Supporting Information

Highly Efficient $g-C_3N_4/SiO_2$ Heterojunction: The role of SiO_2 for the Enhanced Visible Light Photocatalytic Activity

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1. Figure S1: Fourier Transform Infrared Spectrometer (FTIR) of prepared samples.

Figure S1. FT-IR spectra of g-C₃N₄, SiO₂ and g-C₃N₄/SiO₂ samples.

Figure S1 shows the FT-IR spectra of $g-C_3N_4$, SiO₂ and $g-C_3N_4$ /SiO₂ samples. In the FT-IR spectrum of $g-C_3N_4$, the peak at 1629 cm⁻¹ is attributable to the C=N stretching vibration modes, while the peaks at 1228 cm⁻¹,1318 cm⁻¹ and 1403 cm⁻¹ are due to the aromatic C-N stretching.¹⁻³ The peak at 808 cm⁻¹ is related to the s-triazine ring modes.⁴ The peak at 1086 cm⁻¹ is a Si-O-Si antisymmetric stretching and the peak at 796 cm⁻¹ is Si-O symmetric stretching. The peak at 955 cm⁻¹ is caused by the bending vibration of Si-OH. With the increase of SiO₂, a small paek at 1086 cm⁻¹ can be

observed in the $g-C_3N_4/SiO_2$ composites, but the wavenumber has not changed which means the force between $g-C_3N_4$ and SiO_2 is a physical process, rather than chemical action.

2. Fluorescence spectrum (PL) of SiO₂



Figure S2. Fluorescence spectrum (PL) of SiO₂ under excitation by the 266 nm and 325 nm lasers

Under excitation by the 266 nm and 325 nm lasers, SiO_2 has a lot of peaks, indicating its impurity energy level. The valence band maximum is mainly composed

of the by O 2p orbital⁵ and the energy levels of the surface states for quartz are at least 2.19 eV to 3.08 eV above the valence band.

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