

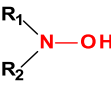
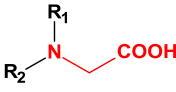
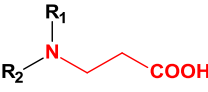
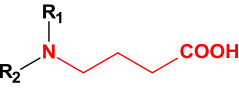
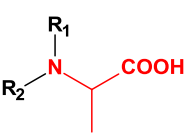
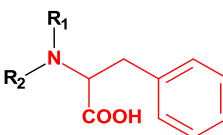
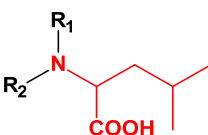
## Intermolecular Interactions and Chiral Crystallization Effects in (1,5,3-Dithiazepan-3-yl)-alkanoic Acids

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### Supporting information

**Table 1** Common occurrence of N–OH and amino acid fragments in the compounds according to found results in Cambridge Structural Database

Structural fragment				
Number of compounds	1856	398	99	24
Structural fragment				
Number of compounds	81	70	34	

**Table 2** Some bond, Å

	1	2a	2b	3	4	5	6	7	8
S1-mean plane	1.139(4)	1.110(3)	-0.710(4)	0.867(4)	0.717(5)	-1.114(3)	1.110 (6) (-1.087(5))	-1.118(3) (1.098(4))	1.126(17)
S5-mean plane	-1.084(3)	-1.105(2)	-0.836(3)	0.815(2)	0.842(6)	1.107(5)	-1.108(4) (1.124(5))	1.126(3) (-1.138(5))	-1.112(16)

**Table 3** Some bonds, Å

	1	2a	2b	3	4	5	6	7	8
C(2)–N(3)	1.461(18)	1.513(16)	1.435(19)	1.437(3)	1.426(2)	1.469(3)	1.519(4)	1.460(4)	1.467(2)
C(4)–N(3)	1.469(2)	1.498(2)	1.433(17)	1.429(3)	1.430(2)	1.482(3)	1.508(6)	1.482(4)	1.476(2)
C(4)–S(5)	1.794(2)	1.788(5)	1.836(15)	1.816(3)	1.846(18)	1.802(2)	1.786(4)	1.806(3)	1.796(16)
C(2)–S(1)	1.814(14)	1.788(17)	1.842(14)	1.829(3)	1.845(16)	1.824(19)	1.794(4)	1.815(3)	1.818(19)
N(3)–C(8)	–	1.493(2)	1.442(17)	1.463(4)	1.461(2)	1.496(2)	1.517(4)	1.487(4)	1.491(2)

**Table 4** Hydrogen bonds and diverse intermolecular contacts

Compound	D–H...A	d(D–H)	d(H...A)	d(D...A)	(D–H...A)	Symmetry code
<b>1</b>	O1–H1...H1	0.870(6)	2.192(2)	2.656(2)	113.05(6)	1–x, 1–y, –z
	C4–H4B...S1	0.952(6)	2.947(2)	3.654(3)	132.11(5)	–1+x, y, z
	O1–H1...O1	0.870(6)	2.656(2)	3.289(3)	130.59(4)	1–x, 1–y, –z
	C7–H7A...O1	0.982(9)	2.509(2)	3.283(3)	135.62(4)	x, –1+y, z
	O1–H1...N3	0.870(3)	2.006(4)	2.850(3)	163.30(2)	1–x, 1–y, –z
<b>2a</b>	C2–H2A...O1	0.920(4)	2.706(2)	3.448(2)	138.30(2)	–2–x, –0.5+y, –2–z
	C8–H8B...O1	1.092(5)	2.154(5)	3.212(2)	162.67(4)	–2–x, –0.5+y, –2–z
	O3–H3B...O2	0.850(4)	1.869(9)	2.701(1)	165.75(9)	x, y, z
	O2–H2...O3	0.820(3)	2.087(8)	2.703(9)	131.81(2)	–3–x, –0.5+y, –2–z

	O3–H3A…O1	0.850(3)	2.636(1)	3.339(2)	140.93(2)	–2–x, –0.5+y, –2–z
	C7–H7A…S1	1.026(5)	2.914(2)	3.919(2)	166.61(1)	–1–x, –0.5+y, –1–z
	C4–H4A…S5	0.934(3)	2.917(1)	3.547(2)	125.93(2)	–2–x, –0.5+y, –1–z
<b>2b</b>	C1'–H1'B…S5	0.960(2)	2.765(5)	3.477(7)	131.57(1)	2–x, –0.5+y, 1.5–z
	C7–H7A…S5	0.935(3)	2.945(8)	3.773(1)	149.27(1)	x, 1.5–y, –0.5+z
	C1'–H1'C…O1	0.960(3)	2.448(5)	3.282(7)	145.17(7)	x, y, z
	O2–H2…O1'	0.830(2)	1.772(5)	2.592(6)	169.35(9)	x, y, z
	C6–H6B…O1	0.966(2)	2.710(6)	3.675(8)	175.55(1)	2–x, 1–y, 2–z
<b>3</b>	C2–H2B…H8B	0.970(1)	2.323(2)	3.115(2)	138.37(3)	–1+x, y, z
	C8–H8A…S5	0.970(5)	2.969(2)	3.899(2)	160.92(1)	x, y, –1+z
	C7–H7A…S1	0.970(6)	2.914(2)	3.815(3)	154.87(1)	–0.5+x, 0.5–y, –0.5+z
	C9–H9A…O2	0.970(5)	2.489(2)	3.453(2)	172.83(3)	1–x, 1–y, 2–z
	C9–H9B…O2	0.970(8)	2.700(2)	3.609(2)	156.36(6)	–1+x, y, z
	O1–H1…O2	0.879(9)	1.771(2)	2.641(2)	170.26(8)	–x, 1–y, 1–z
<b>4</b>	O2–H2C…O1	0.783(6)	1.854(1)	2.631(2)	172.07(8)	–x, 3–y, –z
	C10–H10B…O2	0.991(5)	2.544(1)	3.442(2)	150.55(3)	–x, 2–y, –z
	C9–H9A…O2	1.000(9)	2.686(2)	3.539(2)	143.48(9)	–1+x, y, z
<b>5</b>	C9–H9A…S5	0.980(5)	2.885(2)	3.524(2)	123.65(4)	–1+x, y, z
	C7–H7A…O2	0.888(5)	2.704(2)	3.424(2)	138.93(3)	–x, –0.5+y, 1–z
	C7–H7A…O1	0.888(5)	2.707(2)	3.461(2)	143.31(3)	x, y, –1+z
	C2–H2B…O1	0.971(7)	2.701(1)	3.211(2)	113.36(5)	–x, –0.5+y, –z
	C9–H9B…O1	1.006(5)	2.716(2)	3.354(2)	121.52(5)	–x, –0.5+y, –z
<b>6</b>	C11–H11B…H6A	0.960(6)	2.196(9)	2.984(1)	138.61(9)	–1.5–x, –1–y, –0.5+z
	C13–H13B…H6A	0.960(6)	2.263(9)	3.035(1)	136.91(2)	–1.5–x, –1–y, –0.5+z
	C4–H4B…H2'	0.939(5)	2.328(9)	2.523(1)	90.84(5)	–0.5+x, –1.5–y, –1–z
	C2'–H2'A…S5	0.998(5)	2.970(1)	3.755(2)	136.33(4)	x, y, z
	C13–H13C…S5'	0.960(6)	2.973(2)	3.900(2)	162.60(2)	–0.5+x, –0.5–y, –1–z
	C8–H8…O1'	0.986(5)	2.691(2)	3.594(2)	152.35(6)	x, y, z
	C4–H4A…O1'	0.904(4)	2.353(1)	3.257(1)	178.65(4)	x, y, z
	C2–H2A…O1'	0.970(5)	2.677(2)	3.198(2)	114.04(2)	–0.5+x, –1.5–y, –1–z
	C2'–H2'B…O2	0.989(6)	2.603(1)	3.299(2)	127.41(5)	–0.5+x, –0.5–y, –1–z
	C4'–H4'B…O2	0.970(4)	2.319(1)	3.164(5)	145.16(9)	–1+x, y, z
<b>7</b>	C2–H2A…H2'	0.977(2)	2.398(3)	2.768(5)	102.09(1)	x, y, z
	C4–H4B…O3	0.917(9)	2.645(4)	3.170(4)	117.19(8)	–1+x, y, z
	C6–H6A…O1	0.970(2)	2.624(3)	3.593(5)	176.72(9)	–1+x, y, 1+z
	C7–H7A…O3	1.037(1)	2.642(3)	3.398(4)	129.55(9)	–1+x, y, 1+z
	O1–H1A…O3	0.686(1)	1.977(3)	2.640(4)	162.72(3)	x, –1+y, z
	C7–H7A…O3'	1.037(2)	2.399(3)	3.184(4)	131.62(9)	x, –1+y, 1+z
	O2'–H2'…O2	0.820(1)	2.593(3)	3.141(4)	125.55(1)	x, y, z
	O1'–H1'…O3'	0.820(1)	2.025(4)	2.674(5)	135.72(9)	1+x, y, z
	C6'–H6'B…O3'	0.970(2)	2.396(5)	3.228(7)	143.60(9)	–1+x, y, 1+z
	C9–H9B…S1'	0.970(2)	2.950(4)	3.618(6)	127.02(1)	x, –1+y, 1+z
	C9'–H9'B…S5'	1.083(1)	2.638(3)	3.659(5)	156.90(4)	x, y, z
<b>8</b>	C4–H4B…H7A	1.006(3)	2.373(6)	2.745(7)	100.77(1)	–0.5+x, y, 1.5–z
	C13–H13…S1	0.977(3)	2.942(8)	3.775(9)	143.90(3)	–0.5+x, 0.5–y, 1–z
	O1–H1…C8	0.910(3)	2.639(8)	3.404(1)	142.14(2)	1.5–x, –0.5+y, z
	O1–H1…C4	0.910(3)	2.493(6)	3.257(8)	141.70(1)	1.5–x, –0.5+y, z
	O1–H1…C2	0.910(3)	2.641(7)	3.487(1)	154.97(9)	1.5–x, –0.5+y, z
	C7–H7A…C4	0.979(2)	2.745(7)	3.524(9)	136.87(1)	–0.5+x, y, 1.5–z

**Table 5** S…S contacts vs conformation of dithiazepane cycle

	1	2a	2b	3	4	5	6	7	8
<i>d</i> (S…S), Å	3.598(5)	–	–	–	–	3.380(4)	–	3.596(7)	3.568(8)
Conformation	chair	chair	boat	boat	boat	chair	chair	chair	chair

**Table 6** Crystal data and structure refinement for compounds 1–8

	1	2a	2b	3
<b>I formula</b>	C <sub>4</sub> H <sub>9</sub> N <sub>1</sub> O <sub>1</sub> S <sub>2</sub>	C <sub>6</sub> H <sub>11</sub> NO <sub>2</sub> S <sub>2</sub> ·H <sub>2</sub> O	C <sub>6</sub> H <sub>11</sub> NO <sub>2</sub> S <sub>2</sub> , C <sub>2</sub> H <sub>6</sub> OS	C <sub>7</sub> H <sub>13</sub> NO <sub>2</sub> S <sub>2</sub>
<b>Formula weight</b>	151.24	211.29	271.41	206.30
<b>Crystal</b>	triclinic	monoclinic	monoclinic	monoclinic

<b>system</b>				
<b>Space group</b>	P-1	P2 <sub>1</sub>	P2 <sub>1</sub> /c	P2 <sub>1</sub> /n
<b>a, Å</b>	6.5256(4)	6.6882(3)Å	6.58182(17)	4.9212(5)
<b>b, Å</b>	6.9753(5)	8.6039(4)	16.4019(4)	30.519(2)
<b>c, Å</b>	8.1848(5)	8.0793(4)	11.5843(4)	6.7336(4)
<b>α, °</b>	110.266(6)	90.00°	90.00°	90
<b>β, °</b>	96.816(5)	94.507(5)	98.923(3)	103.709(8)
<b>γ, °</b>	94.924(5)	90.00°	90.00°	90
<b>V, Å<sup>3</sup></b>	343.79(4)	463.48(4)	1235.44(6)	982.53(14)
<b>Z</b>	2	2	4	4
<b>D<sub>calc</sub>/cm<sup>3</sup></b>	1.461	1.514	1.459	1.395
<b>F(000)</b>	160	224	576.0	436
<b>2θ<sub>max</sub>, °</b>	62.34°	62.378	62.2	62.42
<b>Reflections collected</b>	2975	2436	6313	3485
<b>Independent reflections (R<sub>int</sub>)</b>	1957[R(int) = 0.0115]	2054 (R <sub>int</sub> = 0.0084)	3538 (R <sub>int</sub> = 0.0137)	2291 (R <sub>int</sub> = 0.0168)
<b>S</b>	0.857	0.853	1.067	1.057
<b>R<sub>1</sub> [I ≥ 2σ(I)]</b>	0.0274	0.0303	0.0318	0.0536
<b>wR<sub>2</sub></b>	0.1031	0.0888	0.0762	0.1448
<b>Largest diff. peak/hole, e Å<sup>-3</sup></b>	0.34/-0.22	0.44/-0.50	0.30/-0.27	0.50/-0.63
<b>Flack parameter</b>	n/a	-0.13(7)	n/a	n/a

	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>
<b>Empirical formula</b>	C <sub>8</sub> H <sub>15</sub> NO <sub>2</sub> S <sub>2</sub>	C <sub>7</sub> H <sub>13</sub> NO <sub>2</sub> S <sub>2</sub>	C <sub>7</sub> H <sub>13</sub> NO <sub>3</sub> S <sub>2</sub>	C <sub>10</sub> H <sub>19</sub> NO <sub>2</sub> S <sub>2</sub>	C <sub>13</sub> H <sub>17</sub> NO <sub>2</sub> S <sub>2</sub>
<b>Formula weight</b>	221.33	207.30	223.30	249.38	283.4
<b>Crystal system</b>	triclinic	monoclinic	triclinic	orthorhombic	orthorhombic
<b>Space group</b>	P-1	P2 <sub>1</sub>	P1	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	Pbca
<b>a, Å</b>	5.0968(3)	6.4993(6)	6.8616(11)	10.6926(6)	10.5874(4)
<b>b, Å</b>	6.7869(3)	8.6630(6)	8.8688(8)	10.9871(7)	10.2833(3)
<b>c, Å</b>	16.6360(10)	8.5034(5)	8.9458(10)	21.4364(15)	25.2185(10)
<b>α, °</b>	78.801(5)	90	66.923(10)	90	90

$\beta, ^\circ$	87.344(5)	104.738(7)	70.122(13)	90	90
$\gamma, ^\circ$	69.141(5)	90	85.373(10)	90	90
$V, \text{\AA}^3$	527.34(5)	463.01(6)	470.11(11)	2518.3(3)	2745.64(17)
$Z$	2	2	2	8	8
$D_{\text{calc}}/\text{cm}^3$	1.394	1.487	1.577	1.316	1.371
$F(000)$	236	220.0	236.0	1072.0	1200
$2\theta_{\text{max}}, ^\circ$	62.12	62.494	61.838	62.24	62.22
Reflections collected	4589	2474	4183	8525	9252
Independent reflections ( $R_{\text{int}}$ )	3000 ( $R_{\text{int}} = 0.0162$ )	2052 ( $R_{\text{int}} = 0.0178$ )	3248 ( $R_{\text{int}} = 0.0782$ )	6260 ( $R_{\text{int}} = 0.0762$ )	3986 ( $R_{\text{int}} = 0.0181$ )
$S$	0.943	1.108	0.864	1.012	1.110
$R_1 [I \geq 2\sigma(I)]$	0.0384	0.0381	0.0467	0.0602	0.0464
$wR_2$	0.1328	0.1108	0.1182	0.1211	0.1176
Largest diff. peak/hole, $e \text{\AA}^{-3}$	0.74/-0.43	0.42/-0.61	0.43/-0.29	0.75/-0.47	0.32/-0.32
Flack parameter	n/a	-0.02(12)	-0.12(9)	-0.11(9)	n/a

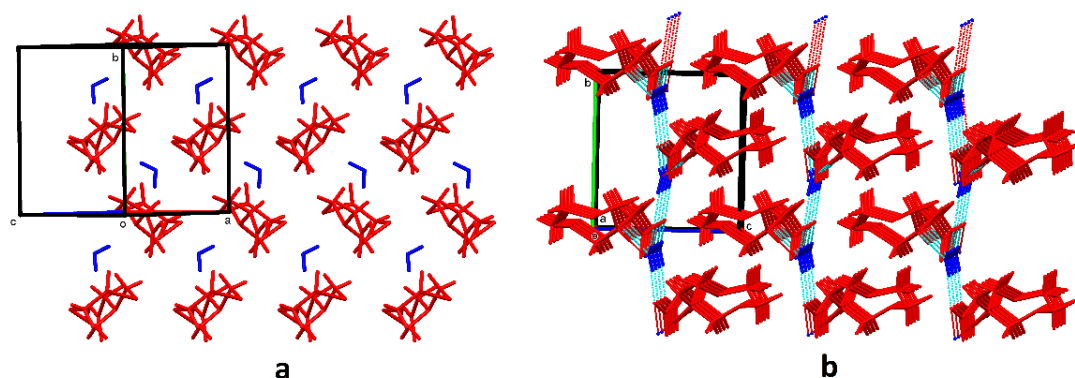
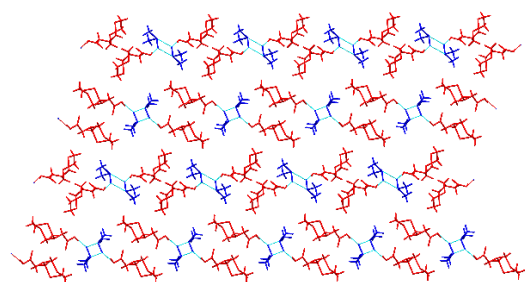
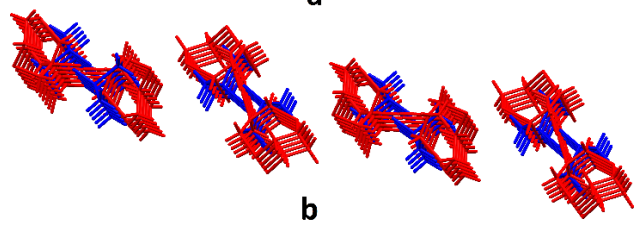


Fig. 1. Projections of clathrate **2a**: (a) along  $a$  axis and (b) along  $b$  axis. The molecules of the main compound are painted with red colour and water molecules are with blue colour.



**a**



**b**

**Fig. 2.** Projections of tetrameric associates **2b** along two orthogonal axes. The molecules of the main compound are in red, whereas DMSO are in blue.