

Anion- and temperature-dependent assembly, crystal structures and luminescent properties of six new Cd(II) coordination polymers based on 2,3,5,6-tetrakis(2-pyridyl)pyrazine

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Supporting information

Table S1 Selected bond lengths [Å] and angles[°] for the compounds 1-6.^a

| Compound 1 | | | |
|-------------------|------------|--------------------|------------|
| Cd(1)-N(4) | 2.314(3) | Cd(1)-N(2) | 2.426(3) |
| Cd(1)-N(3) | 2.390(3) | Cd(1)-N(4)#1 | 2.443(3) |
| Cd(1)-N(1) | 2.403(3) | Cd(1)-Cl(1) | 2.4834(11) |
| N(4)-Cd(1)-N(3) | 96.56(12) | N(1)-Cd(1)-N(4)#1 | 85.26(11) |
| N(4)-Cd(1)-N(1) | 124.95(12) | N(2)-Cd(1)-N(4)#1 | 78.92(10) |
| N(3)-Cd(1)-N(1) | 136.05(10) | N(4)-Cd(1)-Cl(1) | 96.20(8) |
| N(4)-Cd(1)-N(2) | 151.06(11) | N(3)-Cd(1)-Cl(1) | 99.66(9) |
| N(3)-Cd(1)-N(2) | 67.99(9) | N(1)-Cd(1)-Cl(1) | 90.36(8) |
| N(1)-Cd(1)-N(2) | 68.37(9) | N(2)-Cd(1)-Cl(1) | 110.12(7) |
| N(4)-Cd(1)-N(4)#1 | 77.11(12) | N(4)#1-Cd(1)-Cl(1) | 167.70(9) |
| N(3)-Cd(1)-N(4)#1 | 91.42(12) | | |
| Compound 2 | | | |
| Cd(1)-N(5) | 2.367(3) | Cd(1)-O(1) | 2.399(3) |
| Cd(1)-N(3) | 2.367(3) | Cd(1)-N(2) | 2.411(2) |
| Cd(1)-N(1) | 2.376(3) | Cd(1)-O(2) | 2.431(3) |
| Cd(1)-N(7)#1 | 2.357(3) | | |

| | | | |
|-------------------|------------|-------------------|------------|
| N(7)#1-Cd(1)-N(5) | 173.88(10) | N(5)-Cd(1)-N(2) | 80.76(10) |
| N(7)#1-Cd(1)-N(3) | 92.18(10) | N(3)-Cd(1)-N(2) | 68.18(9) |
| N(5)-Cd(1)-N(3) | 87.52(10) | N(1)-Cd(1)-N(2) | 69.28(9) |
| N(7)#1-Cd(1)-N(1) | 84.60(11) | O(1)-Cd(1)-N(2) | 151.52(9) |
| N(5)-Cd(1)-N(1) | 91.46(10) | N(7)#1-Cd(1)-O(2) | 88.49(11) |
| N(3)-Cd(1)-N(1) | 137.04(9) | N(5)-Cd(1)-O(2) | 95.97(11) |
| N(7)#1-Cd(1)-O(1) | 92.80(11) | N(3)-Cd(1)-O(2) | 136.14(10) |
| N(5)-Cd(1)-O(1) | 93.25(11) | N(1)-Cd(1)-O(2) | 86.70(10) |
| N(3)-Cd(1)-O(1) | 83.84(9) | O(1)-Cd(1)-O(2) | 52.34(10) |
| N(1)-Cd(1)-O(1) | 139.03(9) | N(2)-Cd(1)-O(2) | 155.57(10) |
| N(7)#1-Cd(1)-N(2) | 93.44(10) | | |

Compound 3

| | | | |
|-------------------|------------|-------------------|------------|
| Cd(1)-N(4) | 2.348(4) | Cd(1)-N(6)#1 | 2.407(4) |
| Cd(1)-O(1) | 2.354(3) | Cd(1)-N(3) | 2.415(3) |
| Cd(1)-O(2) | 2.403(3) | Cd(1)-N(2) | 2.455(3) |
| Cd(1)-N(1) | 2.405(3) | | |
| N(4)-Cd(1)-O(1) | 97.50(12) | O(1)-Cd(1)-N(3) | 139.56(10) |
| N(4)-Cd(1)-O(2) | 104.78(12) | O(2)-Cd(1)-N(3) | 85.47(9) |
| O(1)-Cd(1)-O(2) | 54.53(9) | N(1)-Cd(1)-N(3) | 134.80(9) |
| N(4)-Cd(1)-N(1) | 93.25(12) | N(6)#1-Cd(1)-N(3) | 90.65(10) |
| O(1)-Cd(1)-N(1) | 85.37(10) | N(4)-Cd(1)-N(2) | 82.25(11) |
| O(2)-Cd(1)-N(1) | 137.34(9) | O(1)-Cd(1)-N(2) | 152.46(10) |
| N(4)-Cd(1)-N(6)#1 | 168.84(12) | O(2)-Cd(1)-N(2) | 152.24(10) |
| O(1)-Cd(1)-N(6)#1 | 91.79(11) | N(1)-Cd(1)-N(2) | 67.20(9) |
| O(2)-Cd(1)-N(6)#1 | 85.64(11) | N(6)#1-Cd(1)-N(2) | 86.66(10) |
| N(1)-Cd(1)-N(6)#1 | 81.31(11) | N(3)-Cd(1)-N(2) | 67.98(9) |
| N(4)-Cd(1)-N(3) | 86.29(11) | | |

Compound 4

| | | | |
|----------------------|------------|------------------|------------|
| Cd(1)-N(7) | 2.323(5) | Cd(2)-N(4) | 2.312(4) |
| Cd(1)-N(10) | 2.363(5) | Cd(2)-N(2) | 2.422(4) |
| Cd(1)-N(4) | 2.366(4) | Cd(2)-N(1) | 2.422(4) |
| Cd(2)-N(12) | 2.261(5) | Cd(2)-N(3) | 2.449(5) |
| Cd(2)-N(7) | 2.300(5) | | |
| N(7)#1-Cd(1)-N(7) | 101.2(3) | N(12)-Cd(2)-N(7) | 90.41(19) |
| N(7)#1-Cd(1)-N(10)#1 | 87.79(18) | N(12)-Cd(2)-N(4) | 167.17(18) |
| N(7)-Cd(1)-N(10)#1 | 169.26(17) | N(7)-Cd(2)-N(4) | 78.04(16) |
| N(7)#1-Cd(1)-N(10) | 169.26(17) | N(12)-Cd(2)-N(2) | 104.44(17) |
| N(7)-Cd(1)-N(10) | 87.79(18) | N(7)-Cd(2)-N(2) | 152.55(16) |
| N(10)#1-Cd(1)-N(10) | 84.0(3) | N(4)-Cd(2)-N(2) | 88.39(15) |
| N(7)#1-Cd(1)-N(4) | 95.67(16) | N(12)-Cd(2)-N(1) | 84.39(19) |
| N(7)-Cd(1)-N(4) | 76.51(16) | N(7)-Cd(2)-N(1) | 137.29(16) |
| N(10)#1-Cd(1)-N(4) | 96.86(16) | N(4)-Cd(2)-N(1) | 100.24(16) |
| N(10)-Cd(1)-N(4) | 92.18(17) | N(2)-Cd(2)-N(1) | 68.24(14) |

| | | | |
|----------------------|-----------|------------------|------------|
| N(7)#1-Cd(1)-N(4)#1 | 76.50(16) | N(12)-Cd(2)-N(3) | 88.85(18) |
| N(7)-Cd(1)-N(4)#1 | 95.67(16) | N(7)-Cd(2)-N(3) | 92.53(16) |
| N(10)#1-Cd(1)-N(4)#1 | 92.18(17) | N(4)-Cd(2)-N(3) | 97.14(16) |
| N(10)-Cd(1)-N(4)#1 | 96.86(16) | N(2)-Cd(2)-N(3) | 65.35(14) |
| N(4)-Cd(1)-N(4)#1 | 167.8(2) | N(1)-Cd(2)-N(3) | 129.56(15) |

Compound 5

| | | | |
|-----------------|------------|--------------------|------------|
| Cd(1)-N(8) | 2.290(4) | Cd(2)-N(10) | 2.306(4) |
| Cd(1)-N(4) | 2.313(4) | Cd(2)-N(4)#1 | 2.338(3) |
| Cd(1)-N(1) | 2.386(3) | Cd(2)-N(13) | 2.369(3) |
| Cd(1)-O(2) | 2.404(3) | Cd(2)-O(5) | 2.370(3) |
| Cd(1)-N(2) | 2.436(3) | Cd(2)-O(4) | 2.432(3) |
| Cd(1)-N(3) | 2.447(3) | Cd(2)-N(11) | 2.435(3) |
| Cd(1)-O(1) | 2.464(3) | Cd(2)-N(12) | 2.435(3) |
| N(8)-Cd(1)-N(4) | 173.02(14) | N(10)-Cd(2)-N(4)#1 | 173.85(13) |
| N(8)-Cd(1)-N(1) | 89.96(14) | N(10)-Cd(2)-N(13) | 89.16(14) |
| N(4)-Cd(1)-N(1) | 95.62(13) | N(4)#1-Cd(2)-N(13) | 93.80(13) |
| N(8)-Cd(1)-O(2) | 83.09(13) | N(10)-Cd(2)-O(5) | 87.83(14) |
| N(4)-Cd(1)-O(2) | 92.96(12) | N(4)#1-Cd(2)-O(5) | 97.54(13) |
| N(1)-Cd(1)-O(2) | 87.35(12) | N(13)-Cd(2)-O(5) | 90.16(12) |
| N(8)-Cd(1)-N(2) | 100.49(13) | N(10)-Cd(2)-O(4) | 94.84(14) |
| N(4)-Cd(1)-N(2) | 85.55(12) | N(4)#1-Cd(2)-O(4) | 86.07(13) |
| N(1)-Cd(1)-N(2) | 67.71(11) | N(13)-Cd(2)-O(4) | 142.65(11) |
| O(2)-Cd(1)-N(2) | 154.69(12) | O(5)-Cd(2)-O(4) | 53.03(11) |
| N(8)-Cd(1)-N(3) | 88.76(14) | N(10)-Cd(2)-N(11) | 83.60(13) |
| N(4)-Cd(1)-N(3) | 90.34(12) | N(4)#1-Cd(2)-N(11) | 90.48(12) |
| N(1)-Cd(1)-N(3) | 134.20(11) | N(13)-Cd(2)-N(11) | 134.30(11) |
| O(2)-Cd(1)-N(3) | 137.75(12) | O(5)-Cd(2)-N(11) | 134.28(12) |
| N(2)-Cd(1)-N(3) | 67.56(11) | O(4)-Cd(2)-N(11) | 83.03(11) |
| N(8)-Cd(1)-O(1) | 83.75(14) | N(10)-Cd(2)-N(12) | 93.40(13) |
| N(4)-Cd(1)-O(1) | 89.28(13) | N(4)#1-Cd(2)-N(12) | 82.73(12) |
| N(1)-Cd(1)-O(1) | 139.71(12) | N(13)-Cd(2)-N(12) | 68.18(11) |
| O(2)-Cd(1)-O(1) | 52.42(12) | O(5)-Cd(2)-N(12) | 158.27(12) |
| N(2)-Cd(1)-O(1) | 152.55(11) | O(4)-Cd(2)-N(12) | 148.06(11) |
| N(3)-Cd(1)-O(1) | 85.56(12) | N(11)-Cd(2)-N(12) | 67.31(11) |

Compound 6

| | | | |
|-------------------|------------|-----------------|------------|
| Cd(1)-N(7)#1 | 2.299(5) | Cd(2)-N(4) | 2.225(4) |
| Cd(1)-N(4) | 2.318(4) | Cd(2)-N(10) | 2.239(5) |
| Cd(1)-N(1) | 2.372(4) | Cd(2)-N(7) | 2.301(5) |
| Cd(1)-N(3) | 2.375(4) | Cd(2)-N(12)#2 | 2.352(5) |
| Cd(1)-N(2) | 2.448(4) | Cd(2)-O(2)#3 | 2.395(4) |
| Cd(1)-O(2) | 2.466(4) | Cd(2)-O(1) | 2.474(4) |
| Cd(1)-O(1) | 2.480(4) | | |
| N(7)#1-Cd(1)-N(4) | 174.80(17) | N(3)-Cd(1)-O(1) | 128.34(13) |

| | | | |
|-------------------|------------|----------------------|------------|
| N(7)#1-Cd(1)-N(1) | 86.88(16) | N(2)-Cd(1)-O(1) | 155.25(14) |
| N(4)-Cd(1)-N(1) | 90.26(16) | O(2)-Cd(1)-O(1) | 52.54(13) |
| N(7)#1-Cd(1)-N(3) | 100.06(16) | N(4)-Cd(2)-N(10) | 160.87(19) |
| N(4)-Cd(1)-N(3) | 85.04(15) | N(4)-Cd(2)-N(7) | 93.09(17) |
| N(1)-Cd(1)-N(3) | 134.54(13) | N(10)-Cd(2)-N(7) | 102.7(2) |
| N(7)#1-Cd(1)-N(2) | 89.59(16) | N(4)-Cd(2)-N(12)#2 | 95.00(17) |
| N(4)-Cd(1)-N(2) | 93.30(14) | N(10)-Cd(2)-N(12)#2 | 96.00(18) |
| N(1)-Cd(1)-N(2) | 67.21(13) | N(7)-Cd(2)-N(12)#2 | 88.88(18) |
| N(3)-Cd(1)-N(2) | 67.97(14) | N(4)-Cd(2)-O(2)#3 | 86.98(15) |
| N(7)#1-Cd(1)-O(2) | 75.41(15) | N(10)-Cd(2)-O(2)#3 | 86.24(17) |
| N(4)-Cd(1)-O(2) | 103.62(14) | N(7)-Cd(2)-O(2)#3 | 76.79(15) |
| N(1)-Cd(1)-O(2) | 133.37(13) | N(12)#2-Cd(2)-O(2)#3 | 165.63(16) |
| N(3)-Cd(1)-O(2) | 91.39(14) | N(4)-Cd(2)-O(1) | 73.91(14) |
| N(2)-Cd(1)-O(2) | 152.21(14) | N(10)-Cd(2)-O(1) | 90.55(19) |
| N(7)#1-Cd(1)-O(1) | 103.50(16) | N(7)-Cd(2)-O(1) | 166.73(15) |
| N(4)-Cd(1)-O(1) | 72.26(14) | N(12)#2-Cd(2)-O(1) | 89.59(17) |
| N(1)-Cd(1)-O(1) | 92.28(13) | O(2)#3-Cd(2)-O(1) | 104.61(14) |

^aSymmetry transformations used to generate equivalent atoms: #1 -x+1,-y+1,-z+1 for compound **1**; #1 x-1,y,z for compound **2**; #1 -x+1/2,y+1/2,-z+3/2 for compound **3**; #1 -x+1,y,-z+5/2 for compound **4**; #1 x,-y+1,z+1/2 for compound **5**; #1 x,-y+1/2,z+1/2 #2 -x+2,-y+1,-z+1 #3 x,-y+1/2,z-1/2 for compound **6**.

Table S2 Selected details of the hydrogen bonds in compounds **2-3**.

| | symmetry code | $d(\text{D-H})/\text{\AA}$ | $d(\text{H}\cdots\text{A})/\text{\AA}$ | $d(\text{D}\cdots\text{A})/\text{\AA}$ | $\angle\text{D-H}\cdots\text{A}/^\circ$ |
|---------------------------|---------------|----------------------------|--|--|---|
| Compound 2 | | | | | |
| C(9)-H(9) \cdots O(3) | x,y-1,z | 0.93 | 2.53 | 3.161(5) | 125 |
| C(10)-H(10) \cdots O(3) | x,y-1,z | 0.93 | 2.66 | 3.221(5) | 119 |
| Compound 3 | | | | | |
| O(3)-H(3A) \cdots O(2) | x, y, z | 0.85 | 2.00 | 2.844(4) | 178.5 |
| O(3)-H(3B) \cdots N(4) | -x,-y+1,-z+1 | 0.85 | 2.16 | 3.007(5) | 175.7 |

Table S3. Structural comparison of the reported Cd-tppz compounds.^b

| Compound | $[\text{Cd}_2(\text{tppz})\text{Cl}_4]^{11\text{h}}$ | $[\text{Cd}_2(\text{tppz})(\text{NO}_3)_4(\text{MeOH})_2]^{11\text{f}}$ | $\{[\text{Cd}_2(\text{tppz})\text{Cl}_4] \cdot \text{EG}\}_n^{11\text{g}}$ | $\{[\text{Cd}_2(\text{N}_3)_2(\text{tppz})(\text{NO}_3)_2] \cdot 2\text{MeOH}\}_n^{11\text{b}}$ | $\{[\text{Cd}_2(\text{N}_3)_4(\text{tppz})]\}_n^{11\text{b}}$ | $\{[\text{Cd}_2(\text{tppz})(\text{pic})\text{Cl}_3]\}_n^{11\text{h}}$ |
|--|--|---|--|---|---|--|
| Cd(II) salt used for synthesis | CdCl_2 | $\text{Cd}(\text{NO}_3)_2$ | CdCl_2 | $\text{Cd}(\text{NO}_3)_2$ | $\text{Cd}(\text{NO}_3)_2$ | CdCl_2 |
| Main ligand used for synthesis | pmpc | tppz | tppz | tppz | tppz | pmpc |
| Coordination mode of the tppz ligand | bis-tridentate μ_2 -bridge | bis-tridentate μ_2 -bridge | bis-tridentate μ_2 -bridge | bis-tridentate μ_2 -bridge | bis-tridentate μ_2 -bridge | bis-tridentate μ_2 -bridge |
| Coordination number of Cd(II) ions | 5-coordinated | 7-coordinated | 6-coordinated | 7-coordinated | 7-coordinated | 6- and 7-coordinated |
| Anion of the Cd(II) salt remained in the structure | Cl^- | NO_3^- | Cl^- | NO_3^- | None | Cl^- |
| Coordination mode of the anion of the Cd(II) salt | μ_1 | μ_1 | μ_1 and μ_2 | μ_1 | None | μ_2 and μ_3 |
| Coordination mode of the azide co-ligand | None | None | None | $\mu_{1,1}\text{-N}_3^-$ | $\mu_{1,1}\text{-N}_3^-$ and $\mu_{1,3}\text{-N}_3^-$ | None |
| Dimensionality / topology of the structure | 0D dinuclear cluster | 0D dinuclear cluster | 1D chain with loops and rods topology | 1D chain with loops and rods topology | 2D (6,3) network | 2D (4,4) network |

^b EG = ethylene glycol; pic = picolinate; pmpc = *N*-(2-pyridyl-methyl) pyridine-2-carbalimine.

References 11b and 11f – 11h are provided in the main text.

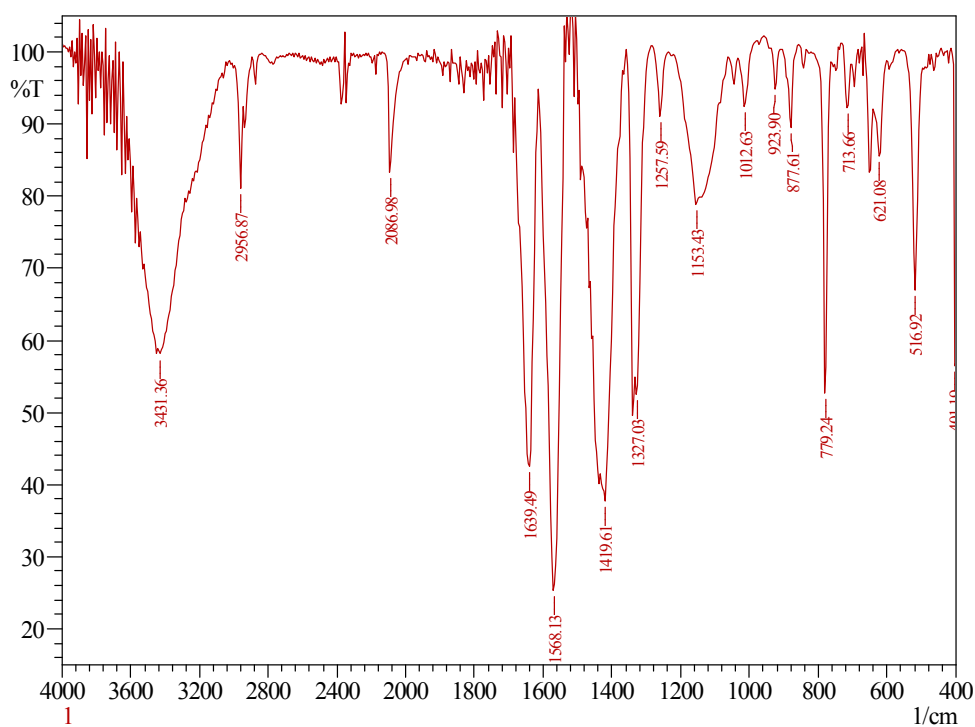


Figure S1. FT-IR spectrum of compound 1.

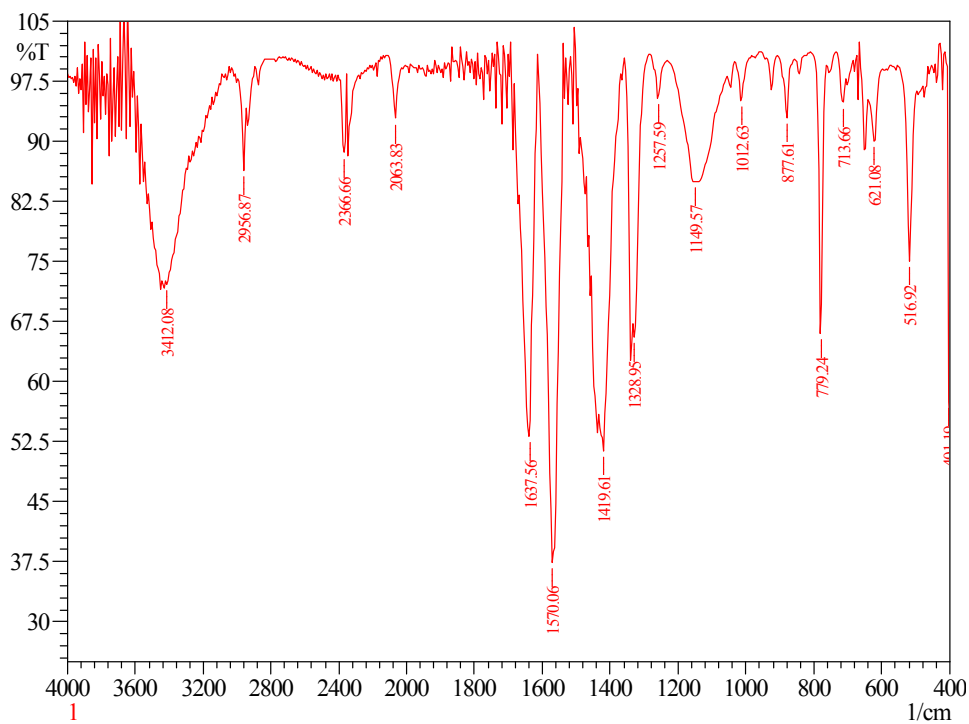


Figure S2. FT-IR spectrum of compound 2.

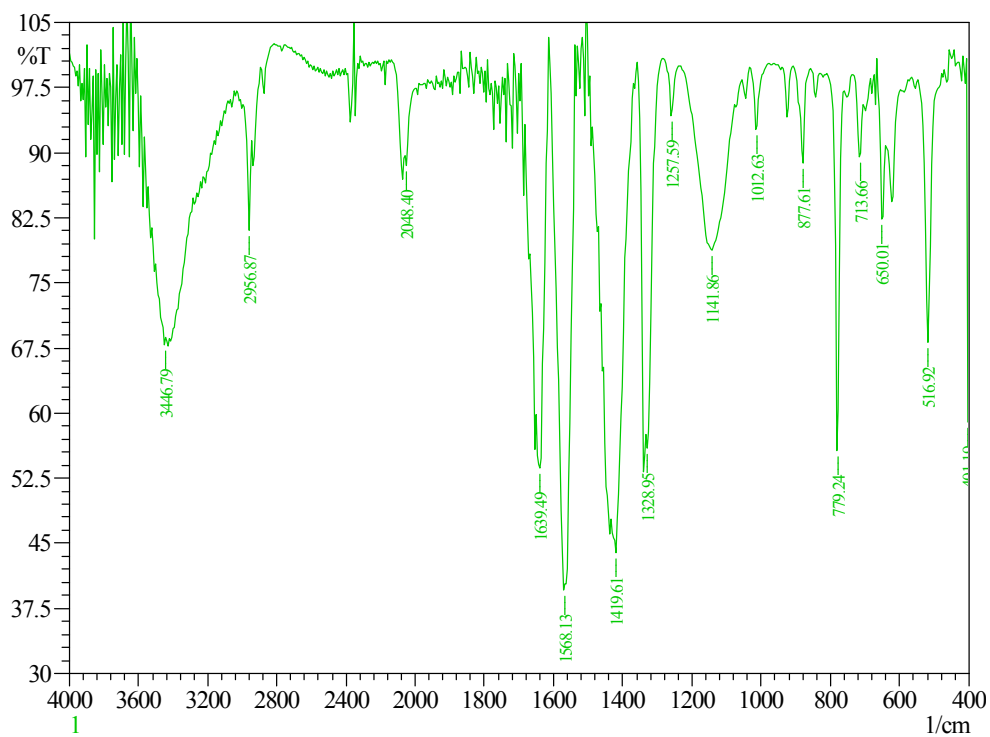


Figure S3. FT-IR spectrum of compound 3.

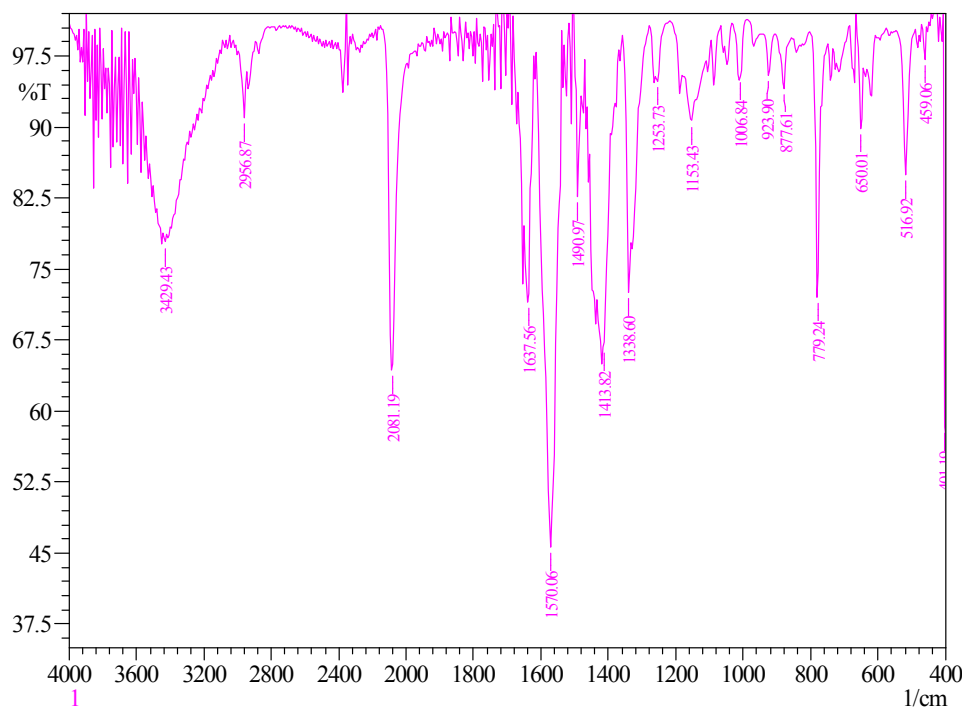


Figure S4. FT-IR spectrum of compound 4.

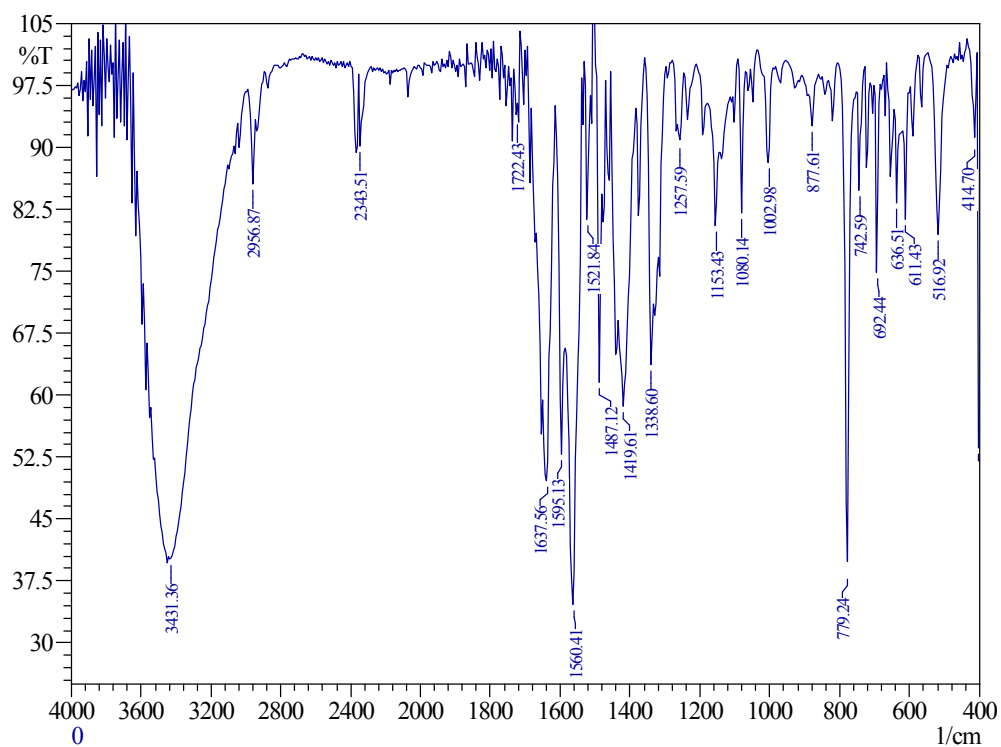


Figure S5. FT-IR spectrum of compound 5.

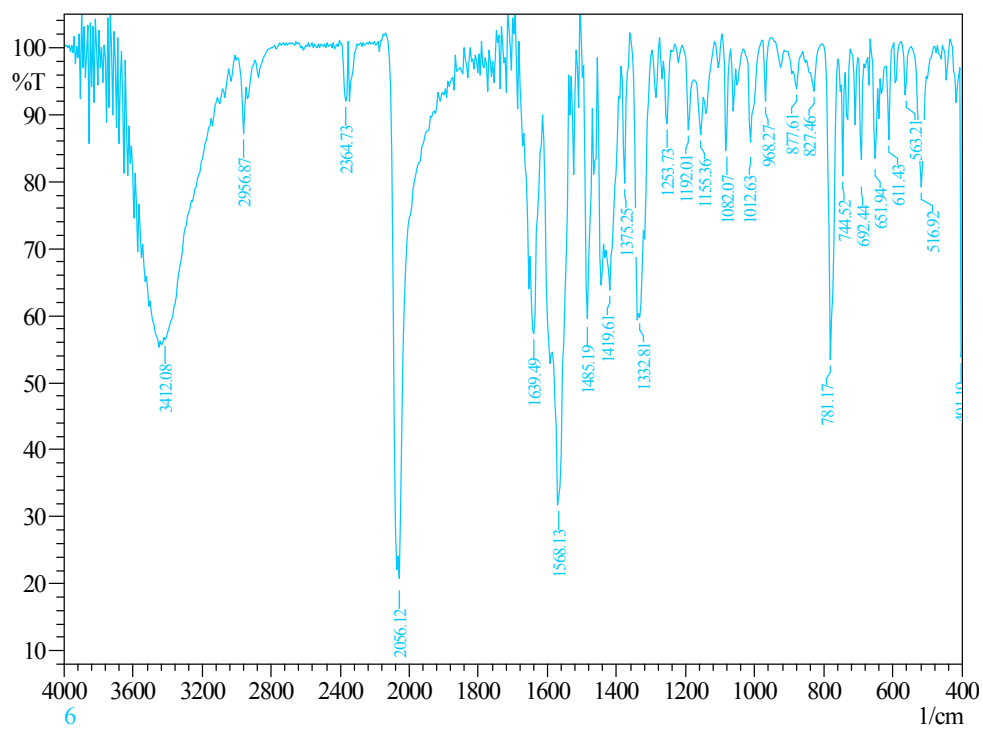


Figure S6. FT-IR spectrum of compound 6.

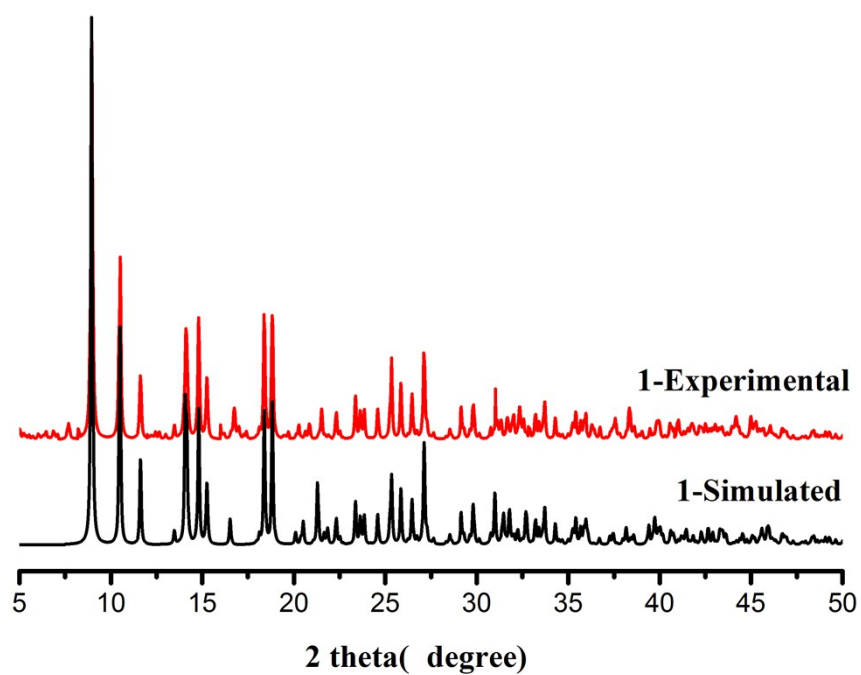


Figure S7 The simulated X-ray powder diffraction patterns (black) and the experimental one (red) of compound 1.

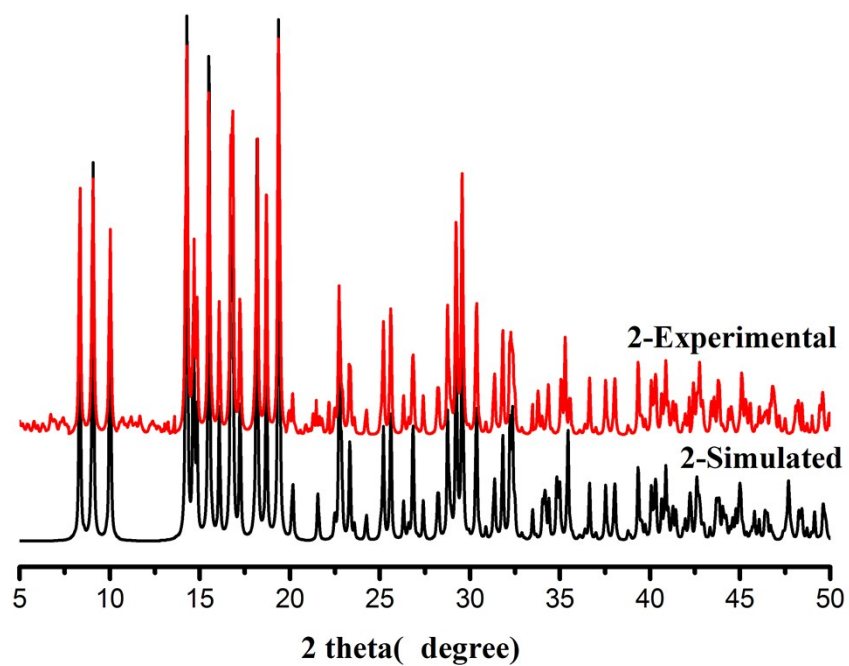


Figure S8 The simulated X-ray powder diffraction patterns (black) and the experimental one (red) of compound 2.

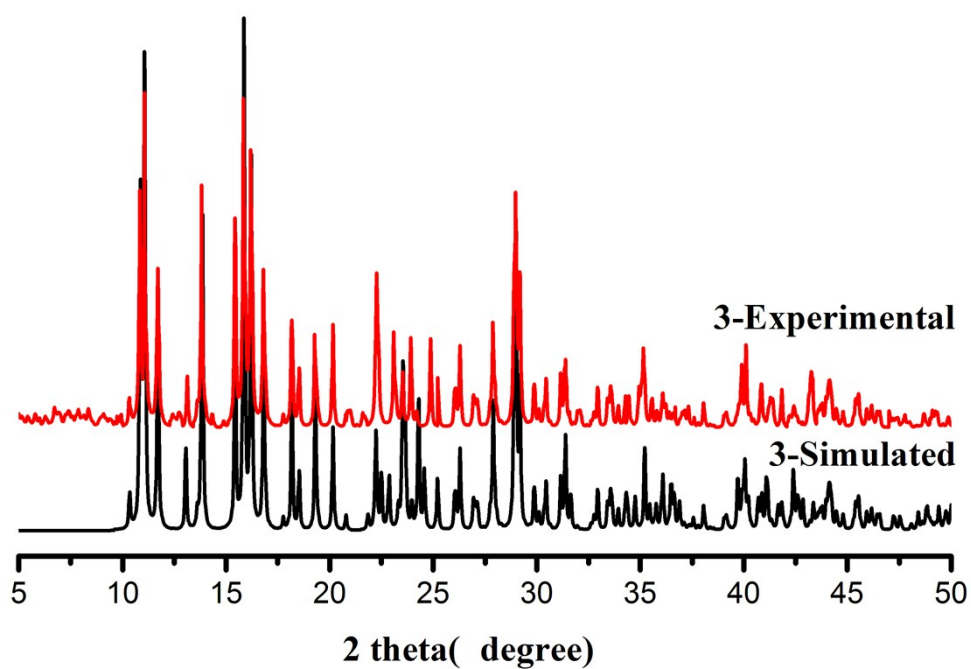


Figure S9 The simulated X-ray powder diffraction patterns (black) and the experimental one (red) of compound 3.

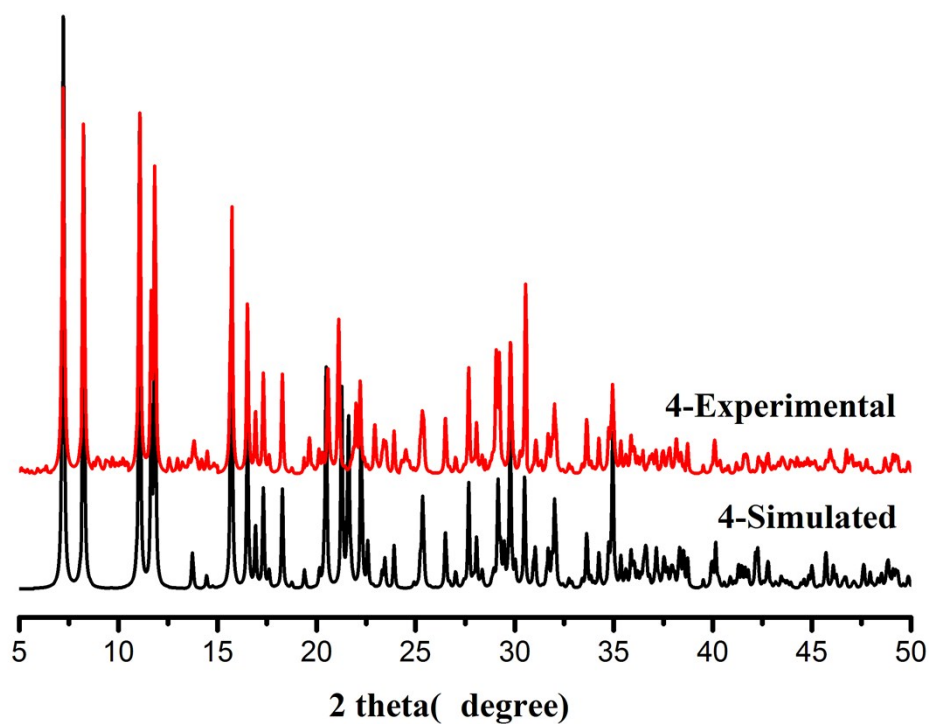


Figure S10 The simulated X-ray powder diffraction patterns (black) and the experimental one (red) of compound 4.

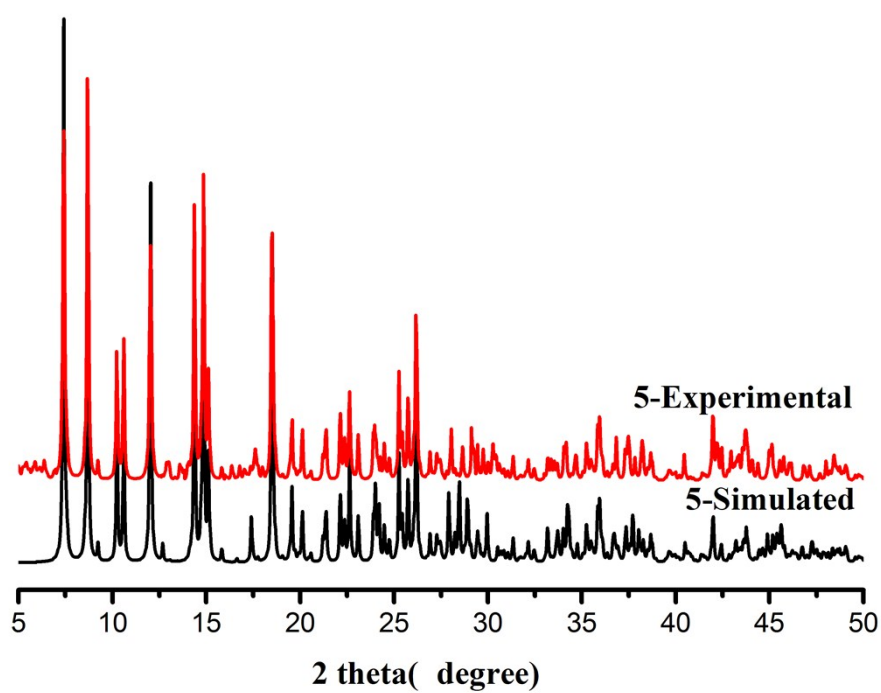


Figure S11 The simulated X-ray powder diffraction patterns (black) and the experimental one (red) of compound 5.

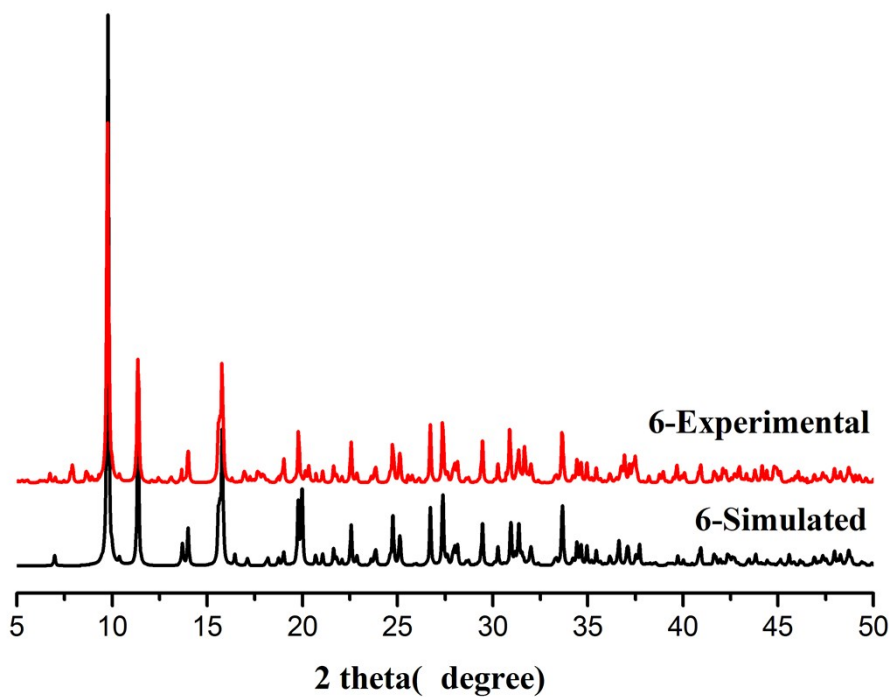


Figure S12 The simulated X-ray powder diffraction patterns (black) and the experimental one (red) of compound 6.