

Electronic Supplementary Information

### Efficiency and Stability of Spectral Sensitization of Boron-Doped-Diamond Electrode through Covalent Anchoring of a Donor-Acceptor Organic Chromophore (P1)

Hana Krysova<sup>a</sup>, Jan Barton<sup>b,c</sup>, Vaclav Petrak<sup>d,e</sup>, Radek Jurok<sup>f</sup>, Martin Kuchar<sup>f</sup>, Petr Cigler<sup>b\*</sup> and Ladislav Kavan<sup>a\*</sup>

<sup>a</sup> J. Heyrovský Institute of Physical Chemistry, v.v.i. Academy of Sciences of the Czech Republic, Dolejskova 3, 18223 Prague 8, Czech Republic

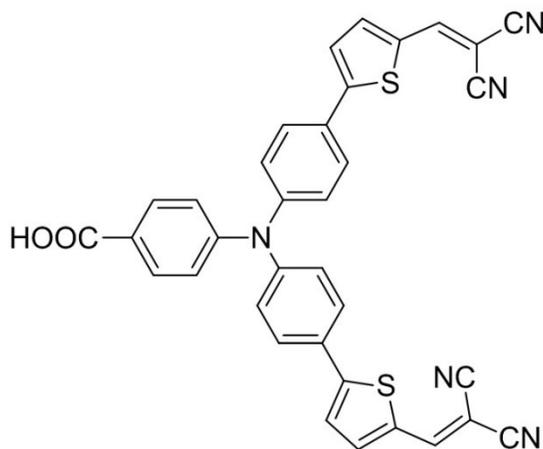
<sup>b</sup> Institute of Organic Chemistry and Biochemistry, v.v.i. Academy of Sciences of the Czech Republic, Flemingovo nam. 2, 166 10 Prague 6, Czech Republic

<sup>c</sup> Faculty of Science, Charles University, Hlavova 2030, 128 40 Prague 2, Czech Republic

<sup>d</sup> Institute of Physics, v.v.i. Academy of Sciences of the Czech Republic, Na Slovance 2, 182 21, Prague 8, Czech Republic

<sup>e</sup> Czech Technical University in Prague, Faculty of Biomedical Engineering, Sítňá 3105, 272 01 Kladno, Czech Republic

<sup>f</sup> Forensic Laboratory of Biologically Active Substances, University of Chemistry and Technology Prague, Technická 5, 166 28 Prague 6 – Dejvice, Czech Republic



**Scheme S1:** Molecular formula of P1 dye

**Table S2a:** Results of the XPS surface analysis (concentrations in atom %). A pronounced drop of the B-content is obvious in the series BDD > BDD+1 > BDD+1 (deprotected) > P1@BDD. The BDD+1 sample shows the F/N ratio = 2.8 (calc. 3). For P1@BDD, the S/N ratio = 0.7 (calc. 0.33).

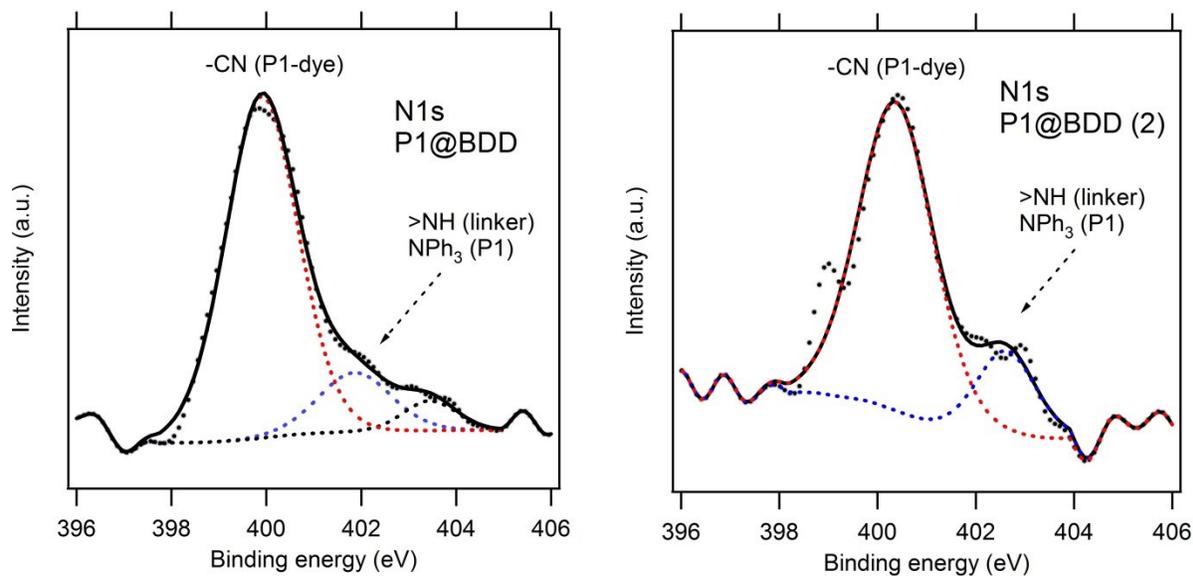
	<b>C 1s</b>	<b>B 1s</b>	<b>O 1s</b>	<b>N1s</b>	<b>S 2p</b>	<b>Cl 2p</b>	<b>F 1s</b>	<b>Si 2s</b>	<b>other*</b>
BDD	94.8	2.5	2.3	-	-	-	-	0.4	-
BDD + 1	84.8	1.6	8.1	0.4	0.2	0.3	1.1	3.3	0.2
BDD + 1 (deprotected)	80.1	1.1	11.5	1.5	0.4	1.0	0.7	1.3	2.4**
P1@BDD (linker 1)	80.1	0.7	10.0	3.9	2.9	0.3	0.4	0.9	0.8
P1@BDD (linker 2)	85.8	1.7	7.7	1.0	0.8	-	0.8	0.7	1.5

\* Other impurities (Ca, Na, P)

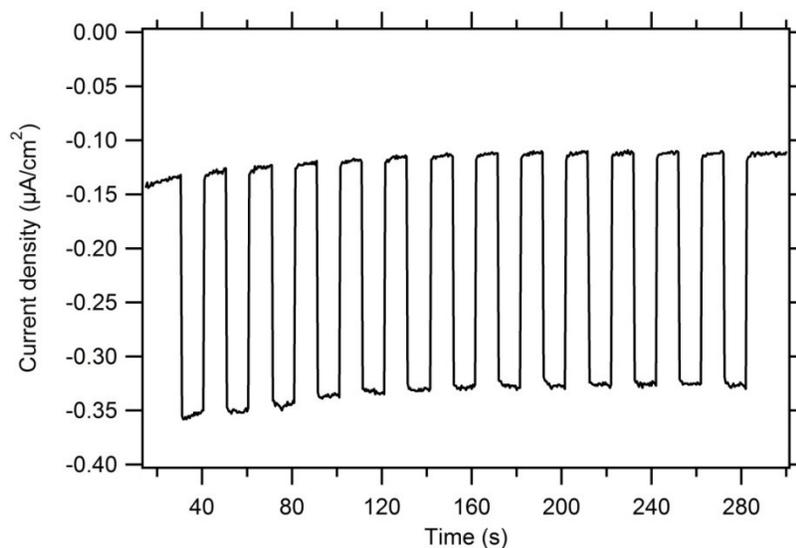
\*\* The main impurity in the BDD+1 (deprotected) sample is Na (1.8 %) originating presumably from the reactants used in the deprotection steps (see Experimental Section in the main text).

**Table S2b:** Results of the XPS surface analysis (binding energies in eV). Very weak peaks are not evaluated.

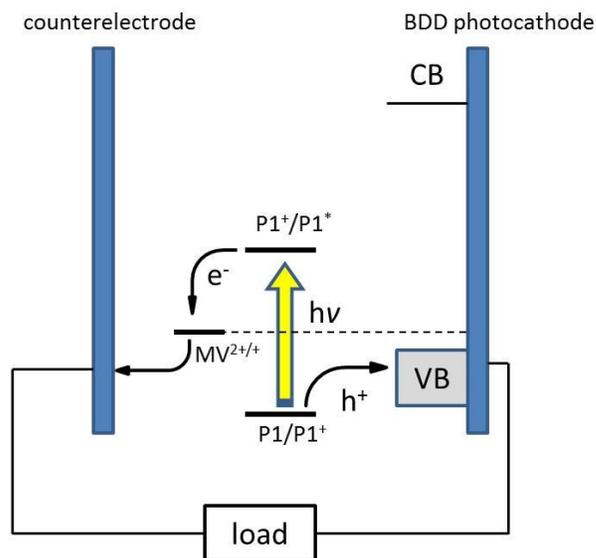
	<b>C 1s</b>	<b>B 1s</b>	<b>O 1s</b>	<b>N1s</b>	<b>S 2p</b>	<b>Cl 2p</b>	<b>F 1s</b>	<b>Si 2s</b>
BDD	284.4	187.6	532.4	-	-	-	-	-
BDD + 1	284.4	186.8	532.0	-	-	199.2	688.4	153.2
BDD + 1 (deprotected)	284.4	186.4	532.4	402.8	168.4	197.6	685.2	153.6
P1@BDD (linker 1)	284.4	187.0	532.0	399.9 402.0	164.4	-	-	153.3
P1@BDD (linker 2)	284.4	186.4	532.0	400.3 402.6	168.8	-	688.0	153.5



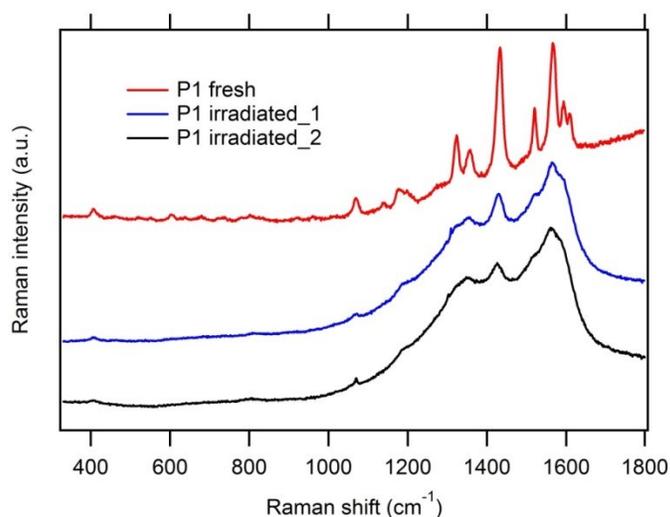
**Figure S3:** Detail of the N1s photoelectron spectra with deconvoluted components. Left chart: the optimized sample P1@BDD (linker **1**). Right chart: P1@BDD (2) (linker **2**). Tentative assignment of photoemission lines to >NH groups from the linker, triphenylamino group from the **P1** dye and the cyano-group from the **P1**-dye is provided.



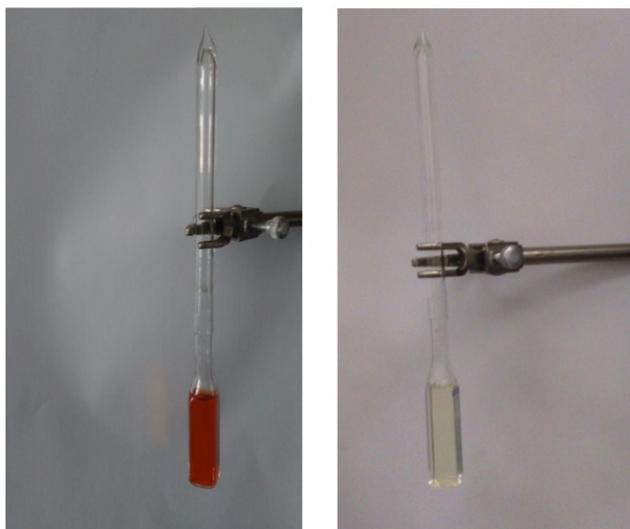
**Figure S4:** Chronoamperometric plot for a BDD electrode sensitized with P1 (anchored via linker 2). Electrolyte solution 0.1 M Na<sub>2</sub>SO<sub>4</sub> containing 5 mM dimethylviologen, applied bias voltage -0.3 V vs. Ag/AgCl. Chopped white light illumination (100 mW/cm<sup>2</sup>; simulated AM1.5G solar spectrum, 10 s dark/light interval).



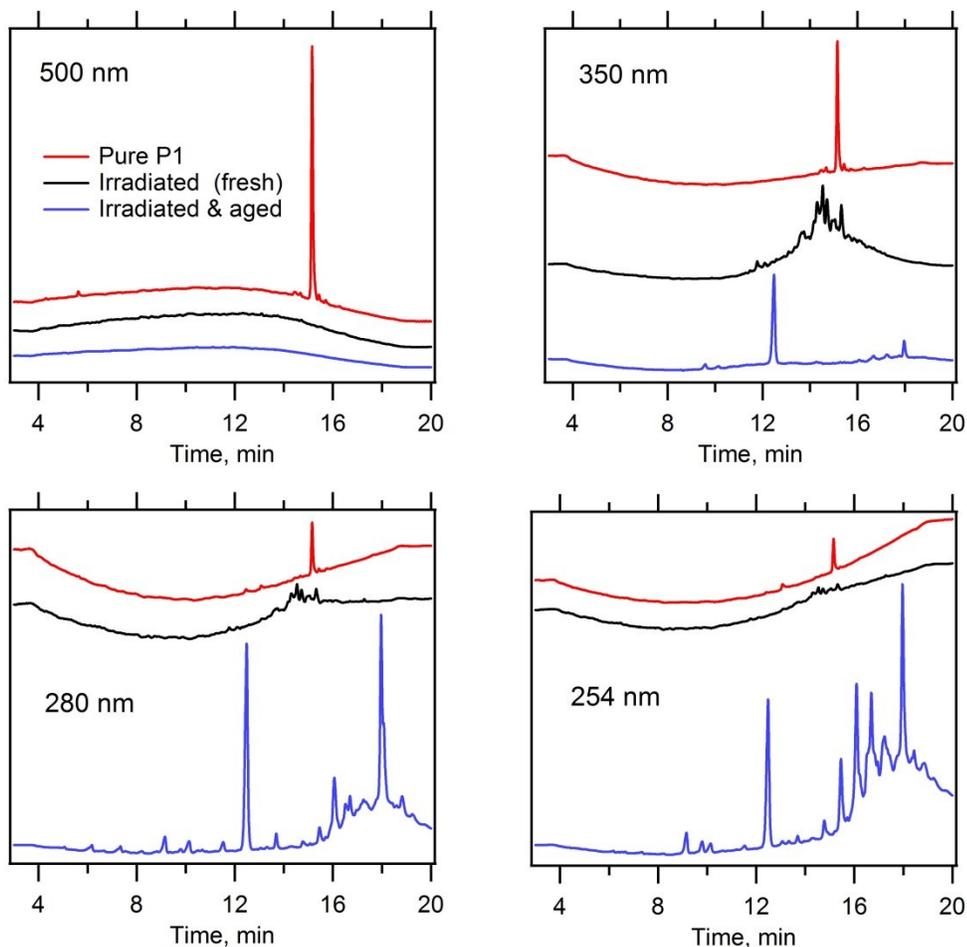
**Figure S5:** Scheme of photocurrent generation in an electrochemical cell with boron-doped diamond photocathode sensitized with P1.



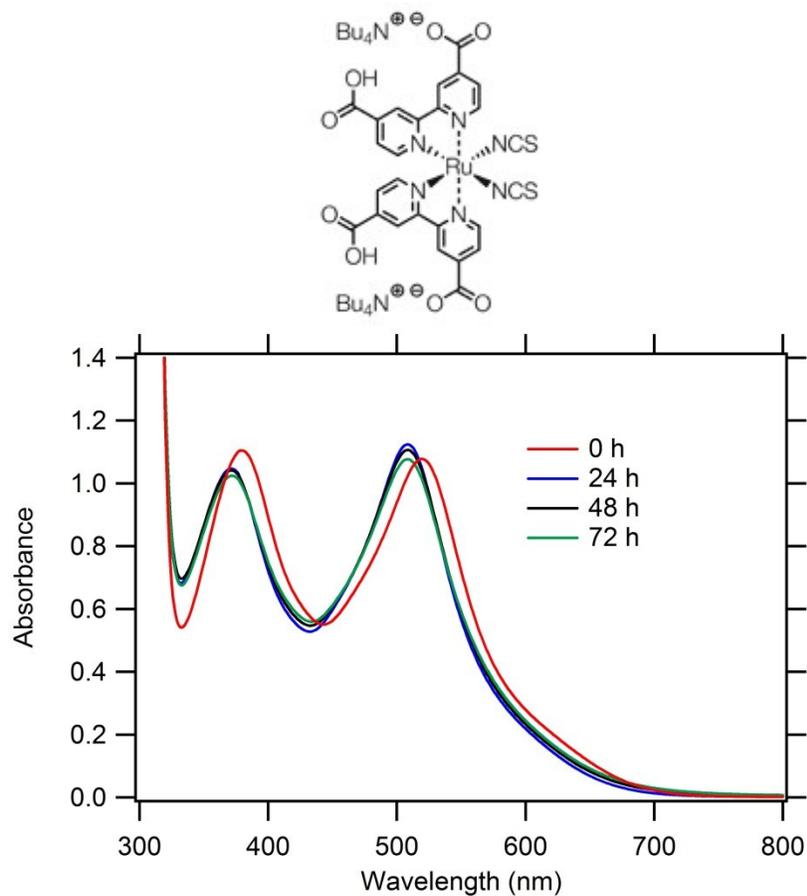
**Figure S6:** Raman spectra (at 488 nm excitation) of pure solid **P1** dye (red curve) and the same **P1** sample which passed long-term irradiation by the 488 nm laser (blue curve). Black curve is for the same sample after subsequent irradiation.



**Figure S7:** Solution of the P1 dye in absolute ethanol (concentration  $10^{-4}$  mol/L) in vacuum-sealed quartz optical cell (1 cm); before irradiation (left image) and after 24 hours of irradiation at 1 sun intensity (right image). The yellowish coloration of the solution after photochemical treatment (right image) comes from the tail of the UV band (see Fig. 7 in the main text). There are no traces of P1 in the illuminated solution detectable by HPLC (cf. Figure S8).



**Figure S8:** HPLC chromatograms of the  $10^{-5}$  mol/L solution of the **P1** dye in absolute ethanol measured shortly ( $\approx$  several hrs) after irradiation for 1 day at 1 sun intensity in a closed quartz cell (black curves). Blue curves are for the irradiated solution, which was stored for 2 months in air at room temperature. Reference chromatograms for freshly made solution of **P1** are shown by red curves. Spectrophotometric detection was carried out at four different wavelengths (500 nm, 350 nm, 280 nm and 254 nm) as it is labeled on each chart.



**Figure S9:** UV-Vis spectrum of the solution of **N719** dye (see the chemical formula on top of the chart) in absolute ethanol. Optical length 1 cm. The spectrum of a fresh solution (red curve) and that after illumination with a white light of 1 sun intensity for 24 hours (blue curve), 48 hours (black curve) and 72 hours (green curve).