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## **Electronic Supplementary Information (ESI)**

## Direct planting of ultrafine $MoO_{2+\delta}$ nanoparticles in carbon nanofibers by electrospinning: self-supported mats as binder-free and long-life anodes for lithium-ion batteries

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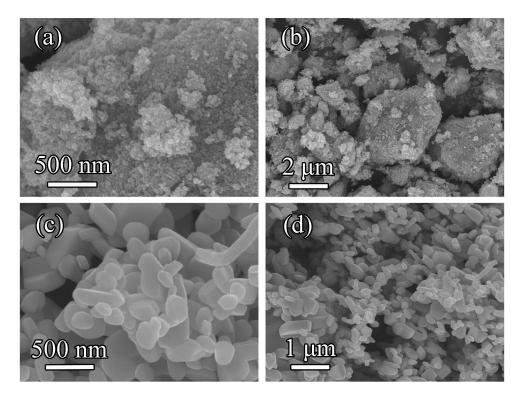
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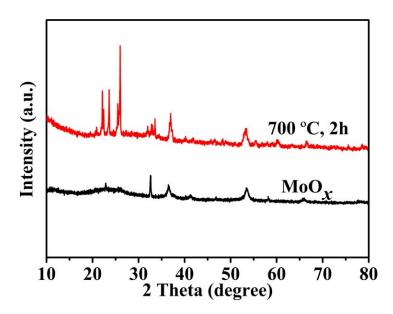
E-mail address: huxl@mail.hust.edu.cn



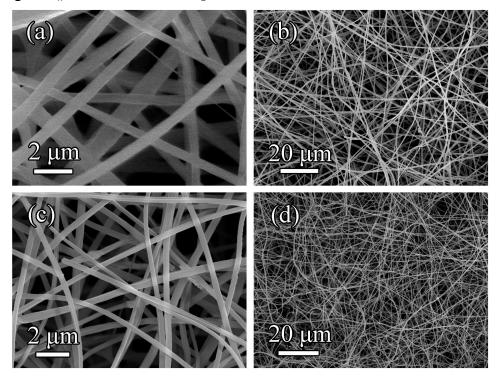
Fig. S1 Photograph of the fresh binder-free C/MoO\_{2+\delta} electrode.



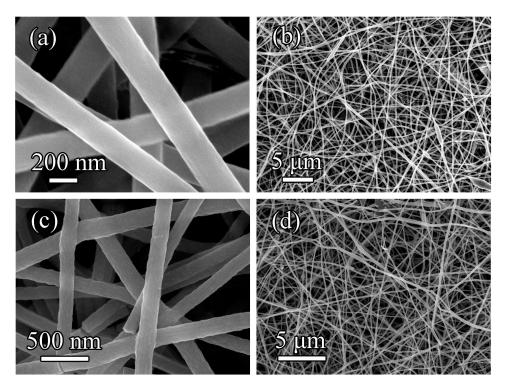
**Fig. S2** SEM images of (a, b)  $MoO_x$  nanoparticles; (c, d) compared particles prepared by heating  $MoO_x$  nanoparticles at 700 °C for 2 h in N<sub>2</sub>.



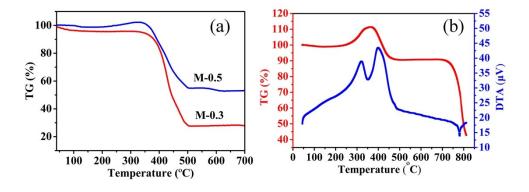
**Fig. S3** XRD patterns of  $MoO_x$  nanoparticles and compared sample prepared by heating  $MoO_x$  at 700 °C for 2 h in N<sub>2</sub>.



**Fig. S4** M-0.3: SEM images of (a,b) electrospun precursor and (c,d) the obtained nanofibers by treating the precursor at 700  $^{\circ}$ C for 2 h in N<sub>2</sub>.



**Fig. S5** M-0.5: SEM images of (a,b) electrospun precursor and (c,d) the obtained nanofibers by treating precursor at 700 °C for 2 h in  $N_2$ .



**Fig. S6** (a) TG curves of M-0.5 and M-0.3 nanofibers synthesized by treating precursors at 700 °C for 2 h in N<sub>2</sub>; (b) TG/DTA curves of the as-prepared C/MoO<sub>2+ $\delta$ </sub> composite nanofibers. (The calculated carbon contents in sample M-0.5 and M-0.3 is 52.5 and 73.2 wt%, respectively). As shown in Fig. S6 (b), the weight change between 200 °C and 600 °C is owing to both the oxidation of MoO<sub>2+ $\delta$ </sub> and the combustion of carbon. The two exothermic peaks between 260-350 °C and 350-500 °C in the DTA profile can be assigned to the oxidation of MoO<sub>2+ $\delta$ </sub> and the combustion of the carbon, respectively. While the endothermic peak between 780-820 °C corresponds to the eliquation and volatilization of MoO<sub>3</sub>.

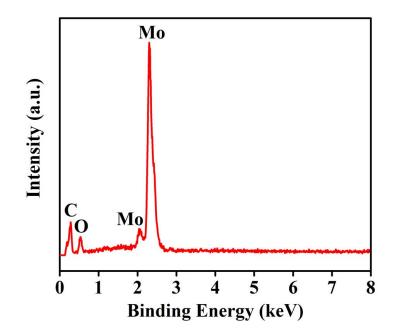
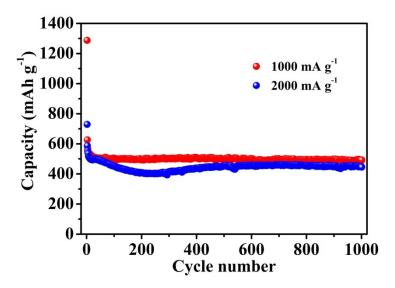


Fig. S7 EDX spectrum of prepared C/MoO<sub>2+ $\delta$ </sub> nanofiber composites.



**Fig. S8** Cycling properties of the binder-free C/MoO<sub>2+ $\delta$ </sub> electrode in the potential range of 3–0.01 V vs. Li/Li<sup>+</sup> at different current densities of 1000 and 2000 mA g<sup>-1</sup>.

Sample	Current density	Cycle number	Specific capacity	References
	(mA/g)		(mAh/g)	
Carbon coated MoO <sub>2</sub>	50	50	762.7	6
Hierarchical MoO <sub>2</sub> nanoarchitecture	200	20	719.1	7
MoO <sub>2</sub> graphene thin film	47.8	100	675.9	9
MoO <sub>2</sub> /MoC heteronanotubes	200	140	790	10
MoO <sub>3-x</sub> nanowire arrays	50	20	630	12
MoO <sub>2</sub> /graphene	500	70	848.6	16
MoO <sub>2</sub> /C	50	350	734	18
Carbon coated MoO <sub>2</sub>	100	30	617.2	19
nanobelts				
rGO wrapped MoO <sub>2</sub> porous nanobelts	5000	1900	420	20
MoO <sub>2</sub> /C nanosheets	500	100	1051	21
3D C/MoO <sub>2+δ</sub> networks	200 2000	250 1000	876.9 447.9	Current work

Table S1. Comparison of battery performance of  $MoO_x/C$  as anode in Li-ion batteries

The references are from the main text.