

MnCoO

Rémi Arras,^{*,†} Thi Ly Le,[‡] Sophie Guillemet-Fritsch,[‡] Pascal Dufour,[‡] and
Christophe Tenailleau[‡]

*CEMES, Université de Toulouse, CNRS, UPS, 29, rue Jeanne-Marvig, F-31055 Toulouse,
France, and CIRIMAT, Université de Toulouse, CNRS, INPT, UPS, 118 Route de
Narbonne, 31062 Toulouse Cedex 9, France*

E-mail: remi.arras@cemes.fr

*To whom correspondence should be addressed

[†]CEMES, Université de Toulouse, CNRS, UPS, 29, rue Jeanne-Marvig, F-31055 Toulouse, France

[‡]CIRIMAT, Université de Toulouse, CNRS, INPT, UPS, 118 Route de Narbonne, 31062 Toulouse Cedex
9, France

1 Co_3O_4 and Mn_3O_4

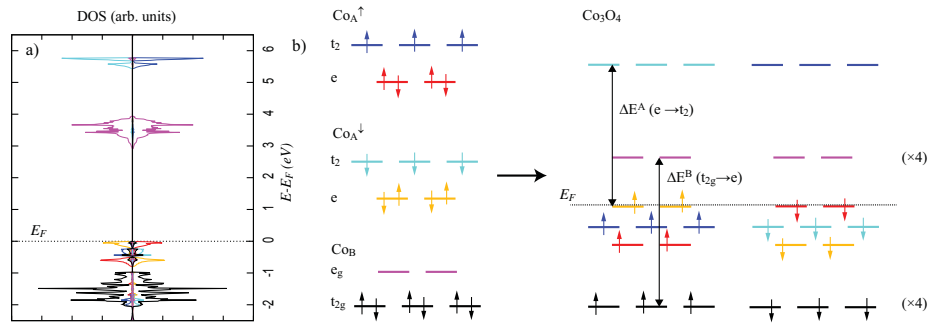


Figure 1S: a) Spin-resolved Co-d contribution to the DOS of Co_3O_4 , b) schematic representation of the electron distribution in the d orbitals.

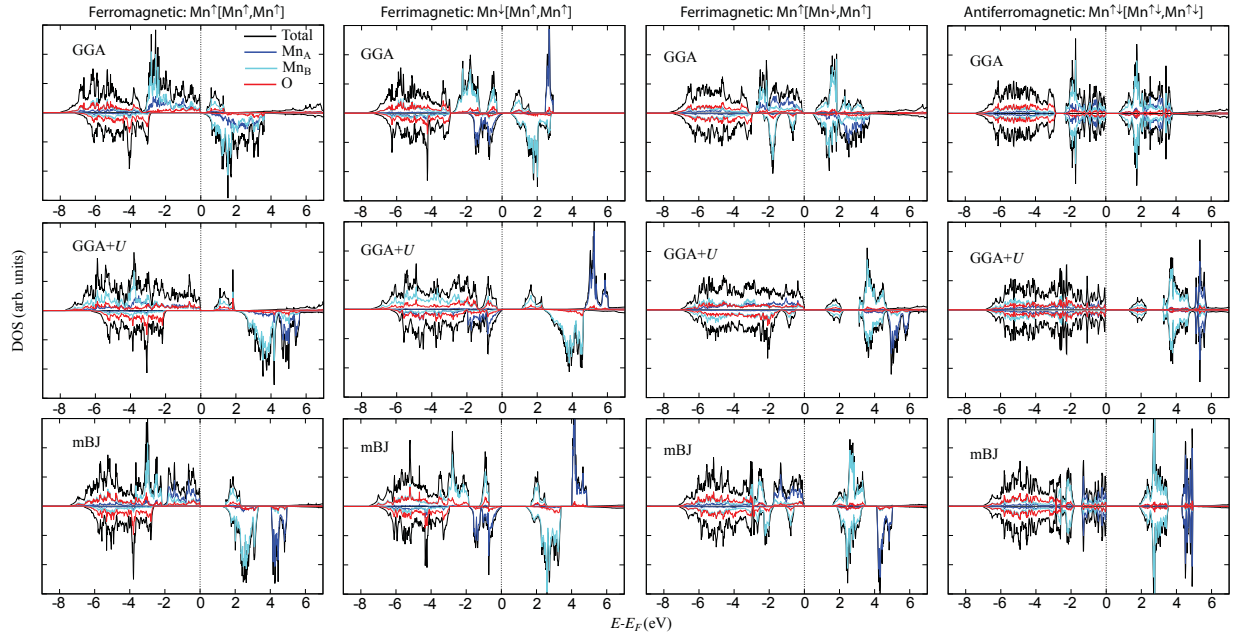


Figure 2S: Spin-resolved DOS for Mn_3O_4 calculated with the 3 approximations and for 3 different magnetic ordering.

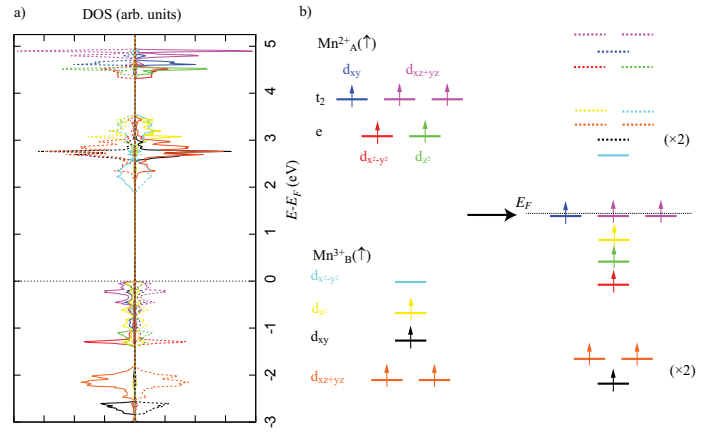


Figure 3S: a) Spin-resolved Mn-d contribution to the DOS of Mn₃O₄, b) schematic representation of the electron distribution in the d orbitals for one spin direction.

2 $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$

2.1 Atomic structure

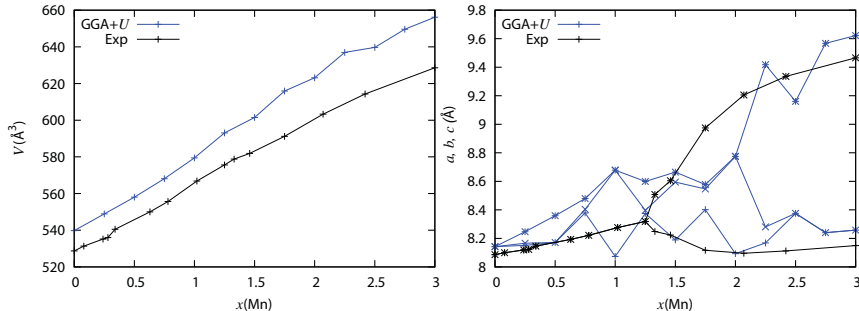


Figure 4S: Comparison of the volume and lattice parameters of $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$ calculated with the GGA+ U method with the experimental measurements.

The GGA calculations may be inadequate to describe correctly the atomic structure of these compounds. It seemed however to be the most reliable solution owing to the problem of computational times and the difficulty to choose a good U_{eff} parameter. We indeed performed a calculations with the GGA+ U method, and found a more complex phase diagram than in GGA, with 3 different orthorhombic structures:

- with $a \sim b < c$ for $0 < x \leq 0.75$,
- with $a < b \sim c$ for $0.75 < x \leq 2.0$,
- with $a \sim b < c$ for $0 < x \leq 0.75$.

The small difference between a , b , and c can be due to the cation distribution, as we suggested it for the GGA calculations. In that case, the effect is much more noticeable. The lattice parameters are in general more sensitive to the cation distribution than with the GGA approximation. Moreover, instead of an underestimation of the volume, the GGA+ U method overestimates it.

However, it is difficult to distinguish, with this method, if the difference found with the measurements are only artefactual, due to a spurious cationic ordering coming from the

supercell description of the lattice, or if the lattice really adopt an orthorhombic shape (more probable here), with $a \neq b \neq c$. Also not discussed, but the angles can also slightly vary from the ideal 90° value (less than 1°). The GGA+ U calculations is not supported by our experimental measurements, but it can be consistent with the orthorhombic phase reported in some studies^{1,2} and calculations performed by other authors on manganites.³

2.2 Electronic structure

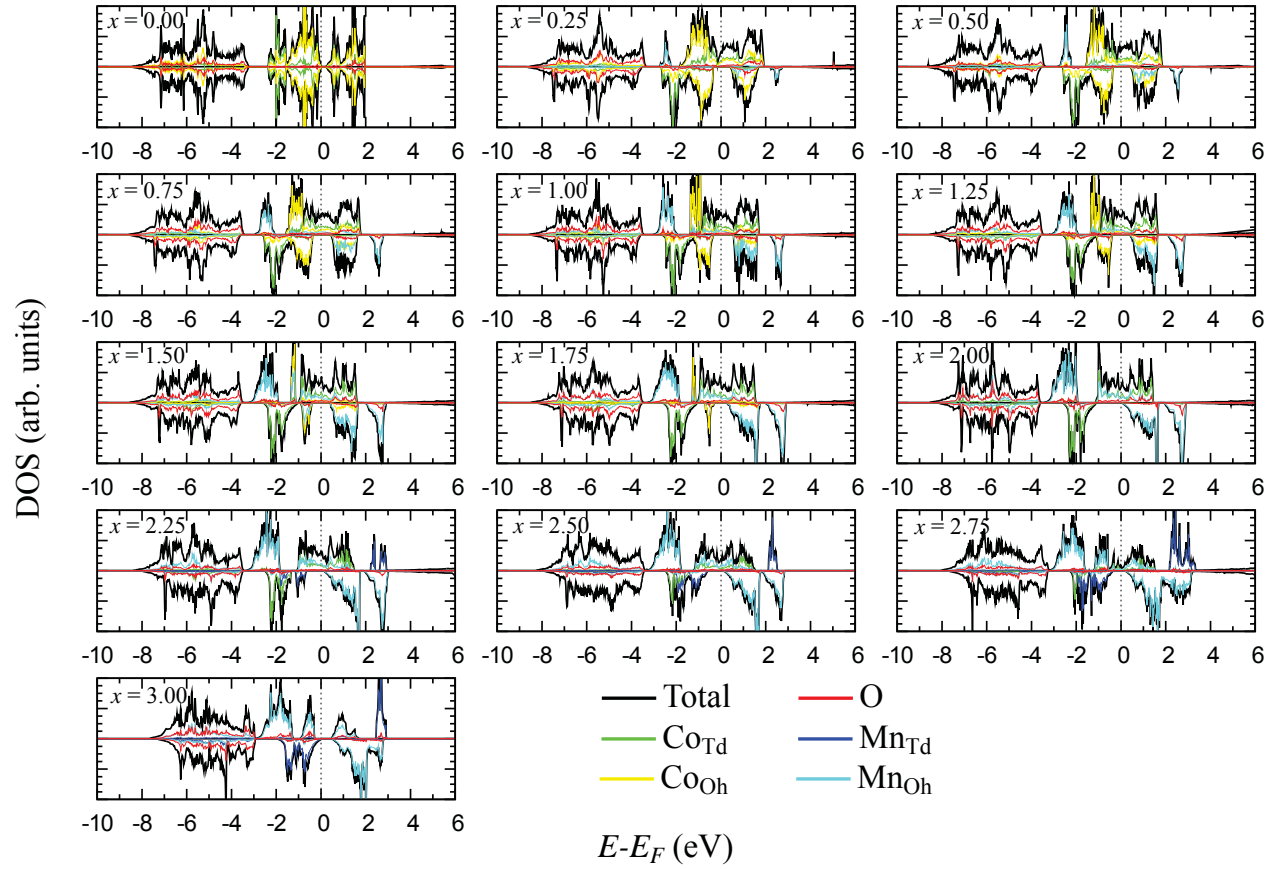


Figure 5S: Spin-resolved DOS for $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$ ($0 \leq x \leq 3$) calculated with the GGA approximation.

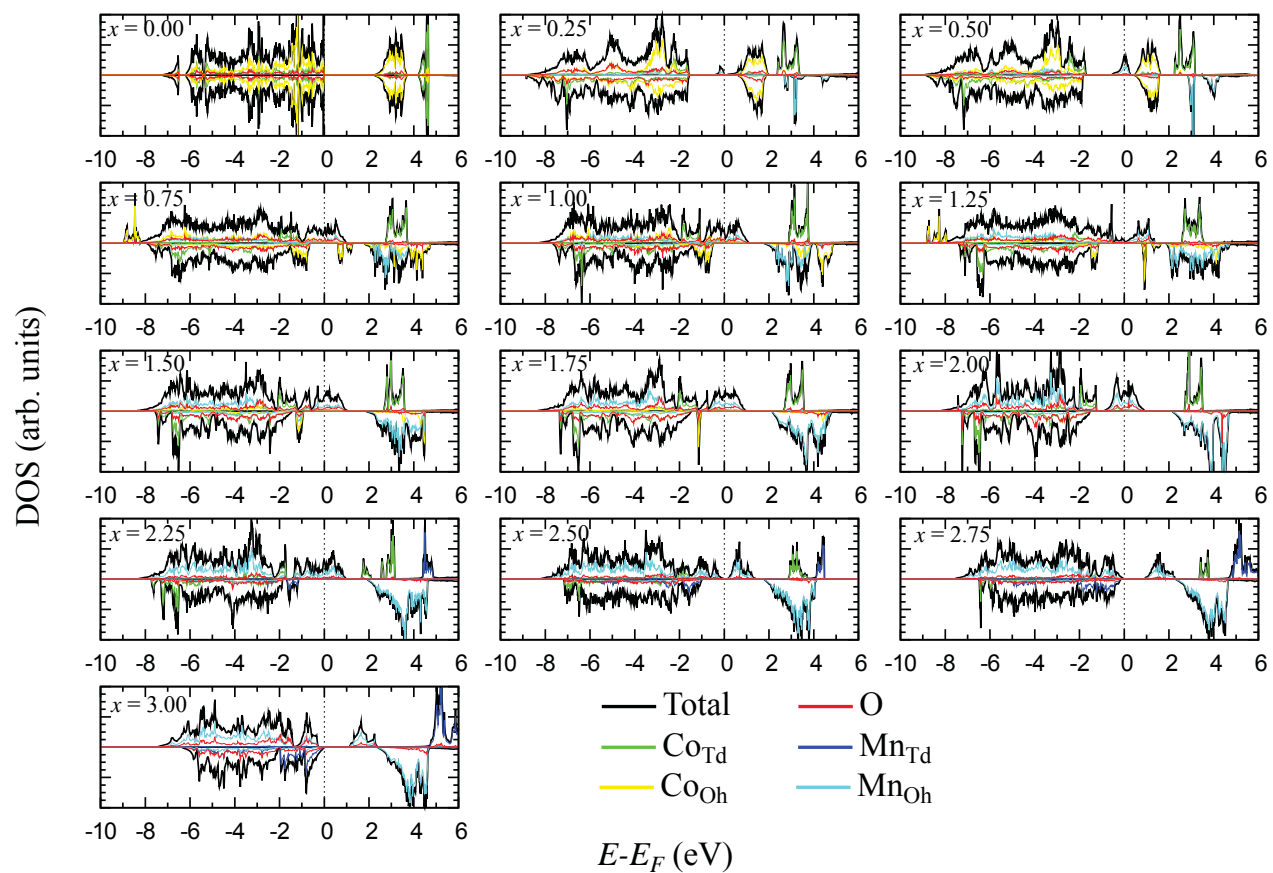


Figure 6S: Spin-resolved DOS for $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$ ($0 \leq x \leq 3$) calculated with the GGA+U approximation.

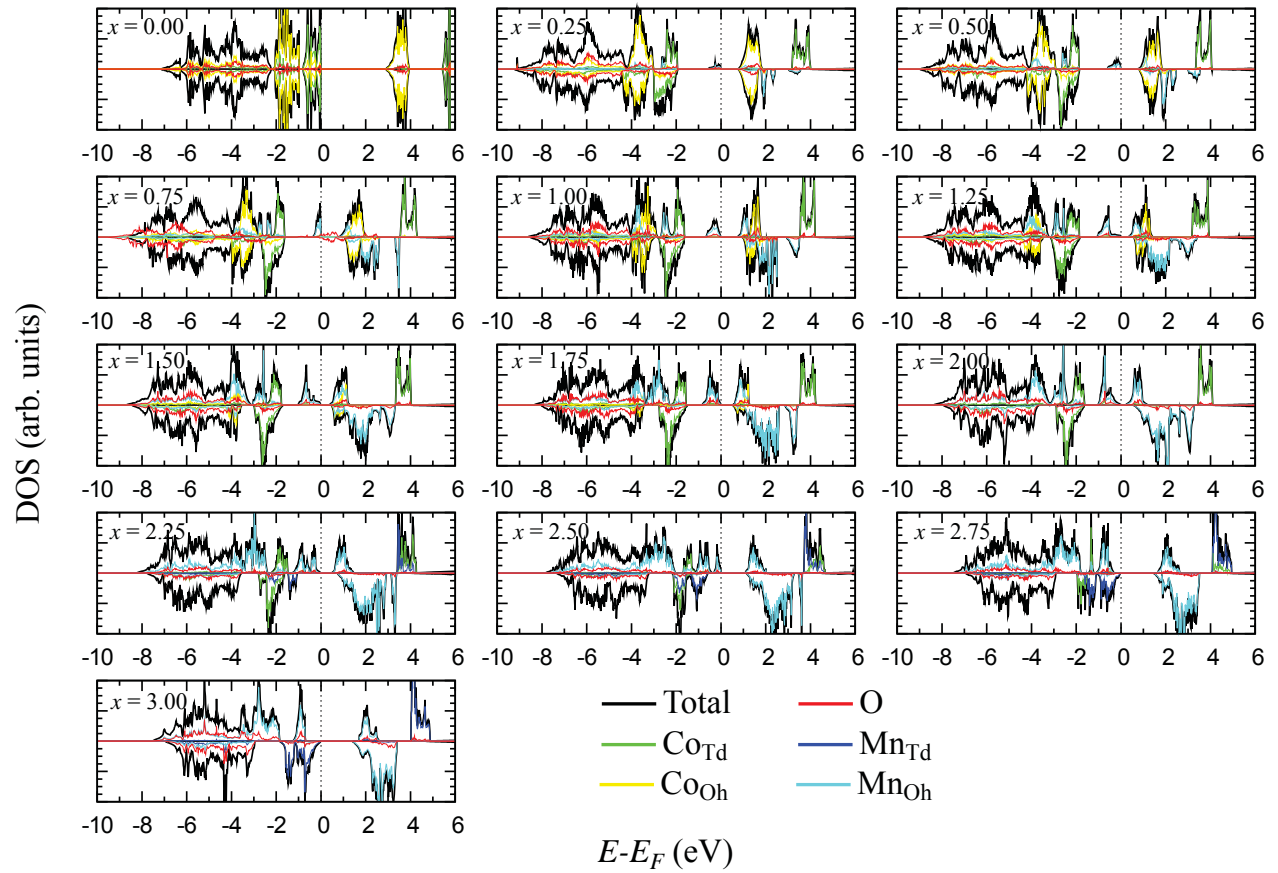


Figure 7S: Spin-resolved DOS for $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$ ($0 \leq x \leq 3$) calculated with the mBJ approximation.

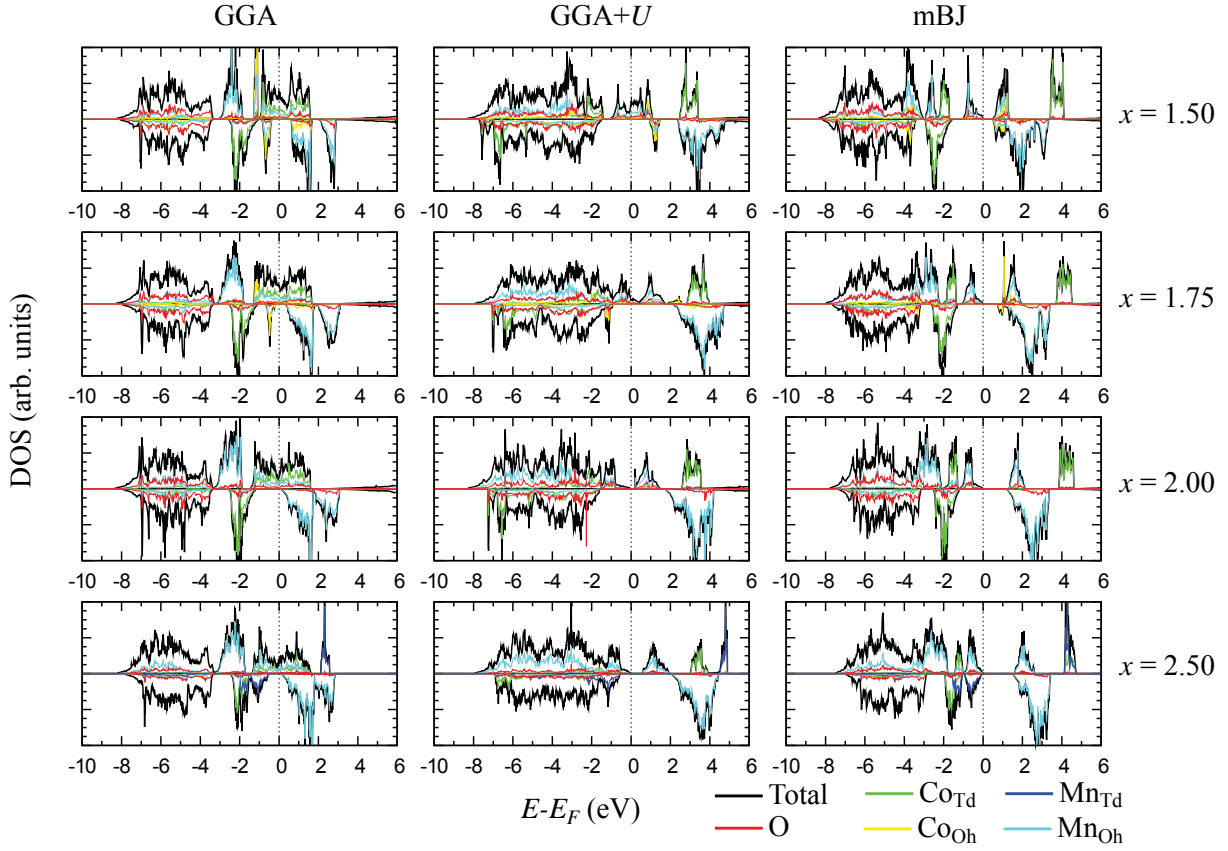


Figure 8S: Spin-resolved DOS for $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$ ($1.5 \leq x \leq 2.5$) calculated with the experimental lattice parameters.

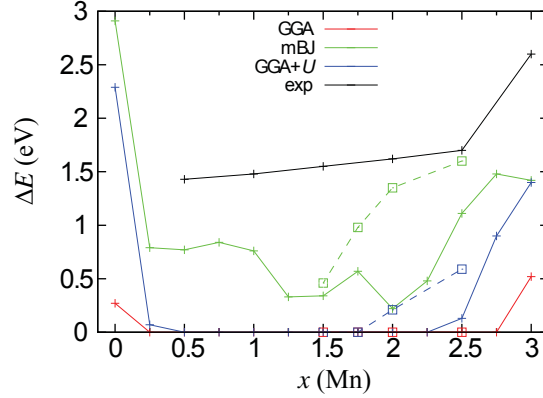


Figure 9S: Band gap energy as a function of x . Solid lines with + symbols corresponds to values obtained with the calculated lattice parameters, while dashed lines with \square symbols are the values calculated with the experimental lattice parameters.

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References

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