

1. Experimental detail:

Compound **1** here is described as an example:

2.5mg compound **1** was dissolved in a saturated solution of NaOH, and the solution was heated to 80°C for 30min, then the solution was cooled to room temperature, about 1.5ml concentrated nitric acid was added into the solution dropwise, now you can obtain a clear solution. Finally, transfer the solution to a 25ml volumetric flask for elemental analysis.

2. Table s1. Hydrogen bonds in compounds **2** and **4**.

Compound <b>2</b>				
		D···A bond length(Å)	D-H···A angle(°)	Symmetry code
1	N(1)-H(1)···O(6 <sub>a</sub> )	3.0169(2)	137.714(5)	a: 0.33333-y, 0.66667+x-y, -0.33333+z
2	N(1)-H(1)···O(14 <sub>b</sub> )	3.0326(3)	145.499(11)	b: 0.33333-x+y, 0.66667-x, -0.33333+z
3	N(2)-H(2)···O(4W)	2.9704(2)	144.985(6)	
4	O(4W)···O(1 <sub>c</sub> )	2.7959(2)		c: -y, x-y, z
5	O(4W)···O(8 <sub>d</sub> )	3.0354(1)		d: -x+y, 1-x, z
6	O(3W)···O(2 <sub>e</sub> )	3.0957(2)		e: 0.33333-y, -0.33333+x-y, -0.33333+z

Compound <b>4</b>				
1	O(4W)···O(2)	3.1096(1)		
2	O(4W)···O(3W <sub>f</sub> )	3.0380(1)		f: 1.5-x, -0.5+y, 0.5-z
3	O(2W)···O(1 <sub>f</sub> )	3.0003(2)		
4	O(3W)···O(2 <sub>g</sub> )	3.1170(3)		g: 1.5-x, 0.5+y, 0.5-z
5	O(3W)···O(3 <sub>h</sub> )	3.0925(2)		h: x, 1+y, z
6	O(1W)···O(4 <sub>g</sub> )	2.9952(1)		
7	O(1W)···O(13 <sub>g</sub> )	3.0954(2)		
8	O(1W)···O(10 <sub>i</sub> )	2.9501(1)		i: x, 1+y, 1+z

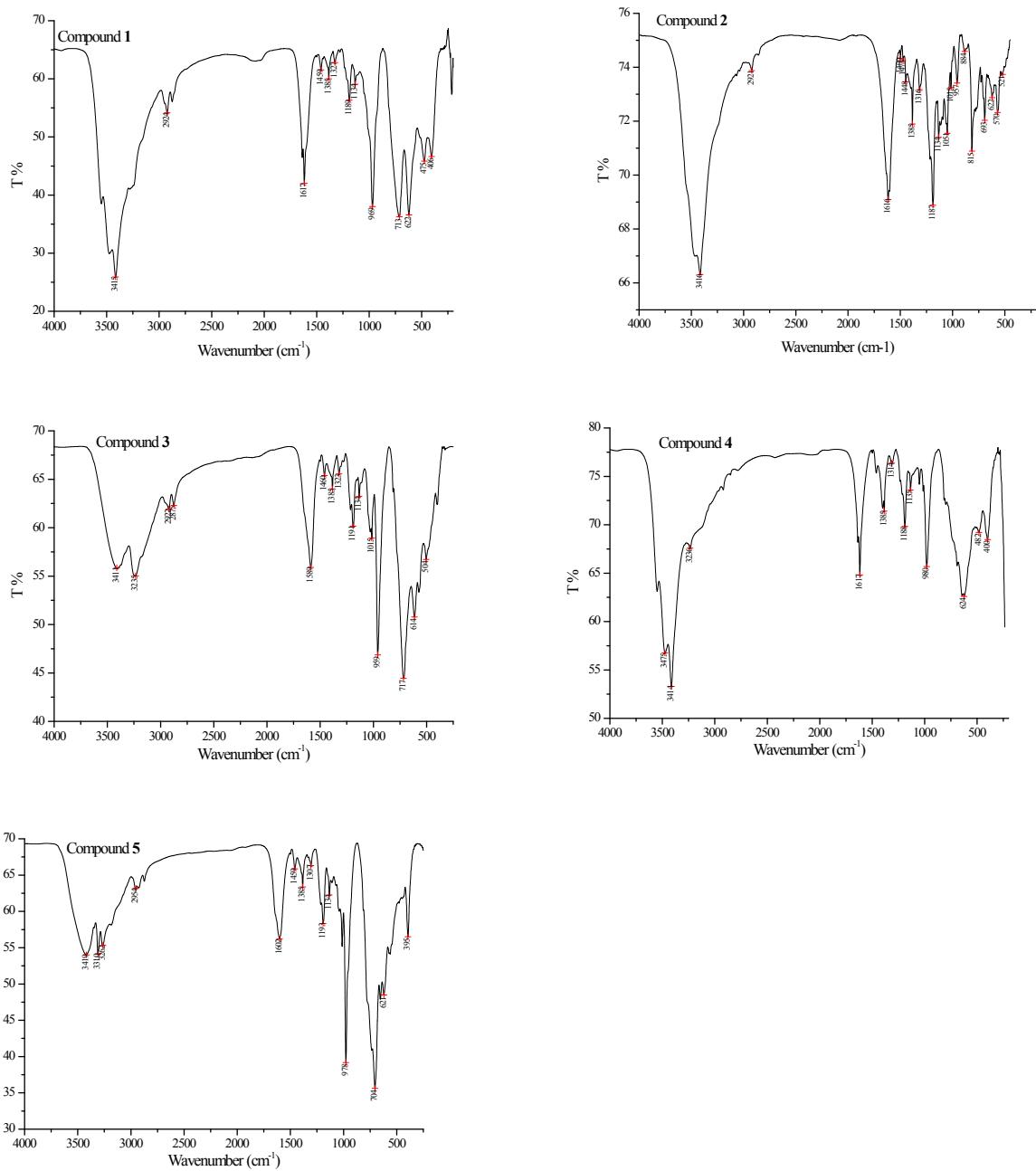


Fig. s1 IR spectra of compounds **1-5**. Infrared spectra for compounds **1-5** were recorded as KBr pellets on a Perkin-Elmer SPECTRUM ONE FTIR spectrophotometer.

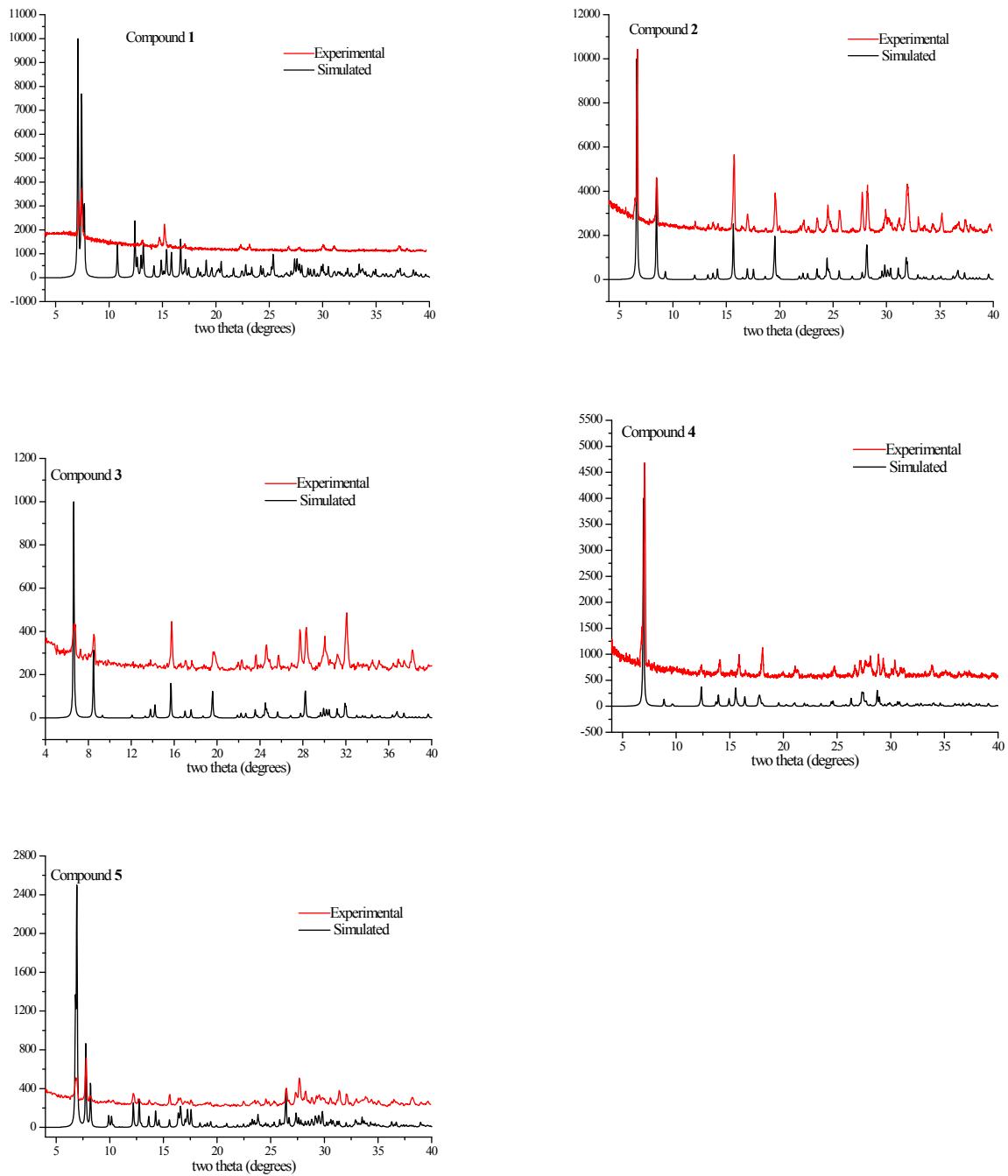


Fig. s2 experimental and simulated XRD patterns of compounds **1-5**. Powder XRD patterns for compounds **1-5** were obtained with a Scintag X1 powder diffractometer system using Cu K $\alpha$  radiation with a variable divergent slit and a solid-state detector.

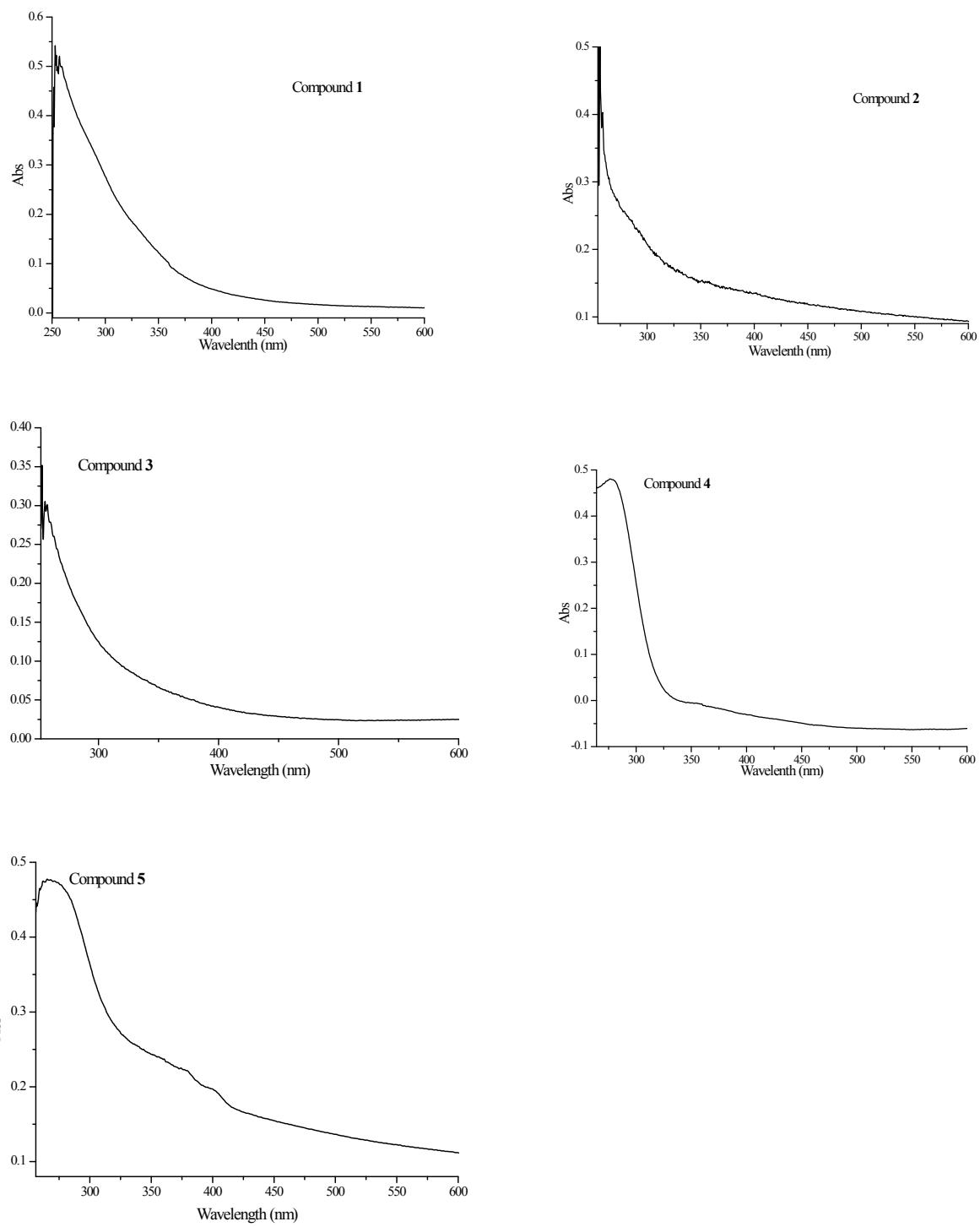
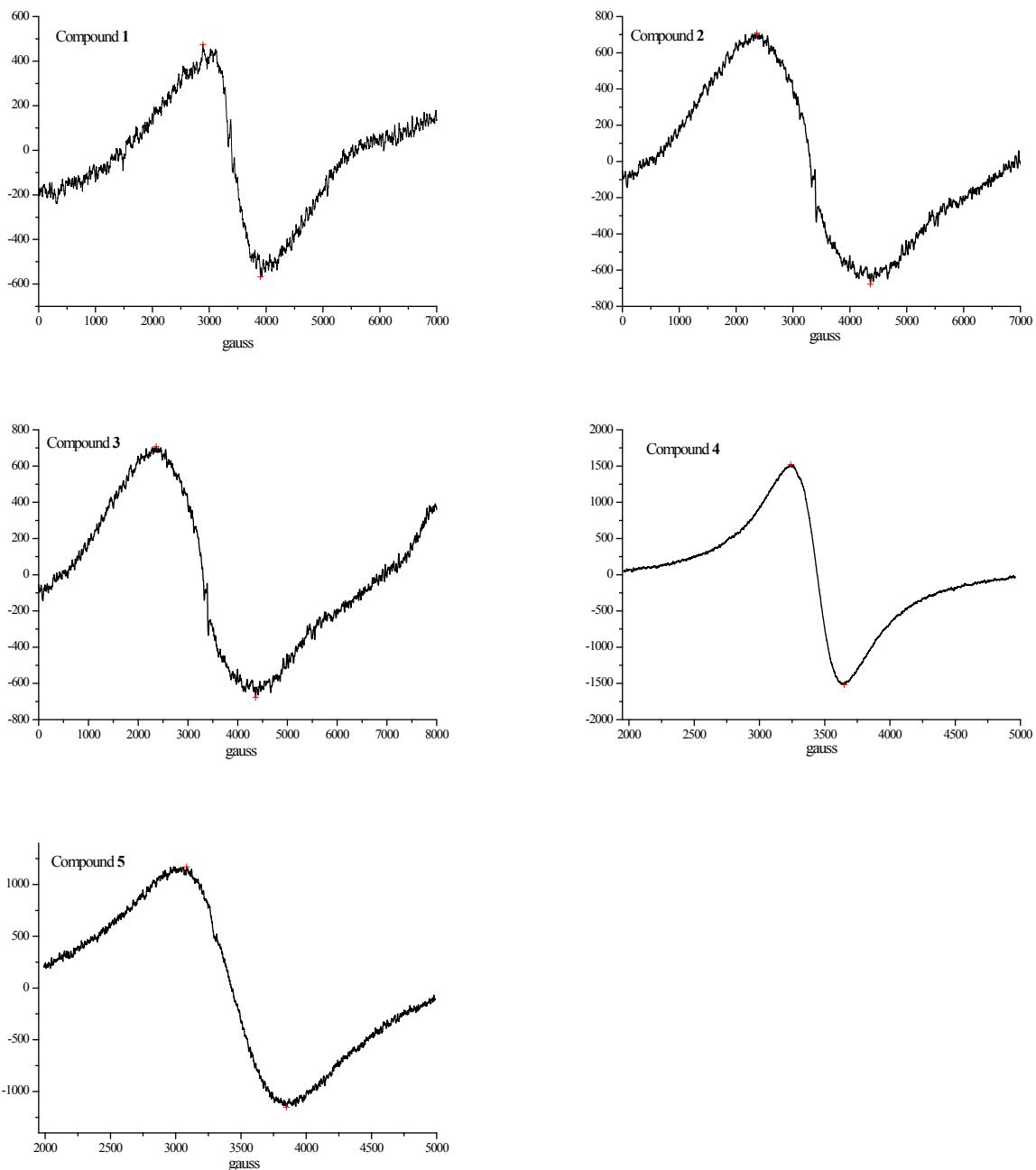


Fig. s3 UV-Vis spectra of compounds **1-5**. UV-vis spectra for compounds **1-5** were recorded on a Shimadzu UV-3100 spectrophotometer.



**Fig. s4 ESR spectra of compounds 1-5.** Electron spin resonance (ESR) spectra for compounds **1-5** were performed on a JEOL JES-FA200 spectrometer operating in the X-band mode. The g value was calculated by comparison with the spectrum of 1,1-diphenyl-2-picrylhydrazyl (DPPH), whereas the spin concentrations were determined by comparing the recorded spectra with that of a Mn marker and DPPH, using the built-in software of the spectrometer.

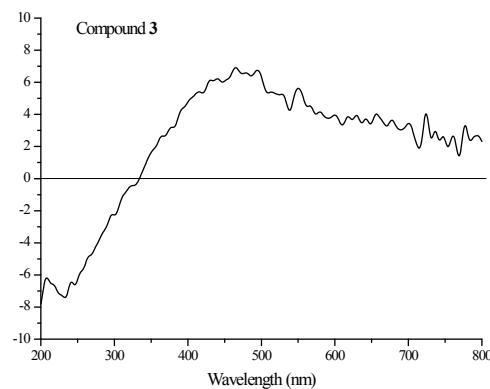
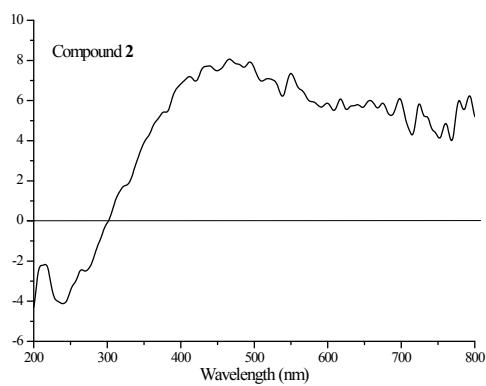
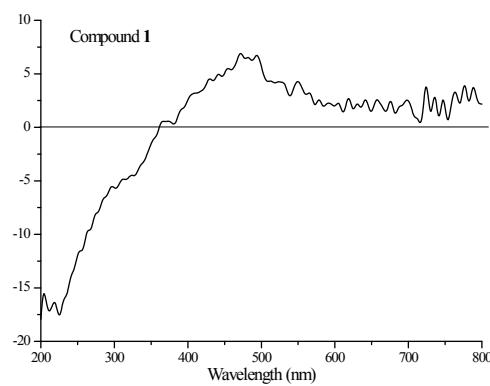


Fig. s5 solid state CD spectra of compounds **1-3**. Solid state circular dichroism (CD) spectra for compounds **1-3** were recorded at room temperature with a Jasco J-810(S) spectropolarimeter.

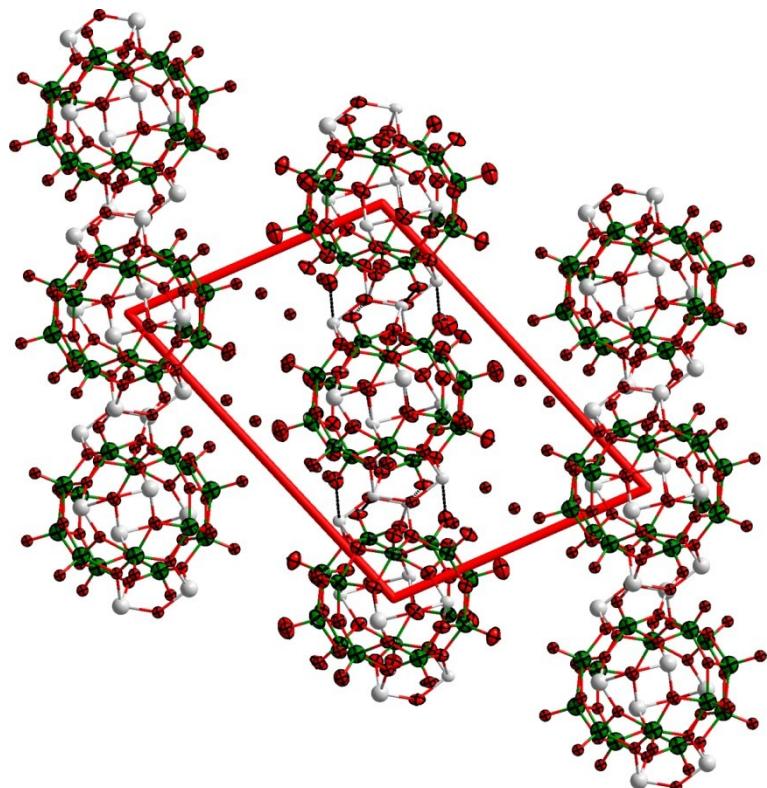


Fig. s6 ball-and-stick representation of the stacking of the inorganic layers in compound 4.