Supporting Information

MnO nanoparticles embedded in carbon matrix as high performance lithium-ion battery anodes: Preparation, Microstructure and Electrochemistry

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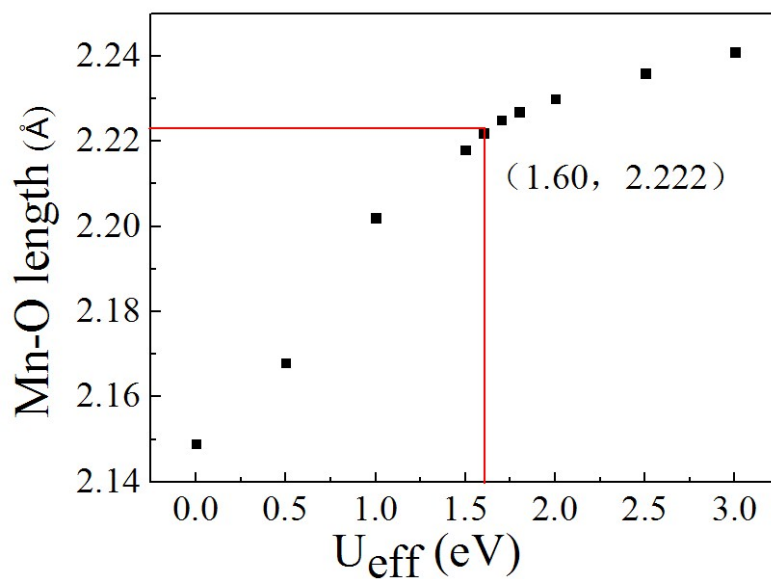Fax: 86-551-63601592

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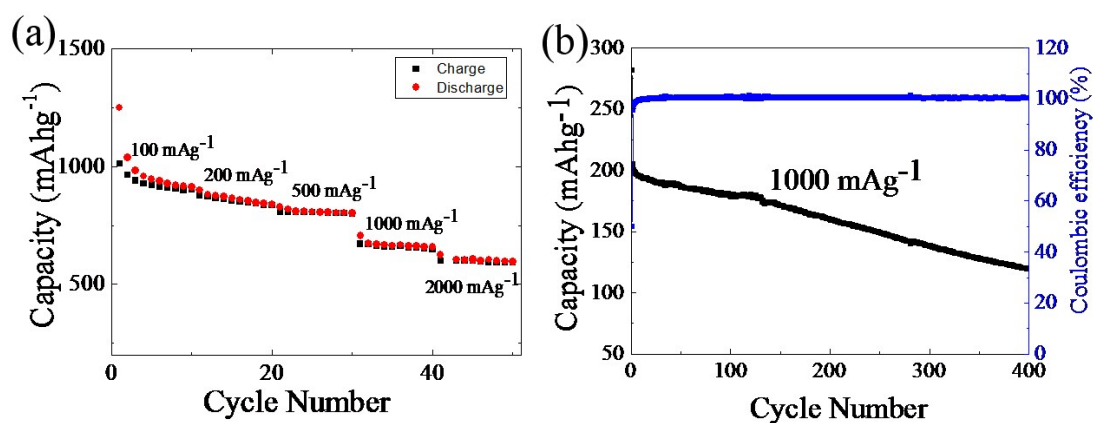Fig. S1 The relationship between Hubbard U value and bond length of Mn−O.

Fig. S2 (a) the rate perfprmance of MnO/C, (b) the discharge capacity and coulombic efficiency (%) of MnO at 1000 mAg^{-1}.

Table S1 Comparison of electrochemical performance of MnO/C with other MnO-based electrodes.

Samples	Current dencity	Capacity	cycle	Ref.
	(mA g^{-1})	(mAh g^{-1})	number	
This work	500	792	50	

	1000	725	400	
MnO@C core–shell	200	770	30	*J. Mater. Chem.*, 2012, 22, 17864. [1]
hollow porous MnO/C	100	702.2	50	*ACS Nano*, 2013, **7**, 7083.[2]
	3000	234.7		
MnO on carbon fibers	1000	575	200	*Sci. Rep*., 2014, **4**, 4229.[3]
carbon/MnO disks	1000	534.6	250	*J. Mater. Chem.*, 2012, **22**, 19190.[4]
MnO/rGO	200	750	100	*J. Mater. Chem. A*, 2015, **3**, 297.[5]
Cu−MnO@C	200	629	100	*J. Phys. Chem. C,* 2014, **118**, 17452.[6]
MnO@C	100	800	80	*Sci. Rep*., 2013, **3**, 2639.[7]
MnO@C nanowires	500	801	200	*Chem-eur. J*., 2013, **19**, 11310.[8]
porous MnO nanoflakes	2460	376.4	55	*J. Mater. Chem.*, 2012, **22**, 9189.[9]

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