

**Mechanochemically synthesized crystalline luminescent 2D coordination  
polymers of La<sup>3+</sup> and Ce<sup>3+</sup>, doped with Sm<sup>3+</sup>, Eu<sup>3+</sup>, Tb<sup>3+</sup>, and Dy<sup>3+</sup>:  
synthesis, crystal structures and luminescence**

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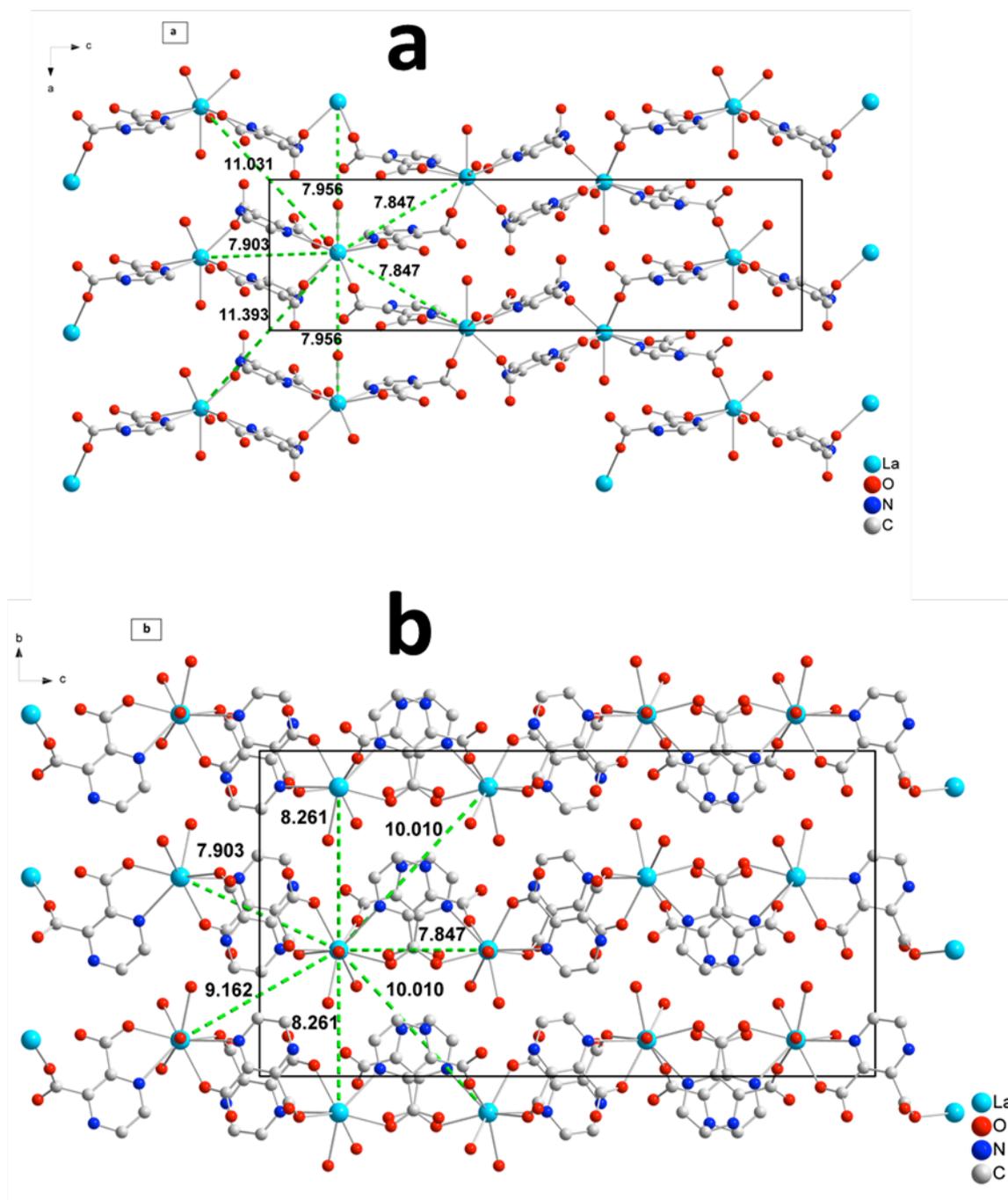
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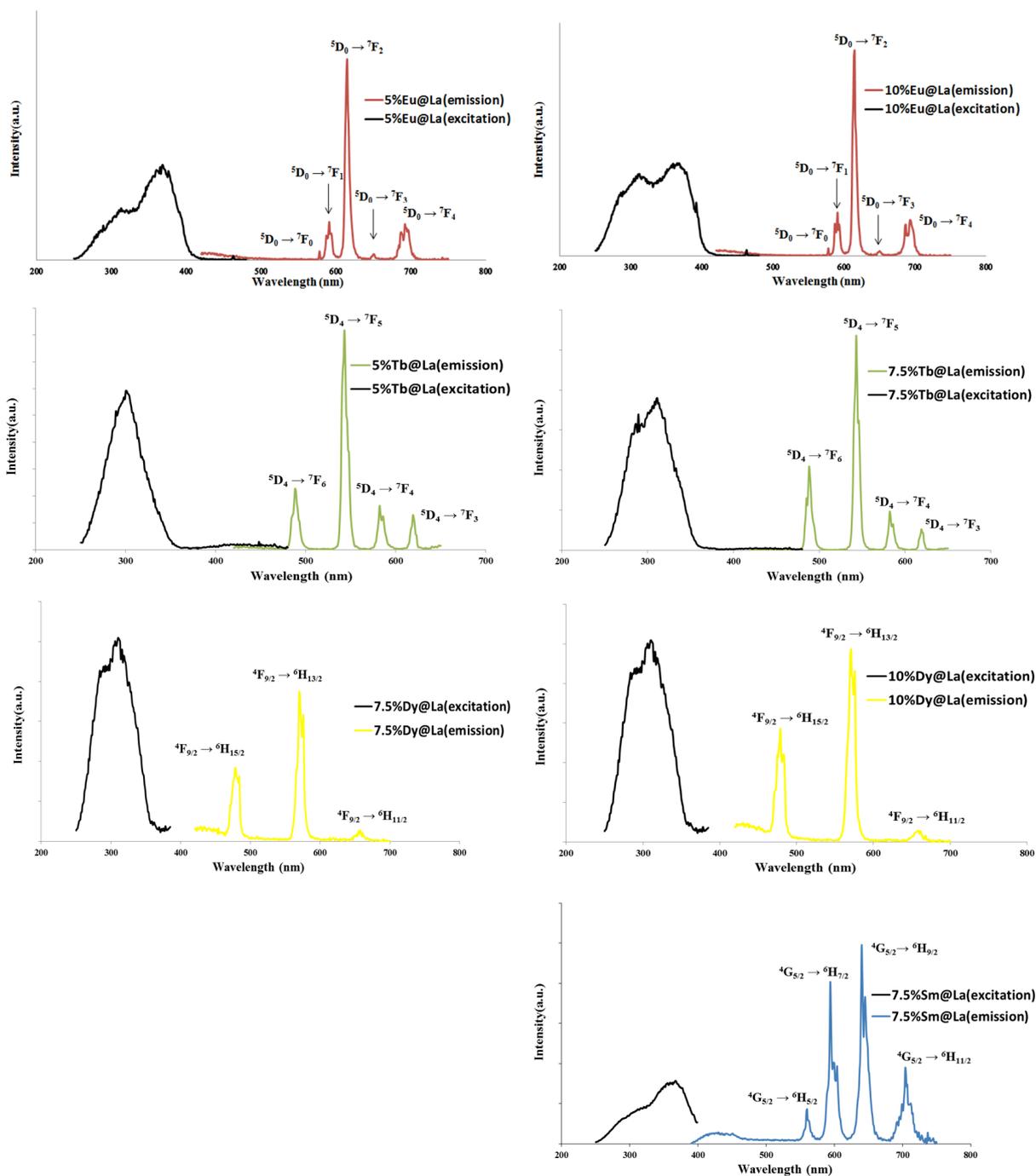
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**Supporting Information**

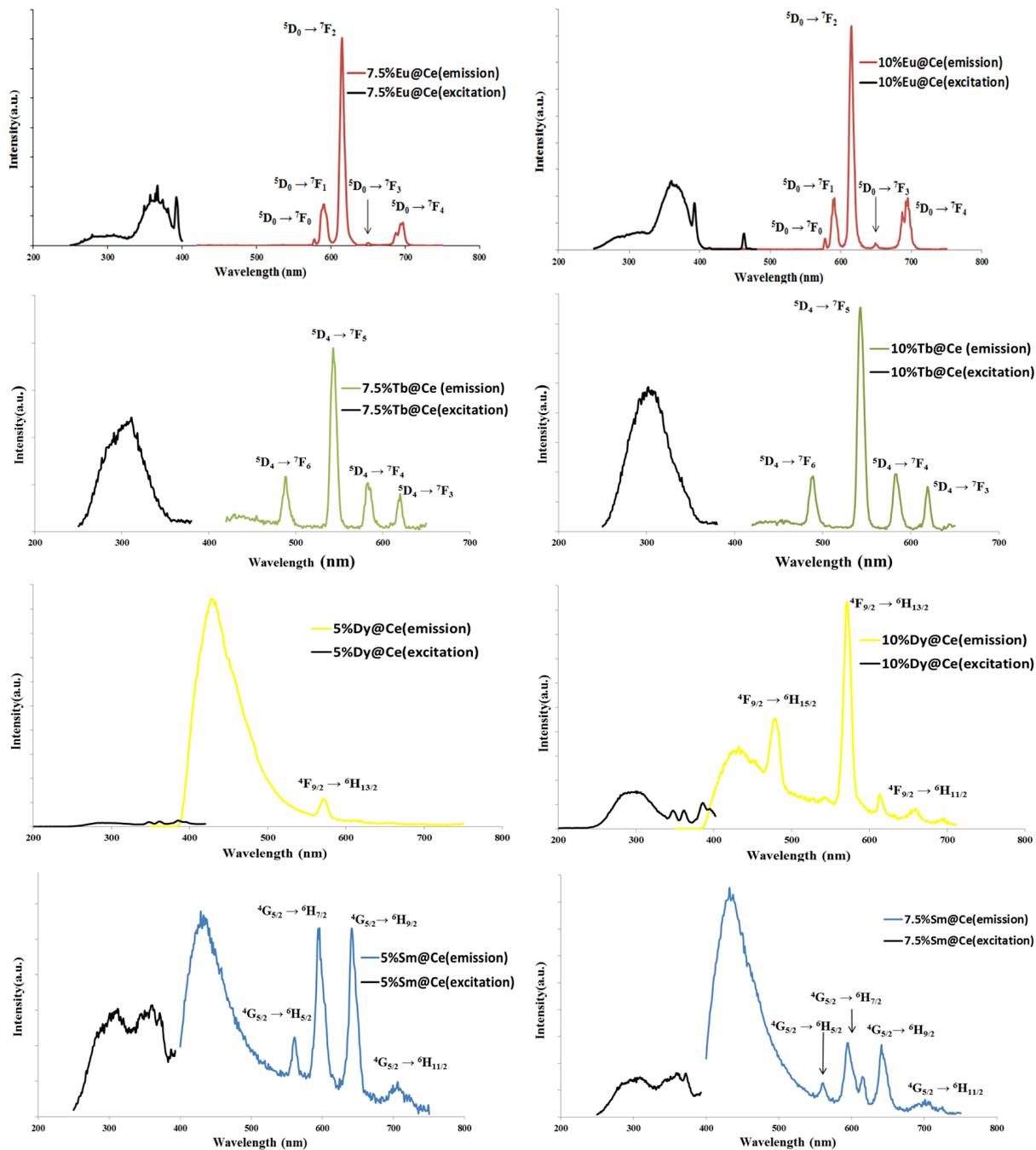


**Fig.S1.** Intermetallic distances between La atoms that belong to a 2D layer (a) and to different 2D layers (b) in  $[\text{La}(\text{pzdc})(\text{Hpzdc})(\text{H}_2\text{O})_3]_n$

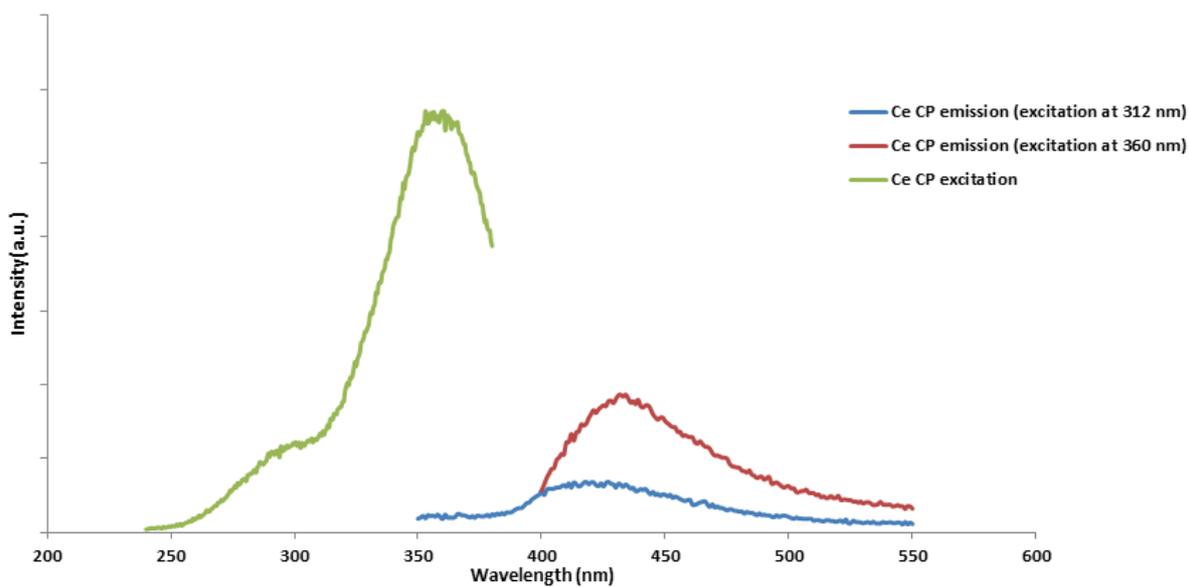
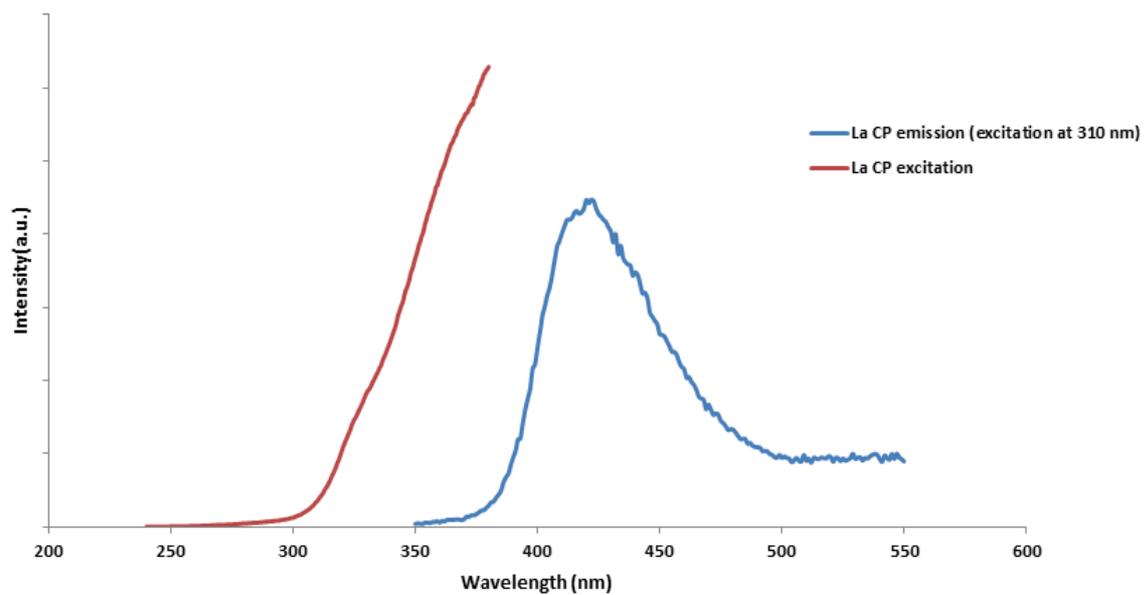
(for compound **2** the Ce-Ce values are slightly shorter ( $\sim 0.02\text{\AA}$ ))



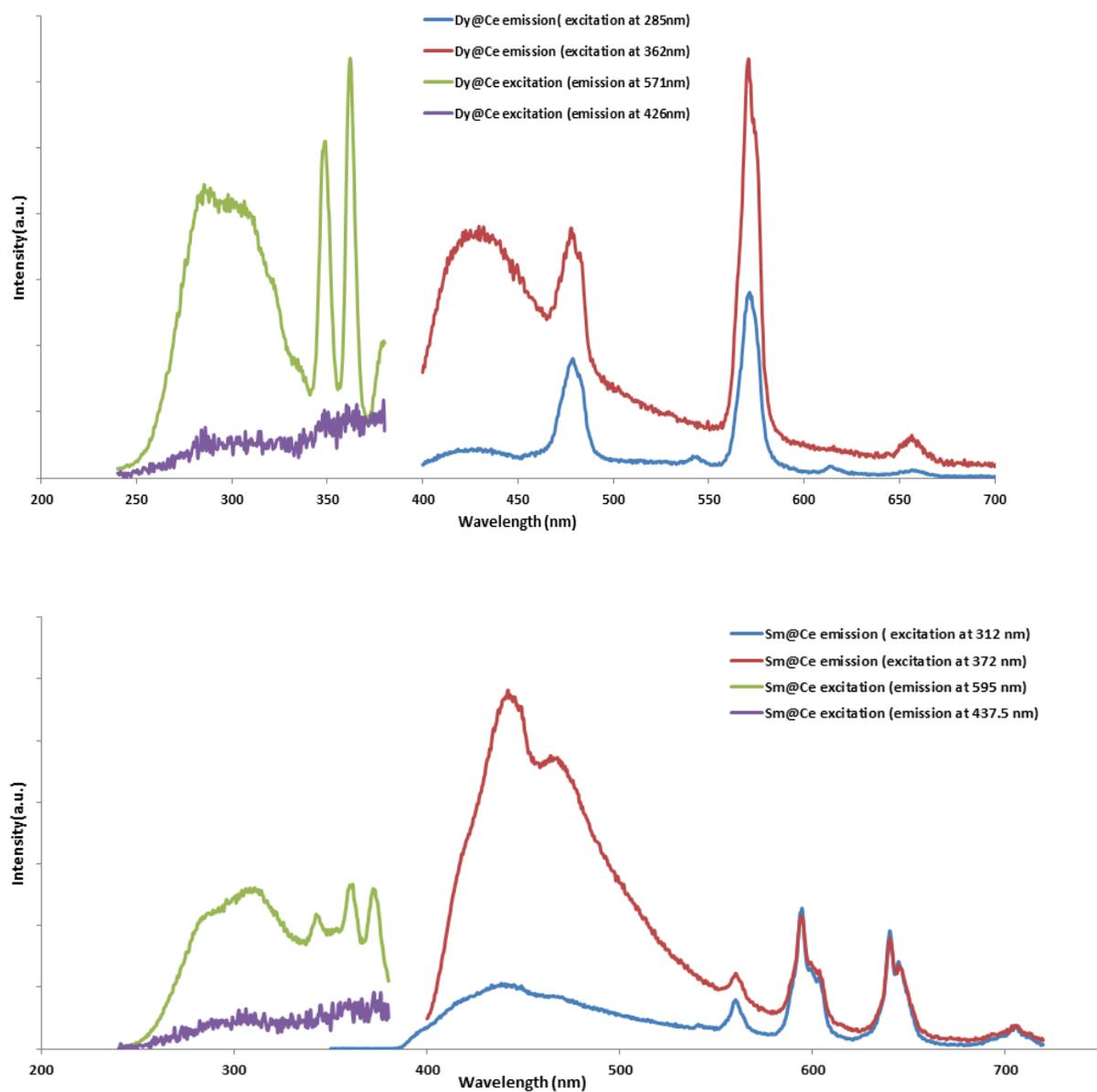
**Fig. S2.** Excitation and emission spectra for  $[\text{La}_{1-x}\text{Ln}'_x(\text{pzdc})(\text{Hpzdc})(\text{H}_2\text{O})_3]_n$ , where  $\text{Ln}' = \text{Sm}^{3+}$ ,  $\text{Eu}^{3+}$ ,  $\text{Tb}^{3+}$ , or  $\text{Dy}^{3+}$  and  $x = 0.050, 0.075$  or  $0.100$ .



**Fig. S3.** Excitation and emission spectra for  $[\text{Ce}_{1-x}\text{Ln}'_x(\text{pzdc})(\text{Hpzdc})(\text{H}_2\text{O})_3]_n$ , where  $\text{Ln}' = \text{Sm}^{3+}$ ,  $\text{Eu}^{3+}$ ,  $\text{Tb}^{3+}$ , or  $\text{Dy}^{3+}$  and  $x = 0.050, 0.075$  or  $0.100$ .



**Fig. S4.** Excitation and emission spectra for [La(pzdc)(Hpzdc)(H<sub>2</sub>O)<sub>3</sub>]<sub>n</sub>, (compound **1**, top graph) and for [Ce(pzdc)(Hpzdc)(H<sub>2</sub>O)<sub>3</sub>]<sub>n</sub>, (compound **2**, bottom graph).



**Fig. S5.** Excitation and emission spectra for  $[\text{Ce}_{0.925}\text{Dy}_{0.075}(\text{pzdc})(\text{Hpzdc})(\text{H}_2\text{O})_3]_n$ , (7.5%Dy@Ce, top graph) and for  $[\text{Ce}_{0.900}\text{Sm}_{0.100}(\text{pzdc})(\text{Hpzdc})(\text{H}_2\text{O})_3]_n$ , (10%Sm@Ce, bottom graph).

**Table S1.** Selected bond lengths [Å] and angles [°] for [Ln(pzdc)(Hpzdc)(H<sub>2</sub>O)<sub>3</sub>]<sub>n</sub> (Ln = La (**1**) and Ce (**2**)).

<b>Bond (Å)</b>	<b>(1)</b>		<b>(2)</b>
La–O3 <sup>i</sup>	2.451(2)	Ce–O3 <sup>i</sup>	2.4283 (16)
La–O12	2.462(2)	Ce–O12	2.4382(15)
La–O2	2.477(2)	Ce–O2	2.4493(15)
La–O14 <sup>ii</sup>	2.462(2)	Ce–O14 <sup>ii</sup>	2.4600(15)
La–N11	2.872(2)	Ce–N11	2.8502(18)
La–O5	2.518(2)	Ce–O5	2.4970(17)
La–O6	2.524(2)	Ce–O6	2.5050(16)
La–O7	2.615(2)	Ce–O7	2.5954(17)
La–N1	2.820(2)	Ce–N1	2.7982(18)
<hr/>			
<b>Angle (°)</b>			
O2–La–O6	83.40(7)	O2–Ce–O6	83.86(6)
O14 <sup>ii</sup> –La–O6	87.26(7)	O14 <sup>ii</sup> –Ce–O6	87.22(6)
O3 <sup>ii</sup> –La–O6	139.39(7)	O3 <sup>ii</sup> –Ce–O6	139.28(6)
O5–La–O6	74.43(8)	O5–Ce–O6	74.45(7)
O3 <sup>i</sup> –La–O7	71.23(7)	O3 <sup>i</sup> –Ce–O7	71.25(5)
O12–La–O7	144.91(7)	O12–Ce–O7	144.89(6)
O5–La–O7	134.05(8)	O5–Ce–O7	133.92(6)
O2–La–O7	71.69(7)	O2–Ce–O7	71.56(6)
O3 <sup>i</sup> –La–O2	86.90(7)	O3 <sup>i</sup> –Ce–O2	87.06(6)
O3 <sup>i</sup> –La–O14 <sup>ii</sup>	76.13(7)	O3 <sup>i</sup> –Ce–O14 <sup>ii</sup>	76.43(6)
O2–La–O14 <sup>ii</sup>	142.57(6)	O2–Ce–O14 <sup>ii</sup>	142.59(6)
O12–La–O5	80.98(7)	O12–Ce–O5	81.13(6)
O14 <sup>ii</sup> –La–O5	133.26(7)	O14 <sup>ii</sup> –Ce–O5	133.22(6)
O6–La–O7	68.36(7)	O6–Ce–O7	68.21(6)
O12–La–O6	134.28(7)	O12–Ce–O6	134.38(5)
O12–La–N1	68.89(7)	O12–Ce–N1	68.73(5)
O2–La–N1	59.84(7)	O2–Ce–N1	60.37(5)
O14 <sup>ii</sup> –La–N1	135.25(6)	O14 <sup>ii</sup> –Ce–N1	135.62(5)
O14 <sup>ii</sup> –La–O7	71.32(7)	O14 <sup>ii</sup> –Ce–O7	71.38(5)
O3 <sup>i</sup> –La–O12	80.44(7)	O3 <sup>i</sup> –Ce–O12	80.54(5)
O12–La–O2	127.92(7)	O12–Ce–O2	128.33(5)
O12–La–O14 <sup>ii</sup>	82.29(7)	O12–Ce–O14 <sup>ii</sup>	82.20(5)
O3 <sup>i</sup> –La–O5	141.96(8)	O3 <sup>i</sup> –Ce–O5	141.86(6)

Symmetry code: (i) x-1/2,y,-z+1/2; (ii) -x,-y+2,-z; (iii) x+1/2,y,-z+1/2

**Table S2.** Hydrogen bonds geometry (Å, °) for [Ln(pzdc)(Hpzdc)(H<sub>2</sub>O)<sub>3</sub>]<sub>n</sub> (Ln = La (1) and Ce (2)).

(1)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O5—H51···O1 <sup><i>i</i></sup>	0.82 (1)	1.93 (1)	2.733 (3)	167 (4)
O5—H52···O13 <sup><i>ii</i></sup>	0.82 (1)	1.95 (1)	2.766 (3)	173 (4)
O6—H61···O11 <sup><i>iii</i></sup>	0.82 (1)	1.91 (1)	2.703 (3)	163 (4)
O6—H62···O4 <sup><i>iv</i></sup>	0.82 (1)	1.93 (1)	2.731 (3)	167 (4)
O7—H71···O1 <sup><i>v</i></sup>	0.82 (1)	2.05 (1)	2.852 (3)	167 (4)
O7—H72···N4 <sup><i>iv</i></sup>	0.82 (1)	2.04 (1)	2.860 (3)	177 (4)
N14—H14···O4 <sup><i>vi</i></sup>	0.87 (1)	1.75 (1)	2.612 (3)	171 (3)

Symmetry codes: (*i*)  $x+1/2, y, -z+1/2$ ; (*ii*)  $-x+1, -y+2, -z$ ; (*iii*)  $-x+1/2, y-1/2, z$ ; (*iv*)  $-x, y-1/2, -z+1/2$ ; (*v*)  $x-1/2, y, -z+1/2$ ; (*vi*)  $-x+1/2, -y+2, z-1/2$ .

(2)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O5—H51···O1 <sup><i>i</i></sup>	0.82 (2)	1.93 (2)	2.734 (2)	167 (3)
O5—H52···O13 <sup><i>ii</i></sup>	0.82 (2)	1.96 (2)	2.777 (3)	173 (3)
O6—H61···O11 <sup><i>iii</i></sup>	0.82 (2)	1.92 (2)	2.714 (2)	162 (3)
O6—H62···O4 <sup><i>iv</i></sup>	0.82 (2)	1.96 (2)	2.737 (2)	158 (3)
O7—H71···O1 <sup><i>v</i></sup>	0.82 (2)	2.06 (2)	2.862 (2)	168 (3)
O7—H72···N4 <sup><i>iv</i></sup>	0.82 (2)	2.05 (2)	2.865 (3)	179 (3)
N14—H14···O4 <sup><i>vi</i></sup>	0.87 (2)	1.74 (2)	2.611 (2)	176 (2)

Symmetry codes: (*i*)  $x+1/2, y, -z+1/2$ ; (*ii*)  $-x+1, -y+2, -z$ ; (*iii*)  $-x+1/2, y-1/2, z$ ; (*iv*)  $-x, y-1/2, -z+1/2$ ; (*v*)  $x-1/2, y, -z+1/2$ ; (*vi*)  $-x+1/2, -y+2, z-1/2$ .

**Table S3.** Decay times of the Ln@La and Ln@Ce (Ln = Sm<sup>3+</sup>, Eu<sup>3+</sup>, Tb<sup>3+</sup>, or Dy<sup>3+</sup>) (all values recorded for solid state samples)

Sample	Decay time [ $\mu$ s]	Sample	Decay time [ $\mu$ s]
La		Ce	
Tb@La5%	811	Tb@Ce5%	739
Tb@La7.5%	822	Tb@Ce7.5%	646
Tb@La10%	817	Tb@Ce10%	727
Eu@La5%	406	Eu@Ce5%	848
Eu@La7.5%	719	Eu@Ce7.5%	152
Eu@La10%	393	Eu@Ce10%	197
Dy@La5%	14.2	Dy@Ce5%	12.4
Dy@La7.5%	14.1	Dy@Ce7.5%	12.7
Dy@La10%	13.4	Dy@Ce10%	12.6
Sm@La5%	30.0	Sm@Ce5%	12.4
Sm@La7.5%	27.3	Sm@Ce7.5%	12.4
Sm@La10%	34.0	Sm@Ce10%	13.6