

Supporting Information

Polymer Fabricated Synthesis of Cerium Oxide Nanoparticles and Applications as a Green Catalyst Towards Multicomponent Transformation with Size Dependent Activity Studies

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Table of Contents

Fig. S1. FESEM images of CeONP.

Fig. S2. TEM and HRTEM images of CeONP.

Fig. S3. FESEM images of CeOX nanoparticles.

Fig. S4. TEM of CeONP synthesized without polymer.

Fig. S5. N₂ adsorption-desorption isotherm and BJH pore size distribution curve of CeO₂- 17R4and CeO₂- P123.

S6. ¹H- NMR data of all nitrostyrene (1A, 1B, 1C) and N-aryl pyrrole (4A, 4B, 4C, 4D, 4E).

S7. ¹H- NMR spectra of all nitrostyrene (1A, 1B, 1C) and N-aryl pyrrole (4A, 4B, 4C, 4D, 4E).

Fig. S8. HRTEM of cerium oxide after catalysis.

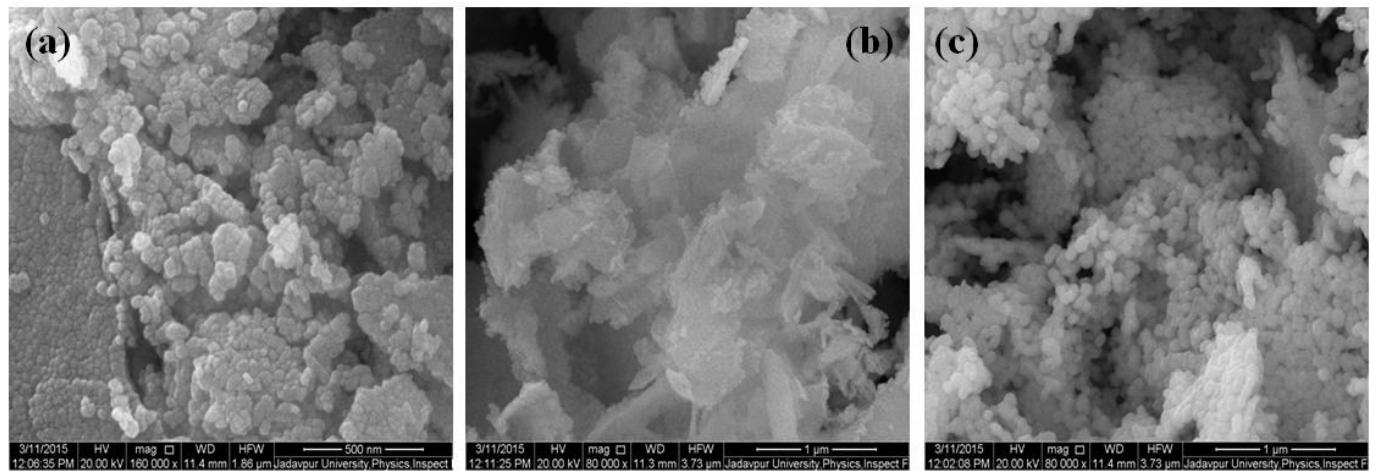


Fig. S1. FESEM images of CeONP, (a) CeO₂ - PVP, (b) CeO₂ - 17R4 and (c) CeO₂ - P123

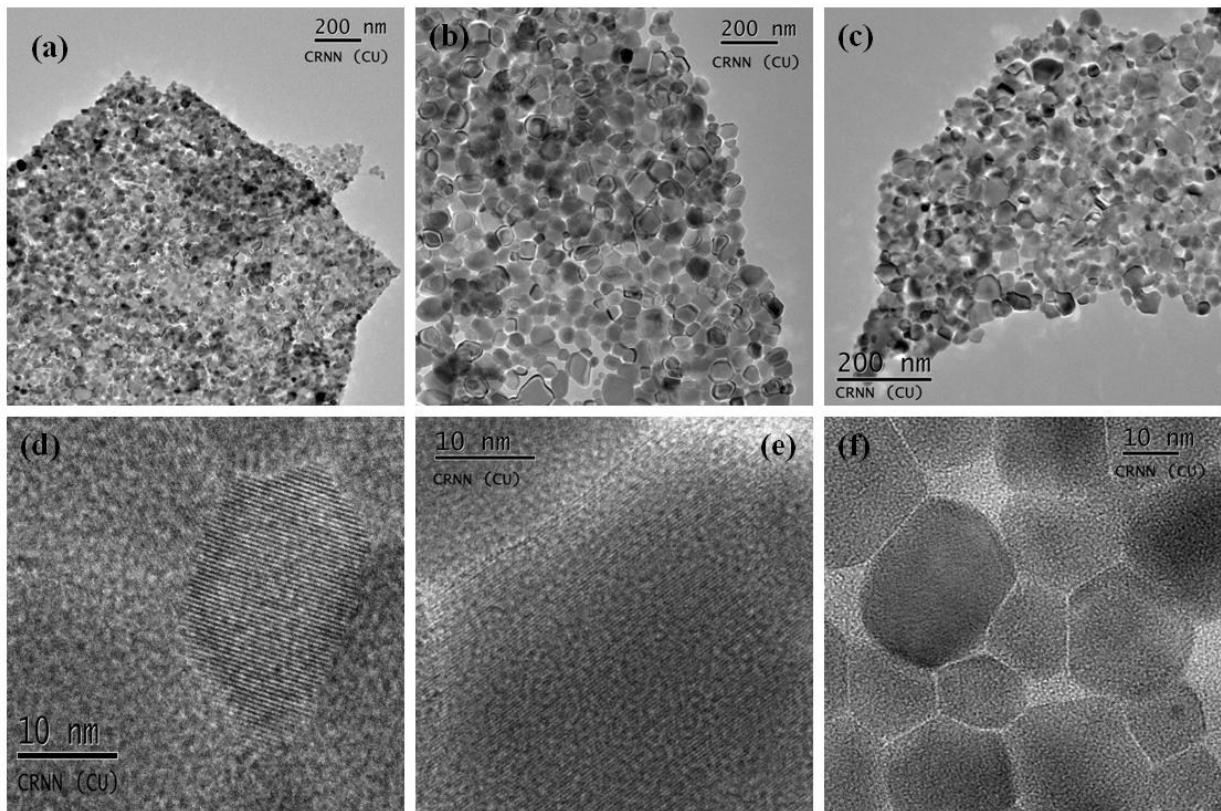


Fig. S2. TEM and HRTEM images of CeONP: (a & d) CeO₂ - 17R4, (b & e) CeO₂ - PVP and (c & f) CeO₂ - P123

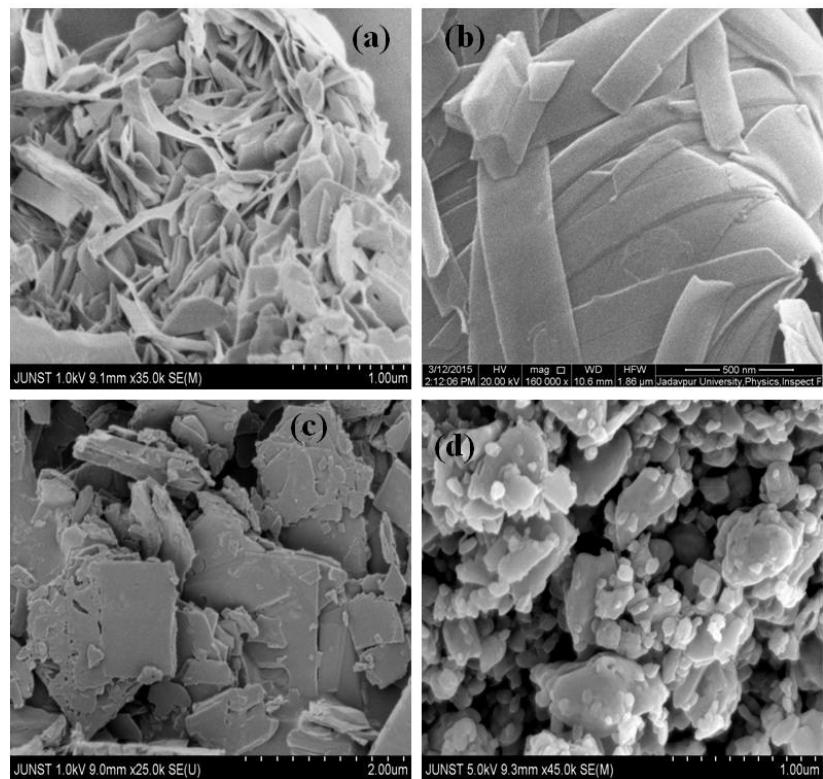


Fig. S3. FESEM images of CeOX nanoparticles, (a) CeOX- PVP (b) CeOX-17R4 and (c) CeOX- P123 (concentration of polymers, $C_{\text{PVP}}=C_{\text{17R4}}=C_{\text{P123}}=1\text{ g/L}$) and (d) CeOX synthesized without polymer

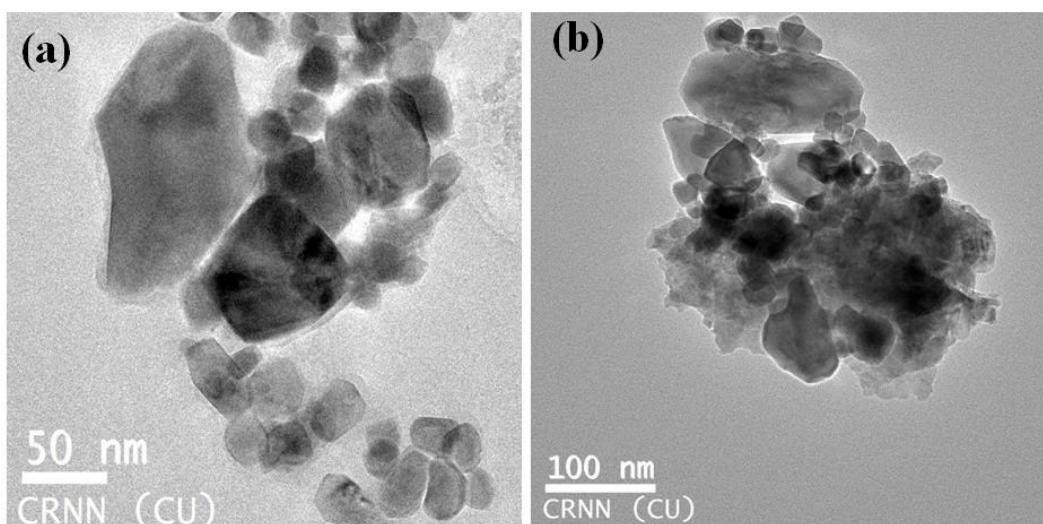


Fig. S4. TEM of CeONP synthesized without polymer

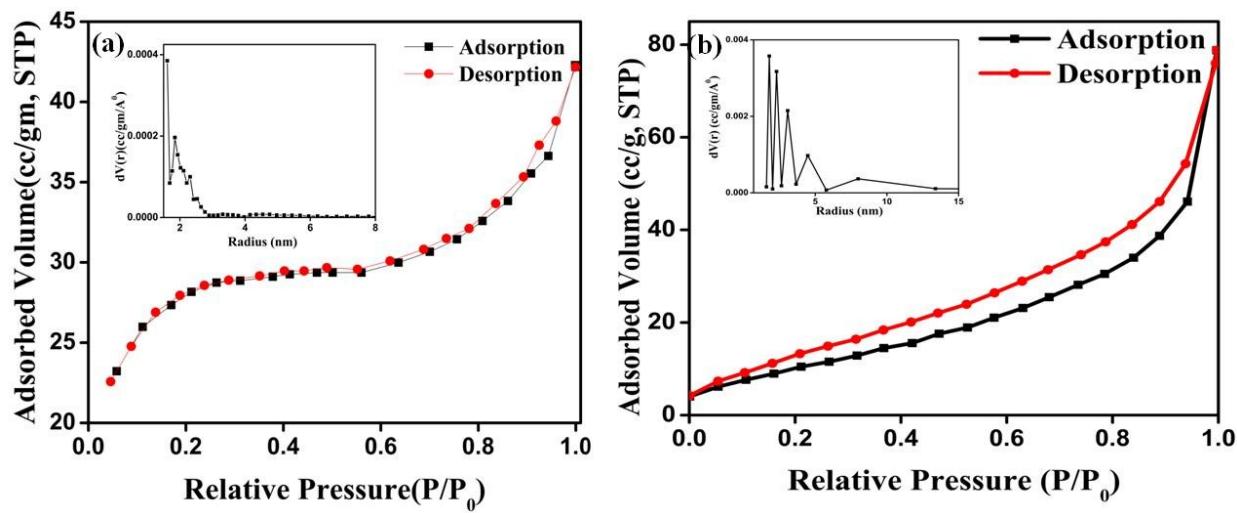


Fig. S5. N₂ adsorption-desorption isotherm and BJH pore size distribution curve (inset) of (a) CeO₂-17R4 and (b) CeO₂-P123

S6:

¹H-NMR data of all nitrostyrene (1A, 1B, 1C) and N-aryl pyrrole (4A, 4B, 4C, 4D, 4E)

(E)-(2-Nitrovinyl)benzene (1A): ¹H NMR (CDCl₃, 300 MHz) δ 7.43–7.49 (m, 2 H), 7.53–7.54 (m, 1 H), 7.57 (s, 2 H), 7.61 (s, 1 H), 8.01 (d, *J* = 13.7 Hz, 1 H) ppm.

(E)-1-Methyl-4-(2-nitrovinyl)benzene (1B): ¹H NMR (CDCl₃, 300 MHz) δ 2.40 (s, 3 H), 7.25 (d, *J* = 8 Hz, 1 H), 7.44 (d, *J* = 8.1 Hz, 2 H), 7.56 (d, *J* = 13.6 Hz, 2 H), 7.98 (d, *J* = 13.6 Hz, 2 H) ppm.

(E)-1-Chloro-4-(2-nitrovinyl)benzene (1C): ¹H NMR (CDCl₃, 300 MHz) δ 7.46 (dd, *J* = 6.6 Hz, 18.8 Hz, 4 H), 7.56 (d, *J* = 13.7 Hz, 1 H), 7.95 (d, *J* = 13.7 Hz, 1 H) ppm.

1-(2-methyl-1,4-diphenyl-1H-pyrrol-3-yl)ethanone (4A): ¹H NMR (CDCl₃, 300 MHz) δ 2.09 (s, 3H, CH₃), 2.42 (s, 3H, CH₃), 6.68 (s, 1H, pyrrole-H), 7.31–7.43 (m, 7H, Ar-H), 7.45–7.50 (m, 3H, Ar-H) ppm.

1-(1-(4-chlorophenyl)-2-methyl-4-p-tolyl-1H-pyrrol-3-yl)ethanone (4B): ¹H NMR (CDCl₃, 300 MHz) δ 2.08 (s, 3H, CH₃), 2.39 (s, 6H, 2CH₃), 6.62 (s, 1H, pyrrole-H), 7.18–7.29 (m, 6H, Ar-H), 7.44–7.48 (m, 2H, Ar-H) ppm.

1-(4-(4-chlorophenyl)-1-(4-methoxyphenyl)-2-methyl-1H-pyrrol-3-yl)ethanone (4C): ¹H NMR (CDCl₃, 300 MHz) δ 2.24 (s, 3H, CH₃), 3.88 (s, 3H, OCH₃), 6.80 (s, 1H, pyrrole-H), 7.00–7.06 (m, 5H, Ar-H), 7.20–7.36 (m, 6H, Ar-H), 7.73 (d, *J* = 8.1 Hz, 2H, Ar-H) ppm.

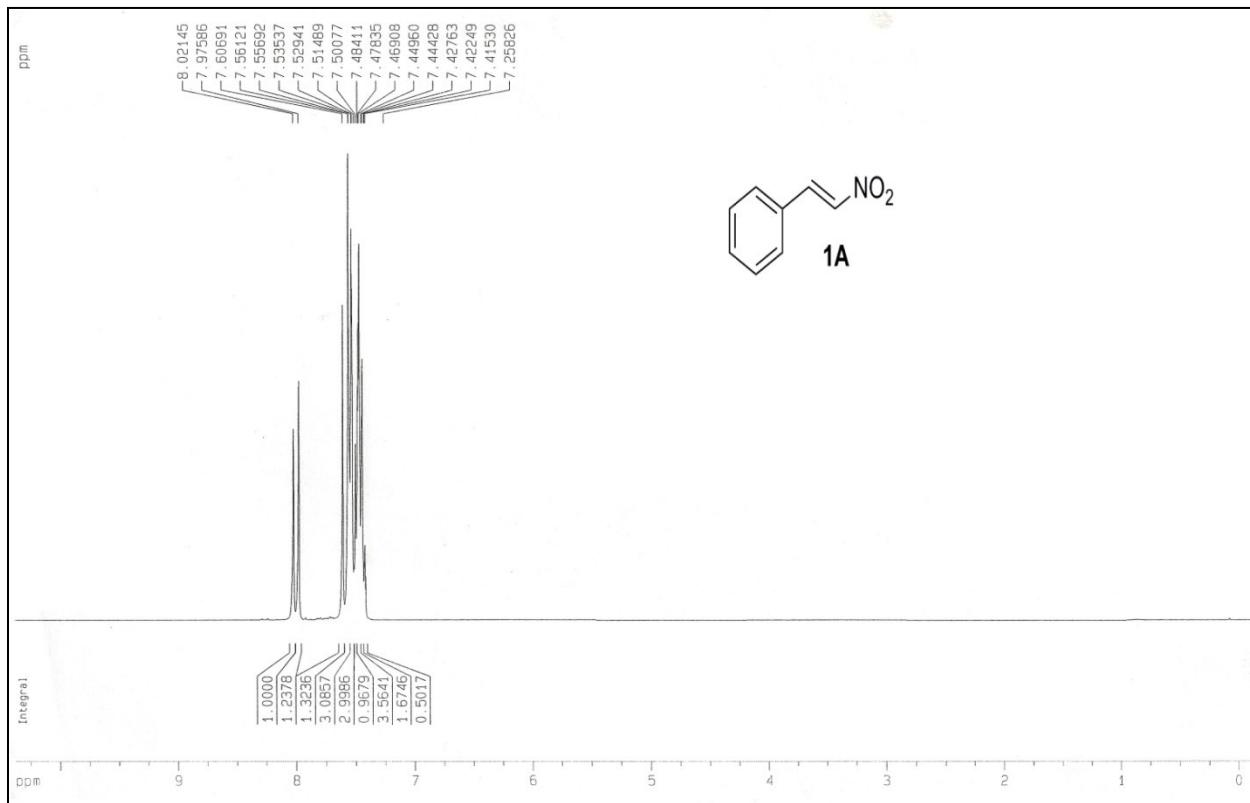
1-(1-(4-methoxyphenyl)-2-methyl-4-phenyl-1H-pyrrol-3-yl)ethanone (4D): ¹H NMR (CDCl₃, 300 MHz) δ 2.07 (s, 3H, CH₃), 2.38 (s, 3H, CH₃), 3.86 (s, 3H, OCH₃), 6.63 (s, 1H, pyrrole-H), 7.0 (d, *J* = 8.9 Hz, 2H, Ar-H), 7.25 (d, *J* = 8.9 Hz, 2H, Ar-H), 7.29–7.35 (m, 1H, Ar-H), 7.37–7.39 (m, 4H, Ar-H) ppm.

1-(2-methyl-1-(naphthalen-2-yl)-4-phenyl-1H-pyrrol-3-yl)ethanone (4E): ¹H NMR (CDCl₃, 300 MHz) δ 2.17 (s, 3H, CH₃), 2.24 (s, 3H, COCH₃), 6.72 (s, 1H, pyrrole-H), 7.32–7.38 (m, 1H, Ar-H), 7.40–7.59 (m, 9H, Ar-H), 7.85 (t, *J* = 7.3 Hz, 2H, Ar-H) ppm .

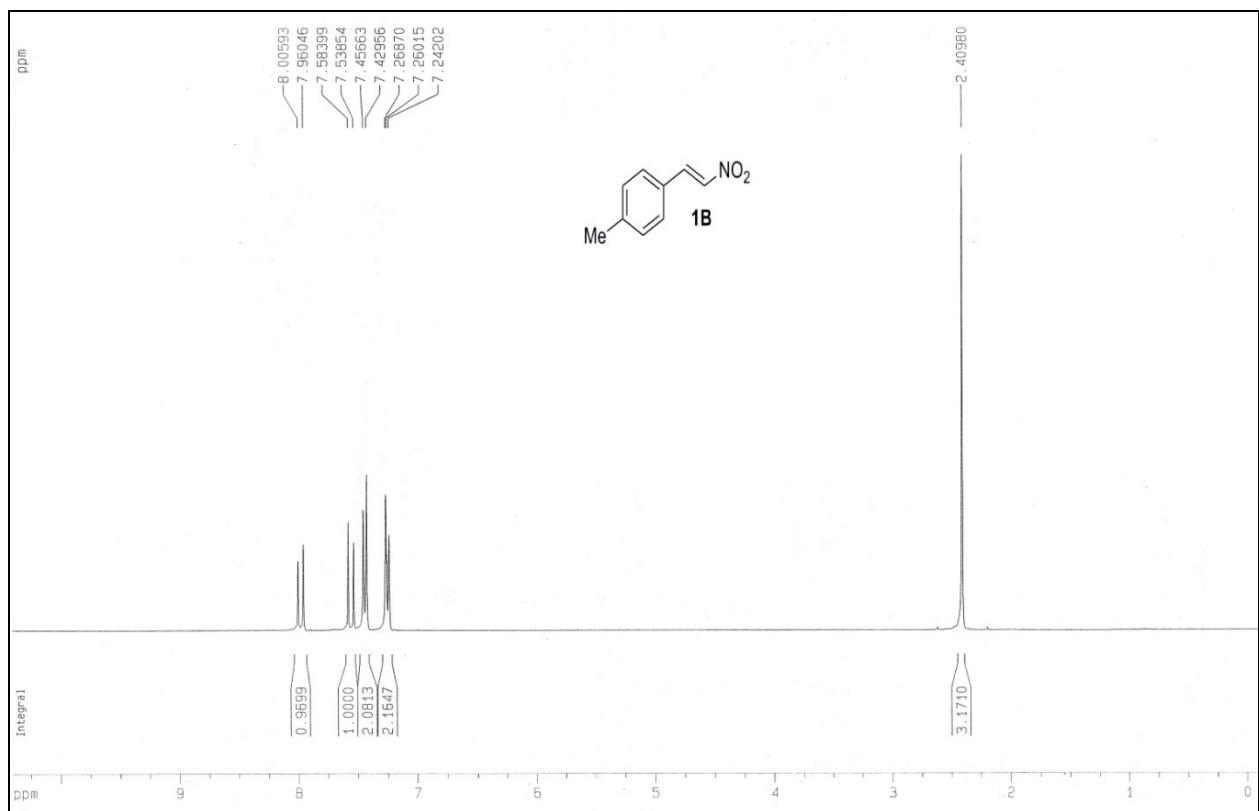
S7.

¹H- NMR spectra of all nitrostyrene (1A, 1B, 1C) and N-aryl pyrrole (4A, 4B, 4C, 4D, 4E)

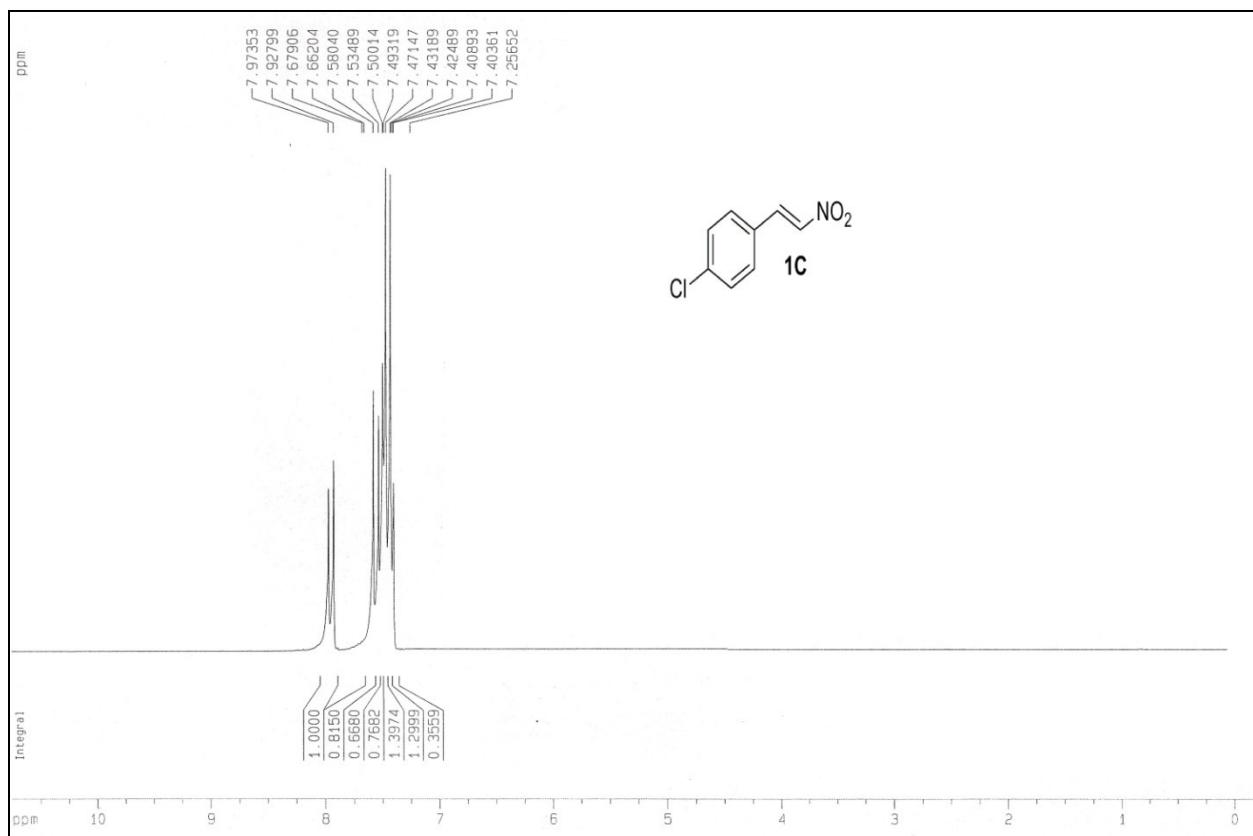
¹H- NMR of (E)-(2-Nitrovinyl)benzene (1A):



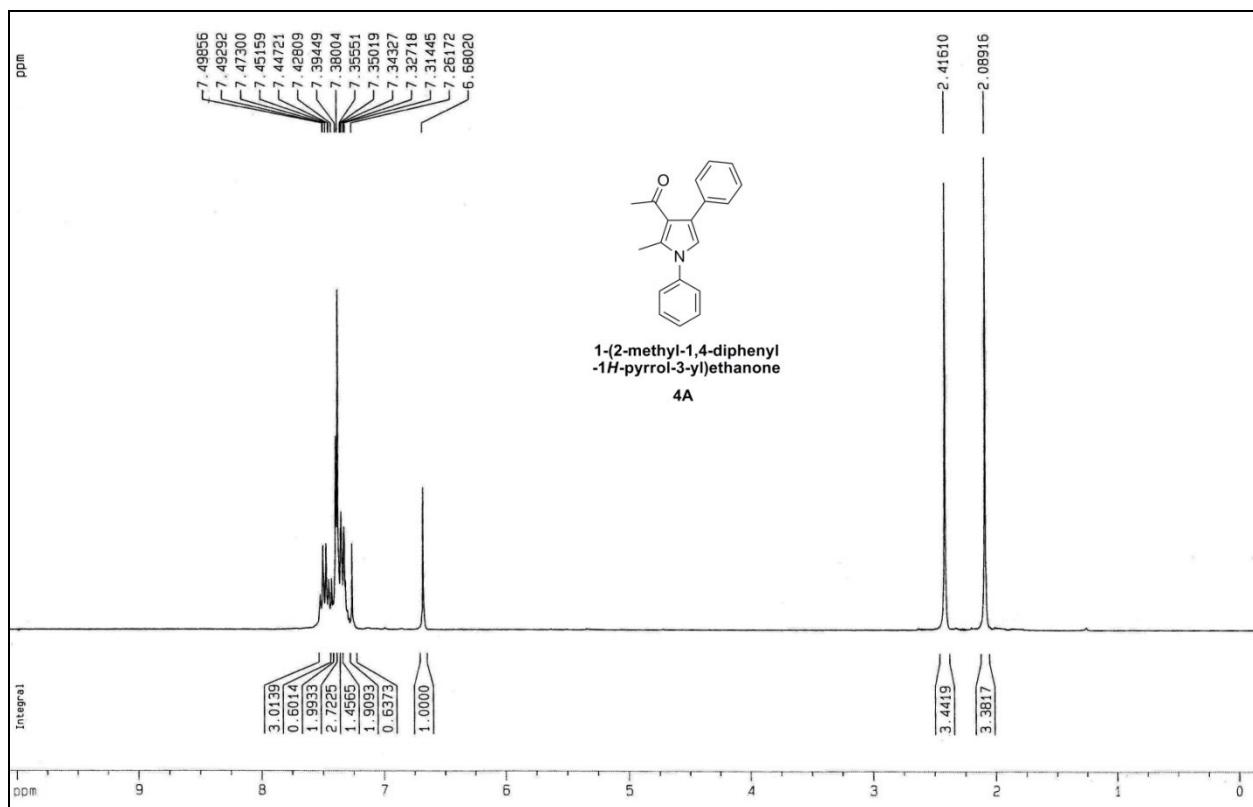
¹H-NMR of (E)-1-Methyl-4-(2-nitrovinyl)benzene (1B):



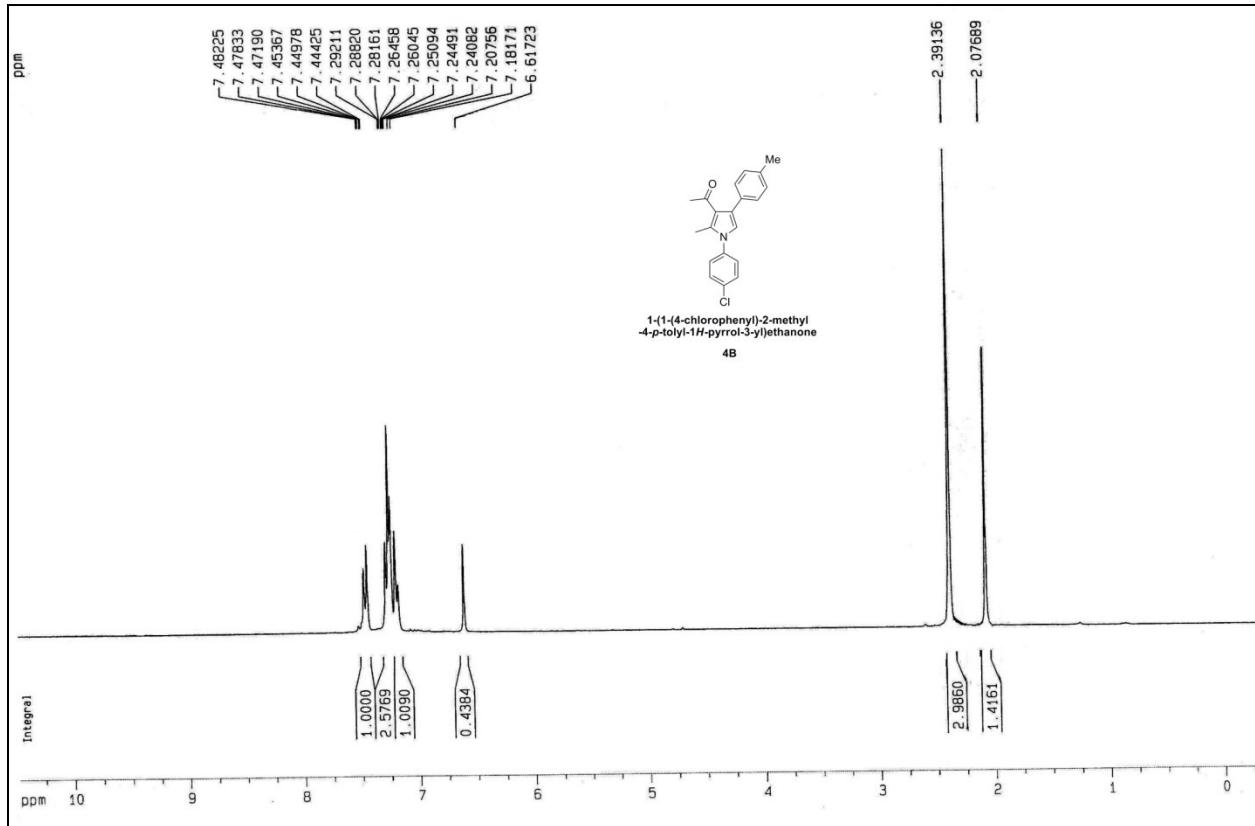
¹H-NMR of (E)-1-Chloro-4-(2-nitrovinyl)benzene (1C):



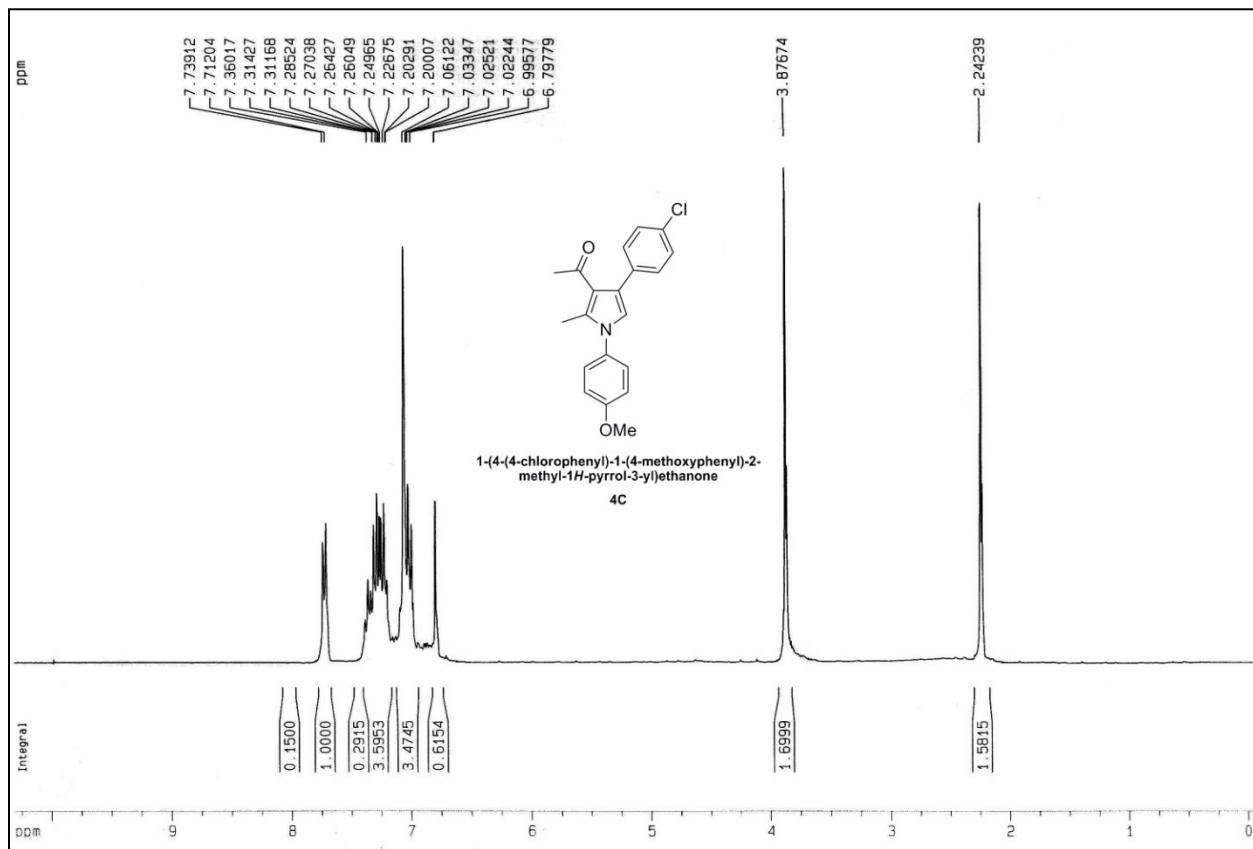
¹H-NMR of 1-(2-methyl-1,4-diphenyl-1*H*-pyrrol-3-yl)ethanone (**4A**):



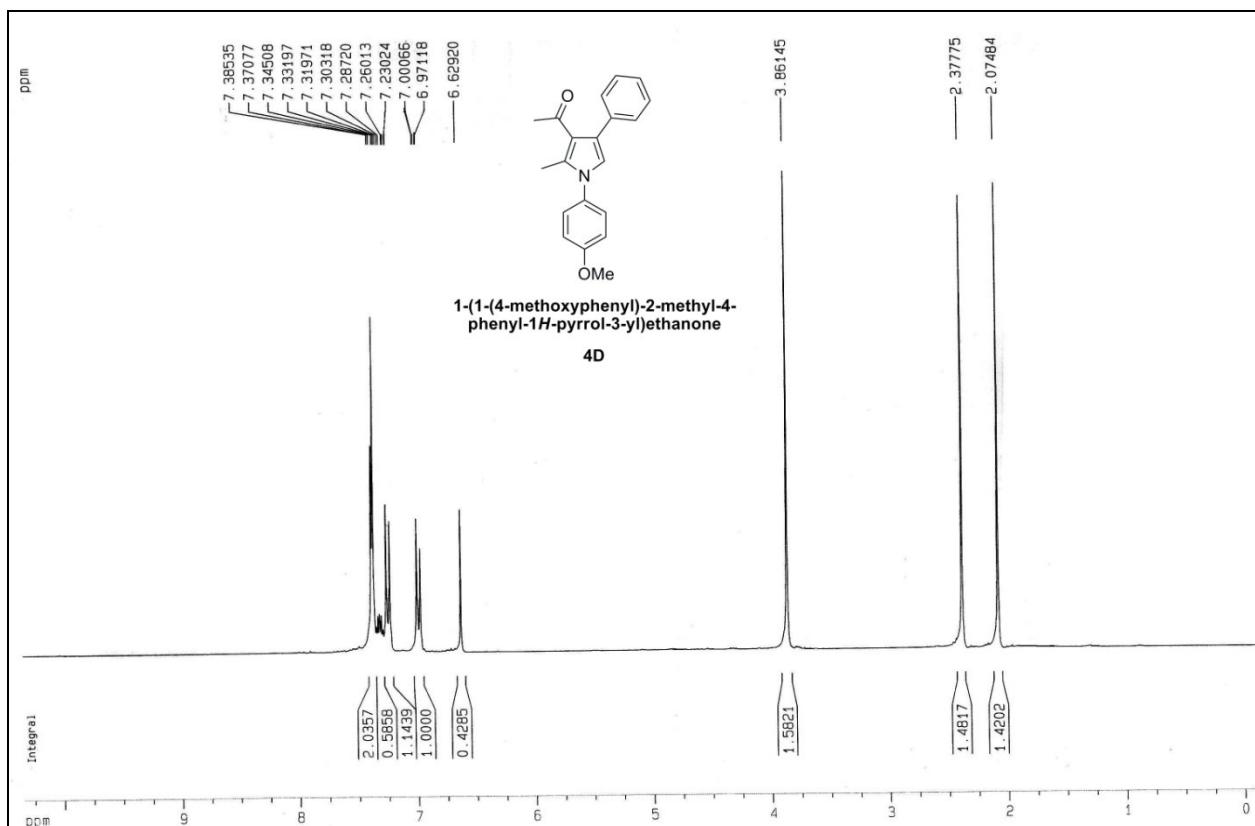
¹H- NMR of 1-(1-(4-chlorophenyl)-2-methyl-4-p-tolyl-1H-pyrrol-3-yl)ethanone (4B):



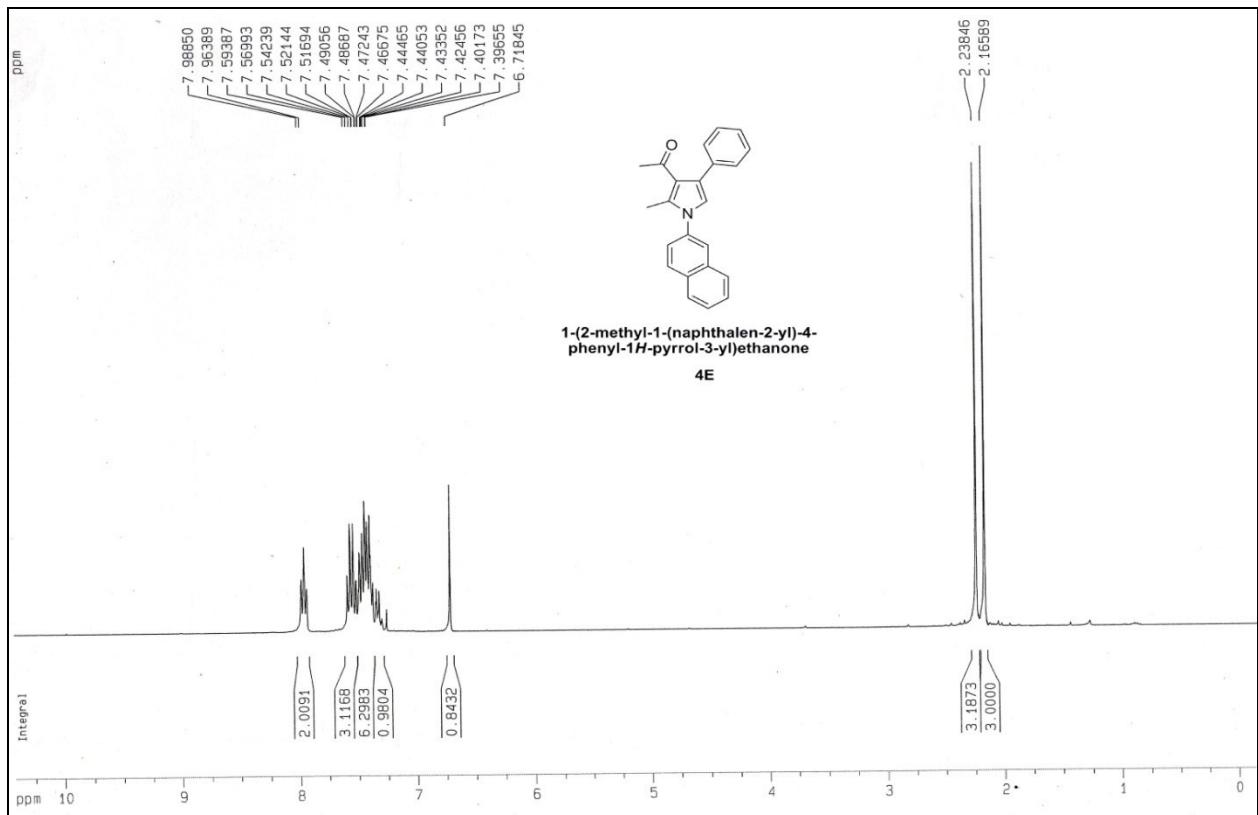
¹H-NMR of 1-(4-(4-chlorophenyl)-1-(4-methoxyphenyl)-2-methyl-1H-pyrrol-3-yl)ethanone (4C):



¹H-NMR of 1-(1-(4-methoxyphenyl)-2-methyl-4-phenyl-1H-pyrrol-3-yl)ethanone (4D):



¹H-NMR of 1-(2-methyl-1-(naphthalen-2-yl)-4-phenyl-1H-pyrrol-3-yl)ethanone (4E):



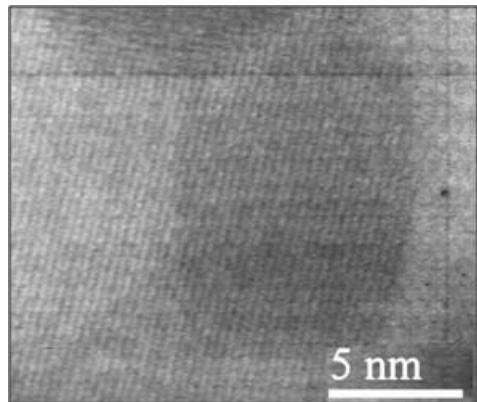


Fig. S8. HRTEM of CeO₂-PVP after catalysis