Supporting information

Vibrational dynamics (IR, Raman, NRVS) and DFT study of new antitumor tetranuclear stannoxane cluster, Sn(IV)-oxo-{di-o-vanillin} dimethyl dichloride

Farukh Arjmand,^{a,*} Surbhi Sharma,^a Mohammad Usman,^a Bogdan M. Leu,^{b,*} Michael Y. Hu,^b Loic Toupet,^c David Gosztola,^d and Sartaj Tabassum^a

S	Table of contents	Page No.
No.		
1	Materials and instrumentation	S2
2	Computational methodology	S2
3	NRVS experimental	S3
4	Raman experimental	S3
5.	DNA binding experimental	S3
6.	Docking experimental	S4
7	Synthesis of 1	S4
8	Description of X-ray Crystal structure	S5
	Table for and stucture refinement data for crystal	S6
	Tables for selected bond lengths and angle of 1	S7, S8
9	Table for Mode Composition Factors, calculated frequency and Mode Description of various types of vibrations in 1	S8-S10
10	Far-IR spectrum of 1	S10
11	ESI Mass spectrum of 1	S11
12	UV-vis and emission spectra of [CT-DNA-1] system	S11, S12
13	In vitro antitumor activity	S12
14	Growth curve showing % control vs. drug or complex concentration (µg/ml)	S13, S14
15	DFT/B3LYP optimized structure of complex 1	S15
16	Tables for DFT/B3LYP calculated bond lengths and angle of 1 in comparison to the X-ray data	S15, S16
17	DFT-optimized structure coordinates of complex 1	S17, S18
18	References	S18, S19

Experimental Materials and instrumentation

o-vanillin, dimethyltin(IV) dichloride and triethylamine were purchased from Sigma Aldrich. All reagents were of the best commercial grade and were used without further purification. Elemental analysis was carried out on Carlo Erba Analyser Model 1106. Molar conductance was measured at room temperature on Eutech con 510 electronic conductivity bridge. Fourier-transform infrared (FTIR) spectra were recorded on an Interspec 2020 and Spectrum Two (Perkin Elmer) FTIR spectrometers. ESI-MS spectra were recorded on Micromass Quattro II triple quadrupole mass spectrometer. NMR spectra were obtained on a Bruker DRX-400 spectrometer with Me₂SO- d_6 as solvent. Electronic spectra were recorded on UV-1700 PharmaSpec UV-vis spectrophotometer (Shimadzu) in DMSO using cuvettes of 1 cm path length and data were reported in λ_{max}/nm .

Computational methodology

All computations reported within were performed using ORCA computational package.¹ Initial coordinates were taken from single crystal X-ray data and used for gas-phase ground state optimization. Gas-phase harmonic vibrational frequencies and IR intensities were calculated based on optimized geometry. The ground state geometry optimization and frequency analyses of the **1** was performed at the hybrid functional B3LYP level of DFT adding def2-TZVP basis set for Sn atom and def2-SVP basis set for all other atoms.² To accelerate the computation, we utilized the resolution of identity (RI) approximation with the decontracted auxiliary def2-TZV/J or def2-SVP/J Coulomb fitting basis sets and the chain-of-spheres (RIJCOSX) approximation to exact exchange as implemented in ORCA.³

The mode composition factor, representing the fraction of the total vibrational kinetic energy of each mode associated with the motion of a Sn atom, is calculated according to:⁴

$$\frac{(mr^2) Sn}{\sum mr^2 all atoms} = e_{Sn}^2$$
(S1)

Where m and r are the atomic mass and atomic displacements, respectively (gray bars in Fig. 3). For each predicted mode we generated a Lorentzian function with the area equal to e_{Sn}^2 and a full-width-at-half-maximum of 10 cm⁻¹. The sum of these functions, for all four Sn atoms, is shown as a red line in Fig. 3.

The NRVS experiment was carried out at beamline 30-ID of the Advanced Photon Source, Argonne National Laboratory⁵ at a temperature of 165 K, with an incident X-ray beam of 23.88 keV, and an energy resolution of 1.3 meV, measured by nuclear forward scattering on ¹¹⁹Sn-enriched tin oxides.⁵ Multiple scans were added to obtain the excitation probability, from which the partial vibrational density of states was extracted using the program PHOENIX.⁶

The Raman experiment was performed at the Center for Nanoscale Materials, Argonne National Laboratory at room temperature using 633-nm excitation from a helium-neon laser with 1-mW Incident power and a Raman microscope (inVia Reflex, Renishaw, Inc.). Scattered light was collected through a 50X objective (Leica, NA = 0.75). The spectra are the result of averaging 100 20-second integrations.

CT–DNA binding experiments were performed in Tris–HCl/NaCl (5:50 mM) buffer at pH 7.2 which included absorption spectral traces, and emission spectroscopy conformed to the standard methods and practices previously adopted by our laboratory.⁷⁻⁹ While measuring the absorption spectra an equal amount of CT–DNA was added to both the compound solution and the reference solution to eliminate the absorbance of the CT–DNA itself, and absorbance of the Tris buffer was subtracted through base line correction.

HEX 8.0 software was used for Molecular docking studies,¹⁰ which is an interactive molecular graphics program for calculating and displaying feasible docking modes of an enzymes and DNA molecule. Structure of the **1** was converted it into PDB format from mol format while the PDB file of the B–DNA dodecamer d(CGCGAATTCGCG)₂ (PDB ID: 1BNA), was downloaded from the protein data bank (http://www.rcsb.org./pdb). Visualization of the docked pose has been done by using CHIMERA (www.cgl.ucsf.edu/chimera), PyMol (http://pymol.sourceforge.net/) and Discovery Studio molecular graphics program.

Synthesis of tetranuclear Sn(IV)-oxo-{di-o-vanillin}dimethyl dichloride

A methanolic solution of dimethyltin(IV) dichloride (0.219 g, 1 mmol) was added drop wise to the solution of o-vanillin (0.152 g, 1 mmol) in the methanol in presence of triethylamine. The resulting mixture was refluxed for about 10 h until a yellow solid was precipitated which was filtered, washed with hexane yielding 67% of crude product. Suitable yellow crystals for single X-ray diffraction were obtained by recrystallizing the product in MeOH/DCM (1:4) mixture.

Yield: (67%); M.p. 197 °C; Anal. Calc. for C₂₄H₃₈Sn₄Cl₂O₈ (%): Calc. C, 28.82; H, 3.83. Found: C, 28.67; H, 3.89. FTIR (KBr pellet, v_{max}/cm^{-1}): 2918 v(C—H), 1633 v(C=O), 1026 v(OCH₃), 1469–446, v(aromatic C=C), 1244–1299 v(phenolic C—O), 522 and 566 v(Sn—C), 585 v(Sn— O), 186, 114 v(Sn—O—Sn).

UV–vis (1 x 10⁻³ M, DMSO, λ_{max} nm): 265 and 341. ¹H NMR (400 MHz, DMSO–*d*₆, ppm): 1.18–1.00 (s, 24H, Sn–CH₃), 3.83 (s, 3H, –OCH₃, vanillin), 6.91–7.39 (m, 6H, aromatic), 9.41 (s, 1H, aromatic CHO). ¹¹⁹Sn–NMR (149.19 MHz, DMSO–*d*₆, ppm): –134, –258 and –389. ESI–MS (DMSO) (*m/z*): 522 [C₁₁H₁₆Sn₂Cl₂O₃]⁺, 151 [C₈H₇O₃]⁺. CCDC: 1403573.

Description of X-ray Crystal structure

Single crystals suitable for X–ray study of the **1** were obtained after slow evaporation of the reaction mixture at room temperature. Single crystal X–ray structural studies of complex was performed on a CCD Oxford Diffraction X caliber Saphir 3 diffractometer employing graphite–monochromated Mo–K α radiation generated from a fine–focus sealed tube (λ = 0.71073Å) at 140(2) K. Data collection strategy was evaluated by using the Crys Alis Pro CCD software. Collections of data were observed by the standard ω scan techniques and were scaled and reduced using Crys Alis Pro RED software. The structure was solved by direct methods using SIR–97¹¹ and refined by least–squares methods on F^2 using SHELXL–97.¹² The positions of all atoms were obtained by direct methods. Anisotropic thermal parameter were assigned to all non–hydrogen atoms and the remaining hydrogen atoms were placed in geometrically constrained position and refined as riding atoms with a common fixed isotropic thermal parameter. The drawing of the complex was realized with PLATON.¹³

CCDC	1403573
formula	C ₂₄ H ₃₈ Sn ₄ Cl ₂ O ₈
Fw (g mol ⁻¹)	1000.28
crystal system	Monoclinic
space group	P 21/m
a (Å)	7.5613(2)
b (Å)	18.8087(3)
c (Å)	11.8932(2)
α (deg)	90
β(deg)	104.691(2)
γ (deg)	90
U (Å ³)	1636.13 (6)
Z	4
ρ_{calc} (g/cm ³)	2.030
$\mu (mm^{-1})$	25.853
F(000)	960.0
crystal size (mm)	0.284 x 0.215 x 0.114
Temp (K)	150 K
measured reflns	16206
unique reflns	3236
θ Range (deg)/ completeness	3.842 to 70.652 deg
(%)	
GOF ^a	1.050
$R^{b}[I > 2\sigma(I)]$	0.0600
w R_2^b (all data)	0.1557
largest diff. peak/hole (e.Å ⁻³)	2.028/-2.450

 Table S1 Crystal and stucture refinement data for 1.

^aGoF is defined as $\{\Sigma[w(F_0^2 - F_c^2)]/(n - P)\}^{1/2}$ where *n* is the number of data and *p* is the number of parameters. ^bR = $\{\Sigma||F_0| - |F_c||/\Sigma|F_0|, wR^2 = \{\Sigma w(F_0^2 - F_c^2)^2 / \Sigma w(F_0^2)^2\}^{1/2}$.

Bond lengths	(Å)
Sn(1)-O(4)	2.013(3)
Sn(1)-C(4)	2.102(5)
Sn(1)-C(3)	2.107(5)
Sn(1)-O(5)	2.140(3)
Sn(1)-O(3)	2.468(4)
Sn(2)-O(4)	2.061(5)
Sn(2)-C(6)	2.097(7)
Sn(2)-C(5)	2.096(7)
Sn(2)-O(3)#1	2.383(3)
Sn(2)-O(3)	2.383(3)
Sn(2)-O(1)	2.460(4)
Sn(2)-O(1)#1	2.460(4)
Sn(3)-O(5)	1.976(5)
Sn(3)-C(2)	2.116(8)
Sn(3)-C(1)	2.119(8)
Sn(3)- $Cl(1)$	2.6052(15)
Sn(3)-Cl(1)#1	2.6052(15)

Table S2 Selected bond lengths (Å) of 1.

 Table S3 Selected bond angles of 1

[deg]
74 16(17)
68.90(14)
142.89(14)
69.98(9)
69.98(9)
139.71(18)
143.76(9)
146.24(13)
73.98(13)
143.76(9)
73.97(13)
146.24(13)
72.29(18)
83.48(3)
83.48(3)
92.50(5)

C(1)-Sn(3)-Cl(1)#1	92.62(4)
Cl(1)-Sn(3)-Cl(1)#1	166.96(7)
Sn(2)-O(3)-Sn(1)	96.13(13)
Sn(1)-O(4)-Sn(1)#1	110.4(2)
Sn(1)-O(4)-Sn(2)	124.72(12)
Sn(1)#1-O(4)-Sn(2)	124.72(12)
Sn(3)-O(5)-Sn(1)#1	128.89(11)
Sn(3)-O(5)-Sn(1)	128.89(11)
Sn(1)#1-O(5)-Sn(1)	101.1(2)
Sn(3)-C(1)-H(1A)	109.5
Sn(3)-C(1)-H(1B)	109.5
Sn(3)-C(1)-H(1C)	109.5

Table S4 Mode composition factors, calculated frequency and mode description of various types of vibrations in **1**.

Type of vibration	Bonds involved	Calculated frequency (cm ⁻¹)	$e_{Sn(1)}^{2}$	$e_{Sn(2)}^{2}$	$e_{Sn(3)}^2$	$e_{Sn(4)}^{2}$	$\sum_{i=1}^{4} e_{Sn(i)}^{2}$	Mode description
Doming	All (O-Sn- O)	37, 108, 119, 129 153,154	0-0.03	0-0.03	0-0.03	0-0.05	0.05-0.07	Perpendicular to the plane
Out-of-	O ₁ -Sn ₂ -O ₆	47	0.00	0.02	0.01	0.00	0.03	Wagging
	$\begin{array}{c} O_1\text{-}Sn2\text{-}O_6\\ O_1\text{-}Sn_2\text{-}O_3\\ O_6\text{-}Sn_2\text{-}O_8\\ O_3\text{-}Sn_1\text{-}O_2\\ Sn_2\text{-}O_3\text{-}Sn_1\\ Sn_2\text{-}O_4\text{-}Sn_1 \end{array}$	51	0.00	0.01	0.00	0.00	0.02	Twisting Wagging Wagging Wagging Wagging Wagging
	$\begin{array}{c} O_1\text{-}Sn2\text{-}O_6\\ O_6\text{-}Sn_2\text{-}O_8 \end{array}$	59	0.00	0.01	0.01	0.01	0.03	Twisting Twisting
	$\begin{array}{c} O1\text{-}Sn2\text{-}O6\\ O_6\text{-}Sn_2\text{-}O_8\\ O_3\text{-}Sn_2\text{-}O_1 \end{array}$	166	0.01	0.00	0.02	0.00	0.03	Twisting
	$\begin{array}{c} O_3\text{-}Sn_2O_1\\ O_1\text{-}Sn_2O_6\\ O_6\text{-}Sn_2O_8\\ O_3\text{-}Sn_1O_2 \end{array}$	175	0.00	0.02	0.02	0.00	0.05	Twisting Twisting Twisting Twisting
	$O_3-Sn_2-O_1$	192	0.02	0.01	0.00	0.01	0.07	Wagging

-	1	1	1	1	1	1		
	$O_1-Sn_2-O_6$ $O_6-Sn_2-O_8$							
	O_3 -Sn ₁ - O_2							
	$\begin{array}{c} O_7\text{-}Sn_4\text{-}O_8\\ O_3\text{-}Sn_1\text{-}O_2\\ O_6\text{-}Sn_2\text{-}O_8 \end{array}$	215	0.00	0.00	0.00	0.02	0.03	Twisting Twisting Wagging
	O ₃ -Sn ₁ -O ₂	221	0.02	0.02	0.00	0.00	0.04	Twisting
	O ₆ -Sn ₂ -O ₈	225	0.02	0.02	0.00	0.02	0.06	Wagging
	O ₆ -Sn ₂ -O ₈	231	0.02	0.02	0.00	0.00	0.04	Wagging
	$O_5-Sn_1-O_4 \\ O_5-Sn_4-O_4$	250	0.02	0.00	0.01	0.01	0.04	Twisting Twisting
	O ₃ -Sn ₂ -O ₁	265	0.01	0.01	0.00	0.00	0.02	Wagging
	$O_5-Sn_1-O_4 \\ O_5-Sn_4-O_4$	289	0.00	0.00	0.00	0.00	0.01	Wagging Wagging
	$Sn_1-O_4-Sn_4$ $Sn_2-O_4-Sn_4$ $Sn_1-O_5-Sn_4$ $Sn_4-O_5-Sn_3$	304	0.03	0.00	0.00	0.00	0.03	Wagging
In-plane	$Sn_2-O_8-Sn_4$ $O_8-Sn_4-O_7$	390	0.00	0.00	0.00	0.00	0.00	Antisymm. stretching
	$\begin{array}{c} O_1\text{-}Sn_2\text{-}O_3\\ O_1\text{-}Sn_2\text{-}O_3\\ Sn_2\text{-}O_8\text{-}Sn_4\\ O_8\text{-}Sn_4\text{-}O_7 \end{array}$	414	0.01	0.00	0.00	0.00	0.01	Scissoring Scissoring Antisymm. stretching Scissoring
	O_1 -Sn ₂ - O_3 O_1 -Sn ₂ - O_3 Sn ₂ - O_3 -Sn ₁ O_8 -Sn ₄ - O_7	449	0.00	0.00	0.00	0.00	0.00	Rocking Symmetrical stretching
	$\begin{array}{c} O_1\text{-}Sn_2O_3\\ O_1\text{-}Sn_2O_3\\ Sn_2\text{-}O_8\text{-}Sn_4\\ O_8\text{-}Sn_4O_7\\ O_1\text{-}Sn_2O_6 \end{array}$	456	0.00	0.00	0.00	0.00	0.00	Rocking Rocking Symmetrical stretching Scissoring
	Sn ₁ -O ₅ -Sn ₄	494	0.00	0.00	0.00	0.00	0.02	Rocking
	$\begin{array}{c} Sn_4\text{-}O_5\text{-}Sn_3\\ Sn_1\text{-}O_5\text{-}Sn_4 \end{array}$	508	0.00	0.00	0.00	0.01	0.01	Antisymm. stretching
	C-Sn ₂ -C C-Sn ₄ -C	517	0.00	0.00	0.00	0.00	0.00	Symmetrical stretching
	C-Sn ₂ -C C-Sn ₄ -C	519	0.00	0.00	0.00	0.00	0.01	Symmetrical stretching

C-Sn ₃ -C	523	0.00	0.00	0.03	0.00	0.03	Symmetrical
							stretching
$Sn_2-O_8-Sn_4$	551	0.00	0.00	0.00	0.00	0.00	Antisymm.
O_8 -Sn ₄ - O_7							stretching
							Scissoring
C-Sn ₃ -C	561	0.00	0.00	0.06	0.00	0.06	Antisymm.
							stretching
$Sn_2-O_8-Sn_4$	563	0.01	0.00	0.00	0.00	0.01	Sym.str.
O_8 -Sn ₄ - O_7							Antisymm. Str.
C-Sn ₁ -C							Antisymm. Str.
O_3 -Sn ₁ - O_2							Antisymm. Str.
							Antisymm.
C-Sn ₄ -C	570	0.03	0.00	0.00	0.01	0.03	stretching
C-Sn ₁ -C							Antisymm.
O_3 -Sn ₁ - O_2							stretching
C-Sn ₄ -C	576	0.00	0.00	0.00	0.06	0.06	Antisymm.
							stretching
C-Sn ₁ -C	587	0.04	0.00	0.00	0.00	0.04	Antisymm. str.
$O_3-Sn_1-O_2$							Rocking
$Sn_1-O_3-Sn_2$							
	500	0.02	0.01	0.00	0.00	0.04	Antinum
C-Sn ₁ -C	590	0.03	0.01	0.00	0.00	0.04	Antisymm.
C Sm C	502	0.00	0.06	0.00	0.00	0.06	Antigumm
C-5112-C	392	0.00	0.00	0.00	0.00	0.00	Antisymm.
Sn O Sn	637	0.01	0.00	0.00	0.00	0.02	Pocking
$Sn_1-O_4-Sn_4$ $Sn_2-O_4-Sn_4$	037	0.01	0.00	0.00	0.00	0.02	Rocking
5112-04-5114							
Sn ₁ -O ₅ -Sn ₄	653	0.01	0.00	0.01	0.00	0.02	Rocking
$Sn_1 O_3 Sn_4$ $Sn_1 O_4 - Sn_4$	055	0.01	0.00	0.01	0.00	0.02	Rooking
Sn ₁ -O ₅ -Sn ₄	666	0.01	0.00	0.01	0.00	0.01	Rocking
$Sn_1-O_4-Sn_4$							0
Sn ₁ -O ₅ -Sn ₄	738	0.00	0.00	0.00	0.00	0.01	Symmetrical
Sn ₁ -O ₄ -Sn ₄							stretching
Sn ₁ -O ₄ -Sn ₂							
Cl-Sn ₃ -Cl	276	0.00	0.00	0.04	0.00	0.04	Antisymm.
							stretching



Fig. S1 Far-IR spectrum of 1.



Fig. S2 ESI mass spectrum of 1.



Fig. S3 UV–vis spectra of **1** in Tris–HCl buffer at pH 7.2 upon addition CT–DNA, $[DNA] = 0.00-4.00 \times 10^{-5} \text{ M}$, $[\mathbf{1}] = 1.0 \times 10^{-5} \text{ M}$. Arrow indicates change in absorbance with increasing concentration of CT–DNA.



Fig. S4 Effect of different concentrations of NaCl on emission spectrum of [CT–DNA–complex 1] system. Arrow indicates the gradual decrease in emission intensity as a function of NaCl concentration.

In vitro antitumor activity

The human cancer cell lines of different histological origin, used *in vitro* antitumor screening of **1** were following *viz.*, MCF–7 (breast), MIA–PA–CA–2 (pancreatic), HeLa (Cervix), and Hep-G2 (Hepatoma). Human malignant cell lines were procured and grown in RPMI–1640 medium supplemented with 10% Fetal Bovine Serum (FBS) and antibiotics to study growth pattern of

these cells. The proliferation of the cells upon treatment with chemotherapy was determined by means of the SRB semi–automated assay. All cell lines were seeded into 96 well plates and cells were counted and cell count was adjusted according to the titration readings so as to give optical density in the linear range (0.5 to 1.8) and were incubated at 37 °C in CO₂ incubator for 24 h. The stock solution of the complexes were prepared as 100 mg/ml in DMSO and four dilutions i.e. 10 µg/ml, 20 µg/ml, 40 µg/ml, 80 µg/ml, in triplicates were tested, each well receiving 90 µL of cell suspension. The plates were labeled properly and were incubated for 48 h. Termination of experiment was done by gently layering the cells with 50 µL of chilled 30% TCA (in case of adherent cells) and 50% TCA (in case of suspension cell lines) for cell fixation and kept at 4 °C for 1h. Plates stained with 50 µL of 0.4% SRB for 20 min. Fig. S5 depicts the antitumor screening data which shows % control growth *vs.* drug concentration (µgml⁻¹) against four human cancer cell lines.





Fig. S5 Growth curve showing % control *vs.* drug or complex concentration (μ g/ml) against three human cancer cell lines MIA–PA–CA–2 (pancreatic), HeLa (cervical), MCF-7 (breast) and Hep-G2 (Hepatoma).

DFT/B3LYP optimized structure

Gas phase optimized structure of tetrastannoxane, complex 1 is depicted in Fig. S6 and calculated bond length, angles and coordinates of complex 1 are given in Table S5 and S6 and Table S7, respectively. The calculated bond lengthes are longer than that of experimental data because the former were optimized in the gas phase and later were in tight crystal lattice. There is a good agreement between DFT/B3LYP optimized structure and experimental data.



Fig. S6 DFT/B3LYP optimized structure of complex 1

 Table S5. Selected bond lengths (Å) of complex 1.

Bond lengths	Experimental	Calculated
Sn(1)-O(4)	2.013(3)	2.033
Sn(1)-C(4)	2.102(5)	2.140
Sn(1)-C(3)	2.107(5)	2.142
Sn(1)-O(5)	2.140(3)	2.101
Sn(1)-O(3)	2.468(4)	2.694
Sn(2)-O(4)	2.061(5)	2.063
Sn(2)-C(6)	2.097(7)	2.134
$\operatorname{Sn}(2)$ -C(5)	2.096(7)	2.135
Sn(2)-O(3)#1	2.383(3)	2.231
Sn(2)-O(3)	2.383(3)	2.581
Sn(2)-O(1)	2.460(4)	2.423
Sn(2)-O(1)#1	2.460(4)	2.364
Sn(3)-O(5)	1.976(5)	2.000
Sn(3)-C(2)	2.116(8)	2.148
Sn(3)-C(1)	2.119(8)	2.147
Sn(3)- $Cl(1)$	2.6052(15)	2.505
Sn(3)-Cl(1)#1	2.6052(15)	2.693

 Bond Angle	Experimental	Calculated
 O(4)-Sn(1)-O(5)	74.16(17)	73.94
O(4)-Sn(1)-O(3)	68.90(14)	64.00
O(5)-Sn(1)-O(3)	142.89(14)	146.61
O(4)-Sn(2)-O(3)#1	69.98(9)	65.91
O(4)-Sn(2)-O(3)	69.98(9)	73.03
O(1)-Sn(2)-O(1)#1	72.29(18)	74.26
O(5)-Sn(3)-Cl(1)	83.48(3)	87.84
O(5)-Sn(3)-Cl(1)#1	83.48(3)	78.96
C(2)-Sn(3)-Cl(1)#1	92.50(5)	96.16
C(1)-Sn(3)-Cl(1)#1	92.62(4)	88.27
Cl(1)-Sn(3)-Cl(1)#1	166.96(7)	166.79
Sn(2)-O(3)-Sn(1)	96.13(13)	96.10
Sn(1)-O(4)-Sn(1)#1	110.4(2)	108.71
Sn(1)-O(4)-Sn(2)	124.72(12)	126.37
Sn(1)#1-O(4)-Sn(2)	124.72(12)	124.92
Sn(3)-O(5)-Sn(1)#1	128.89(11)	125.01
Sn(3)-O(5)-Sn(1)	128.89(11)	132.18
Sn(1)#1-O(5)-Sn(1)	101.1(2)	102.81

 Table S6. Selected bond angles [deg] of complex 1

 Table S7. DFT-optimized structure coordinates of complex 1.

Atom	Х	Y	Z
Sn	0 1/3307000	2 08/000000	7 808475000
Sn	-0.072553000	2.984990000	10 7035/7000
Sn	0.760644000	4.880004000	10.793347000
	0.700044000	7 160227000	4.438107000
	0.088894000	2 216712000	4.92012/000
0	-1.40//30000	0.291595000	240252000
0	-0.238046000	0.381383000	8.309233000
0	-0.702343000	2.434909000	9.933101000
0	-0.423823000	4./515/4000	6.809333000
C	0.231903000	4.332330000	0.380922000
U U	2.803/03000	4.0/1394000	4.00414/000
п	3.000207000	3.477339000	5.009192000
п	5.5/1525000	4.921575000	2.054676000
U U	-0.842416000	4./88134000	3.0340/0000
п	-0.33/403000	3.4988880000	2.2/3003000
П	-1.150858000	3.828333000	2.003333000
U U	1.912014000	2.420230000	8.014921000
п u	2.231/04000	2 242270000	7.094104000 8.167248000
п u	2.301100000	3.342370000	8.10/248000
Г	2.040822000	2 412550000	6.063776000
с ц	-2.020327000	2.412330000	6.903770000
н	-1.802415000	1.757545000	7 711861000
н	-2.049348000	3 326241000	6 619457000
C	1 012494000	4 753150000	11 565443000
н	1.039317000	4 981786000	12 639803000
н	1 397785000	3 738521000	11 392252000
C	-3 080252000	4 771855000	10 470481000
H	-3.628466000	4.933287000	11.408161000
Н	-3.362359000	5.541506000	9.738232000
С	-1.497345000	2.100060000	12.758959000
Н	-1.785653000	1.743386000	13.775400000
С	-1.214611000	1.031789000	11.824966000
С	-1.342386000	-0.294824000	12.319277000
Н	-1.638064000	-0.432723000	13.362708000
С	-1.118235000	-1.386281000	11.509486000
Н	-1.224216000	-2.401989000	11.895470000
С	-0.752416000	-1.180621000	10.161737000
Η	-0.569625000	-2.042059000	9.518281000
С	-0.614694000	0.100941000	9.654434000
С	-0.839489000	1.265329000	10.467371000
С	-0.089457000	-0.659008000	7.423786000
Н	0.166854000	-0.172085000	6.473322000
Н	0.728263000	-1.336698000	7.721689000
Н	-1.020824000	-1.239279000	7.306159000
Н	3.250928000	3.715166000	3.683947000
Н	-1.697672000	5.218295000	3.595863000
Н	1.641468000	5.468985000	11.018376000
H	-3.334303000	3.783751000	10.060858000
Sn	-0.068059000	6.293512000	7.523186000

Cl	0.692126000	2.016746000	4.582253000
0	-1.476969000	6.204271000	12.685632000
0	-0.285665000	8.963165000	8.381615000
0	-0.750566000	6.968542000	10.039958000
С	1.911798000	6.868910000	8.099236000
Η	2.276899000	7.673844000	7.449609000
Η	2.558734000	5.988081000	975660000
Н	1.917395000	7.190022000	9.148947000
С	-2.060770000	6.872856000	6.993084000
Η	-2.043540000	7.677245000	6.246917000
Н	-2.608301000	7.194792000	7.888194000
Η	-2.550666000	5.990493000	6.555446000
С	-1.509866000	7.429928000	12.841421000
Н	-1.788014000	7.804725000	13.852166000
С	-1.236935000	8.461078000	11.874069000
С	-1.349810000	9.807889000	12.318062000
Н	-1.637157000	9.991021000	13.356858000
С	-1.116245000	10.857661000	11.460835000
Н	-1.207462000	11.890506000	11.802665000
С	-0.756107000	10.591088000	10.120011000
Н	-0.565868000	11.427149000	9.446470000
С	-0.633929000	9.291045000	9.656298000
С	-0.872947000	8.166057000	10.522869000
С	-0.049819000	9.987210000	7.433124000
Н	0.214285000	9.483787000	6.493418000
Н	0.784634000	10.636940000	7.748185000
Н	-0.952770000	10.604138000	7.281922000

References

- 1. F. Neese, *Wiley Interdiscip. Rev.: Comput. Mol. Sci. 2*, 2012, 73-78; F. Neese, "Orca." An ab Initio, *Density Functional and Semiempirical Program Package version 2*, 2009.
- 2. C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B:*, *Condens. Matter Mater. Phys.*, 1988, **37**, 785; A. Schaefer, H. Horn and R. Ahlrichs, *J. Chem. Phys.*, 1992, **97**, 2571.
- F. Weigend and R. Ahlrichs, *Phys. Chem. Chem. Phys.*, 2005, 7, 3297-3305; A. Schaefer,
 C. Huber and R. Ahlrichs, *J. Chem. Phys.*, 1994, 100, 5829; S. Grimme, S. Ehrlich and L. Goerigk, *J. Comput. Chem.*, 2011, 32, 1456; S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, 132, 154104.
- 4. B. M. Leu, M. Z. Zgierski, C. Bischoff, M. Li, M. Y. Hu, J. Zhao, S. W. Martin, E. E. Alp and W. R. Scheidt, *Inorg. Chem.*, 2013, **52**, 9948.
- T. S. Toellner, A. Alatas and A. H. Said, *J. Synchrotron Radiat.*, 2011, 18, 605-611; B. M. Leu, M. Sturza, M. Y. Hu, D. Gosztola, V. Baran, T. F. Fassler and E. E. Alp, *Phys. Rev. B.*, 2014, **90**, 104304.

- 6. W. Sturhahn, *Hyperfine Interactions*, 2000, **125**, 149.
- F. Arjmand and M. Muddassir, *Chirality*, 2011, 23, 250–259; I. Yousuf, F. Arjmand, S. Tabassum, L. Toupet, R. A. Khan and M. A. Siddiqui, *Dalton Trans.*, 2015, 44, 10330–10342.
- 8. J. R. Lakowicz and G. Webber, *Biochemistry*, 1973, 12, 4161.
- 9. F. Arjmand, I. Yousuf, M. Afzal and L. Toupet, *Inorg. Chim. Acta*, 2014, **421**, 26–37; M. Chauhan, K. Banerjee and F. Arjmand, *Inorg. Chem.*, 2007, **46**, 3072.
- D. Mustard and D. W. Ritchie, *Proteins: Struct. Funct. Bioinf.*, 2005, 60, 269–274; G. Macindoe, L. Mavridis, V. Venkatraman, M.–D. Devignes and D. W. Ritchie, *Nucl. Acids Res.*, 2010, 38, 445.
- A. Altomare, M. C. Burla, M. Camalli, G. L. Cascarano, C. Giocavazzo, A. Guagliardi, A. G. C. Moliterni, G. Polidori and S. Spagna, J. Appl. Crystallogr., 1999, 32, 115.
- 12. G. M. Sheldrick, SHELX–97, Program for Crystal Structure refinement, University of Göttingen, Germany, 1997.
- 13. A. L. Spek, PLATON Procedure, *A multipurpose Crystallographic Tool, Utrecht University*, Utrecht, The Netherlands, 1998.