

## Supporting Information

### 1. Characterization of the CAU-1 type compounds

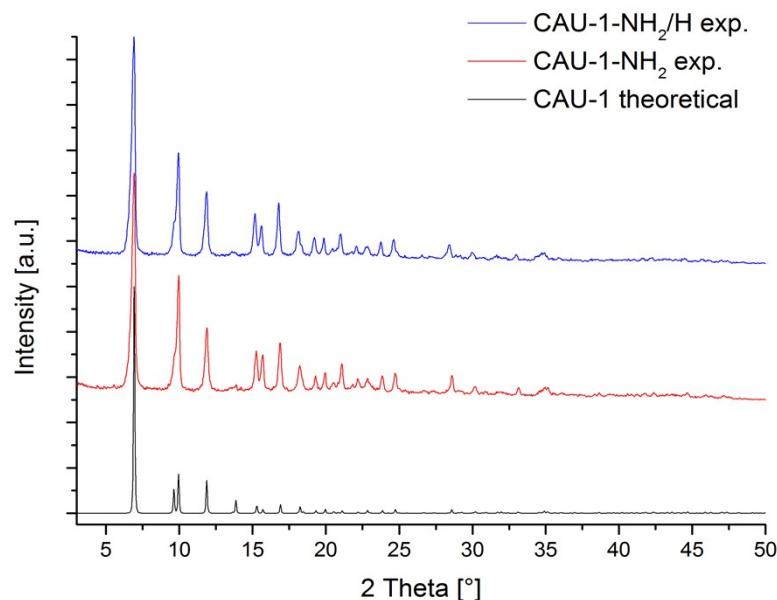


Fig. S1: PXRD patterns of the two CAU-1 samples one containing exclusively aminoterephthalate ions (CAU-1-NH<sub>2</sub>) and the other containing terephthalate and aminoterephthalate ions in a molar ratio 3 to 1 (CAU-1-NH<sub>2</sub>/H).

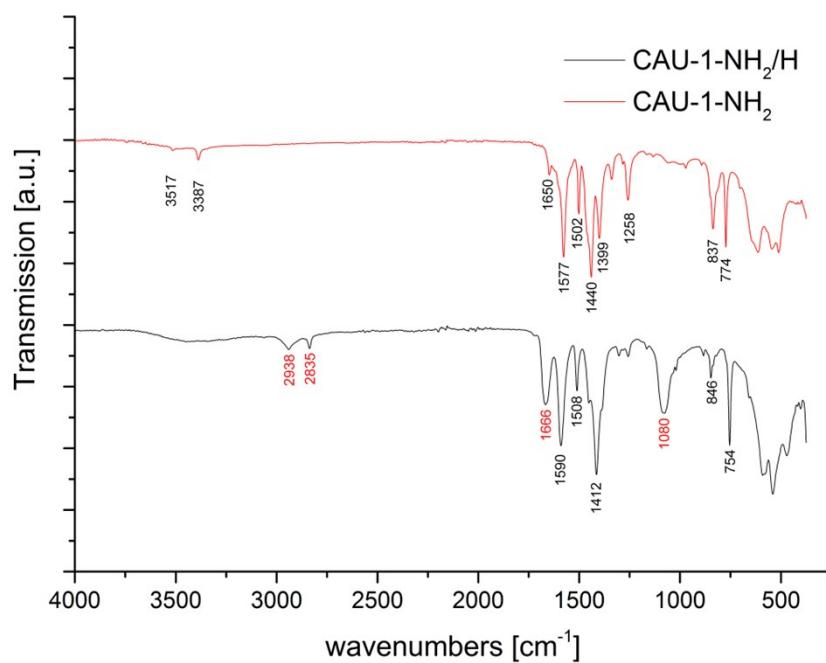


Fig. S2: IR-spectra of the two CAU-1 samples one containing exclusively aminoterephthalate ions (CAU-1-NH<sub>2</sub>) and the other containing terephthalate and aminoterephthalate ions in a molar ratio 3 to 1 (CAU-1-NH<sub>2</sub>/H). Bands corresponding to residual DMF present in the sample are represented in red.

Table S1: Vibrations and corresponding band positions for the IR-spectra of CAU-1-NH<sub>2</sub> and CAU-1-NH<sub>2</sub>/H.

Vibration	Position for	Position for	Comment
	CAU-1-NH <sub>2</sub> [cm <sup>-1</sup> ]	CAU-1-NH <sub>2</sub> /H [cm <sup>-1</sup> ]	
v <sub>as</sub> (NH <sub>2</sub> )	3517	n.v.	
v <sub>s</sub> (NH <sub>2</sub> )	3387	n.v.	
v(C=C) <sub>ring</sub>	1650	overlapped by DMF	
v <sub>as</sub> (COO)	1577	1590	
v(C=C) <sub>ring</sub>	1502	1508	
v <sub>s</sub> (COO)	1440	1412	
v(C=C) <sub>ring</sub>	1399	n.v.	
v(C-NH <sub>2</sub> )	1258	n.v.	
γ(C-H)	837	846	1,2,4- Substitution (CAU-1-NH <sub>2</sub> ) and
(C-H)	774	754	1,4-Substitution (CAU-1-NH <sub>2</sub> /H)
v(C-H)	-----	2938	DMF
v(N-CH <sub>3</sub> )	-----	2835	DMF
v(C=O)	-----	1666	DMF
γ(N-CH <sub>3</sub> )	-----	1080	DMF

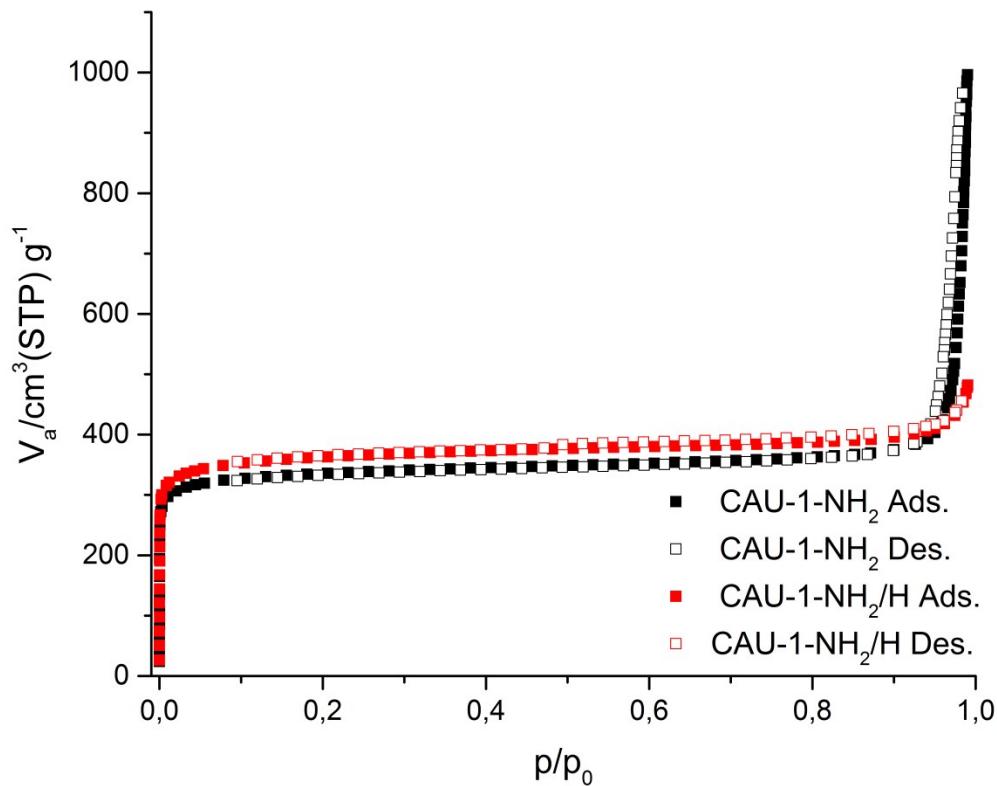


Fig. S3: The N<sub>2</sub>-sorption isotherms (77 K) of the two CAU-1 samples one containing exclusively aminoterephthalate ions (CAU-1-NH<sub>2</sub>) and the other containing terephthalate and aminoterephthalate ions in a molar ratio 3 to 1 (CAU-1-NH<sub>2</sub>/H).

Table S2: Specific BET surface areas and micropore volumes for the samples CAU-1-NH<sub>2</sub> and CAU-1-NH<sub>2</sub>/H compared to the values reported in literature<sup>[16]</sup>.

Compound	spec. BET surface area [m <sup>2</sup> /g]	micropore volume [cm <sup>3</sup> /g]
Literature <sup>[16]</sup>	1530	0.64
CAU-1-NH <sub>2</sub>	1316	0.54
CAU-1-NH <sub>2</sub> /H	1416	0.58

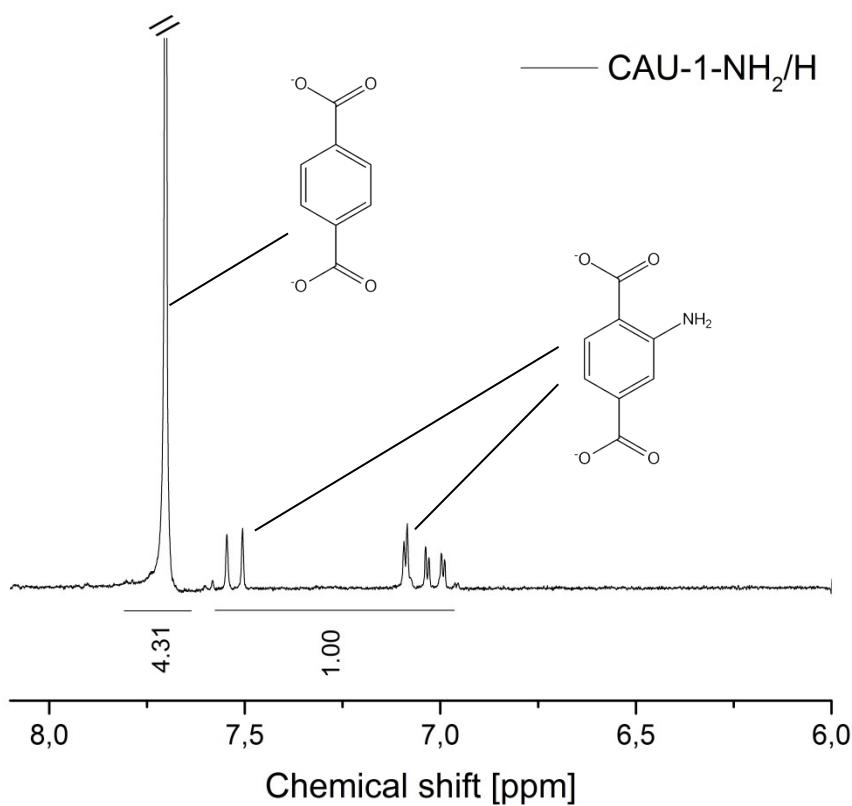


Fig. S4: <sup>1</sup>H-NMR spectrum of CAU-1-NH<sub>2</sub>/H in NaOD/D<sub>2</sub>O (5%) to determine the ratio of terephthalate to aminoterephthalate ions.

### Characterization of the CAU-10 type compounds

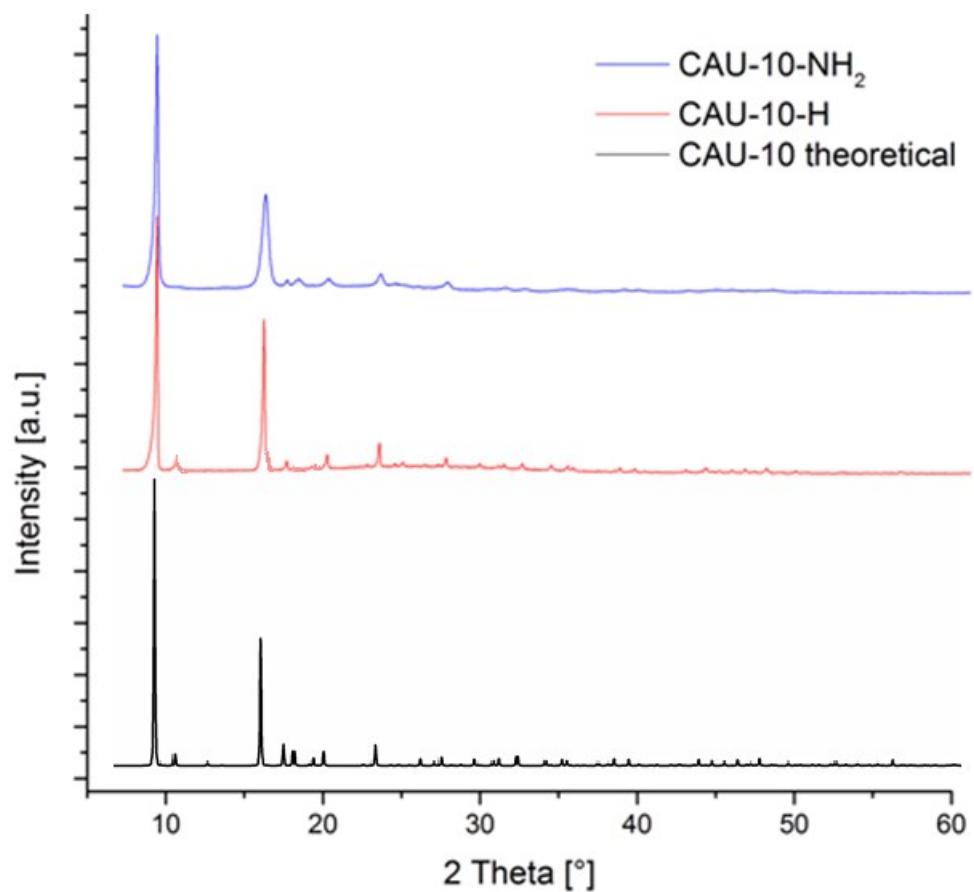


Fig. S5: PXRD patterns of the two CAU-10 samples containing either isophthalate (CAU-10-H) or aminoisoterephthalate (CAU-10-NH<sub>2</sub>) ions.

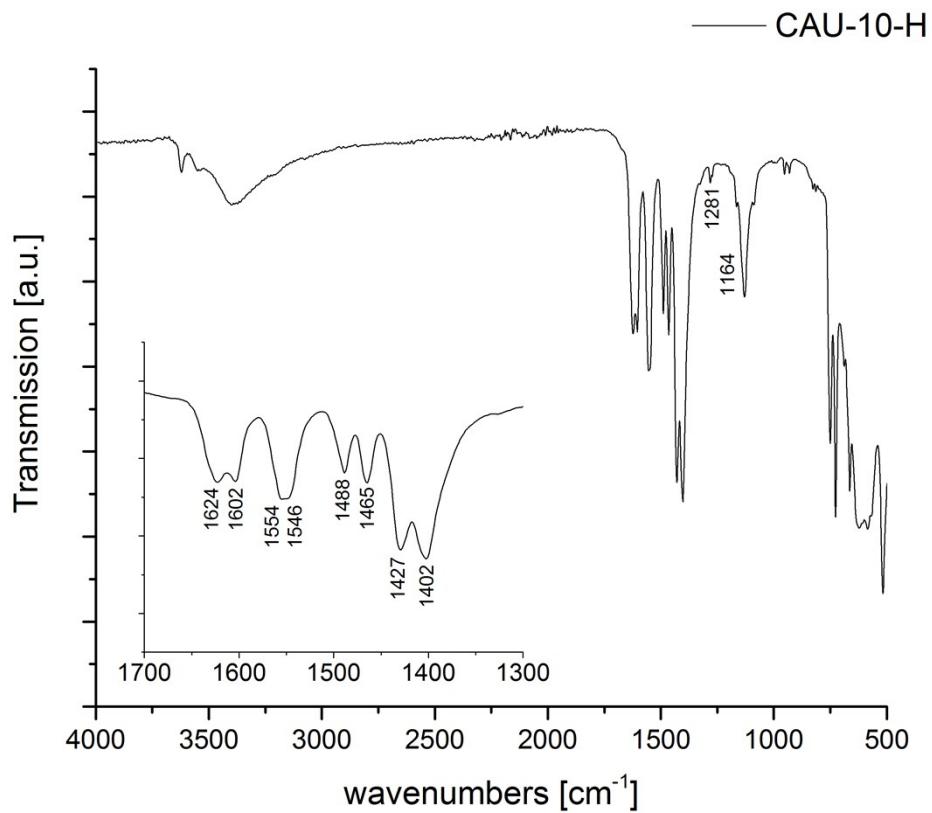


Fig. S6: IR-spectrum of CAU-10-H containing only isoterephthalate ions.

Table S3: Vibrations and corresponding band positions for the IR-spectrum of CAU-10-H.

Vibration	Band position for CAU-10-H [cm <sup>-1</sup> ]
v (COO)	1624
v (C=C)	1602
v <sub>as</sub> (COO)	1554
v <sub>as</sub> (COO)	1546
v (C=C) <sub>ring</sub>	1488
v (C=C) <sub>ring</sub>	1465
v (C=C) <sub>ring</sub>	1427
v <sub>s</sub> (COO)	1402
δ (C-H)	1281
δ (C-H)	1164

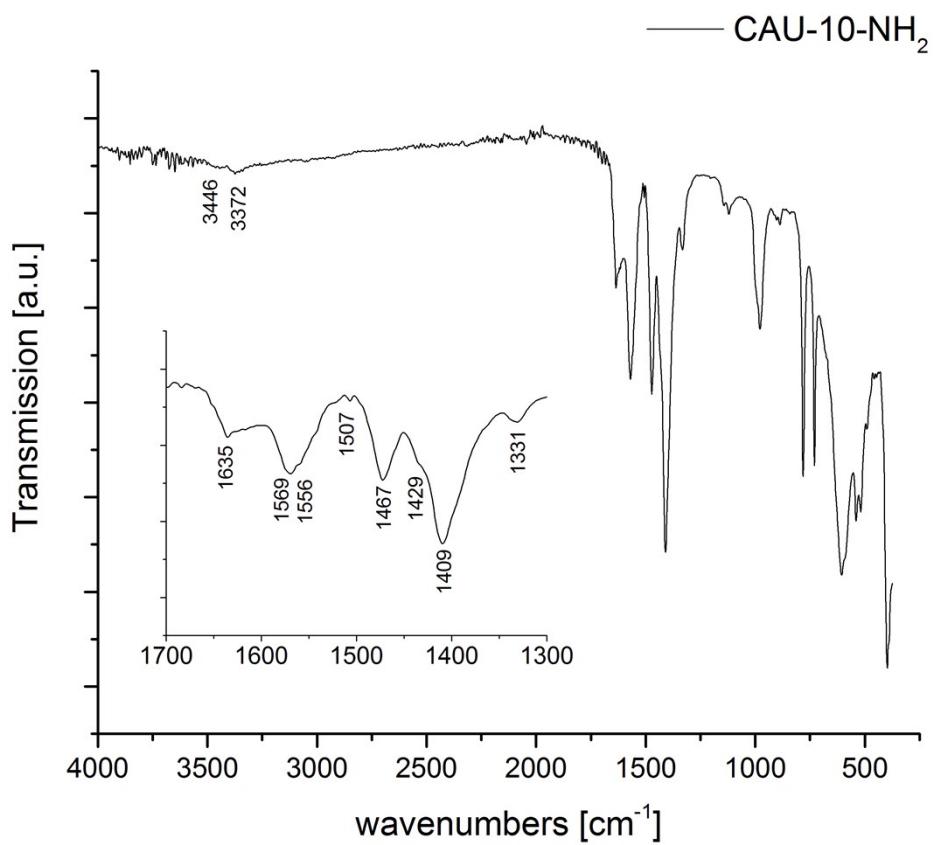


Fig. S7: IR-spectrum of CAU-10-NH<sub>2</sub> containing exclusively aminoisoterephthalate ions.

Table S4: Vibrations and corresponding band positions for the IR-spectrum of CAU-10-H.

Vibration	Position for CAU-10-NH <sub>2</sub> [cm <sup>-1</sup> ]
$\nu_{as}$ (NH <sub>2</sub> )	3372
$\nu_s$ (NH <sub>2</sub> )	3446
$\nu$ (CO <sub>2</sub> )	1635
$\nu_{as}$ (CO <sub>2</sub> )	1569
$\nu_{as}$ (CO <sub>2</sub> )	1556
$\delta$ (NH)	1507
$\nu$ (CC)	1467
$\nu$ (CC)	1429
$\nu_s$ (CO <sub>2</sub> )	1409
$\nu$ (CN)	1331

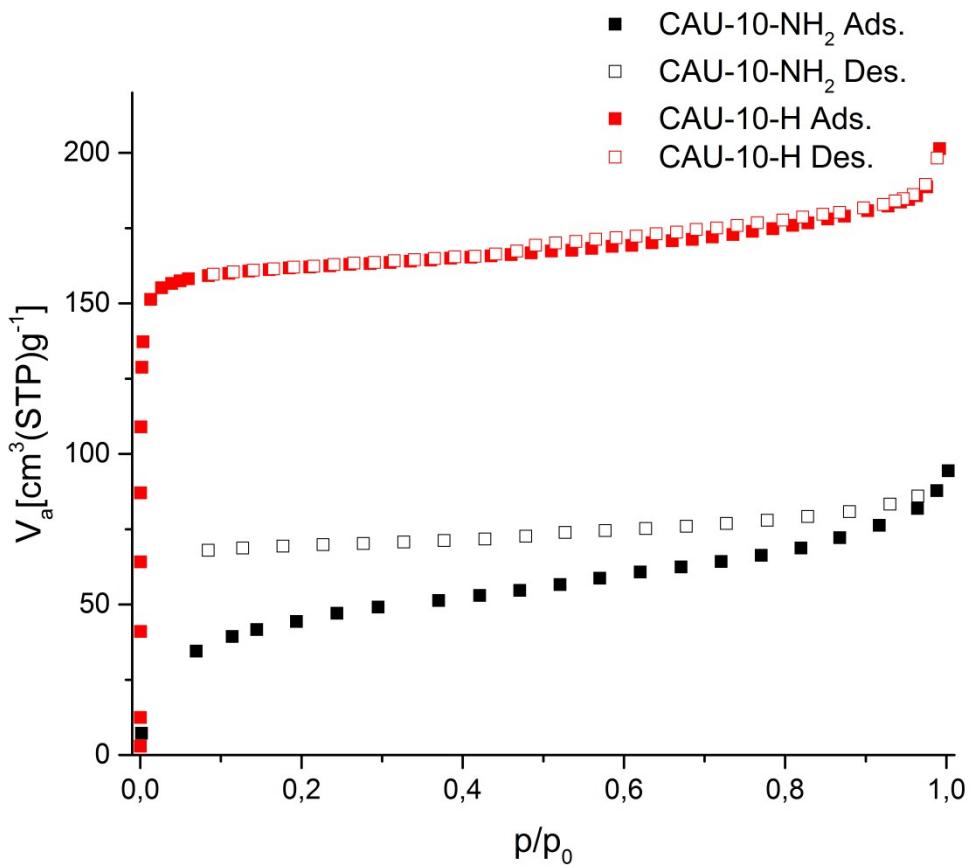


Fig. S8: The results of the N<sub>2</sub> sorption measurements at 77 K of the two CAU-10 samples containing either isophthalate (CAU-10-H) or aminoisoterephthalate (CAU-10-NH<sub>2</sub>) ions.

Table S5: Specific BET surface areas and micropore volumes for the samples CAU-10-H and CAU-10-NH<sub>2</sub> compared to the values reported in literature<sup>[26a]</sup>. The BET evaluation in the case of CAU-10-NH<sub>2</sub> is problematic due to kinetic hindrance.

Compound	spec. BET surface area [m <sup>2</sup> /g]	micropore volume [cm <sup>3</sup> /g]
Literature CAU-10-H <sup>[26b]</sup>	640	0.25
CAU-10-H	656	0.25
CAU-10-NH <sub>2</sub>	[159]	0.08

## 2. Catalytic studies

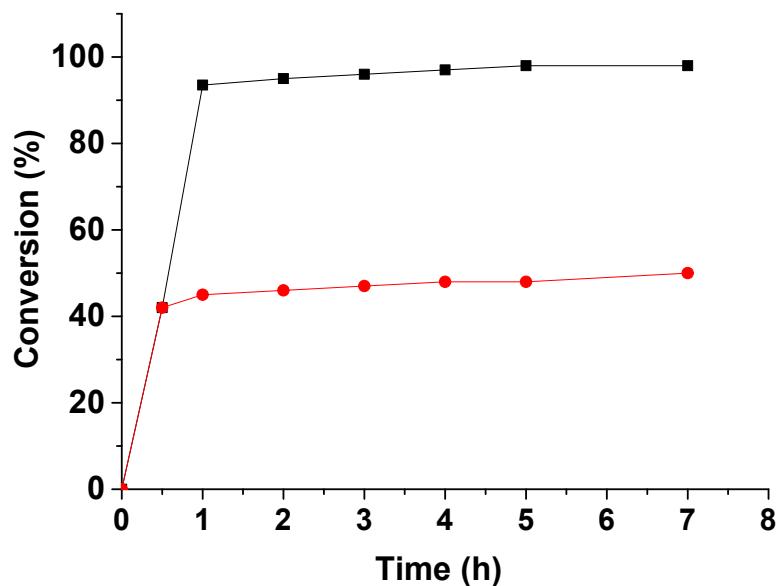


Fig. S9: Result of the hot filtration experiment on CAU-1-NH<sub>2</sub>. Time conversion plots for the reaction between benzaldehyde and malononitrile in the presence of CAU-1-NH<sub>2</sub> (black square) and conversion plot after removal of the catalyst after 30 minutes (red dots).

## 3. Characterisation of the CAU-1 type materials after the catalytic investigation

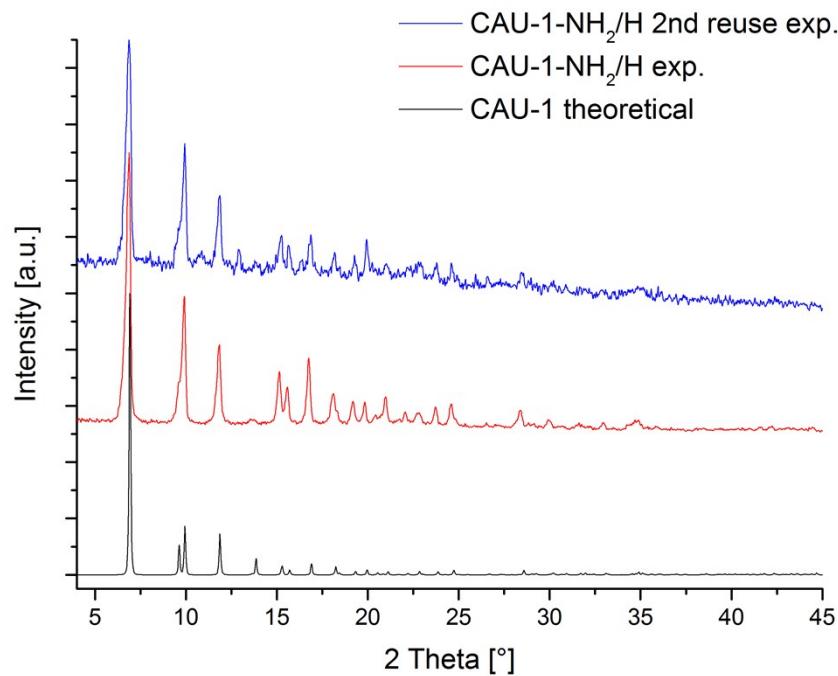


Fig. S10: PXRD patterns of fresh and recovered CAU-1-NH<sub>2</sub>/H catalysts.

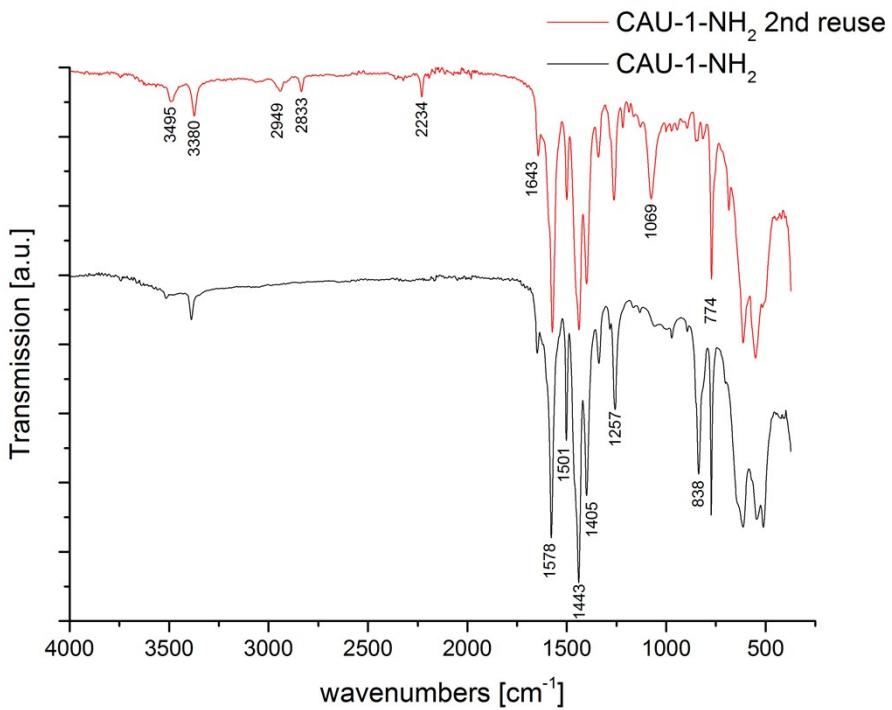


Fig. S11: IR spectra of fresh and recovered CAU-1-NH<sub>2</sub> catalysts. The additionally observed bands in the IR spectrum of the recovered catalyst are due to presence of solvent molecules (ethanol) and the product benzylidenemalonitrile: 1069 cm<sup>-1</sup> (C-O stretching in ethanol), 2234 cm<sup>-1</sup> (CN stretching in product), 2833 cm<sup>-1</sup> (CH stretching in product), 2949 cm<sup>-1</sup> (CH stretching from ethanol).

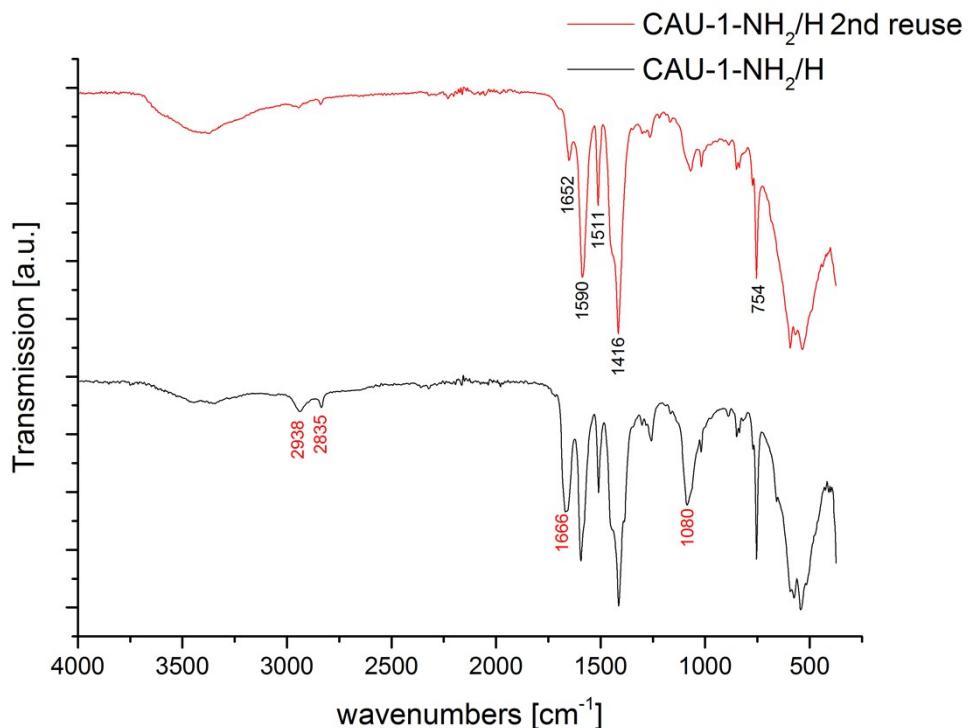


Fig. S12: IR spectra of fresh and recovered CAU-1-NH<sub>2</sub>/H catalysts.