Supplementary Information

Extended Hybrid Architectures Based on Tetra-Co^{II} Sandwiched Polyoxotungstate

Yu Zhu,^a Wei-Hui Fang,^a Qi Wei,^b and Guo-Yu Yang^{*a, b}

 ^a State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, China, E-mail: ygy@fjirsm.ac.cn; Fax: (+86) 591-8371-0051
^b MOE Key Laboratory of Cluster Science, School of Chemistry, Beijing Institute of Technology, Beijing 10008, China. E-mail: ygy@bit.edu.cn; Fax: (+86) 10-6891-8572

The first two authors have equal contributions.

Fig. S1. Powder XRD pattern of compound 1.

Fig. S2. Powder XRD pattern of compound 2.

Fig. S3. Powder XRD pattern of compound 3.

Fig. S4. The asymmetric unit of compound 1.

Fig. S5. The coordination environment of the Co1 and Co2 ions in compound 1.

Fig. S6. The asymmetric unit of compound 2.

Fig. S7. The asymmetric unit of compound 3.

Fig. S8. IR spectrum of compound 1.

Fig. S9. IR spectrum of compound 2.

Fig. S10. IR spectrum of compound 3.

Fig. S11. UV-vis diffuse reflectance spectrum of 1.

Fig. S12. UV-vis diffuse reflectance spectrum of 2.

Fig. S13. UV-vis diffuse reflectance spectrum of 3.

Fig. S14. TG curve of compound 1.

Fig. S15. TG curve of compound 2.

Fig. S16. TG curve of compound 3.

Fig. S17. The X-band EPR spectrum of compound 2.

Fig. S18. The X-band EPR spectrum of compound 3.

Table S1. The BVS values of all the W and O atoms from the asymmetrical unit of compound 1-3.



Fig. S1. Powder XRD pattern of compound **1**, showing the bulk product is in good agreement with the calculated pattern based on the result from single-crystal X-ray diffraction.



Fig. S2. Powder XRD pattern of compound **2**, showing the bulk product is in good agreement with the calculated pattern based on the result from single-crystal X-ray diffraction.



Fig. S3. Powder XRD pattern of compound **3**, showing the bulk product is in good agreement with the calculated pattern based on the result from single-crystal X-ray diffraction.



Fig. S4. The asymmetric unit of compound 1.



Fig. S5. The coordination environment of the Co1 and Co2 ions in compound 1.



Fig. S6. The asymmetric unit of compound 2.



Fig. S7. The asymmetric unit of compound 3.



Fig. S8. IR spectrum of compound 1.



Fig. S9. IR spectrum of compound 2.



Fig. S10. IR spectrum of compound 3.



Fig. S11. UV-vis diffuse reflectance spectrum of 1.



Fig. S12. UV-vis diffuse reflectance spectrum of 2.



Fig. S13. UV-vis diffuse reflectance spectrum of 3.



Fig. S14. TG curve of compound 1.



Fig. S16. TG curve of compound 3.







Fig. S18. The X-band EPR spectrum of compound 3.

Table S1. The BVS values of all the W and O atoms from the asymmetrical unit of compound 1-3.

Compou	d 1:						
Atom	BVS	Atom	BVS	Atom	BVS	Atom	BVS
W1	6.110	W2	6.167	W3	6.182	W4	6.112
W5	6.266	W6	6.209	W7	6.077	W8	6.227
W9	6.099	Co1	2.160	Co2	2.017	Co3	2.379
Co4	2.268						
Compou	d 2 :						
Atom	BVS	Atom	BVS	Atom	BVS	Atom	BVS
W1	6.158	W2	6.214	W3	6.225	W4	6.190
W5	6.261	W6	6.139	W7	6.397	W8	6.185
W9	6.197	Col	2.062	Co2	2.056	Co3	1.879
Compou	nd 3 :						
Atom	BVS	Atom	BVS	Atom	BVS	Atom	BVS
W1	6.213	W2	6.154	W3	6.218	W4	6.222
W5	6.124	W6	6.135	W7	6.133	W8	6.049
W9	6.183	Col	1.973	Co2	1.987	Co3	2.035
Co4	2.454	Co5	2.300				