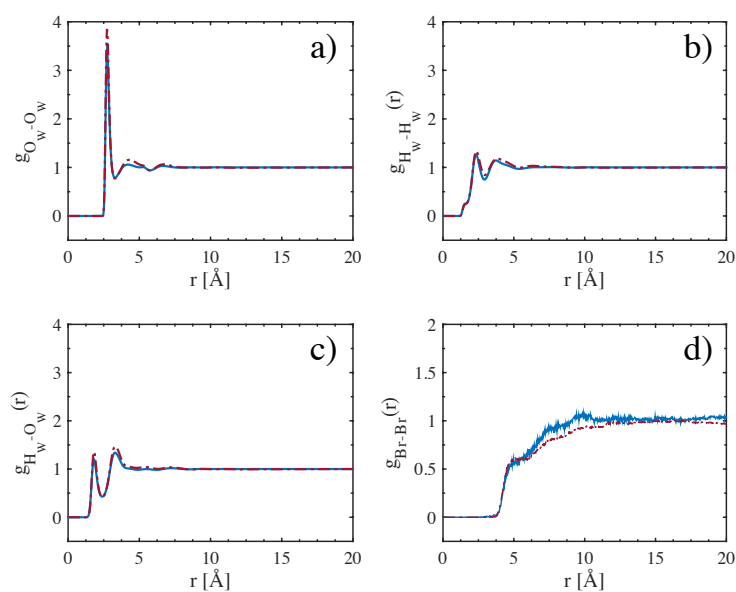


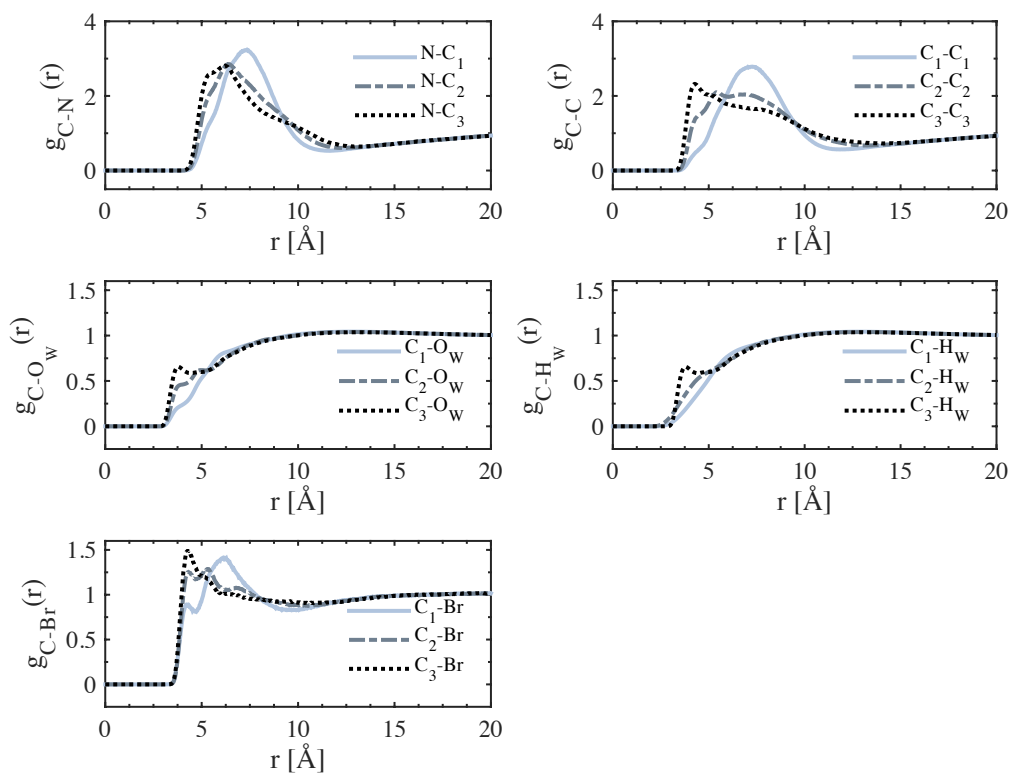
## Supporting Information

### A neutron scattering and modelling study of aqueous solutions of tetramethylammonium and tetrapropylammonium bromide

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**Figure S1:** Radial distributions of the bromide and water for both the TMABr (solid blue line) and TPABr (dot-dashed red line) in water.



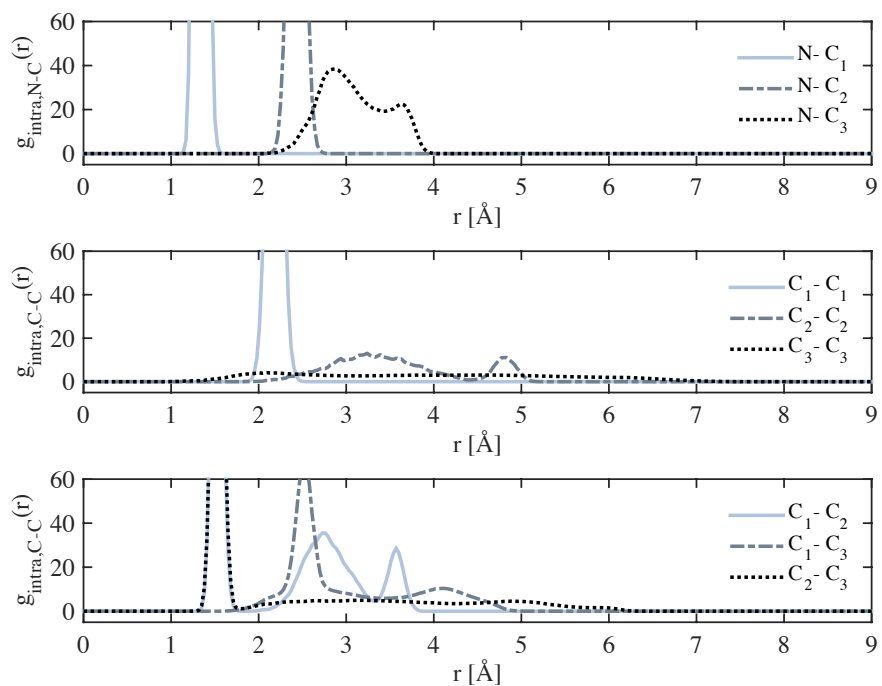
**Figure S2:** Radial distributions of the three different carbons in TPABr in water.

**Table S1:** Coordination number for 0.4 M TMABr in water, between  $R_{\min}$  and  $R_{\max}$ .

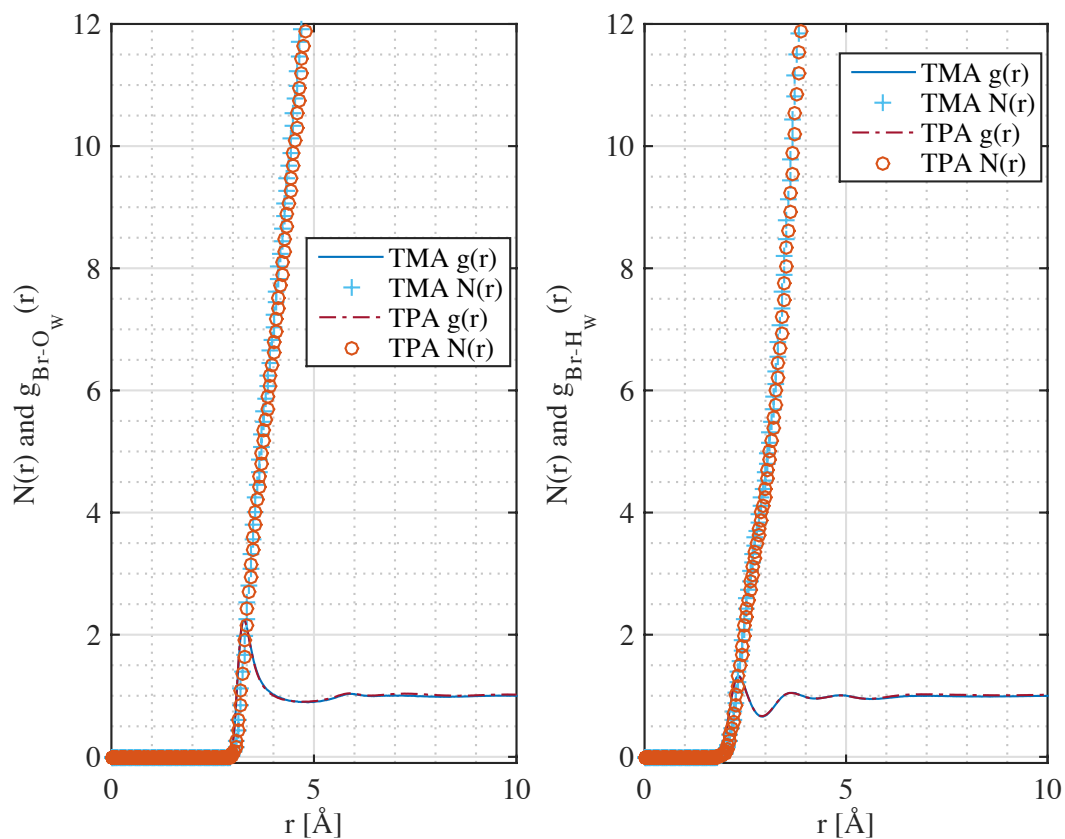
pair-correlation	$R_{\min}$	$R_{\max}$	Number of atoms	Standard error
N - N	4.5	7.5	0.94	$\pm 1.2$
	4.5	10.5	1.9	$\pm 1.8$
N - C <sub>1</sub>	3.6	6.0	1.4	$\pm 2.0$
	3.6	8.5	4.8	$\pm 5.3$
N - Br	3.4	6.5	0.48	$\pm 0.65$
N - O <sub>W</sub>	3.0	6.0	23	$\pm 3.8$
N - H <sub>W</sub>	3.0	6.4	7.3	$\pm 16$
C <sub>1</sub> - C <sub>1</sub>	3.0	4.7	0.36	$\pm 0.80$
	3.0	9.1	5.5	$\pm 5.8$
	3.0	12.5	11	$\pm 9.0$
C <sub>1</sub> - Br	3.0	5.0	0.19	$\pm 0.41$
Br - O <sub>W</sub>	2.7	3.3	1.8	$\pm 1.0$
Br - O <sub>W</sub>	2.7	4.6	11	$\pm 2.2$
Br - H <sub>W</sub>	1.5	3.0	4.6	$\pm 1.3$
Br - Br	3.5	5.9	0.079	$\pm 0.28$
Br - Br	3.5	10.5	1.0	$\pm 0.99$
O <sub>W</sub> - O <sub>W</sub>	2.3	3.3	4.3	$\pm 1.2$
H <sub>W</sub> - H <sub>W</sub>	1.2	3.0	6.3	$\pm 1.6$
O <sub>W</sub> - H <sub>W</sub>	1.3	2.4	2.0	$\pm 0.87$

**Table S2: Coordination numbers for 0.4 M TPABr in water, between  $R_{\min}$  and  $R_{\max}$ .**

pair-correlation	$R_{\min}$	$R_{\max}$	Number of atoms	Standard error
N - N	5.0	10.6	2.0	$\pm 1.3$
N - C <sub>3</sub>	3.9	8.3	4.1	$\pm 2.8$
	3.9	12.6	9.6	$\pm 5.7$
N - C <sub>1</sub>	3.9	11.3	8.3	$\pm 5.4$
N - C <sub>2</sub>	3.9	12.0	8.9	$\pm 5.5$
N - Br	3.9	5.8	0.15	$\pm 0.38$
	3.9	8.6	0.66	$\pm 0.76$
N - O <sub>w</sub>	3.5	7.4	31	$\pm 6.1$
	3.5	10.4	0.00	$\pm 0.00$
N - H <sub>w</sub>	3.2	7.9	0.52	$\pm 4.8$
	3.2	9.5	0.00	$\pm 0.00$
C <sub>3</sub> - C <sub>3</sub>	3.1	5.9	3.1	$\pm 1.6$
	3.1	13.8	13.9	$\pm 6.1$
C <sub>1</sub> - C <sub>1</sub>	3.1	11.7	8.7	$\pm 5.5$
C <sub>2</sub> - C <sub>2</sub>	3.1	6.0	3.6	$\pm 1.4$
	3.1	13.3	13	$\pm 6.0$
C <sub>3</sub> - Br	3.2	4.9	0.085	$\pm 0.29$
	3.2	6.0	0.20	$\pm 0.43$
C <sub>1</sub> - Br	3.2	4.7	0.040	$\pm 0.20$
	3.2	9.5	0.85	$\pm 0.86$
C <sub>2</sub> - Br	3.2	4.8	0.066	$\pm 0.25$
	3.2	6.2	0.22	$\pm 0.45$
Br - H <sub>w</sub>	1.5	3.0	4.3	$\pm 1.3$
Br - Br	5.7	2.0	0.0	$\pm 0.00$
	3.5	8.5	0.39	$\pm 0.60$
	3.5	10.8	0.98	$\pm 0.92$
O <sub>w</sub> - O <sub>w</sub>	2.3	3.3	4.4	$\pm 1.2$
H <sub>w</sub> - H <sub>w</sub>	1.2	3.0	6.44	$\pm 1.7$
O <sub>w</sub> - H <sub>w</sub>	1.3	2.4	1.98	$\pm 0.88$
Br - O <sub>w</sub>	2.7	3.3	1.76	$\pm 1.0$
Br - O <sub>w</sub>	2.7	4.6	10	$\pm 2.4$



**Figure S3:** Internal partial radial distribution functions for nitrogen and carbons within the same TPA<sup>+</sup> molecule.



**Figure S4:** RDFs for bromide-water and the corresponding coordination number,  $N(r)$ .