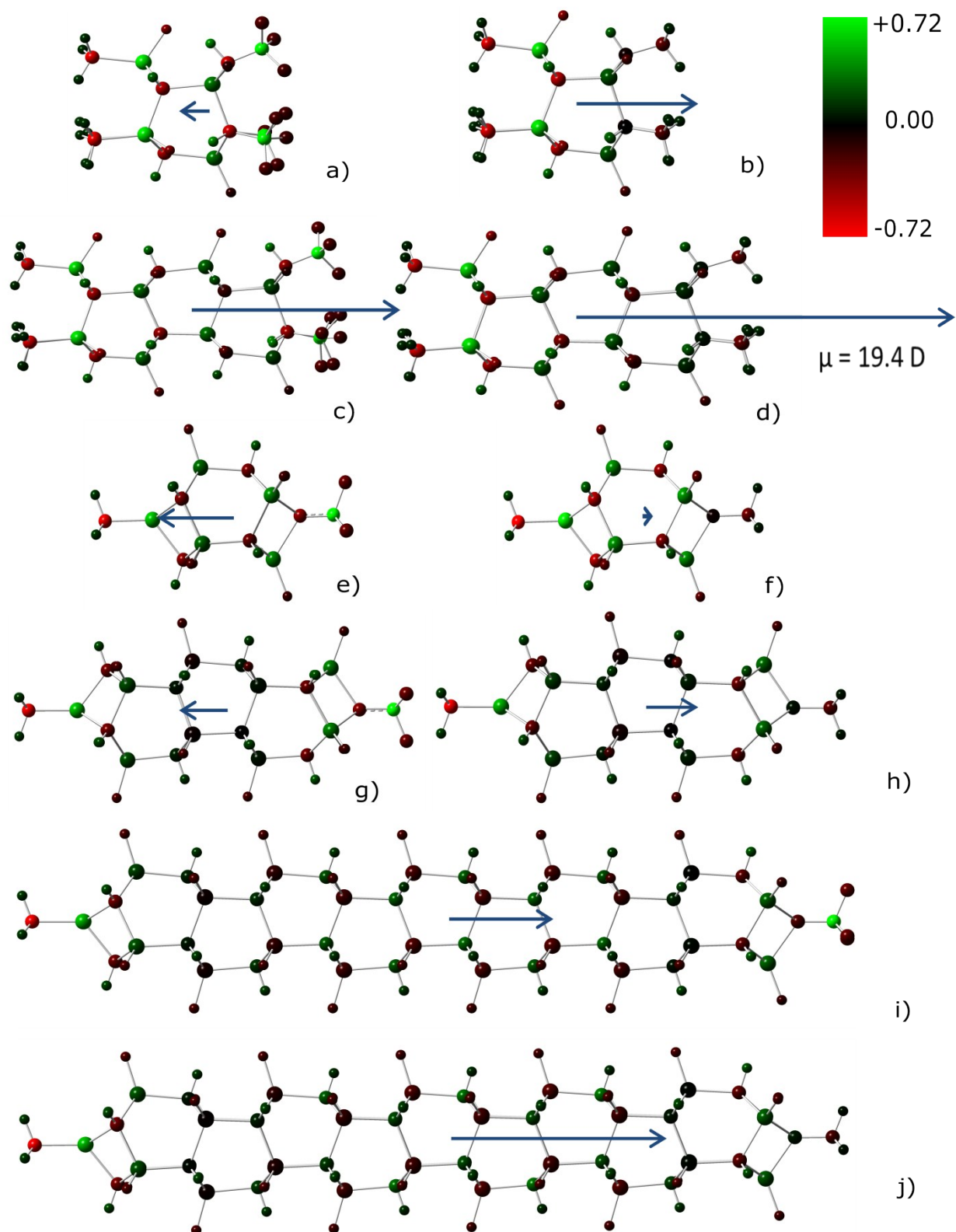


Formal reactions of “capping” of open oligomers with different thermal groups could be written in the following way.

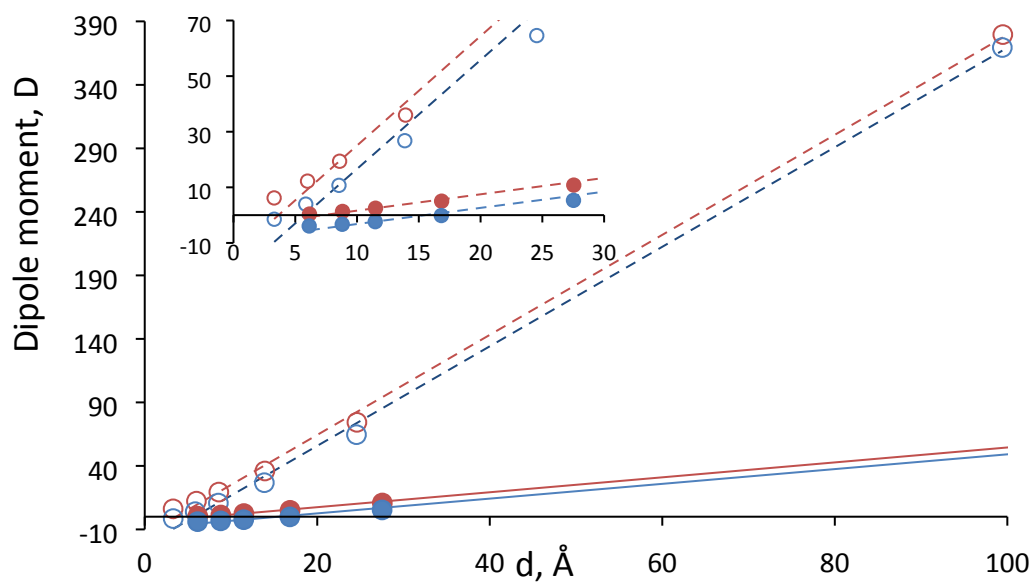


**Table S1** Standard enthalpies  $\Delta H_{298}^0$  (in kcal/mol) of the formal reactions of “capping” oligomers with XGa and NY groups.

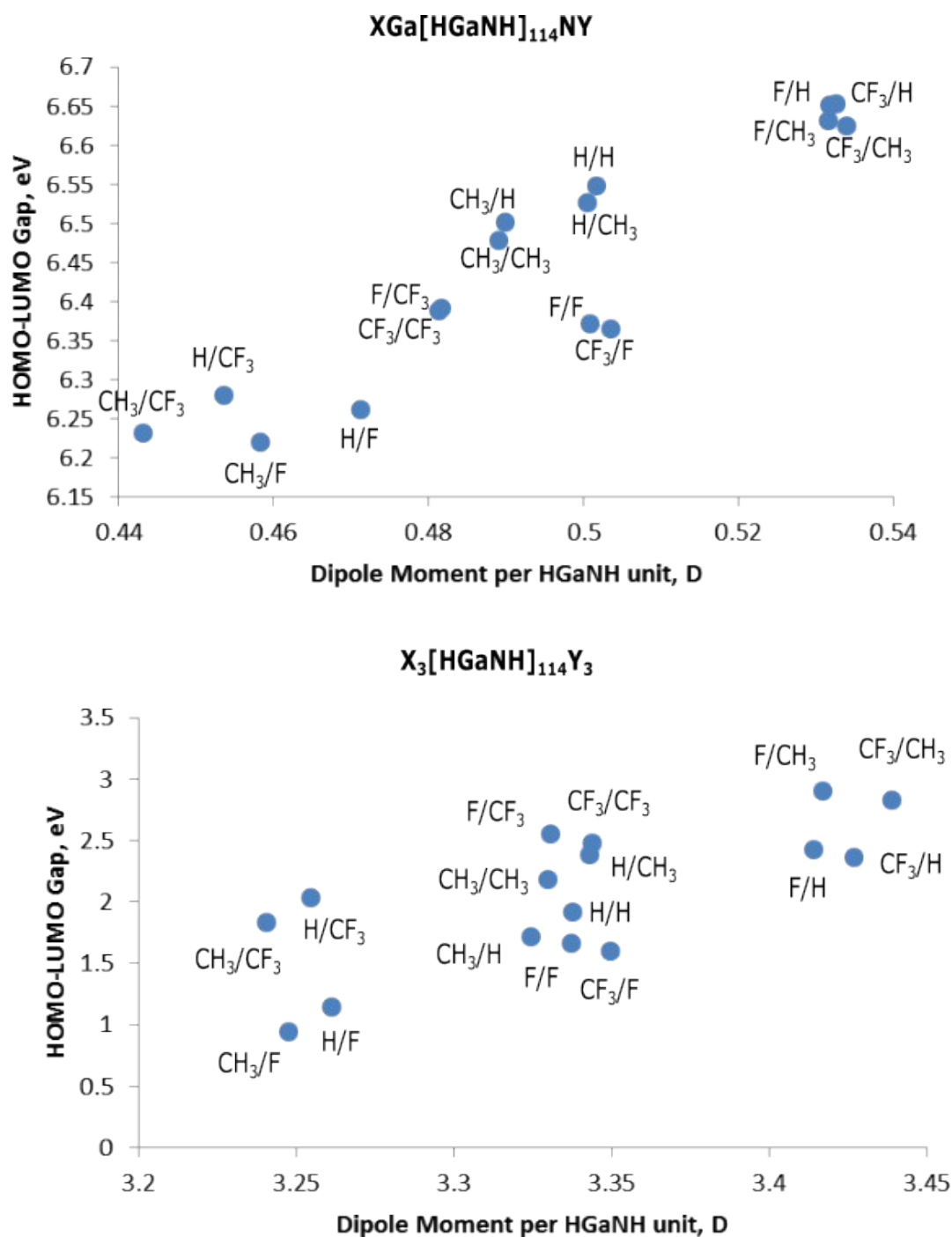
X	Y	<i>n</i> =9	<i>n</i> =30
H	H	-26.24	-51.56
H	CH3	-74.36	-99.60
H	F	-150.12	-179.96
H	CF3	-62.81	-93.75
CH3	H	-49.54	-75.76
CH3	CH3	-97.70	-123.80
CH3	F	-173.09	-204.32
CH3	CF3	-85.63	-118.05
F	H	34.10	13.30
F	CH3	-14.00	-34.75
F	F	-91.22	-115.13
F	CF3	-4.27	-28.62
CF3	H	-37.68	-57.34
CF3	CH3	-85.79	-105.64
CF3	F	-163.04	-186.03
CF3	CF3	-76.20	-99.80



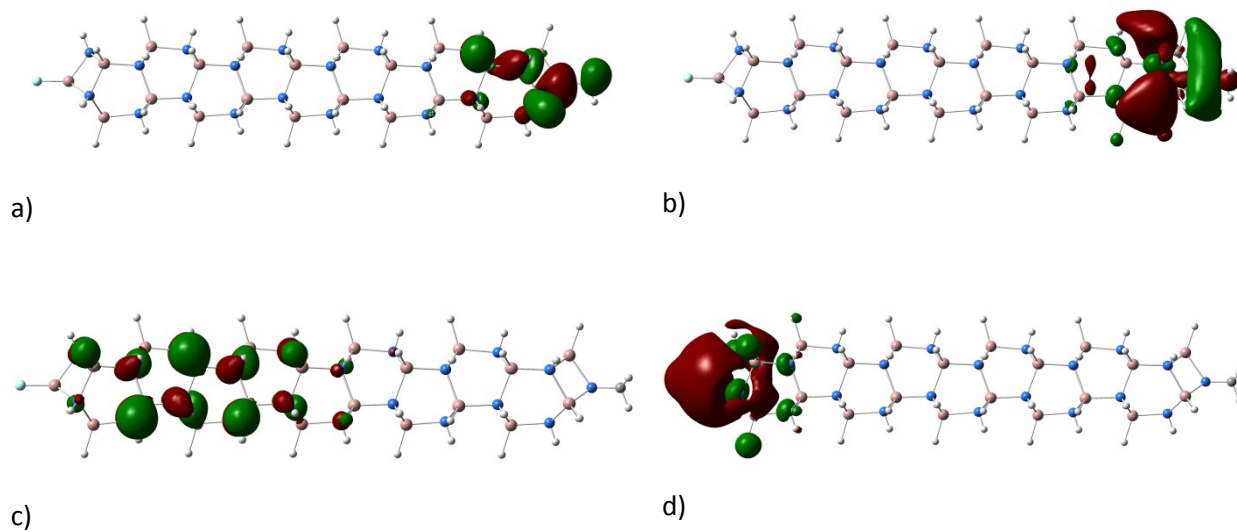
**Figure S1** Dipole moment vectors and ECP charge distribution over atoms of oligomers:  $(\text{CH}_3)_3[\text{HGaNH}]_6(\text{CF}_3)_3$  (a),  $(\text{CH}_3)_3[\text{HGaNH}]_6(\text{CH}_3)_3$  (b),  $(\text{CH}_3)_3[\text{HGaNH}]_{12}(\text{CF}_3)_3$  (c),  $(\text{CH}_3)_3[\text{HGaNH}]_{12}(\text{CH}_3)_3$  (d),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_6\text{NCF}_3$  (e),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_6\text{NCH}_3$  (f),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_{12}\text{NCF}_3$  (g),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_{12}\text{NCH}_3$  (h),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_{12}\text{NCF}_3$  (i),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_{12}\text{NCH}_3$  (j)



**Figure S2** Dipole moments as functions of the lengths of oligomers  $(\text{CH}_3)_3[\text{HGaNH}]_n(\text{CH}_3)_3$  (red open circles),  $(\text{CH}_3)_3[\text{HGaNH}]_n(\text{CF}_3)_3$  (blue open circles),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_n\text{NCH}_3$  (red filled circles),  $\text{CH}_3\text{Ga}[\text{HGaNH}]_n\text{NCF}_3$  (blue filled circles) with  $n=6, 9, 12, 18, 30, 114$ . Lines are linear approximation. Lengths are defined as distances between Ga and N atoms/planes at opposite ends of the oligomer.



**Figure S3** HOMO-LUMO gaps of the XGa[HGaNH]<sub>114</sub>NY and X<sub>3</sub>[HGaNH]<sub>114</sub>Y<sub>3</sub> oligomers with respect of their dipole moment values calculated per HGaNH unit. Tooltips for the particular oligomers are written in X/Y format.



**Figure S4** HOMO (a), LUMO(b) , HOMO-2(c) and LUMO+3 of FGa(HGaNH<sub>30</sub>)NCH<sub>3</sub> oligomer. Relatively to the HOMO energy, LUMO has energy 6.6 eV , LUMO has 6.8 eV and HOMO-2 has energy -0.1 eV. Blue color refers to N, pink to Ga, white to H, gray to C and light blue to F atoms.