Supporting Information

Physicochemical and Tribophysical Properties of Trioctylalkylammonium-bis(salicylato)borate (N888n-BScB) Ionic Liquids: Effect of Alkyl Chain Length

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¹H and ¹³C NMR Characterization of Ionic Liquids

Trioctylhexylammonium bis(salicylato)borate (N8886-BScB): ¹H NMR (ppm): 0.84-0.93 (t, 12H, N(CH₂)₇C<u>H₃</u>, N(CH₂)₅C<u>H₃</u>), 1.25-1.36 (m, 36H, N(CH₂)₂(C<u>H₂</u>)₅, N(CH₂)₂(C<u>H₂</u>)₃), 1.62-1.78 (m, 8H, NCH₂C<u>H₂</u>), 2.82-2.85 (t, 8H, NC<u>H₂</u>), 6.86-6.90 (m, 4H, C₆<u>H₄</u>), 7.39-7.40 (m, 2H, C₆<u>H₄</u>), 7.87-7.89 (dd, 2H, C₆<u>H₄</u>). ¹³C NMR (ppm): 14.01, 14.13, 22.55, 22.61, 23.96, 25.64, 26.17, 26.92, 28.52, 28.99, 29.12, 31.42, 31.62, 31.70, 52.63, 112.61, 114.76, 117.60, 118.91, 119.21, 129.70, 130.67, 132.52, 135.59, 166.52, 170.27, 174.81.

Tetraoctylammonium bis(salicylato)borate (N8888-BScB): ¹H NMR (ppm): 0.86-0.90 (t, 12H, N(CH₂)₇C<u>H</u>₃), 1.20-1.26 (m, 40H, N(CH₂)₂(C<u>H</u>₂)₅), 1.76-1.79 (m, 8H, NCH₂C<u>H</u>₂), 2.87-2.90 (t, 8H, NC<u>H</u>₂), 6.86-6.90 (m, 4H, C₆<u>H</u>₄), 7.39-7.40 (m, 2H, C₆<u>H</u>₄), 7.86-7.88 (dd, 2H, C₆<u>H</u>₄). ¹³C NMR (ppm): 14.10, 21.87, 22.62, 23.82, 25.98, 26.20, 26.74, 28.56, 29.08, 29.19, 29.31, 29.53, 31.65, 31.71, 31.90, 32.68, 52.84, 58.72, 112.58, 114.95, 117.45, 118.51, 119.16, 129.74, 129.90, 135.05, 135.60, 159.52, 161.64, 166.25, 170.27.

Trioctyldecylammonium bis(salicylato)borate (N88810-BScB): ¹H NMR (ppm): 0.82-0.91 (t, 12H, N(CH₂)₇C<u>H₃</u>, N(CH₂)₉C<u>H₃</u>), 1.21-1.43 (m, 44H, N(CH₂)₂(C<u>H₂</u>)₅, N(CH₂)₂(C<u>H₂</u>)₇), 1.76-1.79 (m, 8H, NCH₂C<u>H₂</u>), 2.85-2.88 (t, 8H, NC<u>H₂</u>), 6.75-6.95 (m, 4H, C₆<u>H₄</u>), 7.40-7.43 (m, 2H, C₆<u>H₄</u>), 7.86-7.90 (dd, 2H, C₆<u>H₄</u>). ¹³C NMR (ppm): 14.10, 22.63, 22.70, 26.90, 27.08, 28.92, 29.15, 29.22, 29.32, 29.50, 31.74, 31.90, 32.67, 45.20, 52.69, 53.49, 114.64, 116.33, 117.50, 117.62, 118.67, 119.25, 19.69, 130.65, 132.55, 135.26, 159.55, 161.64, 161.76, 166.70, 174.87.

Trioctyldodecylammonium bis(salicylato)borate (N88812-BScB): ¹H NMR (ppm): 0.85-0.89 (t, 12H, N(CH₂)₇C<u>H</u>₃, N(CH₂)₁₁C<u>H</u>₃), 1.20-1.45 (m, 48H, N(CH₂)₂(C<u>H</u>₂)₅, N(CH₂)₂(C<u>H₂)₉), 1.64-1.86 (m, 8H, NCH₂CH₂), 2.82-2.85 (t, 8H, NC<u>H₂)</u>, 6.86-6.96 (m, 4H, C₆<u>H</u>₄), 7.34-7.44 (m, 2H, C₆<u>H</u>₄), 7.83-7.88 (dd, 2H, C₆<u>H</u>₄). ¹³C NMR (ppm): 14.09, 14.14, 21.88, 22.62, 22.72, 24.00, 25.99, 26.21, 26.88, 28.21, 28.58, 29.02, 29.26, 29.38, 29.48,</u> 29.53, 29.58, 31.65, 31.72, 32.88, 34.02, 52.53, 58.77, 112.61, 114.88, 117.48, 118.57, 119.12, 129.74, 129.90, 135.10, 135.58, 159.56, 161.68, 166.40, 170.27.



Figure S1: FTIR spectrum of N8886-BScB as a representative ionic liquid.



Figure S2. FTIR spectra of N8886-BScB, N8888-BScB, N88810-BScB and N88812-BScB ionic liquids



Figure S3: Thermal decomposition patterns of N8886-BScB, N8888-BScB, N88810-BScB and N88812-BScB ionic liquids at thermal rate of 10 °C.min⁻¹ under the nitrogen flow.



Figure S4: Differential scamming calorimetric pattern of N88810-BScB ionic liquids as a representative ionic liquid.



Figure S5: Storage modulus (G', hollow symbols) and loss modulus (G", filled symbols) as a function of angular frequency for N888n-BScB ionic liquids. Black, red, green and blue color symbols represent N8886-BScB, N8888-BScB, N88810-BScB and N88812-BScB, respectively.



Figure S6: % Reduction of (a) coefficient of friction and (b) wear depth of N888n-BScB (n = 6, 8, 10 and 12) ionic liquids-lubricated aluminium disc compared to PEG 300 lube oil.after 5000 cyclic laps. Lubrication characteristics of PEG 300 lube base oil are plotted for a comparison. A steel ball ($\phi = 2$ mm) was used as counter sliding body. Load: 100 mN, cyclic speed: 50 rpm.



Figure S7: Changes in wear track width on the aluminium disc lubricated with (a) PEG 300 synthetic lube oil, (b) N8886-BScB, (c) N8888-BScB, (d) N88810-BScB and (e) N88812-BScB ionic liquids after 5000 cyclic laps. A steel ball ($\phi = 2 \text{ mm}$) was used as counter sliding body. Load: 100 mN, cyclic speed: 50 rpm.

N8886-BScB,	N8888-BScB,	N88810-BScB,	N88812-BScB,	Vibrational Assignment
cm ⁻¹	cm ⁻¹	cm ⁻¹	cm ⁻¹	
3071, 3042	3070, 3042	3063	3070, 3042	v(C–H), aromatic ring
2957, 2921, 2857	2957, 2928, 2850	2957, 2928, 2850	2957, 2921, 2850	va/vs(C-H), CH ₂ /CH ₃ groups
1687	1680	1701	1701	v(C=O)
1610	1609	1595	1616	va(COO)
1573	1573	1574	1573	v(C=C), aromatic
1467, 1375, 1319	1467, 1375, 1318	1467, 1382, 1318	1467, 1375, 1318	δ C–H, CH ₂ /CH ₃ groups
1270, 1240	1269, 1240	1261, 1240	1269, 1240	US(COO)
1200-900	1200-900	1200-900	1200-900	va(B–O)/v(C–O)
758, 722, 701	758, 722, 695	758, 722, 701	758, 722, 701	US(B-O)

 Table S1: Infrared vibrational assignments of characteristics vibrational features of N888n

 BScB ionic liquids