

Electronic Supplementary Information

Feynman Force Components:

Basis for a Solution to the Covalent vs. Ionic Dilemma

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Table S1. Cartesian coordinates (Å) and total energies (au) for systems investigated in the study calculated at the MP2/aug-cc-pVTZ level of theory).

He ₂ (E= -5.7896308)			
He	0.000000	0.000000	0.000000
He	0.000000	0.000000	3.101452
H ₂ (E= -1.1650231)			
H	0.000000	0.000000	0.000000
H	0.000000	0.000000	0.737437
N ₂ (E= -109.3647998)			
N	0.000000	0.000000	0.000000
N	0.000000	0.000000	1.114093
HF (E= -100.3408907)			
H	0.000000	0.000000	0.000000
F	0.000000	0.000000	0.921838
HCl (E= -460.3151301)			
H	0.000000	0.000000	0.000000
Cl	0.000000	0.000000	1.274785
HBr (E= -2573.2952874)			
H	0.000000	0.000000	0.000000
Br	0.000000	0.000000	1.406605
LiF (E= -107.2702157)			
Li	0.000000	0.000000	0.000000
F	0.000000	0.000000	1.594671
LiCl (E= -467.2589174)			
Li	0.000000	0.000000	0.000000
Cl	0.000000	0.000000	2.048547
LiBr (E= -2580.2432709)			
Li	0.000000	0.000000	0.000000
Br	0.000000	0.000000	2.196729
NaF (E= -261.658502)			
Na	0.000000	0.000000	0.000000
F	0.000000	0.000000	1.998947
NaCl (E= -621.661592)			
Na	0.000000	0.000000	0.000000
Cl	0.000000	0.000000	2.412567
NaBr (E= -2734.6489302)			
Na	0.000000	0.000000	0.000000
Br	0.000000	0.000000	2.553034

Table S2. Feynman force components (au) values for systems investigated in the study (the non-zero components – along the bond axis are given, force orientation refers to molecular geometries from Table S1).

System	Force component	Value	System	Force component	Value
He ₂	$\bar{F}_{edHe1}^{nHe1}$	5.58E-05	HBr	$\bar{F}_{edH}^{nH}$	2.04E-01
	$\bar{F}_{edHe2}^{nHe1}$	1.16E-01		$\bar{F}_{edBr}^{nH}$	4.75E+00
	$\bar{F}_{nHe2}^{nHe1}$	-1.16E-01		$\bar{F}_{nBr}^{nH}$	-4.95E+00
	$\bar{F}_{He2}^{nHe1}$	-5.61E-05		$\bar{F}_{Br}^{nH}$	-2.04E-01
H ₂	$\bar{F}_{edH1}^{nH1}$	1.76E-01		$\bar{F}_{edBr}^{nBr}$	-1.03E+00
	$\bar{F}_{edH2}^{nH1}$	3.39E-01		$\bar{F}_{edH}^{nBr}$	-3.92E+00
	$\bar{F}_{nH2}^{nH1}$	-5.15E-01		$\bar{F}_{nH}^{nBr}$	4.95E+00
	$\bar{F}_{H2}^{nH1}$	-1.76E-01		$\bar{F}_H^{nBr}$	1.03E+00
N ₂	$\bar{F}_{edN1}^{nN1}$	1.49E+00	LiF	$\bar{F}_{edLi}^{nLi}$	-2.38E-01
	$\bar{F}_{edN2}^{nN1}$	9.56E+00		$\bar{F}_{edF}^{nLi}$	3.21E+00
	$\bar{F}_{nN2}^{nN1}$	-1.11E+01		$\bar{F}_{nF}^{nLi}$	-2.97E+00
	$\bar{F}_{N2}^{nN1}$	-1.49E+00		$\bar{F}_F^{nLi}$	2.38E-01
HF	$\bar{F}_{edH}^{nH}$	-1.03E-01		$\bar{F}_{edF}^{nF}$	-9.46E-01
	$\bar{F}_{edF}^{nH}$	3.07E+00		$\bar{F}_{edLi}^{nF}$	-2.03E+00
	$\bar{F}_{nF}^{nH}$	-2.97E+00		$\bar{F}_{nLi}^{nF}$	2.97E+00
	$\bar{F}_F^{nH}$	1.03E-01		$\bar{F}_{Li}^{nF}$	9.46E-01
	$\bar{F}_{edF}^{nF}$	-2.50E+00	LiCl	$\bar{F}_{edLi}^{nLi}$	-1.63E-01
	$\bar{F}_{edH}^{nF}$	-4.62E-01		$\bar{F}_{edCl}^{nLi}$	3.57E+00
	$\bar{F}_{nH}^{nF}$	2.97E+00		$\bar{F}_{nCl}^{nLi}$	-3.40E+00
	$\bar{F}_H^{nF}$	2.50E+00		$\bar{F}_{Cl}^{nLi}$	1.63E-01
HCl	$\bar{F}_{edH}^{nH}$	1.49E-01		$\bar{F}_{edCl}^{nCl}$	-1.06E+00
	$\bar{F}_{edCl}^{nH}$	2.78E+00		$\bar{F}_{edLi}^{nCl}$	-2.34E+00
	$\bar{F}_{nCl}^{nH}$	-2.93E+00		$\bar{F}_{nLi}^{nCl}$	3.40E+00
	$\bar{F}_{Cl}^{nH}$	-1.49E-01		$\bar{F}_{Li}^{nCl}$	1.06E+00
	$\bar{F}_{edCl}^{nCl}$	-1.25E+00	LiBr	$\bar{F}_{edLi}^{nLi}$	-1.45E-01
	$\bar{F}_{edH}^{nCl}$	-1.68E+00		$\bar{F}_{edBr}^{nLi}$	6.24E+00
	$\bar{F}_{nH}^{nCl}$	2.93E+00		$\bar{F}_{nBr}^{nLi}$	-6.09E+00
	$\bar{F}_H^{nCl}$	1.25E+00		$\bar{F}_{Br}^{nLi}$	1.45E-01

System	Force component	Value
LiBr	$\bar{F}_{edBr}^{nBr}$	-1.89E+00
	$\bar{F}_{edLi}^{nBr}$	-4.21E+00
	$\bar{F}_{nLi}^{nBr}$	6.09E+00
	$\bar{F}_{Li}^{nBr}$	1.89E+00
NaF	$\bar{F}_{edNa}^{nNa}$	-6.05E-01
	$\bar{F}_{edF}^{nNa}$	7.54E+00
	$\bar{F}_{nF}^{nNa}$	-6.94E+00
	$\bar{F}_F^{nNa}$	6.05E-01
	$\bar{F}_{edF}^{nF}$	-5.97E-01
	$\bar{F}_{edNa}^{nF}$	-6.34E+00
	$\bar{F}_{nNa}^{nF}$	6.94E+00
	$\bar{F}_{Na}^{nF}$	5.97E-01
NaCl	$\bar{F}_{edNa}^{nNa}$	-4.32E-01
	$\bar{F}_{edCl}^{nNa}$	9.43E+00
	$\bar{F}_{nCl}^{nNa}$	-9.00E+00
	$\bar{F}_{Cl}^{nNa}$	4.32E-01
	$\bar{F}_{edCl}^{nCl}$	-7.57E-01
	$\bar{F}_{edNa}^{nCl}$	-8.24E+00
	$\bar{F}_{nNa}^{nCl}$	9.00E+00
	$\bar{F}_{Na}^{nCl}$	7.57E-01
NaBr	$\bar{F}_{edNa}^{nNa}$	-3.86E-01
	$\bar{F}_{edBr}^{nNa}$	1.69E+01
	$\bar{F}_{nBr}^{nNa}$	-1.65E+01
	$\bar{F}_{Br}^{nNa}$	3.86E-01
	$\bar{F}_{edBr}^{nBr}$	-1.37E+00
	$\bar{F}_{edNa}^{nBr}$	-1.52E+01
	$\bar{F}_{nNa}^{nBr}$	1.65E+01
	$\bar{F}_{Na}^{nBr}$	1.37E+00

Table S3. Electronegativity, force (au) and force ratio data for heteropolar molecules investigated in the study [$\chi(B) - \chi(A)$ is given in Pauling (P), Allred-Rochow (AR) and Allen (A) scales, atom A refers to an atom of partial positive charge (H, Li or Na)].

System	P	AR	A	$\bar{F}_B^{nA}$	$-\frac{F_B^{nA}}{F_{nB}^{nA}}$
HF	1.78	1.90	1.89	0.103	0.0347
HCl	0.96	0.63	0.57	-0.149	-0.0510
HBr	0.76	0.54	0.38	-0.204	-0.0411
LiF	3.00	3.13	3.28	0.238	0.0801
LiCl	2.18	1.86	1.96	0.163	0.0478
LiBr	1.98	1.77	1.77	0.145	0.0238
NaF	3.05	3.09	3.32	0.605	0.0872
NaCl	2.23	1.82	2.00	0.432	0.0481
NaBr	2.03	1.73	1.82	0.386	0.0233