2028 С 9.4115 13.6071 3.5697 С 7.9863 15.7244 4.2392 С 9.0791 15.3010 5.0537 С 12.4733 16.1210 -0.0858 С 12.5960 14.8572 0.5698 С 10.2230 15.8139 -0.0502 С 8.8606 16.3370 -0.3150 С 8.6785 17.6269 -0.8501 Η 9.5459 18.2244 -1.0922 7.3966 18.0844 -1.0530 С Η 7.2225 19.0704 -1.4691 6.2998 17.2696 -0.7047 С С 6.5855 15.9852 -0.1780 С 4.9390 17.6966 -0.8435 4.7426 18.6750 -1.2667 Η 3.9140 16.9024 -0.4351 С Η 2.8874 17.2386 -0.5242 5.4962 15.1309 0.2372 С С 4.1654 15.6125 0.1361 С 3.1387 14.7877 0.6356 2.1075 15.1180 0.5769 Η С 3.4553 13.5826 1.2199 Η 2.7002 12.9387 1.6475 С 4.8037 13.1831 1.2625 С 5.1837 11.9146 1.9294 С 5.9493 9.7497 3.2273 С 4.5692 10.1089 3.1737 С 13.9214 14.2835 0.9626 13.7770 13.3344 1.4761 Η Η 14.5412 14.1129 0.0770 14.4736 14.9653 1.6157 Η 13.6663 16.9267 -0.4836 С Η 13.3516 17.8708 -0.9240 14.3148 17.1239 0.3750 Η 14.2659 16.3745 -1.2149 Η С 6.4414 8.5519 3.9806 Η 7.5239 8.4768 3.8921

6.1768 8.6182 5.0401 Η Η 5.9899 7.6320 3.5968 3.5068 9.2886 3.8321 С Η 2.5288 9.7336 3.6591 3.5098 8.2644 3.4463 Η 3.6829 9.2191 4.9102 Η С 3.6634 12.1354 -3.3572 Η 3.4673 13.0616 -2.8196 Η 3.8676 12.3749 -4.4051 2.7657 11.5109 -3.3370 Η С 4.6747 9.4863 -4.4017 5.1521 8.5238 -4.5748 Η 3.6015 9.3377 -4.2496 Η 4.7864 10.1060 -5.2975 Η С 7.1223 16.8952 4.5946 6.3478 17.0268 3.8408 Η 6.6469 16.7508 5.5693 Η 7.7117 17.8147 4.6609 Η С 9.4680 16.0269 6.3015 Η 10.3405 15.5576 6.7519 9.6917 17.0768 6.0889 Η 8.6463 16.0210 7.0249 Η Ν 9.7887 12.6813 -3.6116 10.8295 12.6214 -2.3897 S 11.7849 13.7188 -2.3088 0 S 8.3565 13.3655 -3.4634 7.3419 12.6465 -4.2139 0 С 11.8024 11.0919 -2.9224 F 10.9886 10.0418 -3.0197 F 12.4014 11.3112 -4.0782 12.7132 10.8516 -1.9760 F С 8.5372 15.0413 -4.3208 9.4963 15.7474 -3.7180 F F 8.8411 14.8660 -5.5946 F 7.3798 15.6993 -4.2156

 $[Am(Me-BTPhen)_2(NTf_2)]^{2+}$ chelate; E(RB3LYP) = -4436.18276125 A.U.

8.1811 12.9193 0.3768 Am 8.0660 13.7550 -2.0846 0 10.2141 12.2051 -1.1022 0 11.4982 14.2369 0.8350 Ν 10.2881 14.6765 0.4838 Ν 11.2100 16.6652 -0.4039 Ν Ν 7.7932 15.5853 -0.0058 5.7327 13.9612 0.7802 Ν

6.3926 11.5928 1.9494 Ν Ν 6.7575 10.4934 2.6108 Ν 4.1268 11.2480 2.5337 Ν 5.4652 11.9853 -1.7623 6.5249 11.4067 -1.1913 N 6.3506 9.5413 -2.6355 Ν Ν 8.6270 10.2601 0.0670 9.7217 11.6644 2.1559 Ν 8.4870 13.9824 2.8397 Ν Ν 7.7932 15.0630 3.1982 9.8609 14.1634 4.7551 Ν С 5.3424 10.1281 -3.2573 С 4.8646 11.3880 -2.7801 С 6.9166 10.2078 -1.6199 С 8.0605 9.5688 -0.9194 С 8.5018 8.2874 -1.3026 8.0048 7.7790 -2.1165 Η С 9.5633 7.7266 -0.6315 9.9325 6.7449 -0.9064 Η С 10.1755 8.4309 0.4256 С 9.6602 9.7141 0.7405 С 11.2688 7.8924 1.1790 Η 11.6562 6.9175 0.9062 11.8019 8.5806 2.2224 С 12.6198 8.1633 2.7984 Η С 10.2366 10.4545 1.8439 С 11.2871 9.8647 2.5946 С 11.7691 10.5787 3.7096 Η 12.5751 10.1629 4.3041 11.1953 11.7813 4.0528 С 11.5126 12.3387 4.9225 Η С 10.1625 12.2924 3.2458 С 9.4737 13.5517 3.6269 С 8.1030 15.6989 4.3211 С 9.1962 15.2477 5.1206 С 12.4226 16.2091 -0.1412 С 12.5664 14.9651 0.5473 С 10.1758 15.8784 -0.0776 С 8.8043 16.3844 -0.3394 С 8.6124 17.6636 -0.8958 9.4748 18.2608 -1.1554 Η С 7.3269 18.1115 -1.0952 7.1446 19.0897 -1.5259 Η С 6.2365 17.2978 -0.7260 6.5299 16.0221 -0.1800 С С 4.8751 17.7209 -0.8678

4.6765 18.6934 -1.3032 Η С 3.8529 16.9293 -0.4491 Η 2.8251 17.2607 -0.5416 С 5.4410 15.1682 0.2470 С 4.1089 15.6466 0.1352 С 3.0799 14.8245 0.6342 Η 2.0492 15.1547 0.5672 С 3.3921 13.6219 1.2249 2.6339 12.9794 1.6485 Η С 4.7405 13.2232 1.2783 С 5.1081 11.9500 1.9481 С 5.8417 9.7694 3.2414 С 4.4662 10.1460 3.1819 С 13.9013 14.4198 0.9478 Η 13.7723 13.4841 1.4892 14.5178 14.2318 0.0635 Η Η 14.4485 15.1270 1.5777 С 13.6024 17.0166 -0.5733 Η 13.2722 17.9366 -1.0516 Η 14.2471 17.2608 0.2764 Η 14.2115 16.4446 -1.2809 С 6.3150 8.5608 3.9892 Η 7.3969 8.4728 3.9048 Η 6.0466 8.6238 5.0480 Η 5.8537 7.6492 3.5975 С 3.3909 9.3364 3.8326 2.4191 9.7938 3.6566 Η 3.3833 8.3135 3.4435 Η 3.5611 9.2614 4.9113 Η С 3.7017 12.0917 -3.4049 3.4931 13.0142 -2.8657 Η 3.9213 12.3373 -4.4483 Η Η 2.8065 11.4632 -3.4017 С 4.7398 9.4506 -4.4438 5.2234 8.4905 -4.6134 Η Η 3.6650 9.2974 -4.3090 Η 4.8632 10.0744 -5.3352 С 7.2719 16.8894 4.6889 Η 6.4928 17.0415 3.9437 6.8034 16.7540 5.6683 Η Η 7.8841 17.7941 4.7511 С 9.6152 15.9596 6.3667 Η 10.4813 15.4685 6.8061 9.8611 17.0048 6.1555 Η Η 8.8013 15.9700 7.0987 Ν 9.8228 12.6513 -3.6072

10.8621 12.6145 -2.3829 S 0 11.8007 13.7269 -2.3045 S 8.3966 13.3538 -3.4858 0 7.3846 12.6315 -4.2370 11.8559 11.0954 -2.9077 С F 11.0570 10.0330 -2.9950 F 12.4489 11.3144 -4.0668 F 12.7722 10.8758 -1.9618 С 8.6020 15.0154 -4.3651 F 9.5511 15.7270 -3.7531 8.9321 14.8198 -5.6296 F F 7.4454 15.6789 -4.2943

 $[Eu(Me-BTPhen)_2(NTf_2)]^{2+}$ monodentate; E(RB3LYP) = -4437.69891049 A.U.

Eu 8.0596 12.9884 0.5936 0 6.6917 15.1228 -2.9830 8.9378 13.0325 -1.6780 0 11.4190 14.0275 0.6574 Ν Ν 10.2048 14.5505 0.4805 11.1318 16.5666 -0.3383 Ν Ν 7.7140 15.5036 0.0751 5.6818 13.9255 0.9581 Ν 6.3970 11.5707 2.0762 Ν 6.7996 10.4501 2.6777 Ν Ν 4.1547 11.1700 2.7124 5.3973 12.1112 -1.5528 Ν Ν 6.5200 11.5852 -1.0597 6.4899 9.8631 -2.6805 Ν Ν 8.7746 10.5563 0.0188 9.7382 11.8530 2.2106 Ν 8.3868 14.0629 2.9780 Ν Ν 7.6532 15.1117 3.3531 9.7239 14.2325 4.9210 Ν С 5.4146 10.4075 -3.2256 С 4.8358 11.5657 -2.6222 С 7.0178 10.4831 -1.6192 С 8.2365 9.8964 -1.0065 С 8.7719 8.6869 -1.4892 8.3000 8.2027 -2.3325 Η С 9.8851 8.1601 -0.8744 Η 10.3229 7.2339 -1.2295 С 10.4532 8.8201 0.2352 С 9.8474 10.0318 0.6480 С 11.5828 8.3093 0.9546 Η 12.0448 7.3912 0.6104

С 12.0585 8.9505 2.0556 Η 12.9041 8.5506 2.6032 10.3550 10.7185 1.8144 С С 11.4461 10.1578 2.5263 С 11.8648 10.8240 3.6957 12.6962 10.4294 4.2692 Η С 11.2037 11.9576 4.1129 Η 11.4808 12.4802 5.0175 С 10.1363 12.4446 3.3367 С 9.3794 13.6467 3.7663 7.9269 15.7273 4.4965 С С 9.0151 15.2815 5.3076 С 12.3427 16.0560 -0.1997 С 12.4881 14.7271 0.3049 С 10.0963 15.7789 -0.0274 С 8.7275 16.3038 -0.2581 С 8.5323 17.5887 -0.7959 Η 9.3937 18.1842 -1.0628 7.2454 18.0442 -0.9747 С Η 7.0635 19.0239 -1.4020 6.1567 17.2391 -0.5864 С С 6.4517 15.9610 -0.0549 С 4.7933 17.6657 -0.6964 4.5878 18.6370 -1.1313 Η С 3.7765 16.8792 -0.2537 Η 2.7484 17.2142 -0.3275 5.3734 15.1220 0.4136 С С 4.0400 15.5990 0.3335 С 3.0274 14.7781 0.8672 1.9944 15.1053 0.8261 Η С 3.3565 13.5768 1.4554 Η 2.6091 12.9332 1.8970 С 4.7056 13.1817 1.4791 С 5.1055 11.9050 2.1192 С 5.9126 9.6853 3.3012 С 4.5315 10.0504 3.3085 С 13.8236 14.0701 0.4603 13.7004 13.0709 0.8748 Η Η 14.3241 13.9895 -0.5093 14.4807 14.6506 1.1144 Η С 13.5199 16.8907 -0.5850 Η 13.1944 17.8915 -0.8619 Η 14.2472 16.9543 0.2293 Η 14.0355 16.4382 -1.4388 С 6.4222 8.4476 3.9734 7.5014 8.3754 3.8485 Η

6.1889 8.4567 5.0422 Η Η 5.9562 7.5513 3.5533 3.4910 9.2067 3.9716 С Η 2.5080 9.6562 3.8452 3.4823 8.1964 3.5510 Η 3.7015 9.1004 5.0406 Η С 3.6053 12.2191 -3.1637 Η 3.3311 13.0685 -2.5407 Η 3.7888 12.5759 -4.1813 2.7671 11.5178 -3.2083 Η 4.8506 9.7950 -4.4640 С 5.3829 8.8770 -4.7056 Η 3.7835 9.5821 -4.3555 Η 4.9544 10.4959 -5.2993 Η С 7.0607 16.8879 4.8770 6.2932 17.0416 4.1204 Η Η 6.5769 16.7166 5.8432 7.6507 17.8042 4.9742 Η 9.3816 15.9622 6.5866 С Η 10.2483 15.4795 7.0339 9.6058 17.0195 6.4147 Η 8.5480 15.9293 7.2955 Η Ν 8.9885 14.3761 -3.8734 9.7839 13.5281 -2.8215 S 11.0432 14.1626 -2.4593 0 S 7.3601 14.3446 -4.0289 6.8125 13.0267 -4.3596 0 С 10.3106 11.9155 -3.6634 F 9.2433 11.1988 -4.0155 F 11.0394 12.1894 -4.7327 F 11.0418 11.1990 -2.7956 С 7.2730 15.3784 -5.6106 F 7.7782 16.5891 -5.3994 F 7.9285 14.7782 -6.5933 F 5.9824 15.4871 -5.9331

 $[Am(Me-BTPhen)_2(NTf_2)]^{2+}$ monodentate; E(RB3LYP) = -4436.17912977 A.U.

8.0526 12.9923 0.6070 Am 6.7198 15.0178 -2.9248 0 8.9991 12.9982 -1.6978 0 Ν 11.4153 14.1263 0.6629 10.1945 14.6328 0.4788 Ν 11.0945 16.6392 -0.3886 Ν Ν 7.6812 15.5660 0.0727 5.6255 13.9829 0.9578 Ν

6.3014 11.6139 2.1056 Ν Ν 6.6796 10.4899 2.7172 4.0438 11.2438 2.6992 Ν Ν 5.3974 12.0275 -1.5567 Ν 6.5267 11.5161 -1.0618 6.5360 9.8072 -2.6948 Ν 8.7963 10.5004 0.0306 Ν Ν 9.7787 11.7974 2.2389 8.4479 14.0205 3.0505 Ν Ν 7.7295 15.0724 3.4465 9.8172 14.1582 4.9714 Ν С 5.4570 10.3396 -3.2440 С 4.8517 11.4808 -2.6332 С 7.0458 10.4257 -1.6232 С 8.2684 9.8481 -1.0054 С 8.8164 8.6499 -1.5016 Η 8.3523 8.1717 -2.3524 С 9.9315 8.1263 -0.8888 10.3789 7.2079 -1.2523 Η С 10.4918 8.7795 0.2284 С 9.8757 9.9825 0.6553 С 11.6272 8.2669 0.9363 Η 12.0914 7.3544 0.5802 12.1060 8.9006 2.0393 С 12.9570 8.5014 2.5789 Η С 10.3896 10.6646 1.8255 С 11.4912 10.1015 2.5217 С 11.9238 10.7594 3.6903 12.7625 10.3604 4.2500 Η С 11.2696 11.8905 4.1230 Η 11.5592 12.4076 5.0267 С 10.1930 12.3828 3.3632 С 9.4496 13.5871 3.8159 С 8.0267 15.6738 4.5913 С 9.1236 15.2088 5.3802 С 12.3113 16.1436 -0.2477 С 12.4749 14.8301 0.2915 С 10.0685 15.8493 -0.0504 С 8.6925 16.3622 -0.2776 С 8.4943 17.6380 -0.8357 9.3540 18.2304 -1.1144 Η С 7.2067 18.0878 -1.0182 7.0214 19.0608 -1.4591 Η 6.1202 17.2853 -0.6191 С С 6.4160 16.0148 -0.0671 С 4.7586 17.7124 -0.7440

4.5580 18.6790 -1.1913 Η С 3.7388 16.9311 -0.3014 Η 2.7110 17.2642 -0.3869 С 5.3312 15.1785 0.4011 С 3.9986 15.6571 0.2995 С 2.9759 14.8435 0.8246 Η 1.9451 15.1749 0.7665 С 3.2913 13.6446 1.4235 Η 2.5351 13.0062 1.8573 С 4.6384 13.2448 1.4682 С 5.0149 11.9655 2.1214 С 5.7714 9.7380 3.3254 С 4.3953 10.1214 3.3053 С 13.8189 14.1945 0.4604 Η 13.7098 13.2069 0.9055 14.3162 14.0906 -0.5088 Η Η 14.4717 14.8025 1.0935 С 13.4767 16.9787 -0.6665 13.1385 17.9675 -0.9701 Η Η 14.2092 17.0741 0.1400 Η 13.9908 16.5066 -1.5106 С 6.2523 8.4956 4.0098 Η 7.3322 8.4078 3.9026 6.0019 8.5116 5.0747 Η Η 5.7804 7.6046 3.5849 С 3.3318 9.2934 3.9517 2.3578 9.7575 3.8101 Η Η 3.3144 8.2835 3.5302 3.5240 9.1834 5.0237 Η С 3.6116 12.1152 -3.1756 3.3170 12.9525 -2.5455 Η 3.7932 12.4854 -4.1887 Η Η 2.7876 11.3980 -3.2319 4.9157 9.7325 -4.4951 С 5.4631 8.8243 -4.7399 Η Η 3.8504 9.5047 -4.4018 Η 5.0208 10.4437 -5.3216 С 7.1778 16.8388 4.9967 Η 6.4024 17.0090 4.2517 6.7044 16.6610 5.9669 Η Η 7.7786 17.7477 5.0969 С 9.5152 15.8719 6.6611 Η 10.3819 15.3744 7.0918 Η 9.7500 16.9282 6.4969 Η 8.6907 15.8417 7.3806 Ν 9.0046 14.3346 -3.8975

S	9.8281 13.5126 -2.8443
0	11.0716 14.1804 -2.4871
S	7.3743 14.2808 -4.0100
0	6.8368 12.9646 -4.3631
С	10.3874 11.9122 -3.6871
F	9.3328 11.1752 -4.0359
F	11.1078 12.1997 -4.7587
F	11.1344 11.2108 -2.8209
С	7.2282 15.3568 -5.5586
F	7.7175 16.5705 -5.3255
F	7.8672 14.7975 -6.5752
F	5.9275 15.4520 -5.8440

 $[Eu(Me-BTPhen)_2(H_2O)]^{3+}; E(RB3LYP) = -2686.29318346$ A.U.

Eu 0.0026 0.0549 -0.1952 0 0.9287 0.1924 -2.6696 3.3995 1.0149 -0.3593 Ν 2.1796 1.5393 -0.5354 Ν Ν 3.1109 3.6017 -1.2227 -0.3193 2.5438 -0.8531 Ν Ν -2.3733 0.9991 0.0704 -1.6879 -1.3545 1.2097 Ν -1.2945 -2.4737 1.8279 Ν -3.9289 -1.7083 1.8758 Ν Ν -2.4301 -0.7611 -2.6240 -1.3686 -1.3081 -2.0173 Ν Ν -1.5382 -3.2481 -3.3594 0.7681 -2.3668 -0.7271 Ν 1.6608 -0.9993 1.4621 Ν 0.3144 1.2561 2.1015 Ν Ν -0.4244 2.3259 2.4171 Ν 1.5634 1.4605 4.0988 С -2.6016 -2.7376 -3.9608 С -3.0596 -1.4332 -3.5797 С -0.9512 -2.5160 -2.4033 С 0.2386 -3.0856 -1.7210 С 0.7499 -4.3379 -2.1066 Η 0.2830 -4.8732 -2.9214 С 1.8360 -4.8494 -1.4299 Η 2.2582 -5.8072 -1.7131 С 2.3857 -4.1369 -0.3447 С 1.7990 -2.8857 -0.0225 С 3.4752 -4.6382 0.4406 Η 3.9235 -5.5851 0.1627 С 3.9250 -3.9525 1.5249

Η	4.7379	-4.3447 2.1251
С	2.2743	-2.1588 1.1333
С	3.3222	-2.7108 1.9132
С	3.7085	-2.0063 3.0716
Н	4.5071	-2.3933 3.6947
С	3.0523	-0.8458 3.4183
Н	3.3011	-0.2957 4.3150
C	2.0240	-0.3726 2.5840
C	1 2645	0.8505 2.9453
C	-0 1996	2 9622 3 5590
C	0.1776	2.5022 5.5550
C	A 327A	3 1139 -1 03/1
C	т.327т Л Л7ЛЛ	1 7568 0 5047
C	4.4/44 2.0702	1.7308 - 0.3947
C	2.0703	2.7970 -0.9097
C	0.7055	3.3390 -1.1791
	0.3220 1.2075	4.041/ -1.0///
П	1.38/3	5.2389 -1.9280
C	-0./60/	5.1216 -1.8320
H	-0.9305	6.11/5 -2.2257
C	-1.8587	4.3242 -1.4502
C	-1.5787	3.0270 -0.9481
С	-3.2147	4.7830 -1.5266
Η	-3.4079	5.7680 -1.9353
С	-4.2378	4.0092 -1.0761
Η	-5.2595	4.3683 -1.1205
С	-2.6660	2.2063 -0.4643
С	-3.9889	2.7147 -0.5119
С	-5.0118	1.9128 0.0341
Η	-6.0376	2.2639 0.0160
С	-4.6986	0.7014 0.6104
Η	-5.4517	0.0739 1.0661
С	-3.3577	0.2776 0.6123
С	-2.9729	-1.0003 1.2621
С	-2.1857	-3.2089 2.4796
С	-3.5655	-2.8214 2.4942
С	5.8132	1.1220 -0.3878
Н	5.6933	0.0838 -0.0835
Н	6.4103	1.1573 -1.3038
Н	6.3831	1.6537 0.3803
С	5.5031	3.9948 -1.2943
Н	5.1728	4.9838 -1.6052
н	6 1253	4 0870 -0 3982
н	6 1425	3 5684 -2 0743
C	-1 6925	-4 4385 3 1751
ч	-0.6166	-4 5378 3 0420
ц	-0.0100	-1 3081 1 7155
11	-1.71.30	-+.570+ +.2455

Η	-2.1830	-5.3348	2.7837
С	-4.6096	-3.6318	3.1867
Η	-5.5891	-3.1758	3.0577
Η	-4.6324	-4.6547	2.7975
Η	-4.3899	-3.7108	4.2566
С	-4.2381	-0.7774	-4.2271
Η	-4.4046	0.2073	-3.7941
Η	-4.0842	-0.6694	-5.3050
Η	-5.1414	-1.3809	-4.0971
С	-3.2814	-3.5380	-5.0205
Η	-2.7717	-4.4890	-5.1612
Η	-4.3266	-3.7263	-4.7535
Η	-3.2965	-2.9935	-5.9702
С	-1.0671	4.1387	3.8783
Η	-1.7973	4.2925	3.0857
Η	-1.5971	3.9884	4.8236
Η	-0.4697	5.0482	3.9925
С	1.1585	3.2245	5.7130
Η	1.9976	2.7448	6.2128
Η	1.3984	4.2772	5.5327
Η	0.2903	3.2133	6.3804
Η	1.8324	0.3988	-2.9352
Η	0.4334	0.0436	-3.4833

 $[Am(Me-BTPhen)_2(H_2O)]^{3+}; E(RB3LYP) = -2684.77367393$ A.U.

Am 0.0052 0.0551 -0.2012 0 0.9493 0.1953 -2.7184 3.4058 1.1108 -0.3916 Ν 2.1785 1.6204 -0.5628 Ν 3.0791 3.7041 -1.2214 Ν Ν -0.3447 2.6107 -0.8558 Ν -2.4202 1.0517 0.0581 Ν -1.7719 -1.3209 1.2167 N -1.4012 -2.4424 1.8452 Ν -4.0253 -1.6411 1.8517 N -2.4090 -0.8496 -2.6624 -1.3462 -1.3839 -2.0458 Ν N -1.5111 -3.3480 -3.3505 Ν 0.7909 -2.4347 -0.7196 1.7067 -1.0543 1.4783 Ν Ν 0.3769 1.2174 2.1575 N -0.3515 2.2877 2.4954 1.6585 1.3928 4.1350 Ν C -2.5743 -2.8503 -3.9625 C -3.0368 -1.5405 -3.6053

С	-0.9262	-2.5977	-2.4063
С	0.2643	-3.1572	-1.7130
С	0.7792	-4.4087	-2.0969
Н	0.3134	-4.9474	-2.9099
С	1.8678	-4.9138	-1.4208
Н	2.2939	-5.8703	-1.7022
С	2.4177	-4.1961	-0.3395
C	1.8275	-2.9457	-0.0158
Ċ	3.5141	-4.6963	0.4358
H	3.9607	-5.6427	0.1537
C	3.9721	-4.0095	1.5150
H	4.7909	-4.3990	2.1089
C	2.3126	-2.2158	1.1381
C	3,3705	-2.7688	1.9062
C	3,7724	-2.0671	3.0606
H	4.5784	-2.4572	3.6721
C	3 1237	-0.9064	3 4178
н	3 3851	-0 3586	4 3122
C	2 0858	-0.4301	2 5972
$\frac{c}{c}$	1 3375	0.7958	2.0799
C	-0 1054	2,9101	3 6403
C	0.9542	2.4556	4 4920
C	4 3022	3 2315	-1 0391
C	4 4697	1 8713	-0.6160
C	2.0493	2.8814	-0.9788
C	0.6747	3 4105	-1 1829
C	0.4883	4 7116	-1 6842
н	1.3512	5.3128	-1.9334
C	-0.7960	5.1834	-1.8432
H	-0.9708	6.1774	-2.2395
C	-1.8908	4.3803	-1.4652
Ċ	-1.6081	3.0846	-0.9576
C	-3.2459	4.8375	-1.5556
Н	-3.4360	5.8215	-1.9680
С	-4.2713	4.0628	-1.1141
Н	-5.2935	4.4187	-1.1686
С	-2.7009	2.2607	-0.4818
С	-4.0241	2.7702	-0.5462
С	-5.0556	1.9719	-0.0121
Н	-6.0798	2.3263	-0.0443
С	-4.7538	0.7608	0.5692
Н	-5.5144	0.1364	1.0162
С	-3.4142	0.3335	0.5884
C	-3.0512	-0.9479	1.2484
C	-2.3111	-3.1624	2.4877
С	-3.6856	-2.7558	2.4801

С 5.8176 1.2540 -0.4155 Η 5.7127 0.2095 -0.1270 6.4160 1.3121 -1.3294 Η Η 6.3777 1.7825 0.3620 5.4650 4.1324 -1.2890 С 5.1201 5.1222 -1.5808 Η Η 6.0910 4.2167 -0.3949 6.1055 3.7285 -2.0801 Η -1.8443 -4.3950 3.1959 С Н -0.7678 -4.5084 3.0803 Н -2.0831 -4.3472 4.2626 Н -2.3401 -5.2868 2.8009 -4.7503 -3.5489 3.1613 С H -5.7218 -3.0812 3.0151 Н -4.7804 -4.5737 2.7778 Н -4.5476 -3.6243 4.2349 C -4.2170 -0.8997 -4.2646 Н -4.3856 0.0925 -3.8499 H -4.0634 -0.8115 -5.3443 Н -5.1189 -1.5029 -4.1231 C -3.2503 -3.6709 -5.0093 Н -2.7385 -4.6233 -5.1316 Н -4.2957 -3.8563 -4.7414 Н -3.2641 -3.1436 -5.9687 -0.9619 4.0870 3.9866 С Η -1.7063 4.2504 3.2094 -1.4742 3.9304 4.9405 Η Н -0.3588 4.9933 4.0972 1.2898 3.1420 5.7736 С 2.1407 2.6589 6.2498 Η 1.5216 4.1980 5.6025 Η Η 0.4359 3.1180 6.4590 Η 1.8374 0.4587 -2.9861 Η 0.4719 -0.0116 -3.5303

 $[Eu(Me-BTPhen)_2(OH)]^{2+}; E(RB3LYP) = -2686.03084884$ A.U.

0.0500 0.0537 -0.2710 Eu 0.8345 0.1556 -2.2441 0 Ν 3.4089 1.2540 -0.0015 2.1902 1.7232 -0.2545 Ν 3.0934 3.6136 -1.3527 Ν Ν -0.3237 2.6142 -0.7831 -2.3829 1.0345 0.0771 Ν N -1.7156 -1.3213 1.2613 N -1.3483 -2.4337 1.8973

Ν	-3.9800	-1.6745	1.8496
Ν	-2.5876	-0.9532	-2.4408
Ν	-1.5147	-1.4784	-1.8551
Ν	-1.4660	-3.2494	-3.4224
Ν	0.7193	-2.4669	-0.6624
Ν	1.6799	-1.0636	1.4789
Ν	0.3294	1.1914	2.1685
Ν	-0.3983	2.2529	2.5152
Ν	1.6622	1.4073	4.1112
С	-2.5250	-2.7342	-4.0277
С	-3.1066	-1.5388	-3.5150
С	-0.9865	-2.5960	-2.3558
С	0.2111	-3.1592	-1.6797
С	0.7595	-4.3831	-2.1111
Η	0.3106	-4.8967	-2.9494
С	1.8554	-4.8895	-1.4506
Η	2.3005	-5.8283	-1.7609
С	2.3965	-4.1906	-0.3524
С	1.7820	-2.9627	0.0029
С	3.5088	-4.6798	0.4071
Η	3.9712	-5.6133	0.1074
С	3.9655	-3.9995	1.4918
Η	4.7962	-4.3830	2.0730
С	2.2820	-2.2258	1.1459
С	3.3550	-2.7687	1.8992
С	3.7635	-2.0603	3.0468
Η	4.5800	-2.4416	3.6499
С	3.1099	-0.9039	3.4054
Η	3.3766	-0.3494	4.2937
С	2.0643	-0.4337	2.5885
С	1.3155	0.7915	2.9715
С	-0.1259	2.8966	3.6444
С	0.9579	2.4694	4.4679
С	4.3143	3.1512	-1.1310
С	4.4758	1.9272	-0.4202
С	2.0657	2.8777	-0.9102
С	0.6924	3.3895	-1.1558
С	0.5023	4.6558	-1.7433
Η	1.3654	5.2365	-2.0358
С	-0.7813	5.1168	-1.9273
Η	-0.9587	6.0869	-2.3779
С	-1.8747	4.3297	-1.5128
С	-1.5849	3.0645	-0.9411
С	-3.2343	4.7662	-1.6319
Η	-3.4317	5.7311	-2.0847
С	-4.2562	3.9974	-1.1707

Η	-5.2816	4.3405 -1.2472
С	-2.6747	2.2340 -0.4705
С	-4.0032	2.7239 -0.5645
С	-5.0290	1.9192 -0.0305
Η	-6.0579	2.2579 -0.0811
С	-4.7161	0.7208 0.5686
Η	-5.4722	0.0894 1.0126
С	-3.3701	0.3091 0.6011
С	-3.0007	-0.9670 1.2673
С	-2.2631	-3.1708 2.5164
С	-3.6365	-2.7868 2.4787
С	5.8208	1.3446 -0.1117
Η	5.7028	0.4237 0.4569
Η	6.3719	1.1231 -1.0307
Η	6.4318	2.0441 0.4663
С	5.4813	3.9396 -1.6351
Η	5.1333	4.8274 -2.1598
Η	6.1326	4.2440 -0.8097
Η	6.0929	3.3381 -2.3146
С	-1.7880	-4.3990 3.2304
Η	-0.7073	-4.4886 3.1327
Η	-2.0456	-4.3609 4.2930
Η	-2.2565	-5.2991 2.8210
С	-4.7080	-3.6012 3.1312
Η	-5.6796	-3.1377 2.9715
Η	-4.7243	-4.6189 2.7292
Η	-4.5266	-3.6904 4.2069
С	-4.3052	-0.8928 -4.1389
Η	-4.5709	0.0061 -3.5850
Η	-4.1063	-0.6202 -5.1797
Η	-5.1624	-1.5726 -4.1450
С	-3.0727	-3.4326 -5.2318
Η	-2.4944	-4.3313 -5.4379
Η	-4.1216	-3.7061 -5.0825
Η	-3.0386	-2.7778 -6.1083
С	-0.9903	4.0690 3.9933
Η	-1.7555	4.2044 3.2306
Η	-1.4778	3.9238 4.9620
Η	-0.3991	4.9866 4.0686
С	1.3263	3.1830 5.7295
Η	2.1922	2.7095 6.1883
Η	1.5533	4.2350 5.5313
Η	0.4937	3.1692 6.4401
Η	1.1761	0.1881 -3.1361

 $[Am(Me-BTPhen)_2(OH)]^{2+}; E(RB3LYP) = -2684.50792474$ A.U.

0.0473 0.0548 -0.2601 Am 0.8563 0.1609 -2.2965 0 Ν 3.3949 1.3466 0.0043 2.1711 1.7993 -0.2558 Ν Ν 3.0517 3.6825 -1.3814 -0.3613 2.6660 -0.7940 Ν -2.4393 1.0820 0.0719 Ν -1.8081 -1.2896 1.2754 Ν -1.4631 -2.4079 1.9141 Ν Ν -4.0837 -1.6138 1.8304 Ν -2.5793 -1.0403 -2.4375 -1.4985 -1.5502 -1.8518 Ν Ν -1.4141 -3.3121 -3.4255 Ν 0.7540 -2.5184 -0.6501 1.7272 -1.1145 1.5095 Ν 0.3959 1.1553 2.2382 Ν Ν -0.3156 2.2222 2.6022 1.7603 1.3435 4.1600 Ν -2.4781 -2.8111 -4.0334 С С -3.0844 -1.6296 -3.5159 С -0.9498 -2.6566 -2.3531 С 0.2525 -3.2084 -1.6730 С 0.8096 -4.4257 -2.1114 Η 0.3656 -4.9372 -2.9532 С 1.9077 -4.9278 -1.4524 Η 2.3601 -5.8616 -1.7670 2.4436 -4.2310 -0.3507 С С 1.8216 -3.0083 0.0137 С 3.5610 -4.7217 0.3995 4.0245 -5.6517 0.0907 Η С 4.0210 -4.0464 1.4848 Η 4.8566 -4.4288 2.0595 С 2.3257 -2.2751 1.1607 С 3.4081 -2.8202 1.9010 С 3.8294 -2.1182 3.0475 Η 4.6529 -2.5035 3.6387 С 3.1817 -0.9633 3.4193 3.4600 -0.4128 4.3064 Η С 2.1266 -0.4890 2.6172 С 1.3908 0.7396 3.0206 С -0.0192 2.8553 3.7310 1.0718 2.4095 4.5353 С 4.2774 3.2357 -1.1557 С С 4.4538 2.0250 -0.4252 2.0319 2.9425 -0.9265 С С 0.6527 3.4406 -1.1758

С	0.4594	4.6991 -	-1.7784
Η	1.3207	5.2790	-2.0772
С	-0.8251	5.1532	-1.9681
Η	-1.0060	6.1175	-2.4297
С	-1.9163	4.3669	-1.5469
С	-1.6254	3.1079	-0.9598
С	-3.2744	4.8031	-1.6787
Н	-3.4680	5.7637	-2.1419
С	-4.2985	4.0385	-1.2177
Н	-5.3238	4.3792	-1.3039
С	-2.7201	2.2797	-0.4881
С	-4.0479	2.7706	-0.5997
С	-5.0818	1.9724	-0.0722
Н	-6.1088	2.3143	-0.1367
С	-4.7800	0.7764	0.5361
Η	-5.5433	0.1493	0.9736
С	-3.4357	0.3610	0.5865
С	-3.0875	-0.9185	1.2612
С	-2.3958	-3.1341	2.5188
С	-3.7632	-2.7306	2.4637
С	5.8055	1.4629 -	-0.1084
Η	5.6983	0.5481	0.4723
Η	6.3598	1.2363	-1.0241
Н	6.4074	2.1773	0.4609
С	5.4353	4.0272 -	-1.6759
Н	5.0772	4.9044	-2.2114
Η	6.0878	4.3485	-0.8579
Η	6.0487	3.4213	-2.3499
С	-1.9469	-4.3703	3.2357
Η	-0.8664	-4.4744	3.1512
Η	-2.2171	-4.3308	4.2952
Η	-2.4227	-5.2629	2.8185
С	-4.8539	-3.5299	3.1029
Η	-5.8172	-3.0537	2.9303
Η	-4.8786	-4.5479	2.7019
Η	-4.6876	-3.6201	4.1811
С	-4.2929	-1.0027	-4.1401
Η	-4.5823	-0.1171	-3.5767
Η	-4.0926	-0.7123	-5.1757
Η	-5.1346	-1.7013	-4.1612
С	-3.0070	-3.5106	-5.2452
Η	-2.4058	-4.3920	-5.4602
Η	-4.0485	-3.8129	-5.0990
Η	-2.9900	-2.8460	-6.1147
С	-0.8651	4.0349	4.1003
Η	-1.6380	4.1852	3.3482

Η	-1.3423	3.8874	5.0737
Η	-0.2622	4.9449	4.1756
С	1.4649	3.1088	5.7976
Η	2.3316	2.6222	6.2410
Η	1.7009	4.1597	5.6046
Η	0.6417	3.0983	6.5190
Η	1.1999	0.1925	-3.1881

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