

Electronic Supporting Information

for

Isomerization of Methoxy Radical Revisited: Impact of Water Dimer

Pradeep Kumar^a, Partha Biswas^b and Biman Bandyopadhyay^a

^a*Department of Chemistry, Malaviya National Institute of Technology Jaipur, J. L. N. Marg, Jaipur – 302017, India.*

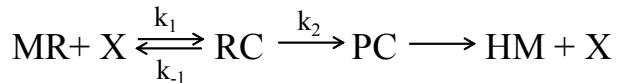
^b*Department of Chemistry, Scottish Church College, 1 & 3 Urquhart Square, Kolkata-700006, India.*

Table of contents

S. No	Caption	Page No.
1	Rate calculations	3
2	Table S1 Relative energies (kcal mol ⁻¹) of all species with respect to the isolated reactants calculated using two different levels of theory	5
3	Table S2 Rate constants (molecule ⁻¹ cm ³ s ⁻¹) in presence of the four catalysts calculated using results obtained from CCSD(T)/aug-cc-pVTZ//MP2/aug-cc-pVTZ level of theory	6
4	Table S3 Concentration of the four catalysts (in molecules cm ⁻³) within 280 K to 320 K temperature range at 0 km altitude	7
5	Table S4 Effective rate constants (s ⁻¹) in presence of the four catalysts within 280 K to 320 K temperature range at 0 km altitude	8
6	Table S5 Effective rate constants (s ⁻¹) in presence of the four catalysts within 280 K to 320 K temperature range at 0 km altitude (MESMER)	9
7	Table S6 Optimized geometries in Cartesian coordinates and normal mode frequencies of all species calculated at MP2/aug-cc-pVTZ level of theory	10
14	Table S7 Absolute energies (in Hartree) of all species involved in isomerization of MR calculated at the two different levels of theory	14
References		15

Rate Calculations:

Isomerization of MR can be written in terms of the bimolecular encounters between MR and catalyst, say X (X = WM, WD, FA and SA) to give HM, as:



In the above reaction scheme RC is assumed to be in equilibrium with MR and X and subsequently RC undergoes unimolecular isomerization through TS to form PC. Under this assumption, the reaction rate (v) can be expressed as:

$$v = \frac{k_1}{k_{-1}} k_2 [\text{MR}][\text{X}] = K_{eq} k_2 [\text{MR}][\text{X}] = k [\text{MR}][\text{X}]$$

$$\text{where } K_{eq} = \frac{k_1}{k_{-1}} \quad \text{and} \quad k = K_{eq} k_2$$

Where, K_{eq} = equilibrium constant of RCs ,

k_2 = rate constant for the unimolecular reaction of RCs leading to PCs.

K_{eq} and k_2 have been computed by using TST as shown below:

$$K_{eq} = \frac{Q_{RC}}{Q_{MR} Q_X} e^{-\frac{(E_{RC} - E_{MR})}{RT}}$$

$$k_2 = \sigma \Gamma \frac{k_B T Q_{TS}}{h Q_{RC}} e^{-\frac{(E_{TS} - E_{RC})}{RT}}$$

Where, \mathcal{Q} = The product of electronic, translational, rotational and vibrational canonical partition functions of the respective species referenced to the zero-point energy (ZPE),

E = ZPE corrected energies of the respective species,

= reaction path number or reaction degeneracy,

Γ = tunneling correction for the reaction (it has been taken into account by assuming unsymmetrical Eckart barrier.^y

k_B = Boltzmann constant

h = Planck constant

T = temperature in Kelvin

R = ideal gas constant.

The inherent assumption behind this approximation is that the time scale to establish the equilibrium is significantly different (fast) as compared to the time scale of second step of reaction (unimolecular conversion). This assumption will work well if the catalyst molecules are present at sufficiently large concentration. Calculations of the rate constants were carried out by use of TheRate program package.¹

It is evident from reaction rate (v) equation that if all other conditions remain same, then $k[X]$ (where, k is the rate constant and $[X]$ is the concentration of the catalyst) can be considered as an effective rate constant:

$$k_{eff} = k[X]$$

To further assess the pre-equilibrium approximation, we have also carried out the rate calculations using MESMER software.² MESMER solves a multi grain master equation to obtain

the phenomenological rate constants which is known as Bartis-Widom phenomenological rate coefficients.

Table S1 Relative energies (kcal mol⁻¹) of all species with respect to the isolated reactants calculated using two different levels of theory

Catalyst	Species	MP2/aug-cc-pVTZ	CCSD(T)/aug-cc-pVTZ
None	MR	0	0
	TS	30.0	32.7
	HM	-14.1	-8.7
WM	MR-WM	-5.1	-5.2
	TS _{WM}	19.7	22.9
	HM-WM	-20.8	-15.4
WD	MR-WD	-8.1	-8.4
	TS _{WD}	5.3	7.8
	HM-WD	-25.8	-20.5
FA	MR-FA	-8.5	-8.8
	TS _{FA}	3.8	4.4
	HM-FA	-24.1	-18.9
SA	MR-SA	-11.4	-11.8
	TS _{SA}	1.3	1.8
	HM-SA	-25.5	-20.3

Table S2 Rate constants (molecule⁻¹ cm³ s⁻¹) in presence of the four catalysts calculated using results obtained from CCSD(T)/aug-cc-pVTZ//MP2/aug-cc-pVTZ level of theory

Temperatures	WM	WD	FA	SA
298	3.55×10^{-29}	2.82×10^{-21}	3.65×10^{-16}	2.80×10^{-13}
259	3.55×10^{-29}	2.82×10^{-21}	6.20×10^{-16}	1.28×10^{-12}
230	9.08×10^{-31}	8.46×10^{-22}	1.40×10^{-15}	7.13×10^{-12}
213	7.08×10^{-32}	3.39×10^{-22}	2.89×10^{-15}	2.74×10^{-11}
216	1.91×10^{-32}	2.15×10^{-22}	2.50×10^{-15}	2.11×10^{-11}
219	2.38×10^{-32}	2.30×10^{-22}	2.18×10^{-15}	1.65×10^{-11}
224	2.97×10^{-32}	2.49×10^{-22}	1.76×10^{-15}	1.11×10^{-11}
235	4.37×10^{-32}	2.85×10^{-22}	1.17×10^{-15}	5.06×10^{-12}
250	1.07×10^{-31}	3.94×10^{-22}	7.61×10^{-16}	2.04×10^{-12}
280	6.56×10^{-30}	1.64×10^{-21}	4.39×10^{-16}	5.16×10^{-13}
290	1.68×10^{-29}	2.22×10^{-21}	3.92×10^{-16}	3.61×10^{-13}
300	4.27×10^{-29}	2.98×10^{-21}	3.60×10^{-16}	2.64×10^{-13}
310	1.06×10^{-28}	3.96×10^{-21}	3.38×10^{-16}	2.00×10^{-13}
320	2.57×10^{-28}	5.20×10^{-21}	3.18×10^{-16}	1.56×10^{-13}

Table S3 Concentration of the four catalysts (in molecules cm⁻³) within 280 K to 320 K temperature range at 0 km altitude. (% H stands for relative humidity)

Catalyst		280	290	298	300	310	320
WM ^a	20% H	5.16×10^{16}	9.56×10^{16}	1.55×10^{17}	1.72×10^{17}	2.92×10^{17}	4.70×10^{17}
	40% H	1.03×10^{17}	1.91×10^{17}	3.09×10^{17}	3.43×10^{17}	5.84×10^{17}	9.40×10^{17}
	60% H	1.55×10^{17}	2.87×10^{17}	4.64×10^{17}	5.15×10^{17}	8.77×10^{17}	1.41×10^{18}
	80% H	2.07×10^{17}	3.82×10^{17}	6.18×10^{17}	6.86×10^{17}	1.17×10^{18}	1.88×10^{18}
	100% H	2.58×10^{17}	4.78×10^{17}	7.73×10^{17}	8.58×10^{17}	1.46×10^{18}	2.35×10^{18}
WD ^a	20% H	8.18×10^{12}	2.36×10^{13}	5.44×10^{13}	6.50×10^{13}	1.63×10^{14}	3.71×10^{14}
	40% H	3.27×10^{13}	9.46×10^{13}	2.18×10^{14}	2.60×10^{14}	6.52×10^{14}	1.48×10^{15}
	60% H	7.36×10^{13}	2.13×10^{14}	4.90×10^{14}	5.85×10^{14}	1.46×10^{15}	3.33×10^{15}
	80% H	1.31×10^{14}	3.78×10^{14}	8.70×10^{14}	1.04×10^{15}	2.60×10^{15}	5.90×10^{15}
	100% H	2.04×10^{14}	5.91×10^{14}	1.36×10^{15}	1.62×10^{15}	4.06×10^{15}	9.24×10^{15}
FA ^b	High	2.61×10^{11}	2.52×10^{11}	2.45×10^{11}	2.43×10^{11}	2.36×10^{11}	2.28×10^{11}
	Average	2.02×10^{10}	1.95×10^{10}	1.90×10^{10}	1.89×10^{10}	1.83×10^{10}	1.77×10^{10}
	Low	2.61×10^{11}	2.52×10^{11}	2.45×10^{11}	2.43×10^{11}	2.36×10^{11}	2.28×10^{11}
SA ^c		3.94×10^8	3.80×10^8	3.70×10^8	3.68×10^8	3.56×10^8	3.45×10^8

^a Reference 3 (concentration of WM and WD has directly been taken from this reference where the authors computed WD concentration from WM concentration and calculated equilibrium constant for WM to WD transformation) ^b Reference 4 ^c Reference 5

Table S4 Effective rate constants (s^{-1}) in presence of the four catalysts within 280 K to 320 K temperature range at 0 km altitude calculated by TheRate. (% H stands for relative humidity)

Catalyst	Conc.	280	290	298	300	310	320
WM	20% H	3.38×10^{-13}	1.61×10^{-12}	5.50×10^{-12}	7.34×10^{-12}	3.10×10^{-11}	1.21×10^{-10}
	40% H	6.76×10^{-13}	3.22×10^{-12}	1.10×10^{-11}	1.46×10^{-11}	6.19×10^{-11}	2.41×10^{-10}
	60% H	1.02×10^{-12}	4.83×10^{-12}	1.65×10^{-11}	2.20×10^{-11}	9.30×10^{-11}	3.62×10^{-10}
	80% H	1.36×10^{-12}	6.43×10^{-12}	2.19×10^{-11}	2.93×10^{-11}	1.24×10^{-10}	4.83×10^{-10}
	100% H	1.69×10^{-12}	8.05×10^{-12}	2.74×10^{-11}	3.66×10^{-11}	1.55×10^{-10}	6.04×10^{-10}
WD	20% H	1.34×10^{-8}	5.25×10^{-8}	1.53×10^{-7}	1.94×10^{-7}	6.46×10^{-7}	1.93×10^{-6}
	40% H	5.37×10^{-8}	2.10×10^{-7}	6.14×10^{-7}	7.76×10^{-7}	2.58×10^{-6}	7.70×10^{-6}
	60% H	1.21×10^{-7}	4.74×10^{-7}	1.38×10^{-6}	1.75×10^{-6}	5.78×10^{-6}	1.73×10^{-5}
	80% H	2.15×10^{-7}	8.41×10^{-7}	2.45×10^{-6}	3.10×10^{-6}	1.03×10^{-5}	3.07×10^{-5}
	100% H	3.35×10^{-7}	1.31×10^{-6}	3.83×10^{-6}	4.83×10^{-6}	1.61×10^{-5}	4.81×10^{-5}
FA	High	1.14×10^{-4}	9.87×10^{-5}	8.95×10^{-5}	8.75×10^{-5}	7.95×10^{-5}	7.26×10^{-5}
	Average	8.87×10^{-6}	7.65×10^{-6}	6.94×10^{-6}	6.79×10^{-6}	6.17×10^{-6}	5.63×10^{-6}
	Low	1.14×10^{-7}	9.87×10^{-8}	8.95×10^{-8}	8.75×10^{-8}	7.95×10^{-8}	7.26×10^{-8}

SA	2.03×10^{-4}	1.37×10^{-4}	1.04×10^{-4}	9.69×10^{-5}	7.10×10^{-5}	5.37×10^{-5}
----	-----------------------	-----------------------	-----------------------	-----------------------	-----------------------	-----------------------

Table S5 Effective rate constants (s^{-1}) in presence of the four catalysts within 280 K to 320 K temperature range at 0 km altitude calculated by MESMER. (% H stands for relative humidity)

Catalyst	Conc.	280	290	298	300	310	320
WM	20% H	3.77×10^{-13}	1.74×10^{-12}	5.81×10^{-12}	7.71×10^{-12}	3.19×10^{-11}	1.22×10^{-10}
	40% H	7.52×10^{-13}	3.47×10^{-12}	1.16×10^{-11}	1.54×10^{-11}	6.34×10^{-11}	2.45×10^{-10}
	60% H	1.13×10^{-12}	5.21×10^{-12}	1.74×10^{-11}	2.31×10^{-11}	9.59×10^{-11}	3.67×10^{-10}
	80% H	1.51×10^{-12}	6.94×10^{-12}	2.32×10^{-11}	3.08×10^{-11}	1.28×10^{-10}	4.89×10^{-10}
	100% H	1.88×10^{-12}	8.68×10^{-12}	2.9×10^{-11}	3.85×10^{-11}	1.6×10^{-10}	6.11×10^{-10}
WD	20% H	1.31×10^{-8}	5.05×10^{-8}	1.46×10^{-7}	1.85×10^{-7}	6.09×10^{-7}	1.80×10^{-6}
	40% H	5.22×10^{-8}	2.02×10^{-7}	5.86×10^{-7}	7.39×10^{-7}	2.44×10^{-6}	7.20×10^{-6}
	60% H	1.18×10^{-7}	4.56×10^{-7}	1.32×10^{-6}	1.66×10^{-6}	5.45×10^{-6}	1.62×10^{-5}
	80% H	2.09×10^{-7}	8.09×10^{-7}	2.34×10^{-6}	2.95×10^{-6}	9.71×10^{-6}	2.87×10^{-5}
	100% H	3.26×10^{-7}	1.26×10^{-6}	3.65×10^{-6}	4.6×10^{-6}	1.51×10^{-5}	4.49×10^{-5}
FA	High	7.93×10^{-5}	7.44×10^{-5}	7.1×10^{-5}	7.12×10^{-5}	6.71×10^{-5}	6.44×10^{-5}
	Average	6.14×10^{-6}	5.76×10^{-6}	5.5×10^{-6}	5.46×10^{-6}	5.20×10^{-6}	5.00×10^{-6}
	Low	7.93×10^{-8}	7.44×10^{-8}	7.1×10^{-8}	7.12×10^{-8}	6.71×10^{-8}	6.44×10^{-8}

SA	2.95×10^{-5}	2.66×10^{-5}	2.45×10^{-5}	2.4×10^{-5}	2.16×10^{-5}	1.95×10^{-5}
----	-----------------------	-----------------------	-----------------------	----------------------	-----------------------	-----------------------

Table S6 Optimized geometries in Cartesian coordinates and normal mode frequencies of all species calculated at MP2/aug-cc-pVTZ level of theory

Species	Cartesian coordinate (Å)			Vibrational frequencies (cm ⁻¹)		
MR	C	0.581872	0.000089	-0.010498		
	H	0.865280	-0.004096	1.050892		
	H	0.999450	0.903630	-0.458822		
	H	0.999079	-0.900437	-0.465240		
	O	-0.794380	0.000046	-0.007980		
HM	C	0.684714	0.027438	-0.066480		
	H	1.113909	0.987547	0.173444		
	H	1.229154	-0.882854	0.109015		
	O	-0.670357	-0.125781	0.021718		
	H	-1.088491	0.736930	-0.057321		
TS	C	-0.628082	0.000001	0.006767		
	H	-1.151053	0.934246	-0.171669		
	H	-1.151052	-0.934234	-0.171728		
	O	0.742190	0.000004	-0.084151		
	H	0.133075	-0.000051	0.976004		
MR-WM	C	-1.304759	-0.496026	-0.011360		
	H	-0.603971	-1.187156	-0.482799		
	H	-2.305564	-0.564287	-0.438409		
	H	-1.365238	-0.771375	1.050999		
	O	-0.824776	0.798388	-0.008100		
	H	1.119029	0.407694	0.017476		
	H	2.646450	0.355046	-0.003530		
	O	1.867007	-0.206359	-0.001346		
HM-WM	C	1.594154	0.453076	-0.049929		
	H	2.647667	0.262332	0.057840		
	H	1.164833	1.393885	0.260395		
	O	0.817172	-0.661979	-0.001150		
	H	-2.475492	-0.249859	0.639227		
	H	-0.114948	-0.389470	0.012611		
	H	-2.324381	0.137155	-0.828992		
	O	-1.874998	0.177916	0.020961		

$T\bar{S}_{\text{WD}}$	C	-1.006422	0.515997	0.008835	-1381.2761	188.7210	373.6247
	H	0.363026	0.769681	0.015035	402.1024	553.3765	617.1818
	H	-1.340918	1.026908	0.923637	821.99	1138.1143	1197.4839
	H	-1.273163	1.033722	-0.925034	1211.1802	1375.31	1519.3544
	O	-0.828801	-0.748636	0.010748	1597.7765	1713.9734	2960.7275
	H	0.860446	-0.683665	-0.087404	3041.2434	3088.2784	3817.0573
	H	1.987319	0.057738	0.739278			
	O	1.509028	0.086090	-0.100563			
MR-WD	C	1.624684	-0.646780	0.278440	49.1017	65.67	66.66
	H	0.804341	-1.286595	-0.080653	120.7245	155.12	197.34
	H	1.429840	-0.490348	1.342557	199.9369	211.02	357.20
	H	2.584594	-1.132864	0.108322	397.2183	574.04	656.67
	O	1.495613	0.535310	-0.419639	843.5776	1042.37	1178.97
	H	-1.423292	2.153358	-0.328180	1394.6224	1455.05	1529.90
	H	-0.163054	1.336766	-0.008796	1646.36	1665.91	2986.39
	H	-1.597196	-0.355781	-0.013842	3081.95	3148.86	3617.99
	O	-1.098996	1.445287	0.235062	3659.85	3900.18	3909.37
	O	-1.523891	-1.313881	-0.166712			
HM-WD	C	1.577060	-0.508740	0.410319	78.43	105.78	142.10
	H	-0.420104	-1.413993	-0.124840	200.42	216.04	228.7773
	H	1.208372	-0.314004	1.409347	274.09	281.06	320.8114
	H	2.474859	-1.088068	0.269558	398.46	550.56	699.1170
	O	1.371977	0.433125	-0.539185	760.90	899.91	1140.4513
	H	-1.583649	2.017965	-0.260571	1251.08	1478.58	1510.4094
	H	0.583193	0.957832	-0.288149	1632.48	1651.60	3159.2826
	H	-1.443934	0.524452	0.106303	3306.63	3489.52	3602.5402
	O	-1.055633	1.406411	0.260073	3669.16	3894.81	3898.1470
	O	-1.381023	-1.296227	-0.211171			
$T\bar{S}_{\text{WD}}$	C	-1.552565	0.225967	0.177216	-1075.77	71.22	186.2543
	H	-0.474998	1.003292	-0.016289	314.6233	318.9303	349.1730
	H	-1.632392	0.400059	1.263376	387.4418	470.8772	539.1420
	H	-2.319457	0.738103	-0.418138	582.0361	599.9233	907.2542
	O	-1.092684	-0.893167	-0.241631	1072.7153	1136.1818	1198.7408
	H	1.889932	-1.540721	-0.454687	1241.9055	1352.1661	1508.8484
	H	0.464068	-1.171925	0.082117	1586.5513	1647.0529	1694.1074
	H	1.244853	0.563144	-0.089663	2876.3894	2947.9467	3052.1656
	O	1.060237	2.003150	0.551133	3101.5656	3831.9969	3884.0190
	O	0.783749	1.459054	-0.197052			

MR-FA	C	-2.052698	0.524993	0.076215			
	H	-1.306125	1.259131	-0.228755	40.3170	86.4580	125.6243
	H	-2.151417	0.606489	1.168719	155.2585	180.6970	209.6524
	H	-3.031596	0.694212	-0.370856	672.2190	807.7323	937.8172
	O	-1.604893	-0.764896	-0.134572	1042.9286	1079.5520	1142.664
	C	1.722305	0.115209	0.020609	1202.5510	1375.1448	1410.067
	O	1.153459	-1.089016	0.059260	1430.1417	1441.4301	1525.887
	H	0.176890	-0.969956	-0.010754	1771.4689	2998.9995	3100.006
	O	1.137058	1.171977	-0.078662	3109.3138	3161.6177	3426.185
	H	2.809616	0.024391	0.092494			
MA-FA	C	-1.800717	0.523674	0.354276			
	H	0.157259	1.114783	-0.120741	92.1393	132.5318	158.0052
	H	-1.429815	0.444542	1.369361	177.8394	210.0715	335.1436
	H	-2.664446	1.132478	0.144673	671.4281	745.0239	801.0968
	O	-1.676894	-0.553924	-0.455107	830.7942	1075.8084	1116.497
	C	1.608167	-0.096982	0.008791	1198.0583	1248.4473	1359.648
	O	0.945221	-1.088904	0.243429	1420.7062	1439.6606	1509.519
	H	-0.863789	-1.024344	-0.190475	1756.5087	3132.0642	3152.884
	O	1.138913	1.127501	-0.204314	3308.5377	3430.4857	3603.395
	H	2.698165	-0.105002	-0.053284			
TS _{FA}	C	-1.686233	0.469583	0.226401			
	H	-0.558927	0.990393	-0.056682	-1581.0941	111.5960	228.2984
	H	-1.532307	0.575994	1.312175	261.0540	389.4417	415.6065
	H	-2.495433	1.074226	-0.184142	544.6929	760.9463	1013.599
	O	-1.456043	-0.683456	-0.308575	1067.4046	1162.5465	1198.660
	C	1.495744	0.080996	0.008453	1242.5259	1294.9177	1317.449
	O	0.958896	-1.091916	0.150798	1406.1732	1476.9912	1541.168
	H	-0.077257	-1.042473	0.019649	1595.5030	1743.4637	2359.819
	O	0.899805	1.154806	-0.162035	2975.1113	3133.1158	3138.370
	H	2.585598	0.062917	0.058366			
MR-SA	C	-2.915023	0.496359	0.011898			
	H	-2.522735	1.068147	-0.830346	21.7927	47.5046	105.8865
	H	-2.494425	0.965032	0.913268	108.4937	137.4555	224.7016
	H	-4.002036	0.510538	0.064505	267.1392	376.4545	424.1798
	O	-2.415190	-0.790996	0.023246	507.4708	536.6131	546.5106
	O	0.131338	-0.989986	-0.786004	767.1543	827.4224	903.2702
	H	-0.813612	-0.942525	-0.472221	911.0901	1035.7410	1151.8127
	O	0.076530	1.252092	0.228191	1162.49	1212.7792	1324.3341
	S	0.951852	0.149227	-0.071781	1391.2265	1437.2317	1456.8997
	O	1.283218	-0.502939	1.357044	1528.8446	3002.8137	3100.7217
	H	2.054365	-1.079815	1.241329	3168.1656	3229.8462	3767.1996
	O	2.178972	0.295933	-0.802405			

HM-SA	C	2.510875	0.680962	0.050198			
	H	0.795983	0.576275	1.022455	39.5454	92.0279	133.9790
	H	2.004345	1.188305	-0.763026	157.3991	193.4639	273.6392
	H	3.315531	1.172627	0.572400	339.8196	382.3827	424.2778
	O	2.654199	-0.660886	-0.028989	503.4366	536.7415	547.1142
	O	-0.008152	-0.880376	-0.871317	657.9004	732.0890	831.6837
	H	1.898349	-1.011354	-0.529868	851.1883	901.8792	1106.0538
	O	-0.139467	0.345177	1.273393	1161.4534	1207.9206	1254.4039
	S	-0.907671	-0.146997	-0.013395	1271.2194	1407.9641	1454.8035
	O	-1.175219	1.21807	-0.807261	1508.6361	3147.9997	3237.1761
	H	-1.951734	1.643462	-0.409544	3304.7289	3686.2906	3759.2855
	O	-2.156985	-0.684880	0.436762			
TS _{SA}	C	-2.450647	0.565668	-0.020888			
	H	-1.457484	0.758118	-0.733978	-1703.5797	35.0188	114.6252
	H	-2.050967	1.184776	0.796835	211.4644	219.0340	337.9861
	H	-3.358948	0.928019	-0.493884	378.8078	409.2606	452.8162
	O	-2.324418	-0.723092	0.097017	479.6697	548.1197	562.8545
	O	-0.008582	-1.075531	0.623297	675.8058	818.4387	900.9640
	H	-1.135673	-0.952674	0.453688	1002.4573	1074.7410	1151.5177
	O	0.036471	0.650509	-1.128048	1196.6273	1220.2323	1253.6144
	S	0.844939	-0.052198	-0.120732	1324.6264	1365.2764	1403.8972
	O	2.110185	-0.610565	-0.504571	1519.71	1602.4591	1743.6568
	O	1.104439	1.098098	0.975020	2993.2428	3186.3530	3765.2460
	H	1.843171	0.807569	1.532654			
WM	O	0.000000	0.000000	0.118189			
	H	0.000000	0.758025	-0.472754	1627.7612	3824.7050	3950.6090
	H	0.000000	-0.758025	-0.472754			
WD	O	-1.385097	0.000601	0.110496			
	H	-1.746597	-0.763119	-0.349950	127.0098	147.1099	154.9316
	H	-1.747854	0.758911	-0.357831	184.1613	360.4078	630.2965
	O	1.512857	-0.000695	-0.121976	1629.3162	1650.2617	3718.7825
	H	0.559405	0.000611	0.048517	3813.9011	3915.1798	3935.2566
	H	1.912972	0.004353	0.751104			
FA	C	0.000000	0.422392	0.000000			
	H	-0.376774	1.447695	0.000000	625.952	674.778	1058.557
	O	1.163418	0.108154	0.000000	1130.131	1301.324	1408.906
	O	-1.034994	-0.439528	0.000000	1794.039	3125.306	3741.349
	H	-0.650614	-1.331055	0.000000			
SA	H	1.683905	-0.014800	1.107699			
	O	0.659484	1.084346	-0.827065	242.8958	328.1320	367.2603
	O	1.034771	-0.687872	0.848480	431.4837	485.9993	531.5543
	S	0.000058	0.000064	-0.159243	542.2237	821.4432	873.6608
	O	-1.035291	0.687175	0.848517	1155.1241	1170.0433	1219.2879
	H	-1.684504	0.013827	1.106801	1469.2375	3760.6034	3764.8107
	O	-0.659005	-1.083655	-0.828259			

Table S7 Absolute energies (in Hartree) of all species involved in isomerization of MR calculated at the two different levels of theory

Species	MP2/aug-cc-pVTZ	CCSD(T)/aug-cc-pVTZ
MR	-114.84396077	-114.885062
TS	-114.79613291	-114.8328949
HM	-114.86634476	-114.8988607
MR-WM	-191.18101248	-191.2357163
TS _{WM}	-191.14159052	-191.1909453
HM-WM	-191.20605728	-191.2519222
MR-WD	-267.52313693	-267.5914694
TS _{WD}	-267.50181805	-267.5656003
HM-WD	-267.55128696	-267.6106135
MR-FA	-304.34423196	-304.4168392
TS _{FA}	-304.32465040	-304.395769
HM-FA	-304.36904354	-304.4328489
MR-SA	-814.28172138	-814.3693943
TS	-814.26141531	-814.3478167
HM-SA	-814.30415433	-814.3829675

Reference:

1. (a) S. Zhang and T. N. Truong, VKLab (version 1.0), University of Utah, 2001;
(b) W. T. Duncan, R. L. Bell and T. N. Truong, *J. Comput. Chem.*, 1998, **19**, 1039.
2. a) D. R. Glowacki, C. H. Liang, C. Morley, M. J. Pilling and S. H. Robertson, *J. Phys. Chem. A*, 2012, **116**, 9545.
b) S. H. Robertson, D. R. Glowacki, C. H. Liang, C. Morley, R. Shannon, M. Blitz, P. W. Seakins, M. J. Pilling, *MESMER (Master Equation Solver for Multi-Energy Well Reactions)*, 2008-2013; an object oriented C++ program implementing master equation methods for gas phase reactions with arbitrary multiple wells. <http://sourceforge.net/projects/mesmer>
3. J. M. Anglada, G. J. Hoffman, L. V. Slipchenko, M. M. Costa, M. F. Ruiz-López and J. S. Francisco, *J. Phys. Chem. A*, 2013, **117**, 10381.
4. A. Razavi, F. Karagulian, L. Clarisse, D. Hurtmans, P. F. Coheur, C. Clerbaux, J. F. Müller and T. Stavrakou, *Atmos. Chem. Phys.*, 2011, **11**, 857.
5. R. P. Turco, P. Hamill, O. B. Toon, R. C. Whitten and C. S. Kiang, *J. Atmos. Sci.*, 1979, **36**, 699.