Electronic Supplementary Information (ESI) for

Critical size of hydrogen-bonded alcohol clusters as effective Brønsted base in solutions

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Contents

- **1. Fluorescence lifetimes**
- 2. CHELPG charges of $AH^+(S_0/S_1)$ and $A(S_0/S_1)$
- 3. MD simulation data
- 4. PES of scanned A(S₀/S₁)…H⁺ bond alteration
- 5. Nature of solvated proton in MeOH dimers

1. Fluorescence lifetimes

Base	[Base] (M)	$\lambda_{\mathrm{mon}}^{a}$ (nm)	rise time (ns)	decay time (ns)	χ^{2}
	0.00 (0.0%) ^b	445	_c	$3.2 (7\%)^d + 9.1 (93\%)$	1.067
MeOH	0.60 (2.4%)	445	-	0.5 (11%) + 8.9 (89%)	1.173
	1.18 (4.8%)	445	-	0.3 (16%) + 7.3 (84%)	1.235
	1.72 (7.0%)	445	-	0.3 (17%) + 6.0 (83%)	1.245
		445	-	0.3 (16%) + 4.8 (84%)	1.211
	2.25 (9.1%)	580	1.6 (1%) + 4.7 (88%)	6.4 (100%)	1.581
	2.75 (11.1%)	445	-	0.4 (13%) + 3.8 (87%)	1.228
EtOH	0.62 (3.6 %)	445	-	0.7 (8%) + 8.1 (92%)	1.106
	0.82 (4.8%)	445	-	0.8 (4%) + 7.5 (96%)	1.222
	1.19 (7.0%)	445	-	0.7 (6%) + 6.0 (94%)	1.094
	1.56 (9.1%)	445	-	0.7 (8%) + 4.7 (92%)	1.084
		580	0.5 (1%) + 4.5 (88%)	6.5 (100%)	1.441
	1.90 (11.1%)	445	-	0.8 (8%) + 4.0 (92%)	1.143
<i>n</i> -PrOH	0.64 (4.8%)	445	-	0.3 (20%) + 8.0 (80%)	1.152
	0.93 (7.0%)	445	-	0.4 (18%) + 6.9 (82%)	1.269
	1.22 (9.1%)	445	-	0.3 (19%) + 5.9 (81%)	1.323
		445	-	0.3 (16%) + 5.1 (84%)	1.326
	1.49 (11.1%)	580	0.9 (1%) + 4.8 (88%)	6.4 (100%)	1.543
	1.74 (13.0%)	445	-	0.3 (17%) + 4.3 (83%)	1.392

Table S1. Fluorescence time constants observed from TCSPC measurements according to the concentration of bases in MeCN

i-PrOH	0.62 (4.8%)	445	-	0.7 (12%) + 8.0 (88%)	1.070
	0.91 (7.0%)	445	-	0.7 (22%) + 6.9 (78%)	1.339
	1.19 (9.1%)	445	-	0.7 (9%) + 5.9 (91%)	1.162
		445	-	0.6 (10%) + 5.1 (90%)	0.943
	1.45 (11.1%)	580	0.6 (1%) + 4.9 (90%)	6.4 (100%)	1.482
	1.70 (13.0%)	445	-	0.7 (10%) + 4.4 (90%)	1.118
<i>n-</i> BuOH	0.52 (4.8%)	445	-	0.9 (7%) + 8.5 (93%)	1.147
	0.76 (7.0%)	445	-	0.6 (9%) + 7.7 (91%)	1.153
	0.99 (9.1%)	445	-	0.6 (12%) + 6.9 (88%)	1.076
	1.21 (11.1%)	445	-	0.9 (8%) + 6.2 (92%)	1.155
	1.43 (13.0%)	445 580	0.9 (1%) + 5.6 (94%)	0.7 (9%) + 5.5 (91%) 6.3 (100%)	1.039 1.597
t-BuOH	0.50 (4.8%)	445	-	1.2 (9%) + 8.8 (91%)	1.079
	0.73 (7.0%)	445	-	0.7 (10%) + 8.3 (90%)	1.123
	0.95 (9.1%)	445	-	0.8 (13%) + 7.7 (87%)	1.091
	1.16 (11.1%)	445	-	0.7 (10%) + 7.2 (90%)	1.129
		445	-	0.6 (19%) + 6.6 (81%)	1.093
	1.36 (13.0%)	580	0.8 (1%) + 6.0 (93%)	6.7 (100%)	1.470
DMSO	0.34 (2.4%)	445	-	0.4 (20%) + 3.2 (80%)	1.006
	0.67 (4.8%)	445	-	0.4 (14%) + 2.0 (86%)	0.964
	0.98 (7.0%)	445	-	0.4 (10%) + 1.4 (90%)	1.050
	1.29 (9.1%)	445	-	0.4 (14%) + 1.1 (86%)	1.007
		580	1.1 (8%)	6.0 (100%)	1.043
	1.60 (11.1%)	445	-	0.5 (24%) + 0.9 (76%)	1.130

^{*a*}Monitored wavelength of fluorescence. ^{*b*}Volume fraction of each base in MeCN. ^{*c*}Instant. ^{*d*}Negative and positive values of initial fractional amplitudes indicate those for rise and decay components, respectively.



Figure S1. Fluorescence kinetic profiles measured at the various concentrations of DMSO in MeCN. The contents of DMSO are given in molar concentration (M) as well as in volume fraction (% v/v) in the panel. Samples were excited at 375 nm and their fluorescence kinetic profiles were monitored at 445 nm. Solid lines are the best-fitted curves to extract kinetic constants, shown in Table S1.



Figure S2. Fluorescence dynamic with MeOD in the MeCN solvent. The sample was excited at 375 nm and its fluorescence kinetic profiles were monitored at 445 and 580 nm. Blue and red lines are the best-fitted curves to extract kinetic constants; at [MeOD] = 2.24 M, $I(t) = 0.157 e^{-t/(0.3 \text{ ns})} + 0.843 e^{-t/(5.4 \text{ ns})}$ at 445 nm while $I(t) = -0.002 e^{-t/(1.0 \text{ ns})} - 0.890 e^{-t/(4.9 \text{ ns})} + e^{-t/(7.1 \text{ ns})}$ at 580 nm.



2. CHELPG charges of $AH^+(S_0/S_1)$ and $A(S_0/S_1)$

Geometries of ground and excited states are optimized at B3LYP/6-31+G* and TD-B3LYP/6-31+G* respectively to calculate the CHELPG values.

3. MD simulation data



Figure S3. (Left column) Radial distribution functions of O and H atoms of methanol molecules with respect to the hydroxyl H atom of AH⁺. (Right column) Radial distribution functions of H and O atoms of methanol molecules with respect to the hydroxyl O atom of AH⁺.

4. PES of scanned $A(S_0/S_1)\cdots H^+$ bond alteration



Figure S4. Potential energy surfaces (PES) of scanned A···H⁺ bond alteration in the ground and excited state with the assistance of a methanol dimer (blue dots correspond to the ground states and red dots correspond to the excited states). Bond lengths are given in Å.

5. Nature of a solvated proton in MeOH dimers

Figure S5. Nature of solvated proton in MeOH dimers surrounded by (a) three MeCN molecules (b) one MeOH and two MeCN molecules (c) two MeOH molecules.