Electronic Supplementary Information

Influence of the cavity dimension on encapsulation of halide within the capsular assembly and side-cleft recognition of sulphate-water cluster assisted by polyammonium tripodal receptor

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Characterization of receptor L₁:



Fig. S1. ESI-mass spectrum of receptor L_1



Fig. S2. FTIR spectrum of receptor L_1



Fig. S3. ¹H NMR spectrum of receptor L_1 in CDCl₃ at 298 K



Fig. S4. ¹³C NMR spectrum of receptor L_1 in CDCl₃ at 298 K

Characterization of receptor L₂:



Fig. S5. ESI-mass spectrum of receptor L_2



Fig. S6. FTIR spectrum of receptor L_2



Fig. S7. ¹H NMR spectrum of receptor L_2 in CDCl₃ at 298 K



Fig. S8. ¹³C NMR spectrum of receptor L_2 in CDCl₃ at 298 K

Characterization of receptor L₃:



Fig. S9. ESI-mass spectrum of receptor L_3



Fig. S10. FTIR spectrum of receptor L_3



Fig. S11. ¹H NMR spectrum of receptor L₃ in CDCl₃ at 298 K



Fig. S12. ¹³C NMR spectrum of receptor L₃ in CDCl₃ at 298 K

Characterization of complex 1a:



Fig. S13. ¹H NMR spectrum of complex 1a in DMSO-d₆ at 298 K



Fig. S14. ¹³C NMR spectrum of complex 1a in DMSO-d₆ at 298 K



Figure S15. Powder X-ray diffraction: simulated pattern from the single–crystal X-ray of complex **1a** (blue), experimental pattern from the crystalline solid of complex **1a** (black).

Characterization of complex 1b:



Fig. S16. ¹H NMR spectrum of complex 1b in DMSO-d₆ at 298 K



Fig. S17. ¹³C NMR spectrum of complex 1b in DMSO-d₆ at 298 K.



Figure S18: Powder X-ray diffraction: simulated pattern from the single–crystal X-ray of complex **1b** (blue), experimental pattern from the crystalline solid of complex **1b** (black).

Characterization of complex 2a:



Fig. S19. ¹H NMR spectrum of complex 2a in DMSO-d₆ at 298 K



Fig. S20. ¹³C NMR spectrum of complex 2a in DMSO-d₆ at 298 K



Figure S21: Powder X-ray diffraction: simulated pattern from the single–crystal X-ray of complex **2a** (blue), experimental pattern from the crystalline solid of complex **2a** (black).

Characterization of complex 2b:



Fig. S22. ¹H NMR spectrum of complex 2b in DMSO-d₆ at 298 K



Fig. S23: ¹³C NMR spectrum of complex 2b in DMSO-d₆ at 298 K



Figure S24: Powder X-ray diffraction: simulated pattern from the single crystal X-ray of complex **2b** (blue), experimental pattern from the crystalline solid of complex **2b** (black).

Characterization of complex 3a:



Fig. S25: ¹H NMR spectrum of complex **3a** in DMSO-d₆ at 298 K.



Fig. S26: ¹³C NMR spectrum of complex 3a in DMSO-d₆ at 298 K.



Figure S27: Powder X-ray diffraction: simulated pattern from the single–crystal X-ray of complex **3a** (blue), experimental pattern from the crystalline solid of complex **3a** (black).



Fig. S28: (a) space-fill view of L_1 receptor from crystallographic b axis showing there is almost no space or no cavity, (b) depicting distances among each L_1 receptor arms, (c) depicting distances among each three arms of two L_1 receptor units in salt 1a, (d) depicting distances among each three arms of two L_1 receptor units in salt 1b, (e) depicting distances among each three arms of two L_2 receptor units in salt 2a and (f) depicting distances among each three arms of two L_2 receptor units in salt 2b.



Fig. S29: Partial 1H NMR spectra (600 MHz, DMSO d6) of (a) free receptor L_1 , (b) addition of excess HCl to L_1 and (c) addition of excess H₂SO₄ to L_1 .



Fig. S30: Partial 1H NMR spectra (600 MHz, DMSO d6) of (a) free receptor L_2 , (b) addition of excess HCl to L_2 and (c) addition of excess H₂SO₄ to L_2 .



Fig. S31: Partial 1H NMR spectra (600 MHz, DMSO d6) of (a) free receptor L_3 and (b) addition of excess HCl to L_3 .

Complex	D-H…A	<i>d</i> (D…H)/Å	<i>d</i> (H…A)/Å	<i>d</i> (D…A)/Å	<d-h…a th="" °<=""><th>Symmetry codes</th></d-h…a>	Symmetry codes
1a	N2-H2C···Cl6	0.90	2.22	3.092(5)	164	x,y,z
	N2-H2D…Cl3	0.90	2.28	3.155(4)	163	х,ү,z
	N4-H4C···Cl3	0.90	2.21	3.077(5)	162	х,ү,z
	N4-H4D…Cl2	0.90	2.21	3.098(5)	170	-х, у-1/2, -z+3/2
	N6-H6C···Cl3	0.90	2.31	3.177(4)	163	х,ү,z
	N6-H6D…Cl2	0.90	2.28	3.140(5)	161	-x+1/2, -y+1, z-1/2
	N9-H9C…Cl1	0.90	2.35	3.237(4)	167	х,ү,z
	N9-H9D…Cl5	0.90	2.18	3.073(5)	171	х,ү,z
	N11-H11C…Cl1	0.90	2.31	3.184(5)	164	х,ү,z
	N11-H11D…Cl4	0.90	2.25	3.094(5)	157	-1/2+x,1/2-y,-z
	N13-H13C…Cl1	0.90	2.25	3.129(5)	167	х,ү,z
	N13-H13D…Cl4	0.90	2.21	3.085(5)	162	-1+x,y,z
	C2-H2B…O2B	0.97	2.56	3.406(11)	146	1-x,1/2+y,3/2-z
	C15-H15…O12	0.93	2.48	3.391(10)	168	-1/2-x,-y,1/2+z
	C35-H35…O1A	0.93	2.51	3.385(10)	157	3/2-x,-y,-1/2+z
	C35-H35…O1B	0.93	2.39	3.296(11)	165	3/2-x,-y,-1/2+z
	C37-H37B…O4B	0.97	2.52	3.382(17)	148	-x,1/2+y,1/2-z
	C50-H50…O4A	0.93	2.50	3.411(9)	167	-1/2+x,1/2-y,-z
1b	N2-H2N…O13	0.86	2.14	2.951(9)	158	x,y,z
	N4-H4N…O13	0.86	2.30	3.063(8)	148	х,ү,z
	N9-H9N…O14	0.86	2.21	2.919(9)	140	х,ү,z
	N11-H11N…O14	0.86	2.30	3.108(9)	156	x,y,z
	N13-H13N…O14	0.86	2.37	3.134(8)	148	x,y,z
	C11-H11B…O21	0.97	2.59	3.529(14)	163	x,y,z
	C19-H19B…O23	0.97	2.55	3.497(10)	164	-x,1-y,1-z
	C23-H23…O18	0.93	2.45	3.283(12)	149	-1+x,y,z
	C28-H28B…O23	0.97	2.53	3.469(9)	163	1-x,-y,1-z
	C44-H44…O2A	0.93	2.46	3.300(2)	151	1-x,1-y,1-z
	C47-H47A…O28	0.97	2.59	3.415(14)	143	х,ү,z
	C54-H54…O16	0.93	2.54	3.368(13)	148	х,ү,z
	C55-H55C…O24	0.96	2.44	3.295(11)	148	-x,1-y,1-z
	C59-H59B…O20	0.96	2.54	3.414(13)	151	х,у,z
		0.00	4 70	2 (14/5)	470	
Za	N2-H2C-+1	0.90	1.72	2.614(6)	1/3	2/3+x,1/3+y,-2/3+z
	N2-H2D····F2	0.90	1./6	2.663(10)	1/8	1-x+y,1-x,z
	C1-H1B…F3	0.97	2.47	3.346(12)	150	2/3-y,1/3+x-y,1/3+z
	C3-H3A…F3	0.97	2.49	3.443(9)	167	х,ү,z

 Table S1. Hydrogen bonding contacts in complexes 1a-b, 2a-b and 3a

2b	N2-H2A…O4A	0.97	1.77	2.738(18)	173	x,1+y,z
	N2-H2A…O4B	0.97	1.78	2.715(13)	162	x,1+y,z
	N3-H3B…O12	0.97	1.75	2.710(10)	168	x,y,z
	N6-H6A…O3	0.90	1.86	2.758(9)	173	x+1, y, z
	N6-H6B…O5	0.90	2.09	2.871(8)	145	х,у,z
	N7-H7A…O3	0.97	2.18	2.932(9)	134	x+1,y,z
	N7-H7B…O9	0.97	1.85	2.806(9)	167	x+1,y,z
	N8-H8C…O9	0.97	2.07	2.910(9)	144	x+1,y,z
	N8-H8D…O5	0.97	1.88	2.841(9)	172	х,у,z
	C2-H2D…O11A	0.97	2.54	3.490(2)	166	х,у,z
	C5-H5…O12	0.93	2.42	3.258(11)	150	x,y,z
	C11-H11B…O7B	0.97	2.58	3.550(2)	177	x,y,z
	C20-H20B…O1	0.97	2.57	3.535(15)	175	x,1+y,z
	C21A-H21C…O14	0.86	2.55	3.260(4)	141	x,1+y,z
	C45-H45…O3	0.93	2.51	3.317(16)	146	-x,1-y,1-z
	C47-H47A…O11A	0.97	2.44	3.320(2)	151	x+1,y,z
	C50-H50…O5	0.93	2.50	3.379(16)	158	1-x,1-y,1-z
	C56-H56C…O9	0.96	2.37	3.316(15)	170	-x,1-y,1-z
3a	N2-H2C···Cl2	0.90	2.24	3.075(10)	153	х,у,z
	N2-H2D…Cl1	0.90	2.28	3.152(10)	164	1+x,1+y,z
	N4-H4C···Cl2	0.90	2.40	3.285(12)	167	х,у,z
	N4-H4D…Cl6	0.90	2.21	3.078(11)	163	х,у,z
	N6-H6C···Cl5	0.90	2.25	3.137(11)	166	1+x,1+y,z
	N6-H6D…Cl3	0.90	2.25	3.110(11)	159	х,у,z
	N9-H9C…Cl6	0.90	2.27	3.143(11)	163	х,у,z
	N9-H9D…Cl1	0.90	2.21	3.078(10)	162	х,у,z
	N11-H11C…Cl4	0.90	2.42	3.312(12)	172	х,у,z
	N11-H11D…Cl5	0.90	2.18	3.051(11)	162	x,-1+y,z
	N13-H13C…Cl3	0.90	2.28	3.160(10)	167	х,у,z
	N13-H13D…Cl4	0.90	2.24	3.080(10)	156	х,у,z
	C21-H21B…O5A	0.97	2.57	3.462(2)	154	-1+x,y,z
	C23-H23…O5A	0.93	2.23	3.120(2)	160	-1+x,y,z
	C23-H23…O5B	0.93	2.50	3.190(2)	160	-1+x,y,z
	C23-H23…O6	0.93	2.57	3.375(18)	145	-1+x,y,z
	C30-H30B…O7A	0.97	2.54	3.450(2)	157	-1+x,y,z
	C32-H32…O7A	0.93	2.39	3.260(2)	155	-1+x,y,z
	C32-H32…O7B	0.93	2.38	3.270(2)	159	-1+x,y,z
	C32-H32…O8	0.93	2.51	3.339(18)	148	-1+x,y,z



Figure S32: Packing motif of crystal structure of complex **1a** (as viewed down the *a*-axis) showing capsular assembly formation.



Figure S33: Packing motif of crystal structure of complex 1b (as viewed down the *a*-axis) showing sidecleft binding of sulphate ions around DMF encapsulated receptor L_1 .



Figure S34: Packing motif of crystal structure of complex **2a** (as viewed down the *c*-axis) showing encapsulation of fluoride ion (F1) supported by H-bonding interactions with exterior fluorides (F2 and F3)



Figure S35: Packing motif of crystal structure of complex 2b (as viewed down the *c*-axis) showing sidecleft binding of sulphate ions around DMF encapsulated receptor L_2 .



Figure S36: Packing motif of crystal structure of complex 3a (as viewed down the *c*-axis) showing linear architecture of chloride via Y-shaped layer like assembly of receptor L_3 .