Supplement

to

pH-Sensitive fluorophores from locked GFP chromophores by a nonalternant analogue of the photochemical *meta* effect

By

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I. Details of Quantum Chemistry Computations

Table S1. Cartesian Coordinates (Å) of MP2/cc-pvdz-optimized ground state structure of $pOH-BDI-OBF_2-NOH-BDI$ (**V**), SA2-CAS(4,3)/cc-pvdz state energies, natural orbitals and occupation numbers at MP2/cc-pvdz structure. All energies in atomic units (hartree).

MP2/c	c-pvdz structure (Å):		
28			
MP2/C	C-PVDZ ENERGY=-981.76	5666344	
В	-1.2828210418	0.1450915535	0.2411792667
С	-0.2693686379	0.0306859887	1.4935578017
С	1.1520849209	0.0704362399	1.3683719946
С	1.9818045787	0.0330903118	2.5186194226
С	1.4217173799	-0.0488115192	3.7935429121
С	0.0197219320	-0.1050311436	3.9310337458
С	-0.8025828865	-0.0676675671	2.7865484312
Н	-1.8930174568	-0.1159996391	2.9105815160
0	-0.4774107288	-0.1948318662	5.1971143101
Н	-1.4438872989	-0.2202375180	5.1259975872
Н	2.0431743181	-0.0785975599	4.6934415159
Н	3.0715656956	0.0634944547	2.4019426860
С	1.7576765184	0.0901635155	0.0566781917
С	1.0223121185	0.0211508994	-1.0966114110
Ν	-0.3797225868	-0.0577143212	-1.0983784564
С	-0.8036754384	-0.0946386881	-2.3531010882
Ν	0.2664391891	-0.0318923464	-3.2005861899
С	1.5076804386	-0.0064045394	-2.4885552259
0	2.6134437162	0.0140000702	-2.9976674304
0	0.1642566752	-0.2789173016	-4.5508242796
Н	0.5296422729	0.5257641063	-4.9593767984
С	-2.2191581221	-0.1468030536	-2.8012264206
Н	-2.5923019177	0.8811808390	-2.9459590007
Н	-2.2931343912	-0.6901393141	-3.7536723637
Н	-2.8187621612	-0.6290425267	-2.0185574872
Н	2.8490842653	0.1342998789	-0.0503622942
F	-1.8668511482	1.4185144517	0.1056299405
F	-2.2728009015	-0.8571770850	0.2322244934

SA2-CAS(4,3)/cc-pvdz State Energies @ above structure: !MCSCF STATE 1.1 Energy -979.021497453775 !MCSCF STATE 2.1 Energy -978.855937958705

SA2-CAS(4,3)/cc-pvdz state-averaged natural orbitals and occupation numbers:



Table S2. Cartesian Coordinates (Å) of MP2/cc-pvdz-optimized ground state structure of pH-BDI-oBF₂-NOH-BDI (**VI**), SA2-CAS(4,3)/cc-pvdz state energies, natural orbitals and occupation numbers at MP2/cc-pvdz structure. All energies in atomic units (hartree).

MP2/cc-pvdz structure (Å): 27

MP2/CC-PVDZ ENERGY=-906.71474686

В	-1.3551224641	0.1063183456	0.5025427701
С	-0.4063670570	-0.0020758939	1.8040583473
С	1.0194265180	0.0388217401	1.7545644556
С	1.8000532690	-0.0017925332	2.9371434849
Н	2.8936789245	0.0304366059	2.8639659254
С	1.1741121867	-0.0891421446	4.1860652183
С	-0.2307502138	-0.1459010380	4.2557924664
С	-0.9991807821	-0.1047821695	3.0773233703
Н	-2.0917483904	-0.1534255200	3.1486545898
Н	-0.7257296288	-0.2219624094	5.2303771337
Н	1.7747018707	-0.1193277431	5.1011601411
С	1.6965533840	0.0736847282	0.4708805179
С	1.0230539273	0.0162674036	-0.7172078885
Ν	-0.3766595213	-0.0636977833	-0.7942497950
С	-0.7329811759	-0.0801419562	-2.0688334276
Ν	0.3791136812	0.0019164207	-2.8615537352
С	1.5809160489	0.0110991571	-2.0857096183
0	2.7125019072	0.0361140750	-2.5309254216
0	0.3478469461	-0.2315554723	-4.2175725845
Н	0.6900292793	0.5946845839	-4.6023426914
С	-2.1238361646	-0.1282327517	-2.5880263899
Н	-2.5040426405	0.9015562961	-2.6967824027
Н	-2.1440383824	-0.6244295121	-3.5681591571
Н	-2.7530713004	-0.6543876762	-1.8584647766
Н	2.7918656515	0.1225042082	0.4222556604
F	-2.3179535308	-0.9171436990	0.4213466346
F	-1.9556023194	1.3684695772	0.3418675063

 SA2-CAS(4,3)/cc-pvdz state energies @ above structure:

 !MCSCF STATE 1.1 Energy
 -904.152278845704

 !MCSCF STATE 2.1 Energy
 -903.979074392594

SA2-CAS(4,3)/cc-pvdz natural orbitals and occupation numbers:



Table S3. Cartesian Coordinates (Å) of MP2/cc-pvdz-optimized ground state structure of pO^- -BDI-oBF₂-NOH-BDI (**IX**), SA2-CAS(4,3)/cc-pvdz state energies, natural orbitals and occupation numbers at MP2/cc-pvdz structure. All energies in atomic units (hartree).

MP2/cc-pvdz ground state structure (Å): 27 MP2/CC-PVDZ ENERGY=-981.23472746 В -1.2887613285 0.1648270579 0.2626397101 С -0.3048509807 0.0274210866 1.5409679915 С 1.1463627031 0.0960656044 1.4080500134 С 1.9763178052 0.0718515104 2.5904154613 3.8431307562 С 1.4281010460 -0.0432991641 С -0.0272585067 -0.1520545753 4.0421499200 С -0.8300722550 -0.1055392457 2.8088504322 Η -1.9166576684 -0.1753533611 2.9497756172 0 -0.5338495038 -0.2704716112 5.1765346901 Η 2.0529596509 -0.0685209385 4.7443913912 Η 3.0665086804 0.1357962687 2.4612251143 1.7350716855 0.1140963869 0.1449158110 С 1.0000155165 0.0515337991 -1.0606944010 С -0.3931955071 -0.0330507832 -1.0627025667 N -0.8399299171 -0.1182252350 -2.3275617939 С 0.2525916052 -0.0784918601 -3.1284184389 Ν -2.4182078162 С 1.4692692619 0.0047452235 -3.0530926064 2.5514112828 -0.0012018118 Ο 0.3113957705 -0.2037358778 -4.4983740548 Ο 1.3020036978 -0.1900453326 -4.5665317146 Н -2.2535433876 -0.1566805007 -2.7893670145 С 0.8621593019 -2.9813954217 Η -2.6332639119 Η -2.3289169251 -0.7394053654 -3.7213294554 -0.6134889566 -2.0011203172 Η -2.8670644671 0.1551151314 0.0471219183 Η 2.8288672900 F -1.8785399034 1.4504514174 0.1458001694 F -2.3109001647 -0.8222079138 0.2315410490

SA2-CAS(4,3)/cc-pvdz state energies (a.u.) @ above structure: !MCSCF STATE 1.1 Energy -978.485966095080 !MCSCF STATE 2.1 Energy -978.331240446186

SA2-CAS(4,3)/cc-pvdz natural orbitals and occupation numbers:



Table S4. Cartesian Coordinates (Å) of MP2/cc-pvdz-optimized ground state structure of $pOH-BDI-OBF_2-NO^--BDI$ (**VIII**), SA2-CAS(4,3)/cc-pvdz state energies, natural orbitals and occupation numbers at MP2/cc-pvdz structure; SA3-CAS(6,4)/cc-pvdz natural orbitals and occupation numbers. All energies in atomic units (hartree).

MP2/cc-pvdz ground state structure (Å): 27

MP2/CC-PVDZ	ENERGY=-981	.23018773
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В	-1.2588388060	0.0000123507	-0.1708443727
С	-0.2465488269	-0.000003392	-1.4453225744
С	1.1771320756	0.0000013187	-1.3315802790
С	1.9881804768	0.000039525	-2.4946410175
С	1.4164827726	0.000031270	-3.7716535019
С	0.0149204336	-0.0000016419	-3.8960417277
С	-0.7893129686	-0.0000073487	-2.7435520025
Н	-1.8834760029	-0.0000147957	-2.8569454194
0	-0.5136188383	-0.0000055079	-5.1706567309
Н	-1.4751656221	-0.000008408	-5.0545719682
Н	2.0319365393	0.000096085	-4.6775253996
Н	3.0806315745	0.0000105752	-2.3887205775
С	1.7865198626	0.000006937	-0.0108477582
С	1.0107772636	-0.0000233950	1.1173146444
N	-0.3698108839	-0.0000485514	1.1128147456
С	-0.7714010124	-0.0000232477	2.4301269096
Ν	0.2554999752	0.0000219254	3.3051835416
С	1.4867273510	-0.0000092182	2.5250407871
0	2.6147577463	-0.0000122818	2.9910865469
0	0.2151799430	0.0000233270	4.5869928101
С	-2.1843019456	-0.0000367544	2.8818783564
Н	-2.7068518849	-0.8870363663	2.4884862068
Н	-2.1755553422	-0.0005674396	3.9818255606
Н	-2.7065977423	0.8875425842	2.4894390703
Н	2.8757420683	0.0000177051	0.1146946427
F	-2.1007001769	-1.1540981849	-0.1662613976
F	-2.1006217007	1.1541733426	-0.1662008568

SA2-CAS(4,3)/cc-pvdz state energies @ above structure: !MCSCF STATE 1.1 Energy -978.457618378426 !MCSCF STATE 2.1 Energy -978.355140116831

SA2-CAS(4,3)/cc-pvdz natural orbitals and occupation numbers:



 SA3-CAS(6,4)/cc-pvdz state energies @ above structure:

 !MCSCF STATE 1.1 Energy
 -978.464306206420

 !MCSCF STATE 2.1 Energy
 -978.340379936465

 !MCSCF STATE 3.1 Energy
 -978.276515722728

SA3-CAS(6,4)/cc-pvdz natural orbitals & occupation numbers:



Table S5. Cartesian Coordinates (Å) of MP2/cc-pvdz-optimized ground state structure of pH-BDI-oBF₂-NO⁻-BDI (NO^- base conjugate to **VI**, not explicitly named in the text), SA2-CAS(4,3)/cc-pvdz state energies, natural orbitals and occupation numbers at MP2/cc-pvdz structure. SA3-CAS(6,4)/cc-pvdz natural orbitals and occupation numbers. All energies in atomic units (hartree).

MP2/cc	-pvdz ground state	structure (Å):	
26			
MP2/C	C-PVDZ ENERGY=-906.	17982590	
В	-1.3292129570	0.0001502548	-0.4242195494
С	-0.3929321932	0.000004214	-1.7559939079
С	1.0357519565	0.0000492978	-1.7233582366
С	1.7905240672	-0.0000042838	-2.9243337707
Н	2.8868639677	0.0000373286	-2.8676975154
С	1.1479951904	-0.0001200467	-4.1686426405
С	-0.2599027572	-0.0002012129	-4.2222178935
С	-1.0035478657	-0.0001450604	-3.0284761963
Н	-2.0993416485	-0.0002155049	-3.0805826339
Н	-0.7713232025	-0.0003087224	-5.1927513066
Н	1.7379923440	-0.0001543048	-5.0930714270
С	1.7200646766	0.0001041013	-0.4377198943
С	1.0100016273	0.0000061069	0.7325326347
Ν	-0.3667960828	-0.0001232616	0.8088223013
С	-0.6912682865	-0.0001452210	2.1475412472
N	0.3838050511	-0.0000632660	2.9622206444
С	1.5671973306	0.0000061698	2.1120721670
0	2.7203781924	0.0000576941	2.5107257960
0	0.4182845835	-0.0001077091	4.2442035961
С	-2.0760124095	-0.0002187604	2.6788846647
Н	-2.6194789051	-0.8870578959	2.3146855239
Н	-2.0056586998	-0.0007001608	3.7765484039
Н	-2.6192580056	0.8871166039	2.3155435640
Н	2.8146781289	0.0001935323	-0.3745793661
F	-2.1660719840	1.1552449802	-0.3652471208
F	-2.1665949270	-1.1545722479	-0.3653598010

SA2-CAS(4,3)/cc-pvdz state energies @ above structure: !MCSCF STATE 1.1 Energy -903.589936277323 !MCSCF STATE 2.1 Energy -903.490042645950

SA2-CAS(4,3)/cc-pvdz natural orbitals & occupation numbers:



 SA3-CAS(6,4)/cc-pvdz state energies @ above structure:

 !MCSCF STATE 1.1 Energy
 -903.596407595891

 !MCSCF STATE 2.1 Energy
 -903.475424305193

 !MCSCF STATE 3.1 Energy
 -903.406293051620

SA3-CAS(6,4)/cc-pvdz natural orbitals & occupation numbers:





Figure S1. Geometries of species for which theoretical data is reported in Table 3 of the main text. Geometries were determined by optimization on the ground state of an MP2/cc-pvdz model (see text). Heavy (i.e. not H) bond lengths are shown in Å. The figure shows why we chose to denote the NB bond as coordinate (dative) in the text. Deprotonation of the NOH leads to a ~0.1 Å contraction of the NB bond. Since NOH is also expected to stabilize the covalent bonding form (see figure), we deduced that the protonated species was best drawn as having a dative NB bond.

MP2/cc-pvdz Ground State Species

II. Spectroscopic and other data

	V		IV		VI	
Solvent	Abs	Ext	Abs	Ext	Abs	Ext
	max	coef	max	coef	max	coef
H ₂ O/MeOH	371	18000	416	22500	409	19000
H ₂ O	369	18000	411	22000	405	n.d.ª
EtOH	368	20500	421	23500	407	22000
CH ₃ CN	365	22500	407	24000	402	22000

II.1 Optical properties

a - is not soluble enought



II.2 Titration curves (absorbtion)

Figure 1. Absorption spectra of compound V at different pH in phosphate buffer (50 mM).



Figure 2. Absorption spectra of compound IV at different pH in phosphate buffer (50 mM).



Figure 5. Absorption spectra of compound V at different pH in water-methanol mixture.



Figure 6. Absorption spectra of compound **IV** at different pH in watermethanol mixture.



Figure 7. Absorption spectra of compound **VI** at different pH in watermethanol mixture.





Figure 9. Emission spectra of compound IV at different pH in water.



Figure 10. Emission spectra of compound IV at different pH in phosphate buffer.

III. Summary of SVD Analysis

KinTek Explorer software (<u>www.kintekcorp.com</u>, KinTek Corp, Austin, TX) was used to deconvolute changes in spectra as a function of pH based upon singular value decomposition analysis in order to derived the pH dependence and spectra of individual species. Fitting by nonlinear regression included weighting to give preference to the most significant Eigenvectors. Fitting to each model was based upon numerical integration of the rate equations to derive the endpoint of the reaction at each pH. The figures below show the original pH dependence of the emission spectra and the derived emission spectra and pH dependence of each species.















IV. Kamlet-Taft analysis of quantum yield data

In order to better understand the dependence of the quantum yield data for IV in Table 1, we performed a multivariate linear regression of logQY for this species against the Kamlet-Taft solvatochromic parameters for the solvents described. The Kamlet-Taft parameters α , β , and π^* for the solvents were taken from reference 34. The resulting fit had correlation coefficient R=0.99 and was found to be $\log QY = 1.0 - 0.1\alpha - 1.6\beta - 1.1\pi^*$. The parameters α , β , and π^* are measures of the relative hydrogen bond donation (acidity), hydrogen bond acceptance (basicity) and polar solvation power of the solvents used. For more detailed background see reference 34. We emphasize that there are only five data points for IV in Table 1 against which the fit was taken. It should be interpreted with extreme caution. With this in mind, the analysis does support the assertion that increases in both solvent basicity (β) and polarity (π^*) both kill the QY more effectively than do increases in the solvent acidity (α).