

Supporting Information

Is the chemistry of lawrencium peculiar?

Wen-Hua Xu^{a,b} and Pekka Pyykkö*^b

^a Key Laboratory of Synthetic and Natural Functional Molecule Chemistry of the Ministry of Education, College of Chemistry and Molecular Science, Northwest University, 710127 Xi'an, China. E-mail: xuwenhua.qf@gmail.com.

^b Department of Chemistry, University of Helsinki, POB 55 (A. I. Virtasen aukio 1), 00014 Helsinki, Finland. E-mail: Pekka.Pyykko@helsinki.fi

Table S 1 Excitation energies (in eV) of the Lr atom from the $7p_{1/2}$ ground state to the $6d_{3/2}$ excited state. Non-relativistic (NR) reference values in parentheses. 'DF': Single- jj -configuration Dirac-Fock. 'PW'= Present work.

Ref. ¹	DF(PW)	MCDF ²
0.156	0.626 (-2.297)	0.165

Table S 2 Cartesian coordinates of optimized structures of the hydrides studied. Lengths in Å, angles in degree. ZORA2c, PBE, TZ2P. ADF outputs bonding energy instead of total energy by default.

LuH, Bond energy = -15.97329756 a.u.			
Lu	0.000000	0.000000	-0.001116
H	0.000000	0.000000	1.893789
LrH, Bond energy = -206.85080914 a.u.			
Lr	0.000000	0.000000	-0.001003
H	0.000000	0.000000	1.958639
LuH ₂ , Bond energy = -16.10361981 a.u.			
Lu	0.000000	-0.054547	0.023366
H	1.601334	-0.872929	0.681107
H	-1.601334	-0.872929	0.681107
LrH ₂ , Bond energy = -206.96678954 a.u.			
Lr	0.000000	-0.021740	-0.003001
H	1.606069	-0.889333	0.694291
H	-1.606069	-0.889333	0.694291
LuH ₃ , Bond energy = -16.25528212 a.u.			
Lu	0.101430	0.000000	-0.000485
H	0.641310	-1.596787	0.921607
H	0.641310	1.596787	0.921607
H	0.640251	0.000000	-1.844728
LrH ₃ , Bond energy = -207.11349376 a.u.			
Lr	-0.131551	0.000000	0.002248
H	0.592051	-1.559353	-0.897596
H	0.592051	1.559353	-0.897596
H	0.591449	0.000000	1.802944

Table S 3 Cartesian coordinates of optimized structures of the trichlorides and monocarbonyls studied. Lengths in Å, angles in degree. ZORA2c, PBE, TZ2P.

LuCl ₃ , Bond energy = -16.42018352 a.u.			
Lu	0.000000	0.000000	0.000000
Cl	1.194041	2.068139	0.000000
Cl	1.194041	-2.068139	0.000000
Cl	-2.388081	0.000000	0.000000
LrCl ₃ , Bond energy = -207.25782680 a.u.			
Lr	0.000000	0.000000	0.186717
Cl	1.167489	2.022151	-0.462759
Cl	1.167489	-2.022151	-0.462759
Cl	-2.334978	0.000000	-0.462759
LuCO, Bond energy = -16.40676579 a.u.			
Lu	-0.000223	-0.000402	0.032047
C	-0.003583	-0.006479	2.329114
O	-0.005290	-0.009566	3.495924
LrCO, Bond energy = -207.28264400 a.u.			
Lr	-0.000373	-0.000674	-0.026497
C	-0.003576	-0.006466	2.357064
O	-0.005147	-0.009307	3.526518

Table S 4 Cartesian coordinates of optimized structures of $[(\text{Cp}')_3\text{M}]^-$, $\text{Cp}' = \text{C}_5\text{H}_4\text{SiMe}_3$. Lengths in Å. TPSSh, def2-TZVP, COSMO.

Total energy = -2814.630460158 a.u.			
Lr	-0.4531807	-0.2877815	-0.3841826
Si	2.5802096	-2.9304156	-1.0528258
Si	2.0802987	3.0248059	0.0947735
Si	-3.9540284	0.3293947	1.5668960
C	0.2290485	-1.9104392	-2.4903269
C	-1.1747461	-1.9705142	-2.4221018
C	-1.5232380	-2.6562810	-1.2281307
C	-0.3294503	-3.0177081	-0.5754978
C	0.7895600	-2.5244246	-1.3226513
C	2.9363403	-4.6833893	-1.6860098
C	3.6896134	-1.7463548	-2.0158931
C	3.0627326	-2.8974186	0.7714167
C	-0.6684139	2.4337808	-0.7723118
C	-1.3360849	1.7976800	-1.8483001
C	-0.3410147	1.2058229	-2.6726486
C	0.9086478	1.4143503	-2.0635294
C	0.7301831	2.1882723	-0.8705046
C	2.7703558	4.4599559	-0.9331810
H	3.5738928	4.9836262	-0.4029943
H	1.9870162	5.1900257	-1.1623422
H	3.1752174	4.1005165	-1.8849338
C	1.4231671	3.7618492	1.7039779
H	1.0867530	2.9926185	2.4034095
H	0.5806644	4.4353520	1.5165840
H	2.2074854	4.3451149	2.1988398
C	3.5227294	1.8686100	0.4725898
H	3.1953822	1.0047218	1.0574469
H	4.3064472	2.3920013	1.0318659
H	3.9712404	1.4886526	-0.4507429
C	-1.5705242	-1.3616476	1.8794222
C	-0.2047481	-1.2371670	2.1798833
C	0.1063043	0.1476018	2.2414814
C	-1.0847416	0.8643913	2.0003536
C	-2.1457418	-0.0585471	1.7222078
C	-4.2289983	2.0812044	0.9210123
C	-4.7774192	0.2279481	3.2729597
C	-4.8401501	-0.9027370	0.4450075
H	0.7990789	-1.4297090	-3.2732384
H	-1.8702756	-1.5630911	-3.1436794
H	-2.5284945	-2.8658925	-0.8888632
H	-0.2675356	-3.5420465	0.3680112
H	2.7050711	-4.7703941	-2.7529133
H	2.3263386	-5.4221093	-1.1552677
H	3.9886704	-4.9561702	-1.5469713
H	3.5665960	-0.7211807	-1.6569098
H	3.4558105	-1.7599342	-3.0853127
H	4.7433890	-2.0236986	-1.9027734
H	2.9371220	-1.8939141	1.1874783
H	4.1085145	-3.1981587	0.9011940
H	2.4453572	-3.5844082	1.3593670
H	-1.1525863	2.9986630	0.0107774
H	-2.3963260	1.8330537	-2.0560805
H	-0.5231326	0.6419304	-3.5759214
H	1.8579474	1.0600243	-2.4435877
H	-2.0962896	-2.2958037	1.7349096
H	0.4978495	-2.0477861	2.3132317
H	1.0745366	0.5670870	2.4755401
H	-1.1751340	1.9401403	1.9922529
H	-3.7345847	2.8222985	1.5574550
H	-5.2978764	2.3209382	0.9039283
H	-3.8385745	2.1974039	-0.0931065
H	-4.3210606	0.9378153	3.9709097
H	-4.6675460	-0.7734411	3.7022842
H	-5.8484280	0.4534589	3.2172911
H	-4.4438247	-0.8518184	-0.5731842
H	-5.9159037	-0.6972162	0.4100481
H	-4.7095181	-1.9295118	0.8021117

References

- 1 E. Eliav, U. Kaldor and Y. Ishikawa, *Phys. Rev. A*, 1995, **52**, 291–296.
- 2 S. Fritzsche, C. Z. Dong, F. Koike and A. Uvarov, *Eur. Phys. J. D*, 2007, **45**, 107–113.