Electronic Supplementary Information for

Thermodynamic and Redox Properties of Graphene Oxides for Lithium-Ion Battery Applications: First Principles Density Functional Theory Modeling Approach

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1. Determination of the atomic charges

All the atomic charges in our study are determined by the Mulliken population analysis combined with the density functional theory (DFT) method at the B3LYP level of theory and 6-31G(d,p) basis set in the Jaguar package. Figures S1-S illustrate the cluster models utilized to calculate the atomic charges and their atomic charges.



Pristine graphene

Туре	Charge
H1	0.16
C1	-0.17
C2	0.04
C3	0.04
C4	-0.04
C5	-0.03
C6	0.005

Figure S1: The atomic charges for the pristine graphene.

Graphene with 1 carbonyl group



Туре	Charge
01	-0.45
C1	0.33
C2	0.08
C3	0.08
C4	0.004
C5	0.004
C6	0.0004

Figure S2: The atomic charges for the graphene with 1 carbonyl group.



Graphene with 2 locally distributed carbonyl groups

Figure S3: The atomic charges for the graphene with 2 locally distributed carbonyl groups.



0.33

0.30 0.31

-0.21

-0.16

-0.15

-0.22

Graphene with 3 locally distributed carbonyl groups

Figure S4: The atomic charges for the graphene with 3 locally distributed carbonyl groups.

C1

C2

C3 C4

C5

C6

C7



Graphene with 4 locally distributed carbonyl groups

Туре	Charge
01	-0.42
O2	-0.35
O3	-0.31
O4	-0.37
C1	0.32
C2	0.33
C3	0.25
C4	0.29
C5	-0.17
C6	-0.14
C7	-0.08
C8	-0.04

Figure S5: The atomic charges for the graphene with 4 locally distributed carbonyl groups.



Graphene with 2 uniformly distributed carbonyl groups

Figure S6: The atomic charges for the graphene with 2 uniformly distributed carbonyl groups.



Graphene with 3 uniformly distributed carbonyl groups

Figure S7: The atomic charges for the graphene with 3 uniformly distributed carbonyl groups.



C34

C35

C41

C42

C43

C44

C45

-0.01

0.02

-0.20

-0.06

0.07

0.01 -0.009

Graphene with 4 uniformly distributed carbonyl groups

Figure S8: The atomic charges for the graphene with 4 uniformly distributed carbonyl groups.

04

C1

C2

C3

C4

-0.44

0.32

0.32

0.32

0.32

Graphene with 1 hydroxyl group



Туре	Charge
01	-0.47
H1	0.36
C1	0.26
C2	-0.02
C3	-0.07
C4	0.002
C5	0.005
C6	-0.002

Figure S9: The atomic charges for the graphene with 1 hydroxyl group.



Graphene with 2 locally distributed hydroxyl groups

Туре	Charge
01	-0.58
O2	-0.57
H1	0.46
H2	0.43
C1	0.28
C2	0.26
C3	-0.29
C4	-0.14
C5	-0.03

Figure S10: The atomic charges for the graphene with 2 locally distributed hydroxyl groups.

C4 O1 C1 C5 O2 C2 C6 O3 C3 C7 H3

Graphene with 3 locally distributed hydroxyl groups

Туре	Charge
01	-0.52
O2	-0.63
O3	-0.61
H1	0.43
H2	0.46
H3	0.41
C1	0.27
C2	0.27
C3	0.30
C4	-0.26
C5	-0.13
C6	-0.12
C7	-0.27

Figure S11: The atomic charges for the graphene with 3 locally distributed hydroxyl groups.



Graphene with 4 locally distributed hydroxyl groups

Туре	Charge
01	-0.51
O2	-0.60
O3	-0.63
O4	-0.60
H1	0.41
H2	0.45
H3	0.46
H4	0.41
C1	0.17
C2	0.17
C3	0.26
C4	0.29
C5	-0.02
C6	-0.09
C7	-0.14
C8	-0.26

Figure S12: The atomic charges for the graphene with 4 locally distributed hydroxyl groups.



0.23

0.23

-0.02

-0.08

0.005

0.008

0.01

C1 C2

C3

C4

C5

C6

C7

Graphene with 2 uniformly distributed hydroxyl groups

Figure S13: The atomic charges for the graphene with 2 uniformly distributed hydroxyl groups.



Graphene with 3 uniformly distributed hydroxyl groups

Figure S14: The atomic charges for the graphene with 3 uniformly distributed hydroxyl groups.



Graphene with 4 uniformly distributed hydroxyl groups

Туре 01 02 03 -0.50 C33 0.08 04 C34 -0.45 0.02 0.35 H1 C35 -0.03 H2 0.36 C41 -0.07 0.37 H3 C42 -0.01 H4 0.36 C43 0.001 C1 0.25 C44 0.004 0.003 C2 0.24 C45 C3 0.28 C4 0.24

Figure S15: The atomic charges for the graphene with 4 uniformly distributed hydroxyl groups.