

Identification and H(D)-bond Energies of C–H(D)···Cl Interactions in Chloride–Haloalkane Clusters: Combined X-ray Crystallographic, Spectroscopic, and Theoretical Study

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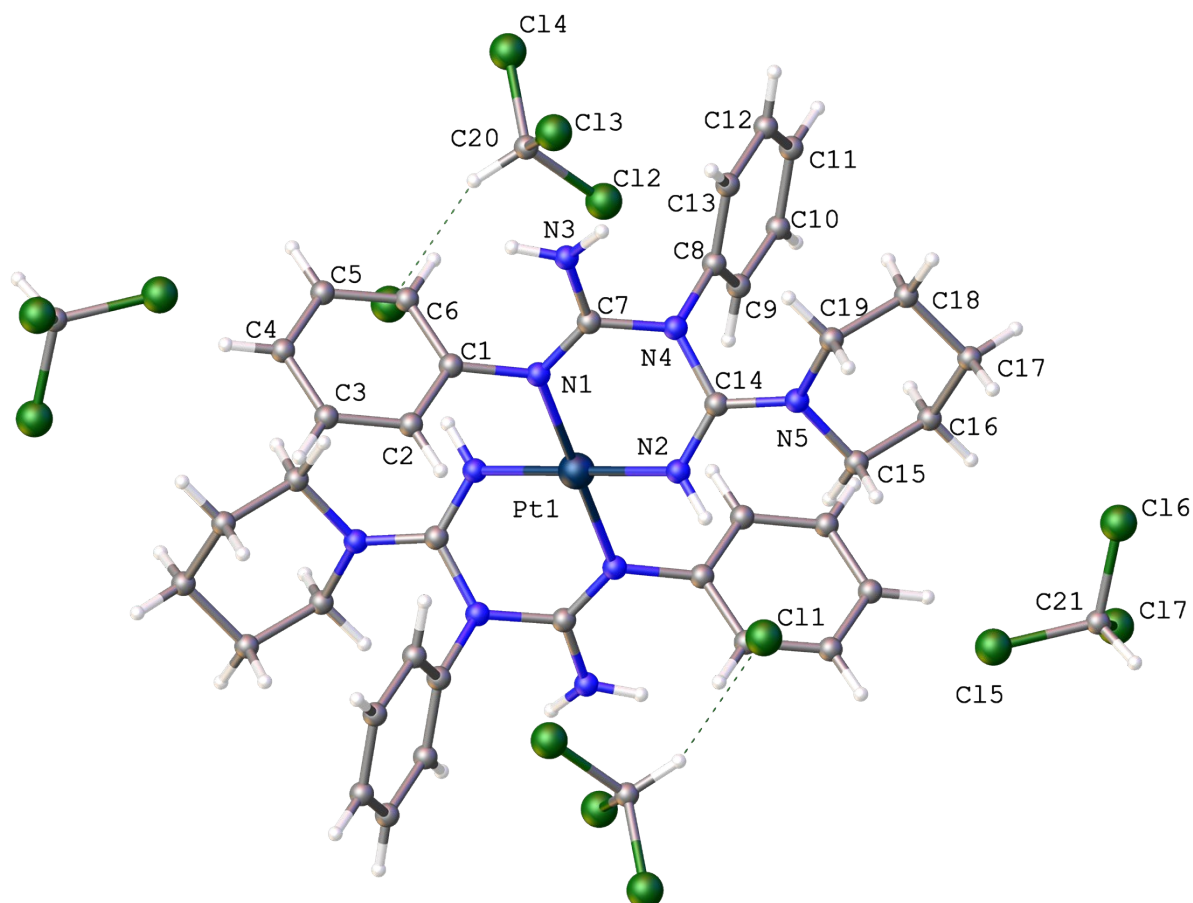
Table S1. Crystallographic and refinement data.

Compound	[1][Cl ₂ (CHBr ₃) ₄]	[1][Cl ₂ (CH ₂ Cl ₂) ₂]	[1][Cl ₂ (C ₂ H ₄ Cl ₂) ₂]	[1][Cl ₂ (CDCl ₃) ₄]
Empirical formula	C ₄₂ H ₅₀ Br ₁₂ Cl ₂ N ₁₀ Pt	C ₄₀ H ₅₀ Cl ₆ N ₁₀ Pt	C ₄₂ H ₅₄ Cl ₆ N ₁₀ Pt	C ₄₂ H ₄₆ Cl ₁₄ D ₄ N ₁₀ Pt
Formula weight	1919.83	1078.69	1106.74	1386.31
T (K)	100(2)	100(2)	100(2)	100(2)
λ (Å)	0.71073	0.71073	0.71073	0.71073
Cryst syst	Orthorhombic	Triclinic	Monoclinic	Orthorhombic
Space group	Pbca	P $\bar{1}$	P2 ₁ /c	Pbca
<i>a</i> (Å)	14.2649(5)	12.9443(5)	9.3446(5)	13.1946(2)
<i>b</i> (Å)	18.1921(5)	13.0808(5)	9.4052(6)	17.8886(2)
<i>c</i> (Å)	23.1012(7)	15.1903(6)	26.2178(17)	23.6588(3)
α (deg)	90	109.653(2)	90	90
β (deg)	90	113.249(2)	92.274(3)	90
γ (deg)	90	92.700(2)	90	90
<i>V</i> (Å ³)	5995.0(3)	2176.92(15)	2302.4(2)	5584.25(13)
<i>Z</i>	4	2	2	4
ρ_{calc} (Mg/m ³)	2.127	1.646	1.596	1.649
μ (Mo K α) (mm ⁻¹)	10.467	3.634	3.438	3.223
No. reflns.	35250	21401	14065	87106
Unique reflns.	7682	7299	4156	8161
GOOF (F ²)	1.115	1.076	1.160	1.024
R _{int}	0.0514	0.0841	0.0858	0.0402
R1 ^a (<i>I</i> ≥ 2 σ)	0.0587	0.0601	0.0735	0.0288
wR2 ^b (<i>I</i> ≥ 2 σ)	0.1140	0.1129	0.1266	0.0545

$$^a R1 = \Sigma||F_o| - |F_c||/\Sigma|F_o|. \quad ^b wR2 = [\Sigma[w(F_o^2 - F_c^2)^2]/\Sigma[w(F_o^2)^2]]^{1/2}.$$

Table S2. Selected bond lengths [\AA] and angles [$^\circ$] for solvates.

Solvate	[1][Cl ₂ (CDCl ₃) ₄]	[1][Cl ₂ (CHBr ₃) ₄]	[1][Cl ₂ (C ₂ H ₄ Cl ₂) ₂]	[1][Cl ₂ (CH ₂ Cl ₂) ₂]
Pt(1)–N(2)	2.001(2)	2.009(6)	2.000(9)	2.019(6)
Pt(1)–N(2B)				1.999(6)
Pt(1)–N(1)	2.015(2)	2.006(6)	2.037(9)	1.997(7)
Pt(1)–N(1B)				2.014(6)
N(1)–C(7)	1.303(3)	1.301(9)	1.263(14)	1.319(10)
N(2)–C(14)	1.295(3)	1.298(9)	1.282(13)	1.294(10)
N(3)–C(7)	1.336(3)	1.329(9)	1.328(14)	1.324(10)
N(4)–C(14)	1.404(3)	1.399(9)	1.428(12)	1.411(10)
N(4)–C(7)	1.413(3)	1.416(9)	1.436(13)	1.410(10)
N(5)–C(14)	1.346(3)	1.348(9)	1.353(12)	1.329(9)
N(2)–Pt(1)–N(1)	86.48(8)	85.8(2)	83.9(4)	84.0(3)
C(7)–N(1)–Pt(1)	122.96(17)	124.0(5)	124.3(8)	123.8(6)
C(14)–N(2)–Pt(1)	122.88(17)	123.0(5)	122.7(7)	122.4(6)
C(14)–N(4)–C(7)	121.7(2)	121.7(6)	119.4(9)	119.3(7)
N(1)–C(7)–N(4)	121.2(2)	120.9(6)	119.3(11)	118.4(8)
N(2)–C(14)–N(4)	120.3(2)	120.8(6)	119.7(9)	119.0(7)

**Figure S1.** Molecular structure of [1][Cl₂(CDCl₃)₄] with an atom numbering scheme.

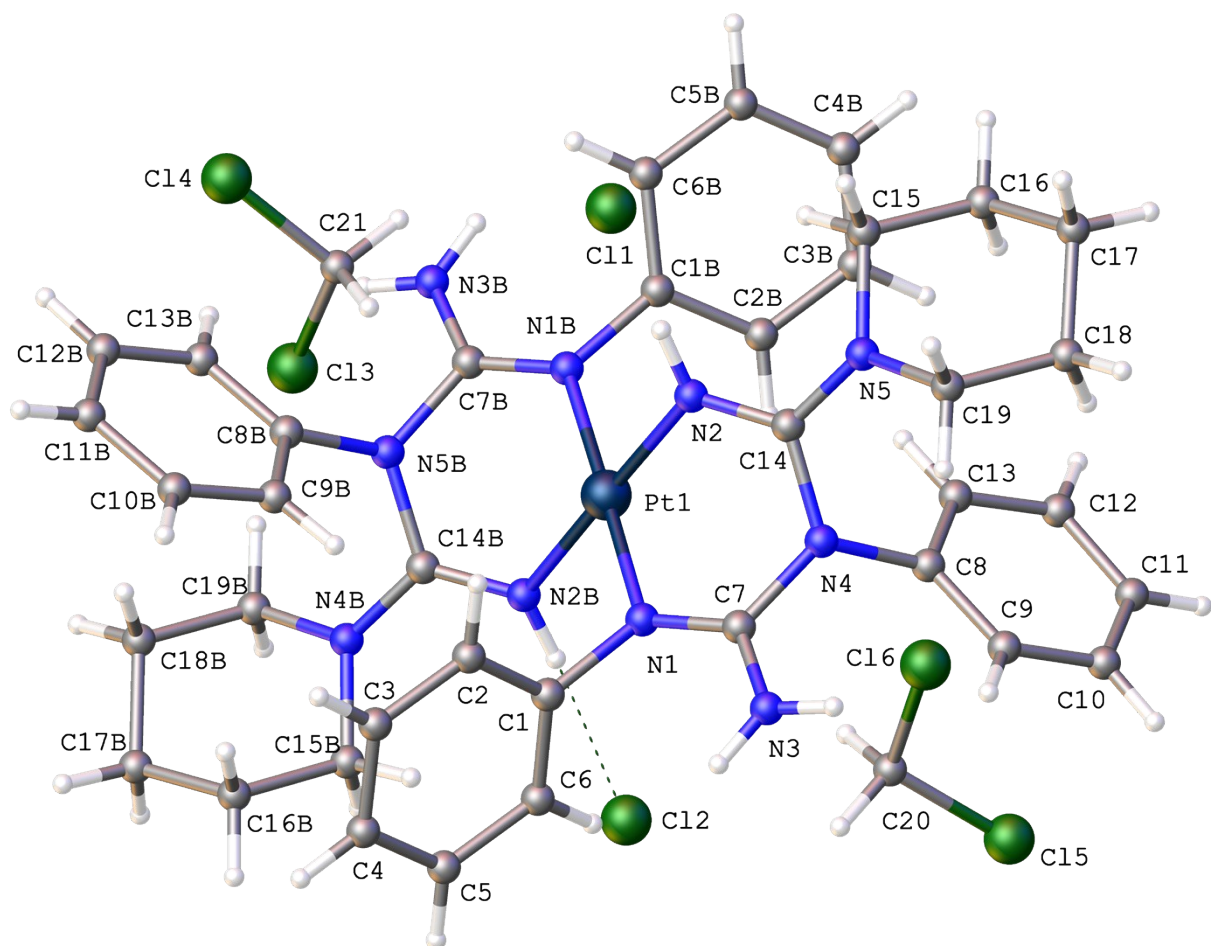


Figure S2. Molecular structure of $[1][Cl_2(CH_2Cl_2)_2]$ with an atom numbering scheme.

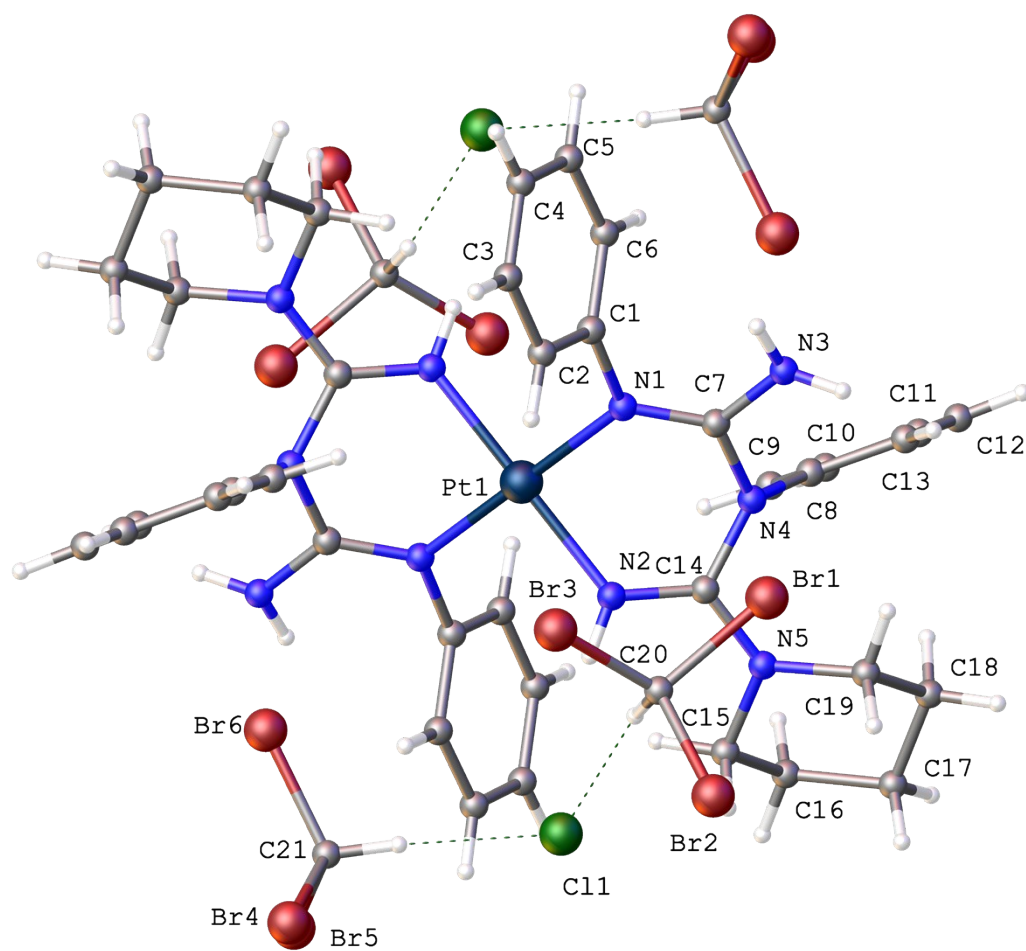


Figure S3. Molecular structure of [1][Cl₂(CHBr₃)₄] with an atom numbering scheme.

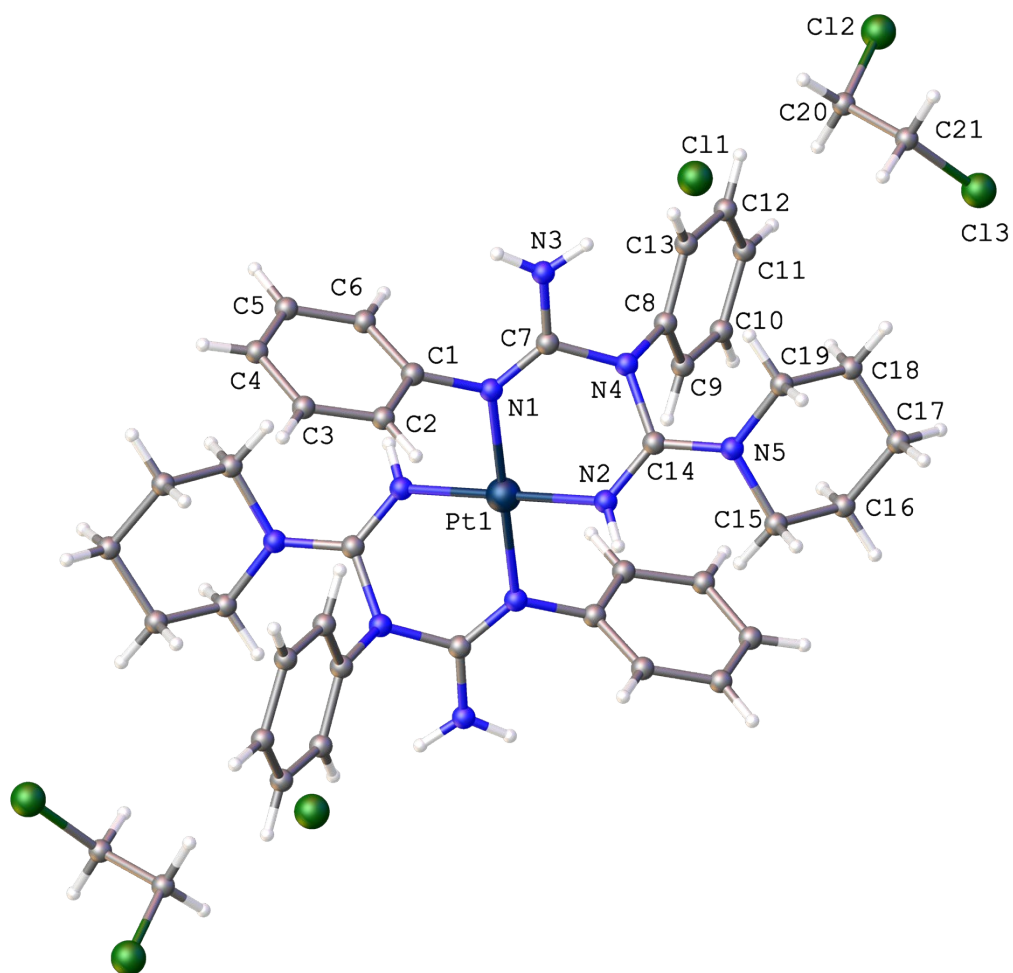


Figure S4. Molecular structure of $[1][Cl_2(C_2H_4Cl_2)_2]$ with an atom numbering scheme.

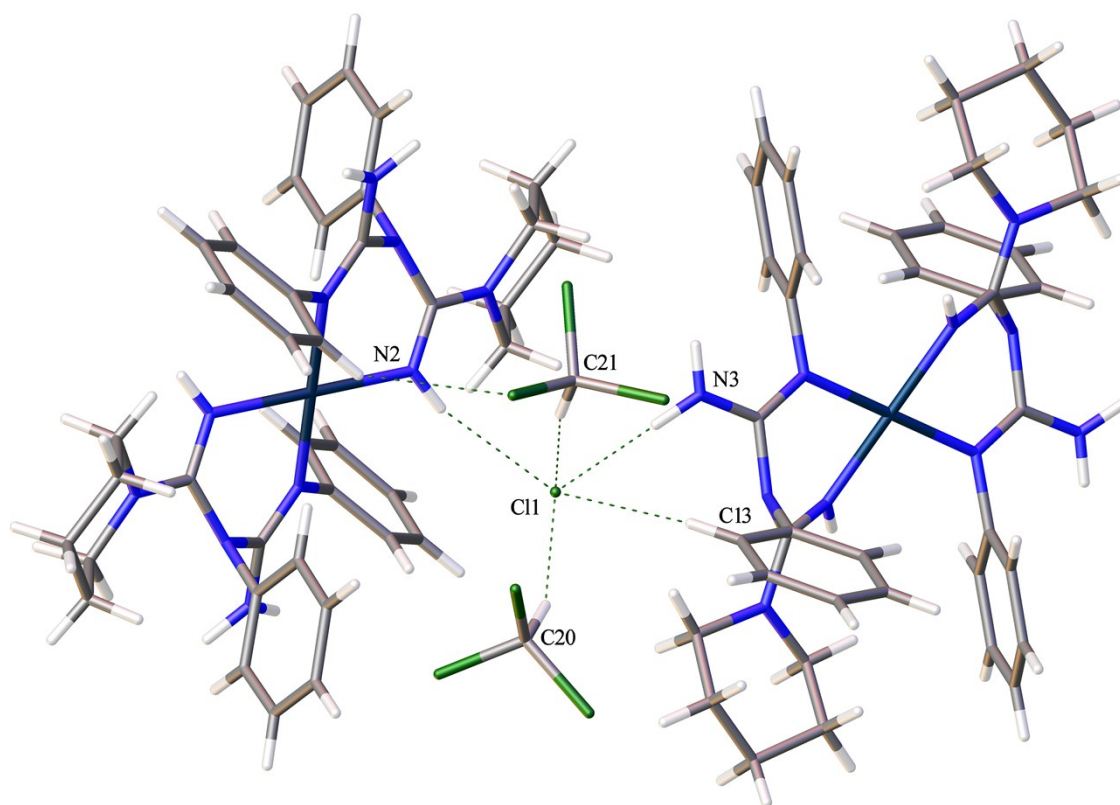


Figure S5. Hydrogen bonds in the structure of $[1][Cl_2(CDCl_3)_4]$ showing a pyramidal environment of Cl^- anion and the $[Cl(CDCl_3)_2]^-$ cluster.

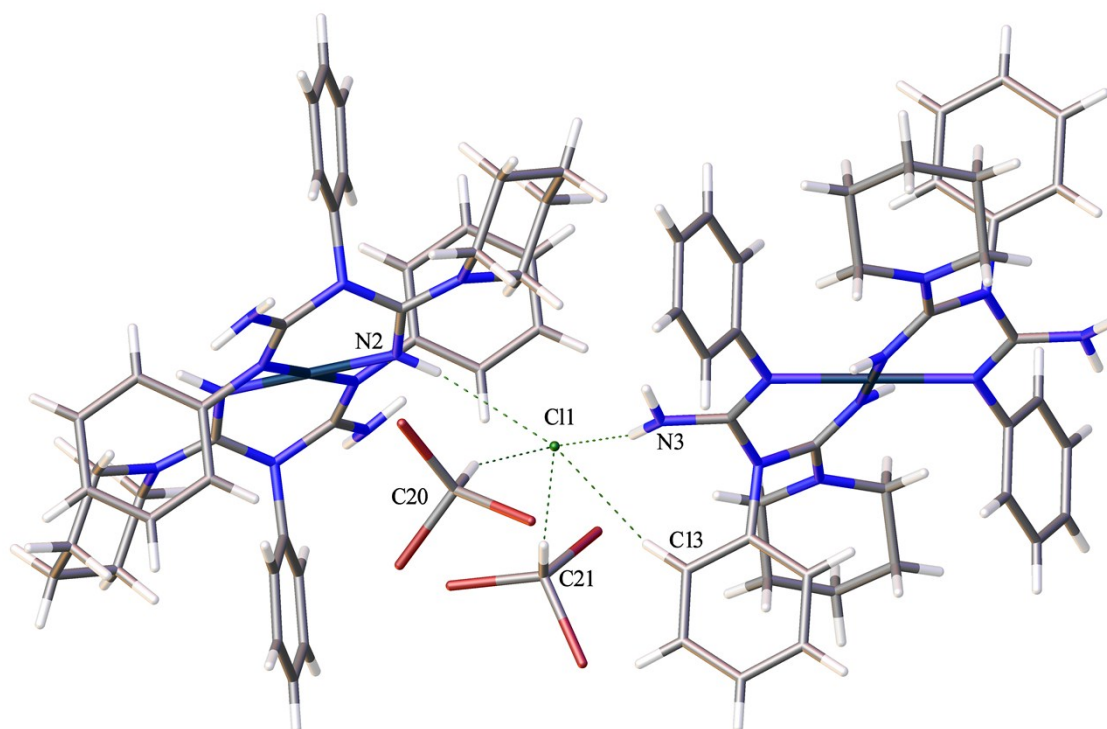


Figure S6. Intermolecular hydrogen bonds in the structure of $[1][Cl_2(CHBr_3)_4]$ showing a pyramidal environment of Cl^- anion and the $[Cl(CHBr_3)_2]^-$ cluster.

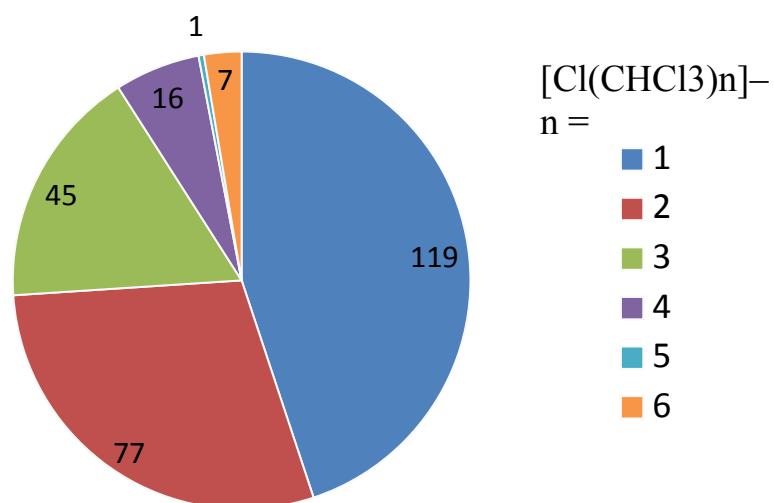


Figure S7. The distribution of the chloride-chloroform clusters $[\text{Cl}(\text{CHCl}_3)_n]^-$ depending on the number of H-bonded chloroform molecules n (according to the CCDC database).

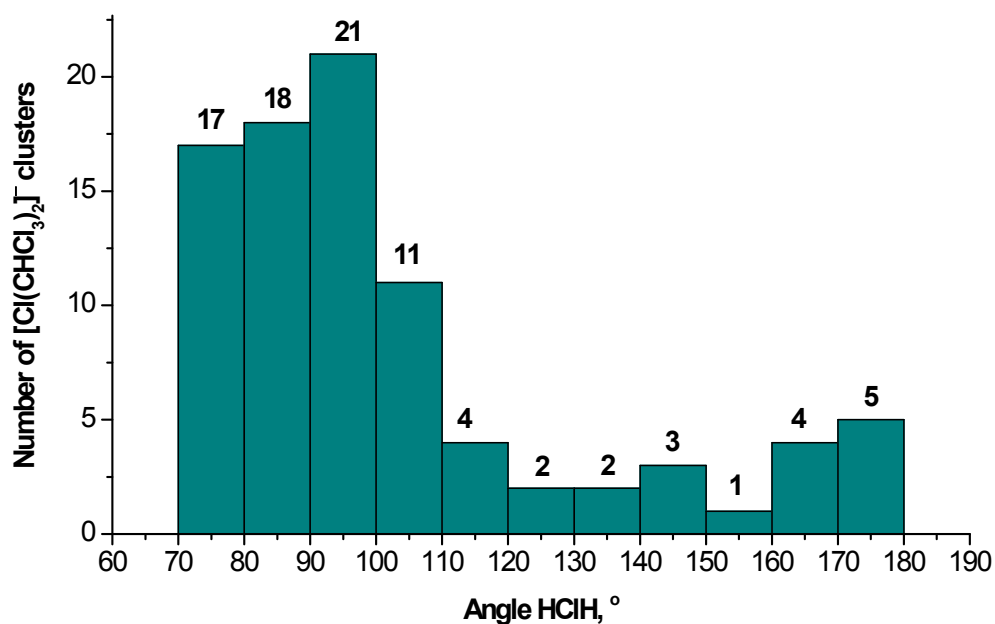


Figure S8. The distribution of dimeric chloride-chloroform clusters $[\text{Cl}(\text{CHCl}_3)_2]^-$ depending on the value of HClH angle (according to the CCDC database).

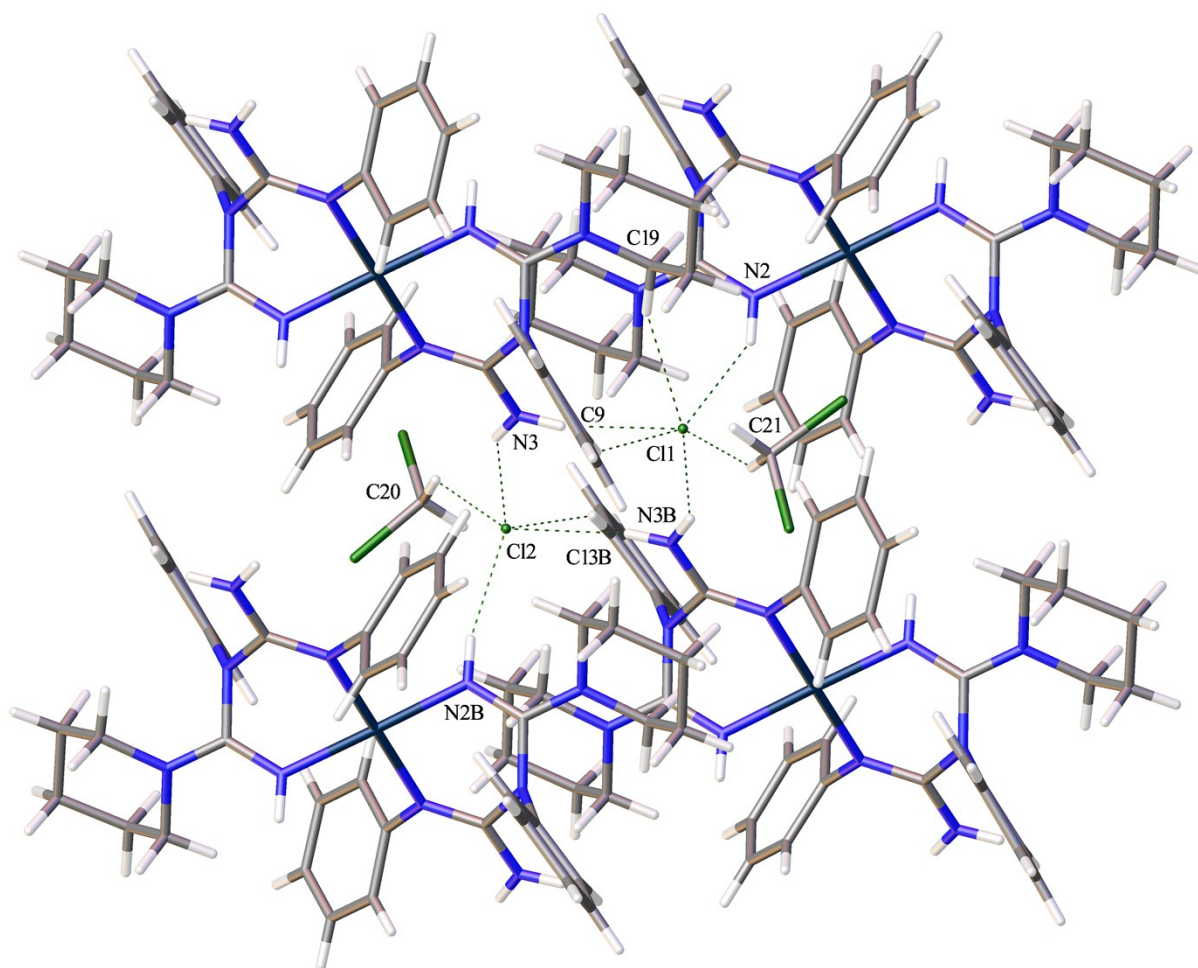


Figure S9. Intermolecular hydrogen bonds in the structure of $[1][Cl_2(CH_2Cl_2)_2]$ showing the asymmetric environment of the Cl^- anions.

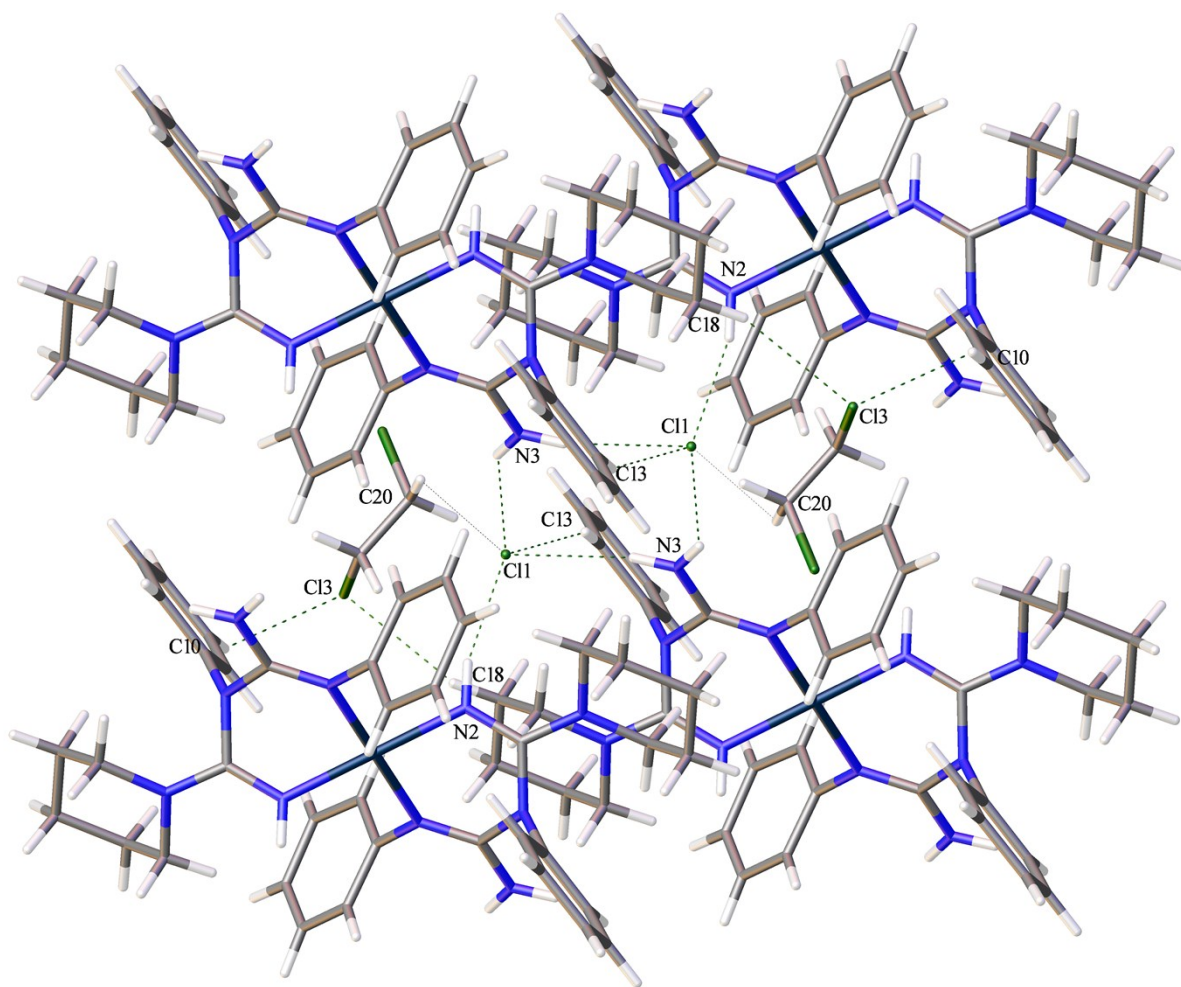


Figure S10. Intermolecular hydrogen bondings in the structure of $[1][Cl_2(C_2H_4Cl_2)_2]$ showing the pyramidal environment of the Cl^- anions.

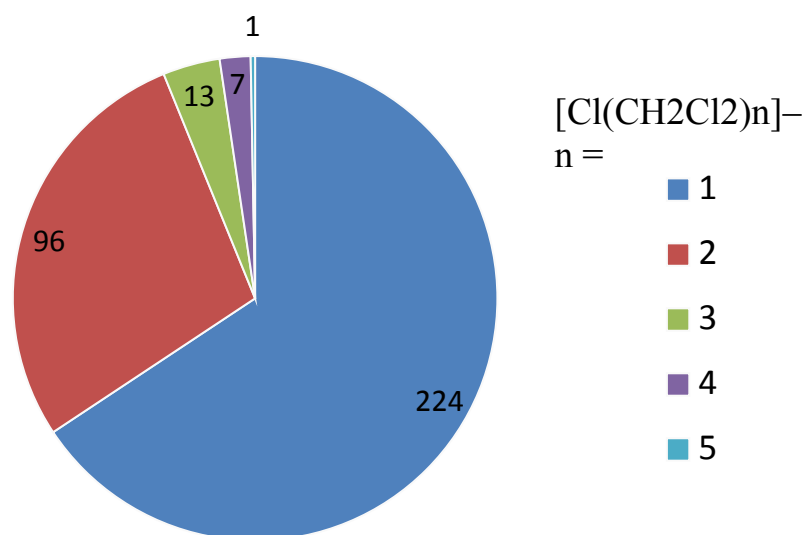


Figure S11. The distribution of the chloride-dichloromethane clusters $[\text{Cl}(\text{CH}_2\text{Cl}_2)_n]^-$ depending on the number of H-bonded chloroform molecules n (according to the CCDC database).

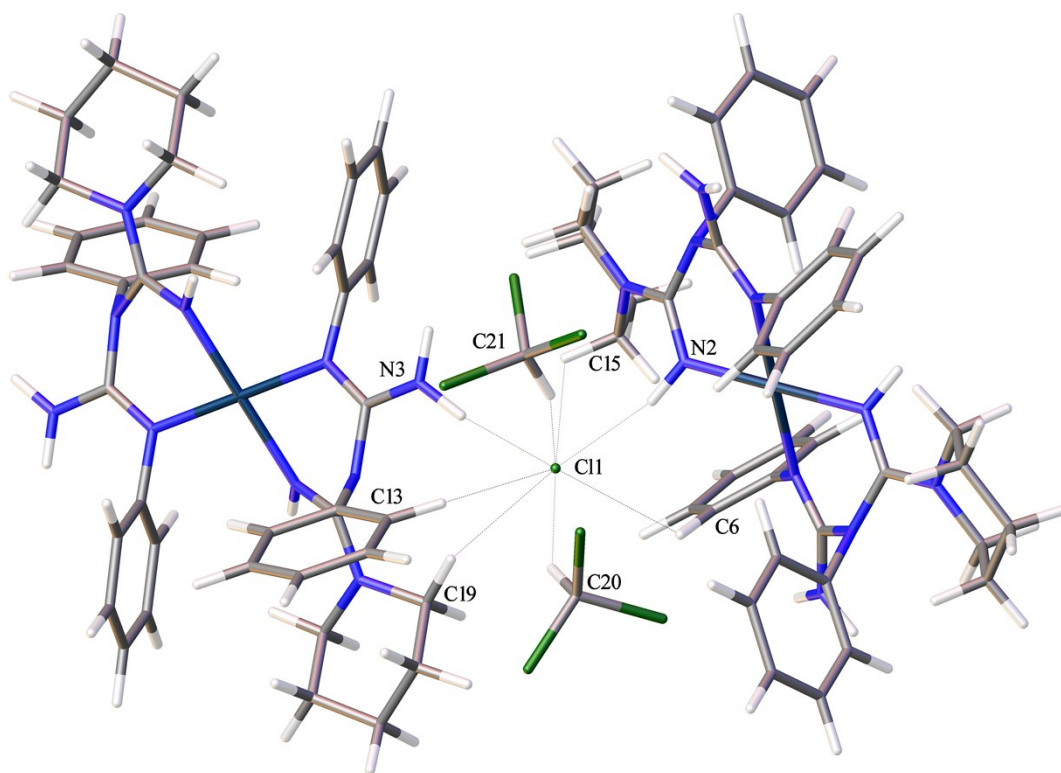


Figure S12. Graphical representation of hydrogen bondings formed by the Cl^- ion in the $[\mathbf{1}]_2[\text{Cl}(\text{CHCl}_3)_2]$ system according to DFT calculations.

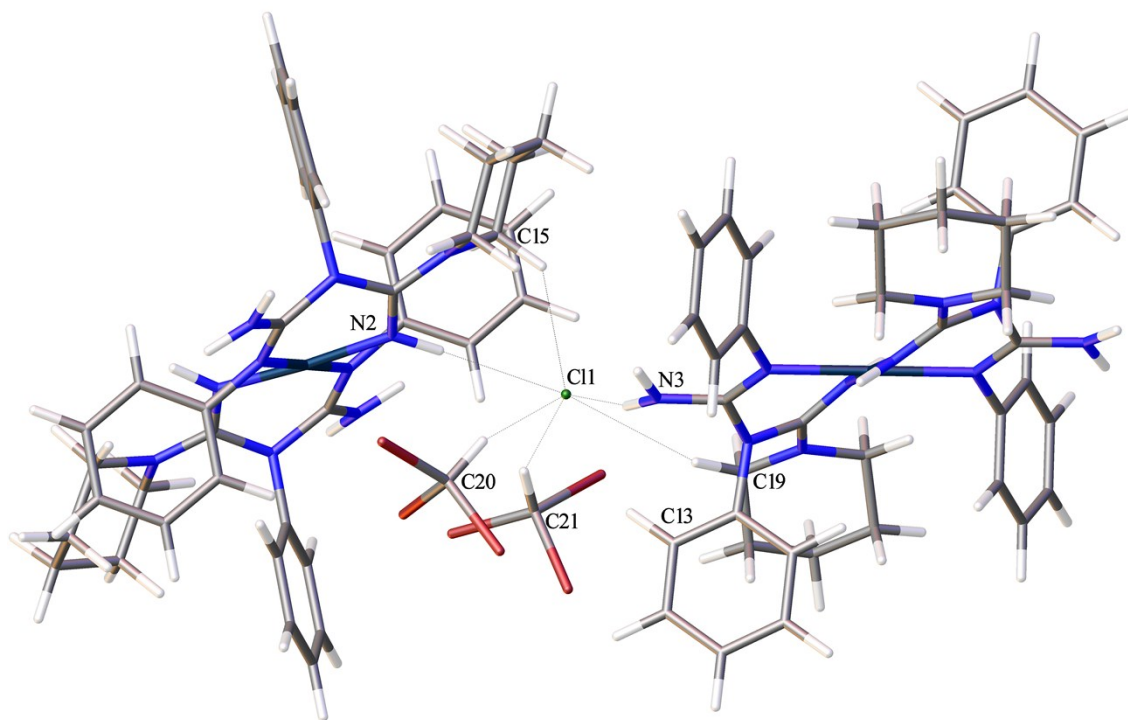


Figure S13. Graphical representation of hydrogen bondings formed by the Cl^- ion in the $[\mathbf{1}]_2[\text{Cl}(\text{CHBr}_3)_2]$ system according to DFT calculations.

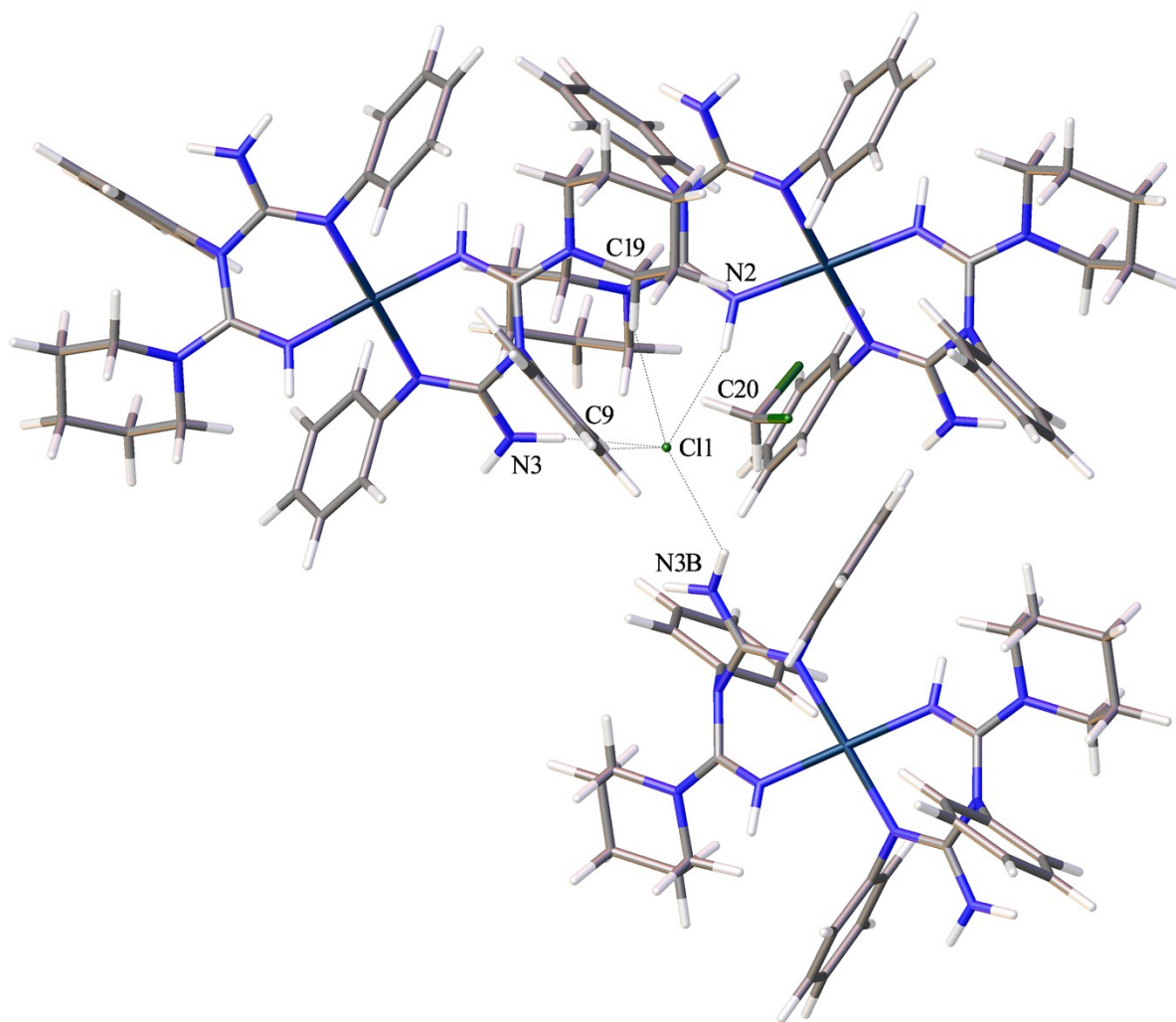


Figure S14. Graphical representation of hydrogen bondings formed by the Cl^- ion in the $[\mathbf{1}]_3[\text{Cl}(\text{CH}_2\text{Cl}_2)]$ system according to DFT calculations.

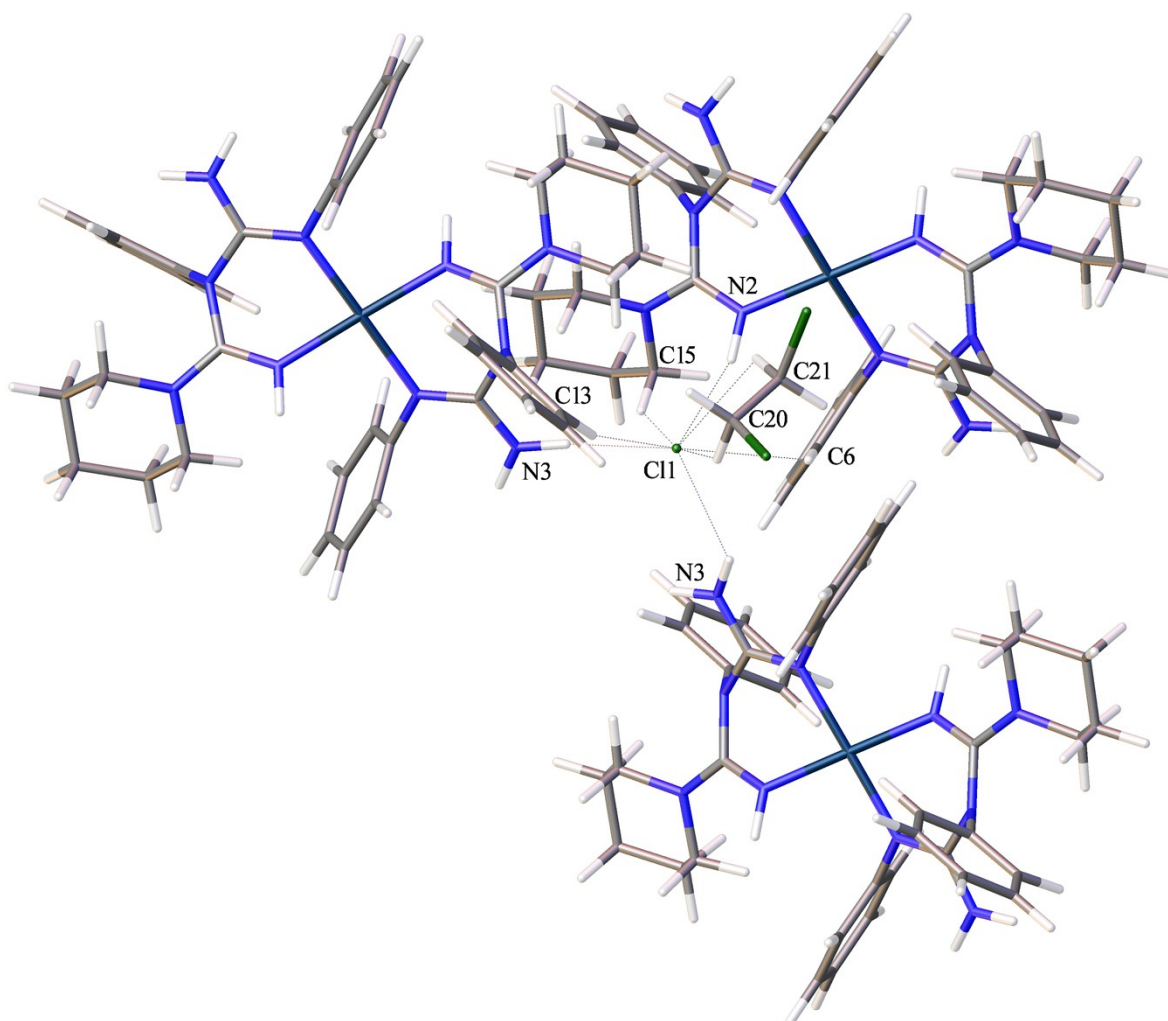


Figure S15. Graphical representation of hydrogen bondings formed by the Cl⁻ ion in the [1]₃[Cl(C₂H₄Cl₂)] system according to DFT calculations.

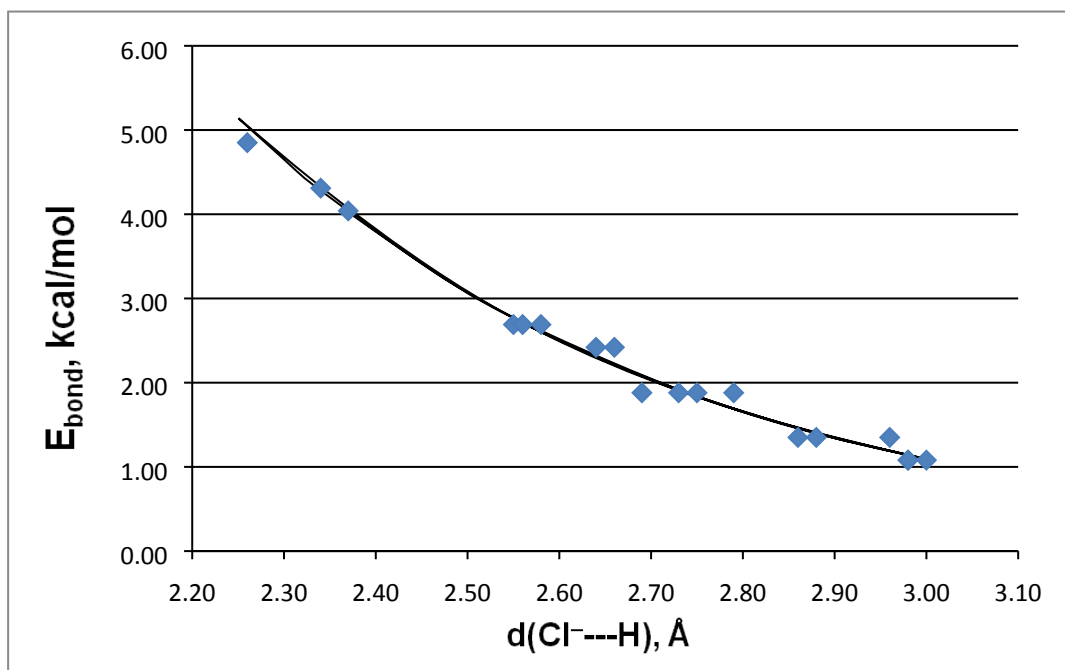


Figure S16. The correlation between theoretically calculated Cl⁻···H distances (Å) and energies of the corresponding contacts E_{bond}^b (kcal/mol).

Table S3. Experimentally determined and theoretically calculated selected bond lengths, values of the density of all electrons – $\rho(\mathbf{r})$, Laplacian of electron density – $\nabla^2\rho(\mathbf{r})$, energy density – H_b , potential energy density – $V(\mathbf{r})$, and Lagrangian kinetic energy – $G(\mathbf{r})$ (Hartree) at the bond critical points (3, –1), corresponding to HBs in $[\mathbf{1}]_2[\text{Cl}(\text{CHCl}_3)_2]$, $[\mathbf{1}]_2[\text{Cl}(\text{CHBr}_3)_2]$, $[\mathbf{1}]_3[\text{Cl}(\text{CH}_2\text{Cl}_2)]$, and $[\mathbf{1}]_3[\text{Cl}(\text{C}_2\text{H}_4\text{Cl}_2)]$ as well as energies of these bonds E_{bond} (kcal/mol) defined by two methods.

Contact	Bond lengths, Å		$\rho(\mathbf{r})$	$\nabla^2\rho(\mathbf{r})$	H_b	$V(\mathbf{r})$	$G(\mathbf{r})$	E_{bond}^a	E_{bond}^b
	X-ray	Theory							
$[\mathbf{1}]_2[\text{Cl}(\text{CHCl}_3)_2]$									
Cl \cdots H–C(20)Cl ₃	2.43	2.37	0.020	0.062	0.001	–0.014	0.015	4.39	4.04
Cl \cdots H–C(21)Cl ₃	2.44	2.55	0.014	0.043	0.001	–0.009	0.010	2.82	2.69
Cl \cdots H–N(3)H	2.20	2.08	0.034	0.087	–0.003	–0.027	0.024	8.47	6.46
Cl \cdots H–N(2)	2.56	2.27	0.023	0.068	0.000	–0.017	0.017	5.33	4.58
Cl \cdots H–C(19)H	3.05	2.88	0.007	0.025	0.001	–0.004	0.005	1.26	1.35
Cl \cdots H–C(15)H	2.97	2.96	0.007	0.024	0.001	–0.004	0.005	1.26	1.35
Cl \cdots H–C(13)	2.73	2.56	0.014	0.044	0.001	–0.009	0.010	2.82	2.69
Cl \cdots H–C(6)	3.14	2.79	0.009	0.034	0.002	–0.005	0.007	1.57	1.88
$[\mathbf{1}]_2[\text{Cl}(\text{CHBr}_3)_2]$									
Cl \cdots H–C(20)Br ₃	2.43	2.26	0.025	0.069	–0.001	–0.019	0.018	5.96	4.85
Cl \cdots H–C(21)Br ₃	2.36	2.34	0.021	0.063	0.000	–0.015	0.016	4.71	4.31
Cl \cdots H–N(3)H	2.29	2.16	0.028	0.075	–0.001	–0.022	0.020	6.90	5.38
Cl \cdots H–N(2)	2.71	2.68	0.010	0.033	0.001	–0.006	0.007	1.88	1.88
Cl \cdots H–C(15)H	2.87	2.58	0.014	0.046	0.001	–0.009	0.010	2.82	2.69
Cl \cdots H–C(19)H	3.43	3.00	0.006	0.019	0.001	–0.003	0.004	0.94	1.08
Cl \cdots H–C(13)	2.83	2.86	0.008	0.025	0.001	–0.004	0.005	1.26	1.35
$[\mathbf{1}]_3[\text{Cl}(\text{CH}_2\text{Cl}_2)]$									
Cl \cdots H–C(20)HCl ₂	2.58	2.98 3.00	Appropriate bond critical points (3, –1) were not found						

Cl \cdots H-N(3)H	2.40	2.24	0.024	0.066	-0.001	-0.018	0.017	5.65	4.58
Cl \cdots H-N(3B)H	2.45	2.52	0.014	0.047	0.001	-0.009	0.011	2.82	2.96
Cl \cdots H-N(2)	2.66	2.66	0.010	0.034	0.001	-0.006	0.007	1.88	1.88
Cl \cdots H-C(19)H	2.91	2.64	0.012	0.040	0.001	-0.007	0.009	2.20	2.42
Cl \cdots H-C(9)	2.79	2.69	0.011	0.035	0.001	-0.006	0.007	1.88	1.88
[1] ₃ [Cl(C ₂ H ₄ Cl ₂)]									
Cl \cdots H-C(20)H	2.83	2.73	0.010	0.035	0.002	-0.006	0.007	1.88	1.88
Cl \cdots H-C(21)H		2.98	0.006	0.021	0.001	-0.003	0.004	0.94	1.08
Cl \cdots H-N(3)H	2.57	2.30	0.021	0.060	0.000	-0.015	0.015	4.71	4.04
Cl \cdots H-N(3')H	2.52	2.59	0.011	0.040	0.001	-0.007	0.009	2.20	2.42
Cl \cdots H-N(2)	2.52	2.45	0.015	0.048	0.001	-0.010	0.011	3.14	2.96
Cl \cdots H-C(15)H	3.07	2.66	0.013	0.041	0.001	-0.008	0.009	2.51	2.42
Cl \cdots H-C(13)	2.82	2.75	0.011	0.034	0.001	-0.006	0.007	1.88	1.88
Cl \cdots H-C(6)		2.98	0.007	0.023	0.001	-0.003	0.004	0.94	1.08

^a $E_{\text{bond}} = -V(\mathbf{r})/2$, see ref.³⁴ in the main text

^b $E_{\text{bond}} = 0.429G(\mathbf{r})$, see ref.³⁵ in the main text

Table S4. Cartesian atomic coordinates of the calculated equilibrium structures.

Nuclear charges of elements are shown in the second column.

Structure	Charge	X	Y	Z
{[1] ₂ -(CHCl ₃) ₂ -Cl}	78	5.077307	-1.133332	0.627857
	7	5.125520	0.208747	2.178750
	7	7.100879	-0.926969	0.675583
	1	7.689867	-1.733314	0.485021
	7	5.812378	2.237254	3.076030
	7	7.066766	1.320127	1.361229
	1	6.574602	2.871278	3.262159
	1	5.004840	2.265754	3.686669
	7	9.043893	0.361115	0.559318
	6	4.133491	0.075956	3.206562
	6	4.289740	-0.915647	4.176070
	1	5.179458	-1.544657	4.159963
	6	3.312270	-1.078129	5.150532
	1	3.437546	-1.839740	5.917551
	6	2.180949	-0.262651	5.153029
	1	1.422174	-0.391738	5.922424
	6	2.026814	0.715103	4.175961
	1	1.138181	1.343463	4.156201
	6	3.003273	0.890235	3.197343
	1	2.872065	1.625513	2.402033
	6	5.962699	1.210888	2.222745
	6	7.198607	2.540559	0.574270
	6	6.763549	2.493992	-0.746835
	1	6.331569	1.568137	-1.131514
	6	6.872700	3.626285	-1.544575
	1	6.534974	3.595379	-2.578098
	6	7.392342	4.804034	-1.014220
	1	7.467461	5.693284	-1.635608
	6	7.815047	4.846666	0.310652
	1	8.226827	5.764383	0.724155
	6	7.732776	3.708843	1.106742
	1	8.117715	3.737027	2.126783
	6	7.745704	0.191401	0.867616
	6	9.732254	-0.521571	-0.391889
	1	8.989795	-1.110820	-0.943481
	1	10.379792	-1.217376	0.168046
	6	10.552222	0.309268	-1.370775
	1	9.855971	0.892285	-1.996326
	1	11.093171	-0.372899	-2.039910
	6	11.512640	1.247705	-0.652949
	1	12.267293	0.657170	-0.108567
	1	12.060818	1.861408	-1.377310
	6	10.749242	2.126770	0.329016
	1	11.433833	2.761611	0.905860
	1	10.063227	2.795050	-0.215720
	6	9.949743	1.262952	1.288528
	1	10.631124	0.625148	1.876987
	1	9.368144	1.851976	2.001397

	7	5.068354	-2.454149	-0.944568
	7	3.055731	-1.359778	0.605394
	1	2.442329	-0.565474	0.831065
	7	4.080185	-4.153263	-2.201536
	7	3.098041	-3.567897	-0.193386
	1	3.216666	-4.627478	-2.420666
	1	4.775390	-4.074593	-2.934084
	7	1.224350	-2.765852	0.957729
	6	6.126384	-2.311373	-1.900790
	6	6.284063	-1.103867	-2.582285
	1	5.555192	-0.308943	-2.424148
	6	7.365406	-0.932918	-3.437934
	1	7.485185	0.007880	-3.972477
	6	8.292099	-1.960142	-3.614209
	1	9.133390	-1.824232	-4.290378
	6	8.136767	-3.160385	-2.927604
	1	8.859848	-3.962614	-3.058407
	6	7.057381	-3.338832	-2.063981
	1	6.938791	-4.268810	-1.506750
	6	4.125244	-3.335189	-1.136558
	6	3.209334	-4.817585	0.547589
	6	3.765413	-4.765720	1.824309
	1	4.105932	-3.807111	2.221660
	6	3.873562	-5.932256	2.572488
	1	4.303069	-5.896056	3.571226
	6	3.445965	-7.144971	2.038725
	1	3.537335	-8.058478	2.621754
	6	2.901791	-7.190827	0.758749
	1	2.565046	-8.137179	0.341717
	6	2.773973	-6.024428	0.011394
	1	2.316766	-6.060837	-0.976761
	6	2.451369	-2.504350	0.459791
	6	0.690989	-2.015990	2.101350
	1	1.471385	-1.358937	2.499144
	1	-0.127236	-1.363362	1.749784
	6	0.213469	-2.986494	3.172272
	1	1.087698	-3.532835	3.563185
	1	-0.203446	-2.409873	4.008939
	6	-0.804263	-3.972586	2.618371
	1	-1.705782	-3.420947	2.297249
	1	-1.124068	-4.679672	3.393852
	6	-0.220187	-4.713310	1.421778
	1	-0.969580	-5.369076	0.957327
	1	0.617070	-5.352783	1.742606
	6	0.273070	-3.719215	0.382020
	1	-0.591097	-3.134044	0.006522
	1	0.733946	-4.211441	-0.479789
	17	3.160294	2.805569	-2.034991
	17	3.861470	3.679737	0.686350
	17	2.206096	5.386664	-1.025657
	6	2.626981	3.723316	-0.598675
	1	1.749571	3.199423	-0.205870
	78	-5.807392	0.649770	-0.139283
	7	-3.954018	-0.157842	-0.494917
	7	-4.803000	1.974219	1.038085

	1	-5.221493	2.314485	1.899963
	7	-1.664693	0.077619	-0.886792
	7	-2.870318	1.944977	-0.288606
	1	-0.808296	0.519877	-0.513889
	1	-1.583032	-0.895393	-1.158136
	7	-3.130981	3.584007	1.342284
	6	-3.852357	-1.574331	-0.656756
	6	-4.084571	-2.411496	0.435995
	1	-4.316295	-1.965726	1.403471
	6	-4.019819	-3.791903	0.273847
	1	-4.200854	-4.443897	1.126885
	6	-3.722399	-4.341191	-0.973171
	1	-3.674379	-5.421490	-1.095115
	6	-3.491906	-3.505085	-2.061834
	1	-3.264733	-3.927229	-3.038800
	6	-3.562240	-2.121304	-1.908738
	1	-3.385892	-1.456951	-2.756362
	6	-2.856658	0.564919	-0.552312
	6	-2.477657	2.863786	-1.351545
	6	-3.492284	3.619505	-1.939076
	1	-4.523206	3.487827	-1.605049
	6	-3.177199	4.540371	-2.930686
	1	-3.965682	5.134583	-3.387619
	6	-1.857176	4.688891	-3.349264
	1	-1.612484	5.403063	-4.132225
	6	-0.854258	3.922921	-2.764602
	1	0.181338	4.030384	-3.086319
	6	-1.156642	3.015932	-1.752662
	1	-0.362129	2.450003	-1.268091
	6	-3.638266	2.494203	0.743300
	6	-3.987527	4.578463	1.996524
	1	-5.037668	4.357135	1.769376
	1	-3.851263	4.509680	3.088982
	6	-3.629160	5.972569	1.496561
	1	-3.885407	6.032405	0.425964
	1	-4.255975	6.706260	2.020586
	6	-2.148764	6.273200	1.690532
	1	-1.918503	6.291816	2.768657
	1	-1.907619	7.269846	1.302156
	6	-1.299576	5.207634	1.007269
	1	-0.230240	5.368145	1.202083
	1	-1.436089	5.255904	-0.084375
	6	-1.686247	3.826043	1.506009
	1	-1.480210	3.753099	2.588079
	1	-1.121967	3.022618	1.020957
	7	-7.667269	1.457048	0.193348
	7	-6.787554	-0.667383	-1.329356
	1	-6.308381	-1.055797	-2.137397
	7	-9.977548	1.342003	0.401843
	7	-8.842451	-0.585817	-0.185472
	1	-10.835687	0.857447	0.187087
	1	-10.034198	2.303468	0.714394
	7	-8.369150	-2.345807	-1.649595
	6	-7.708075	2.862251	0.479410
	6	-7.466551	3.777666	-0.543676

	1	-7.306705	3.412075	-1.557737
	6	-7.430698	5.136509	-0.251850
	1	-7.251451	5.855526	-1.048640
	6	-7.636621	5.580792	1.053502
	1	-7.615928	6.645581	1.275391
	6	-7.881375	4.663059	2.071155
	1	-8.050246	5.007286	3.089400
	6	-7.913381	3.298400	1.789062
	1	-8.093767	2.565976	2.577549
	6	-8.787245	0.787520	0.127969
	6	-9.397839	-1.456207	0.846625
	6	-8.495401	-2.105476	1.685870
	1	-7.424452	-1.940248	1.550951
	6	-8.972824	-2.955178	2.676871
	1	-8.273693	-3.467933	3.333699
	6	-10.343554	-3.138450	2.837572
	1	-10.716452	-3.798318	3.617392
	6	-11.238903	-2.481839	1.999010
	1	-12.309614	-2.629726	2.118260
	6	-10.767527	-1.645734	0.991514
	1	-11.474570	-1.176314	0.306999
	6	-7.961493	-1.190318	-1.095650
	6	-7.403449	-3.344246	-2.128341
	1	-6.401644	-3.081468	-1.765893
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	6	-7.797708	-4.721862	-1.612784
	1	-7.699590	-4.717110	-0.514571
	1	-7.082654	-5.459302	-2.000891
	6	-9.224597	-5.077731	-2.005421
	1	-9.295210	-5.153599	-3.102685
	1	-9.499222	-6.061382	-1.606761
	6	-10.188879	-4.009554	-1.505233
	1	-11.212229	-4.212837	-1.845368
	1	-10.207253	-4.004133	-0.404052
	6	-9.765705	-2.641492	-2.011453
	1	-9.821764	-2.616974	-3.112940
	1	-10.405096	-1.836807	-1.640508
	17	0.613361	-1.894004	-2.344701
	17	2.999261	-0.721630	-3.552850
	17	0.439778	0.675808	-3.729067
	6	1.468755	-0.354649	-2.737652
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	7	4.561631	4.690852	-1.738875
	1	3.861384	5.412667	-1.809756
	1	5.351757	4.718633	-2.371190
	7	1.529933	3.175002	1.244177
	7	3.438599	3.941361	0.143562
	6	6.387260	2.652211	-1.653134
	6	6.349678	1.599533	-2.567867

	1	5.507584	0.907818	-2.546893
	6	7.400185	1.436827	-3.463357
	1	7.370462	0.615502	-4.177473
	6	8.477159	2.322262	-3.458374
	1	9.288768	2.195845	-4.171705
	6	8.508848	3.372521	-2.545255
	1	9.346867	4.066413	-2.537790
	6	7.466662	3.537267	-1.634209
	1	7.484971	4.347191	-0.903051
	6	4.489175	3.757448	-0.775528
	6	3.508935	5.160583	0.936931
	6	3.019584	6.373026	0.463101
	1	2.527118	6.434268	-0.508262
	6	3.119107	7.510251	1.259412
	1	2.741451	8.461102	0.890351
	6	3.685454	7.426498	2.527322
	1	3.753274	8.315413	3.150130
	6	4.163146	6.207022	3.000143
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	6	4.085426	5.071741	2.202244
	1	4.461055	4.108528	2.554664
	6	2.726846	2.867682	0.707849
	6	0.557876	4.060615	0.601640
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	1	1.039771	4.564781	-0.242681
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	1	0.735764	5.745006	1.929633
	1	-0.818111	5.632981	1.090074
	6	-0.611744	4.303621	2.801874
	1	-1.008547	5.016732	3.535103
	1	-1.458194	3.672249	2.480512
	6	0.462666	3.423902	3.426728
	1	0.067150	2.855937	4.279714
	1	1.282864	4.052964	3.810413
	6	1.020278	2.440143	2.411806
	1	1.849658	1.867247	2.842841
	1	0.256269	1.716212	2.086656
	7	5.189321	-0.167052	2.168653
	7	7.261525	1.023894	0.800882
	1	7.880070	1.821533	0.680647
	7	5.990576	-2.147585	3.101881
	1	6.821695	-2.683080	3.305412
	1	5.223588	-2.191962	3.762708
	7	9.137945	-0.361747	0.572646
	7	7.107486	-1.264278	1.277661
	6	4.185394	-0.059149	3.189011
	6	4.322228	0.894236	4.198302
	1	5.217927	1.514435	4.231628
	6	3.311473	1.032614	5.143697
	1	3.418346	1.765483	5.941065
	6	2.167748	0.235942	5.075723
	1	1.383221	0.351308	5.820970
	6	2.034975	-0.705131	4.060317
	1	1.148002	-1.330249	3.973940
	6	3.046231	-0.856114	3.116311

	1	2.960284	-1.593525	2.321120
	6	6.046860	-1.147024	2.206231
	6	7.028229	-2.313069	0.270335
	6	7.183990	-3.658634	0.587486
	1	7.395597	-3.974548	1.606974
	6	7.071418	-4.615131	-0.420128
	1	7.184013	-5.667755	-0.171310
	6	6.826487	-4.229725	-1.733945
	1	6.741287	-4.982930	-2.513480
	6	6.700122	-2.879367	-2.048038
	1	6.509658	-2.566873	-3.073540
	6	6.794907	-1.921037	-1.047911
	1	6.695057	-0.861593	-1.286576
	6	7.853369	-0.136218	0.891323
	6	9.933742	-1.476517	1.113408
	1	10.716953	-1.031310	1.749133
	1	9.294598	-2.085242	1.758347
	6	10.570918	-2.289058	-0.001160
	1	9.780812	-2.791979	-0.581048
	1	11.193957	-3.074519	0.445079
	6	11.398040	-1.385231	-0.906261
	1	11.828799	-1.958069	-1.735833
	1	12.246801	-0.973373	-0.336472
	6	10.541366	-0.243749	-1.435275
	1	11.136530	0.448494	-2.045327
	1	9.742900	-0.640639	-2.083803
	6	9.892590	0.544028	-0.303526
	1	9.214274	1.298326	-0.721935
	1	10.655092	1.066221	0.297162
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	6	2.047879	0.221543	-2.153769
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	35	2.628114	1.980633	-2.726591
	35	0.981881	-0.626885	-3.512212
	35	3.541726	-0.897624	-1.630094
	6	2.044890	-3.152161	0.476735
	1	1.834253	-2.075892	0.447935
	35	1.304086	-3.912294	-1.126982
	35	1.210201	-3.848399	2.076741
	35	3.956521	-3.455489	0.561406
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	7	-5.265484	-1.593325	1.014659
	1	-5.627113	-1.969205	1.886471
	7	-2.379322	0.488257	-1.009922
	1	-1.491544	0.110086	-0.638232
	1	-2.393820	1.468991	-1.262013
	7	-3.583502	-3.215123	1.131432
	7	-3.417461	-1.482993	-0.419192
	6	-4.759298	1.905091	-0.952448
	6	-5.028071	2.797648	0.086124
	1	-5.134163	2.411974	1.100089
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	6	-1.770552	-2.403509	-2.038531
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	6	-2.538476	-3.930713	-3.750613
	1	-2.324086	-4.563116	-4.609277
	6	-3.838155	-3.844081	-3.257043
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	6	-4.110235	-3.026138	-2.167571
	1	-5.124947	-2.931174	-1.774224
	6	-4.120283	-2.098107	0.619222
	6	-2.143330	-3.532992	1.115534
	1	-1.816963	-3.562546	2.169137
	1	-1.589560	-2.714047	0.646214
	6	-1.893773	-4.884828	0.464913
	1	-2.158338	-4.834482	-0.602998
	1	-0.820854	-5.106184	0.516796
	6	-2.708552	-5.966724	1.159977
	1	-2.563281	-6.937658	0.671078
	1	-2.353331	-6.082283	2.197918
	6	-4.182043	-5.589001	1.168643
	1	-4.785181	-6.332752	1.706534
	1	-4.561983	-5.546482	0.134136
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	1	-5.460477	-3.951633	1.748375
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	7	-7.624366	0.752510	-1.334723
	1	-7.248430	1.100980	-2.212904
	7	-10.512132	-1.344779	0.752979
	1	-11.405036	-0.951316	0.496027
	1	-10.493117	-2.308019	1.066063
	7	-9.379244	2.268550	-1.618525
	7	-9.535268	0.619667	0.025669
	6	-8.157570	-2.732944	0.684297
	6	-7.944141	-3.650597	-0.343618
	1	-7.858070	-3.290739	-1.368718
	6	-7.853138	-5.004934	-0.043114
	1	-7.699947	-5.726721	-0.842797
	6	-7.969589	-5.441648	1.275667
	1	-7.909336	-6.503614	1.503812
	6	-8.175426	-4.521046	2.298927
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	1	-8.426746	-2.426713	2.799230
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	6	-9.346657	3.221240	2.676375
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	1	-7.947462	2.226861	1.357469
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	1	-12.353353	3.926667	-1.489390
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	1	-10.769337	5.898134	-1.618895
	1	-10.719987	4.906902	-3.074622
	6	-8.990946	4.675555	-1.806910
	1	-8.398516	5.430316	-2.340958
	1	-8.733773	4.754303	-0.737296
	6	-8.575825	3.294197	-2.295474
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	1	-5.229739	3.955080	2.378461
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	7	-3.401161	5.224262	1.261954
	7	1.487561	5.245734	1.155583
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	1	0.324385	1.782762	1.667403
	7	3.798593	5.505596	1.133116
	1	3.695828	6.459215	0.804155
	1	4.668997	5.289786	1.602650
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	6	1.415540	8.800977	0.033360
	1	1.589396	9.794350	0.442076
	6	1.033226	8.646878	-1.296489
	1	0.914667	9.521031	-1.932804
	6	0.808509	7.371687	-1.813148
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	6	0.971794	6.248343	-1.008933
	1	0.792147	5.243336	-1.394860
	6	2.681600	4.839332	1.499490
	6	3.856950	3.714864	3.325171
	6	3.351622	3.860550	4.614095
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	1	6.287301	3.889022	6.320731
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	1	7.175421	3.632446	4.020781
	6	5.228482	3.617618	3.104470
	1	5.617174	3.456986	2.096660
	6	2.027851	2.557224	2.203389
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	6	3.815051	0.867613	1.994809
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	6	-8.700192	-2.497762	-3.719673
	1	-7.672492	-2.494216	-3.349298
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	6	-4.465685	0.115534	-2.908262
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	6	-1.914820	2.480216	4.416014
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	1	-4.889880	10.199495	-1.210150
	6	-3.016956	9.239041	-0.706978
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	1	-1.376896	8.130024	0.153302
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	7	3.733362	5.582790	1.660339
	1	3.649785	6.567647	1.433298
	1	4.540667	5.321377	2.211832
	7	2.751460	3.613137	2.399499
	7	2.432472	1.306424	2.479720
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	6	1.376398	8.978019	0.768044
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	1	5.517924	3.417030	6.937245
	6	3.570072	3.487450	6.022109
	1	3.038814	3.486517	6.971118
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	1	1.766083	3.589604	4.828082
	6	1.908684	2.514080	2.159330
	6	3.730843	0.859313	1.970845
	1	3.522464	0.234339	1.080070
	1	4.304149	1.730754	1.630151
	6	4.483778	0.036154	2.999073
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	6	3.614172	-1.114829	3.489858
	1	3.412127	-1.803450	2.648665

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	6	2.295461	-0.572724	4.026091
	1	2.490083	0.077470	4.893963
	1	1.639470	-1.384141	4.370489
	6	1.561317	0.228323	2.962665
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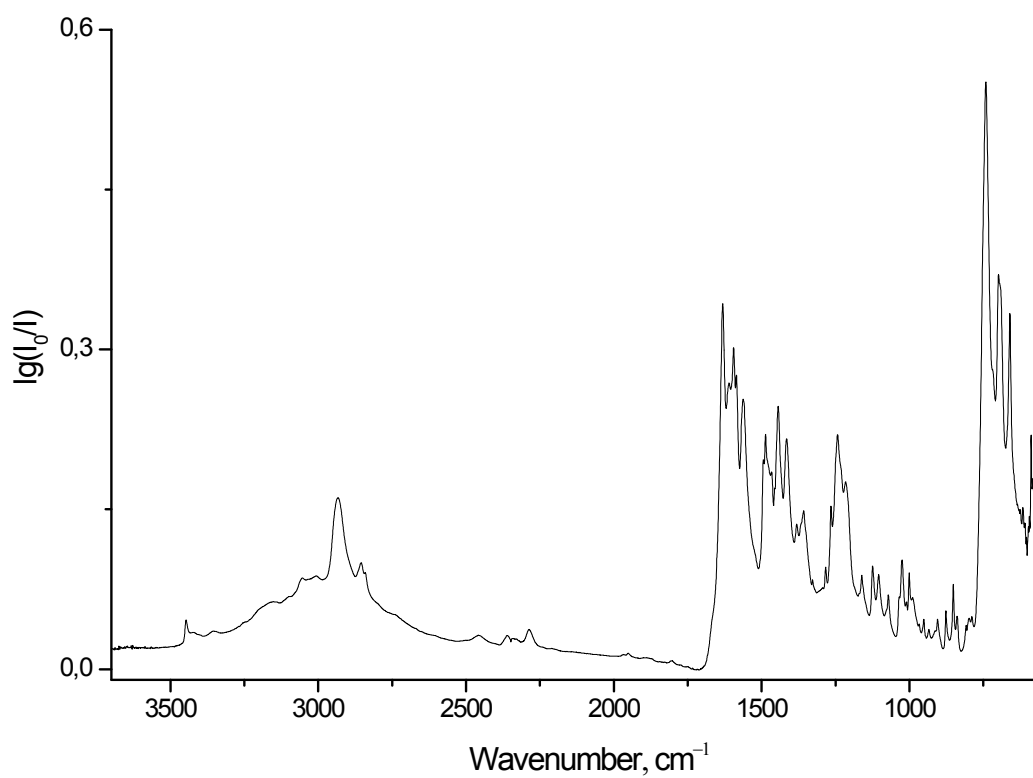


Figure S17. ATR IR spectrum of $[1][\text{Cl}_2(\text{CHCl}_3)_4]$.

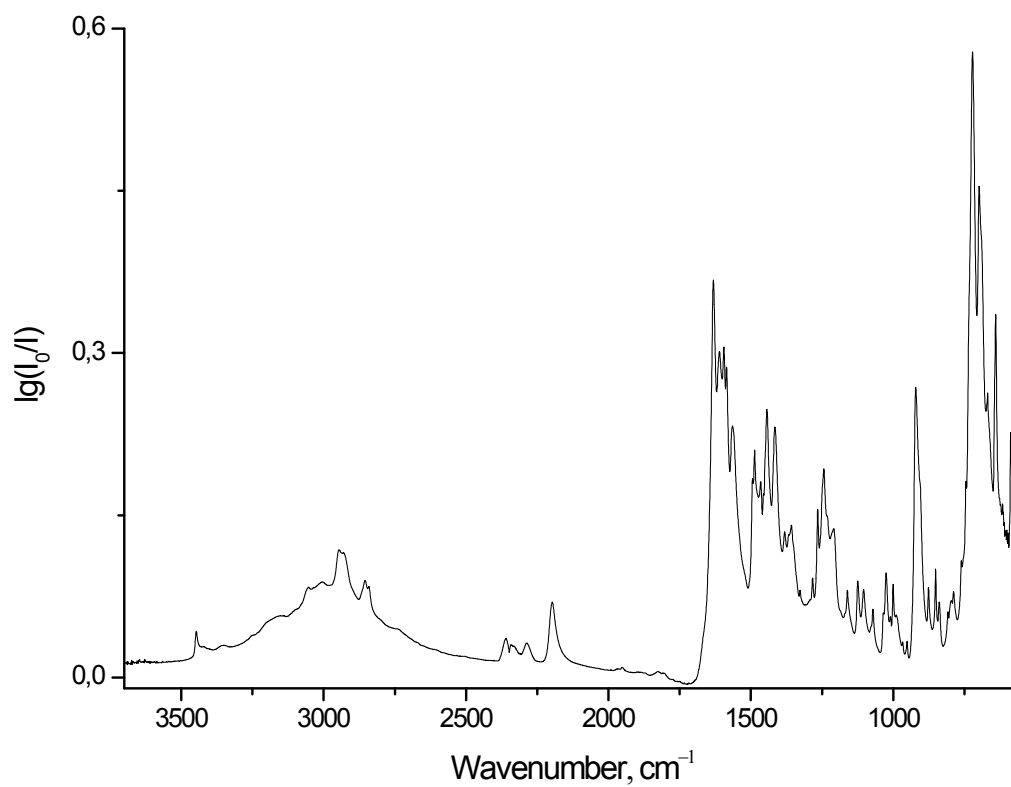


Figure S18. ATR IR spectrum of $[1][\text{Cl}_2(\text{CDCl}_3)_4]$.

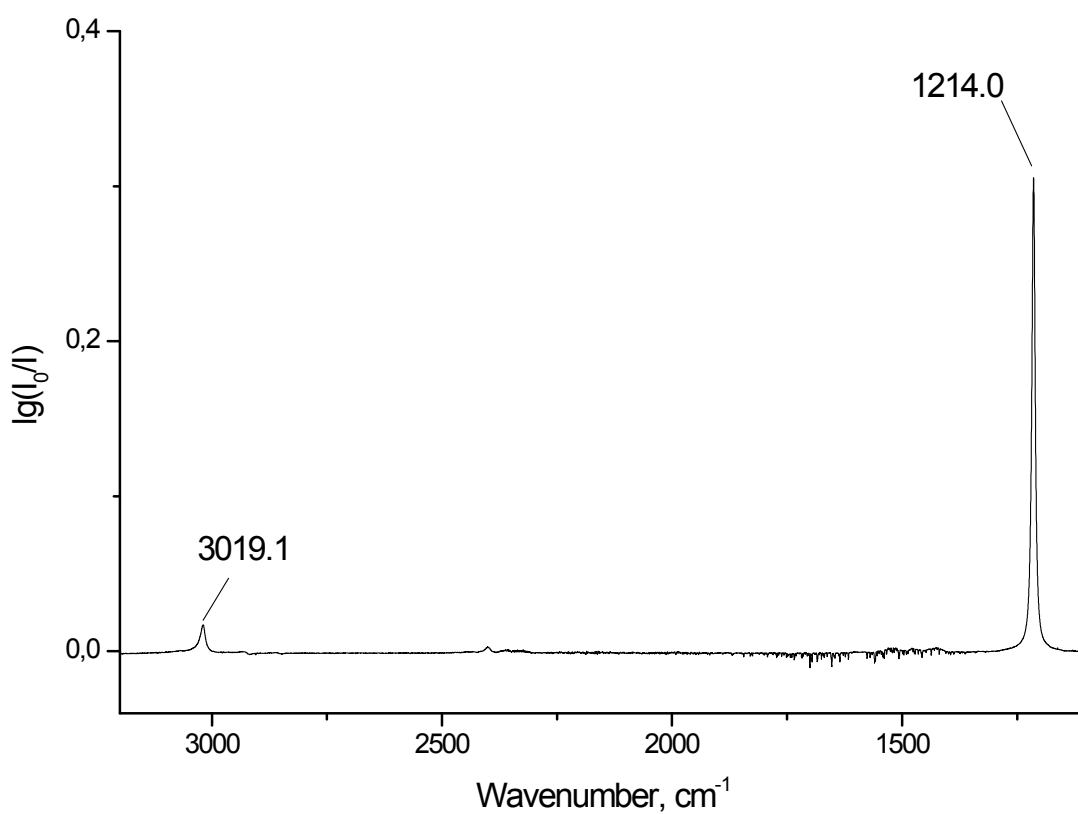


Figure S19. ATR IR spectrum of neat CHCl_3 .

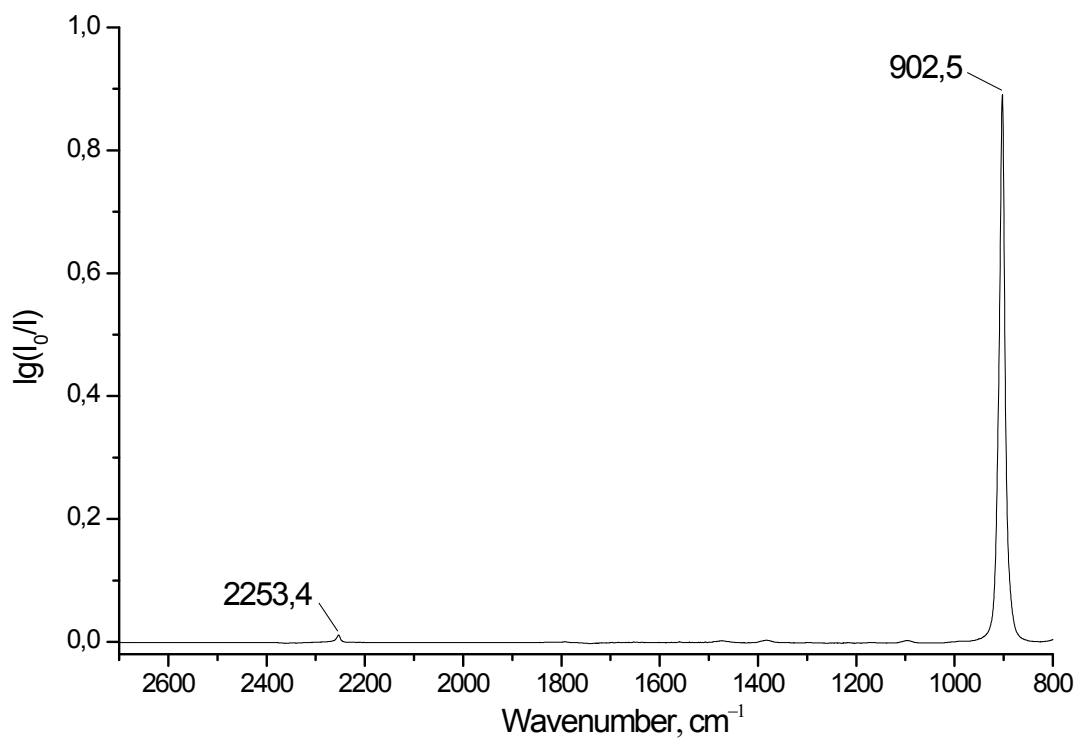


Figure S20. ATR IR spectrum of neat CDCl_3 .

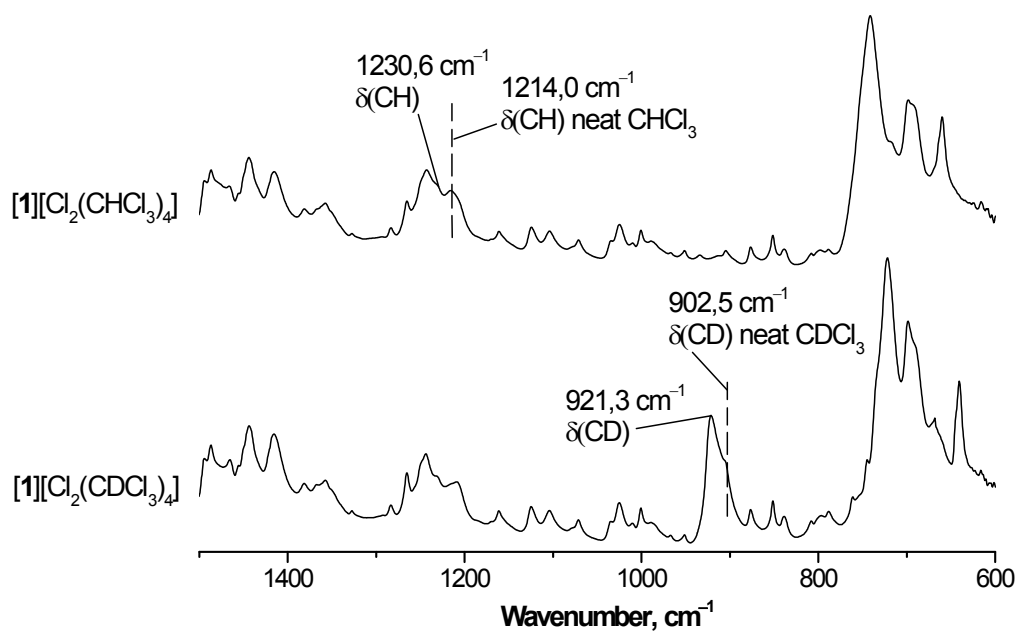


Figure S21. ATR FTIR spectra of $[1][Cl_2(CHCl_3)_4]$ and $[1][Cl_2(CDCl_3)_4]$ in the range 1500–600 cm^{-1} .

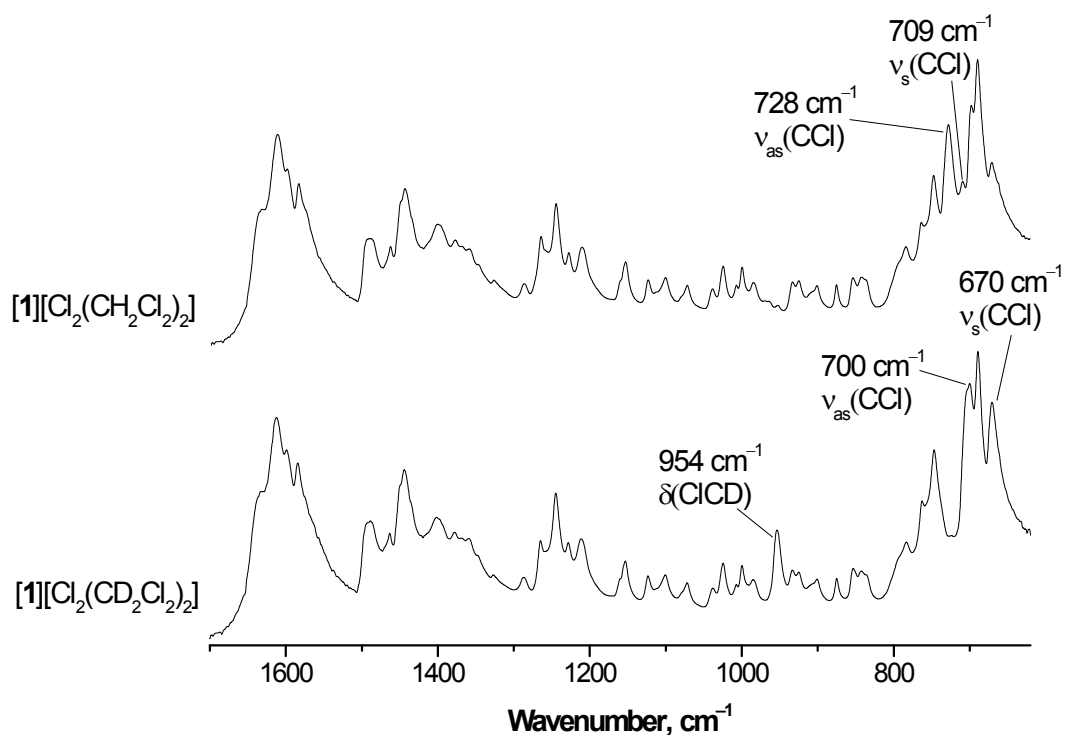


Figure S22. ATR FTIR spectra of $[1][Cl_2(CH_2Cl_2)_2]$ and $[1][Cl_2(CD_2Cl_2)_2]$ in the range 1700–600 cm^{-1} .

Table S5. Crystallographic parameters of C–H···Cl[−] hydrogen bonds formed between chloride anion and co-crystallized 1,2-dichloroethane molecules (according to CCDC database).

Compound	$d(\text{H}\cdots\text{A}), \text{Å}$	$d(\text{D}\cdots\text{A}), \text{Å}$	$\angle(\text{DHA}), \text{deg}$	Ref.
$[(\text{C}_{65}\text{H}_{54}\text{OsO}_3\text{P}_3\text{S})\text{Cl}]\cdot 2\frac{1}{2}\text{C}_2\text{H}_4\text{Cl}_2\cdot \text{H}_2\text{O}$	2.81	3.581	137.2	1
$[(\text{C}_{30}\text{H}_{42}\text{PrU}_2\text{S})\text{Cl}]\cdot \text{C}_2\text{H}_4\text{Cl}_2$	2.81	3.577	136.2	2
$[(\text{NBu}_4)(\text{C}_{11}\text{H}_9\text{BF}_2\text{N}_2\text{O}_2)\text{Cl}]\cdot \frac{1}{2}\text{C}_2\text{H}_4\text{Cl}_2$	2.70	3.682	169.7	3
$[(\text{C}_6\text{H}_{12}\text{Cl}_3\text{N}_4\text{S})\text{Cl}]\cdot \text{C}_2\text{H}_4\text{Cl}_2$	2.59	3.506	152.8	4
$[(\text{C}_{44}\text{H}_{52}\text{N}_4)\text{Cl}_2]\cdot 4\text{C}_2\text{H}_4\text{Cl}_2$	2.74	3.531	139.2	5
$[(\text{C}_{59}\text{H}_{48}\text{OsCl}_2\text{P}_3)\text{Cl}]\cdot 2\text{C}_2\text{H}_4\text{Cl}_2\cdot 2\text{H}_2\text{O}$	2.90	3.596	128.1	6
	2.77	3.406	122.9	
$[(\text{C}_{91}\text{H}_{77}\text{RuClP}_6)\text{Cl}_2]\cdot 2\text{C}_2\text{H}_4\text{Cl}_2\cdot 3\text{H}_2\text{O}$	2.72	3.658	162.0	7
$[(\text{C}_{56}\text{H}_{58}\text{PtP}_6)\text{Cl}_2]\cdot \text{C}_2\text{H}_4\text{Cl}_2$	2.67	3.412	133.7	8
$[(\text{C}_6\text{HCl}_9\text{N}_{10}\text{P}_3)\text{Cl}]\cdot \text{C}_2\text{H}_4\text{Cl}_2$	2.70	3.471	135.2	9

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