

Electronic Supplementary Information

Dithieno[3,2-*b*:2',3'-*d*]pyrrole based, NIR absorbing, solution processable, small molecule donor for efficient bulk heterojunction solar cells.

Manohar Reddy Busireddy,^a Venkata Niladri Raju Mantena,^a Narendra Reddy Cheredy,^{a,*} Balaiah Shanigaram,^b Bhanuprakash Kotamarthi,^{b,*} Subhayan Biswas,^c Ganesh Datt Sharma^{c,*} and Jayathirtha Rao Vaidya^{a,d*}

^aCrop Protection Chemicals Division, ^bInorganic & Physical Chemistry Division, ^dAcSIR, CSIR-Indian Institute of Chemical Technology, Hyderabad-500 007, India

^cDepartment of Physics, The LNM Institute of Information Technology, Jamdoli, Jaipur, India.

Corresponding author Tel.: +91 40 27193933; +91 40 27193174

E-Mail: chereddynarendra@gmail.com; jrao@ict.res.in; gdsharma273@gmail.com

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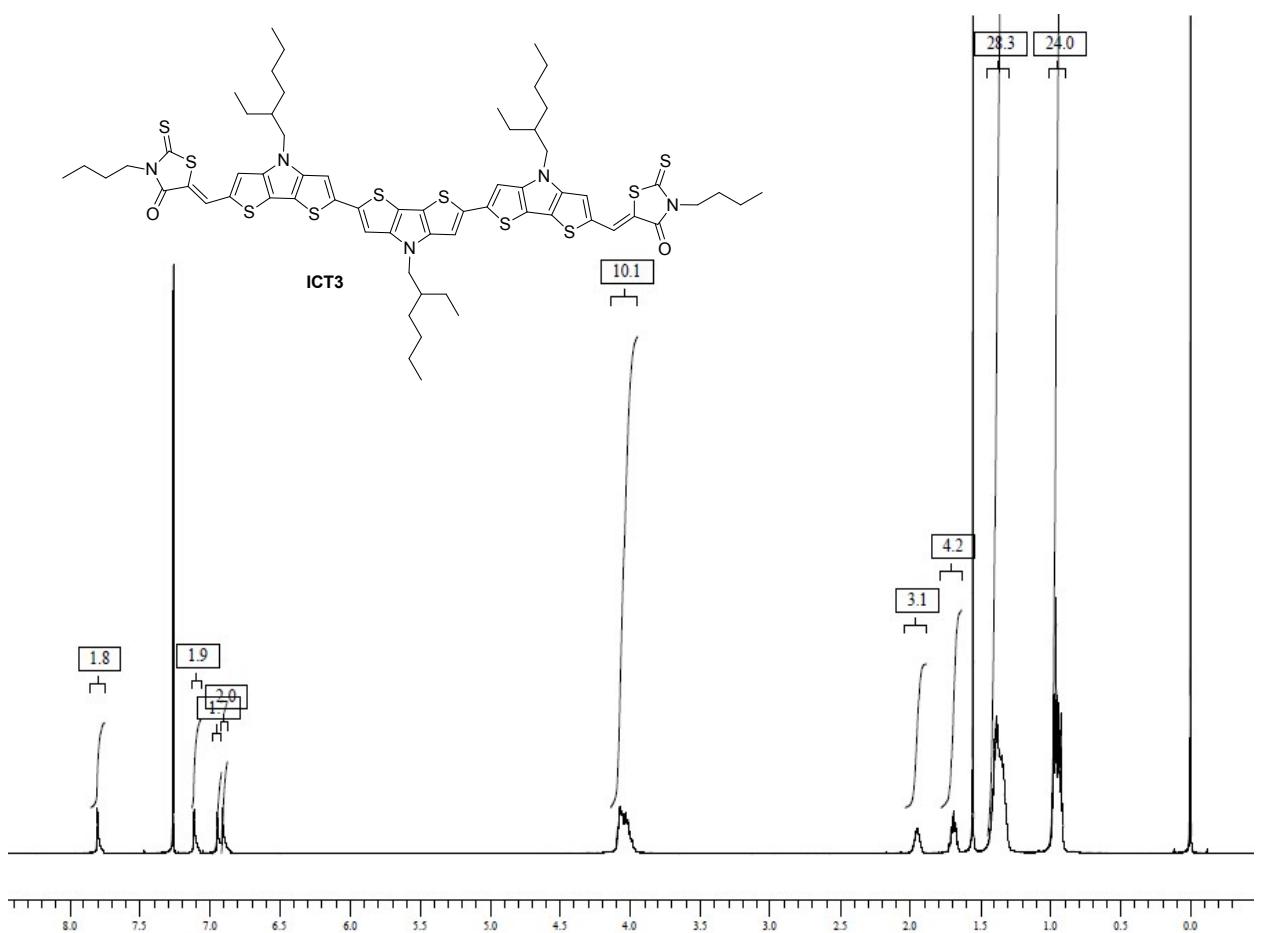


Fig. S1. ¹H NMR spectrum of ICT3

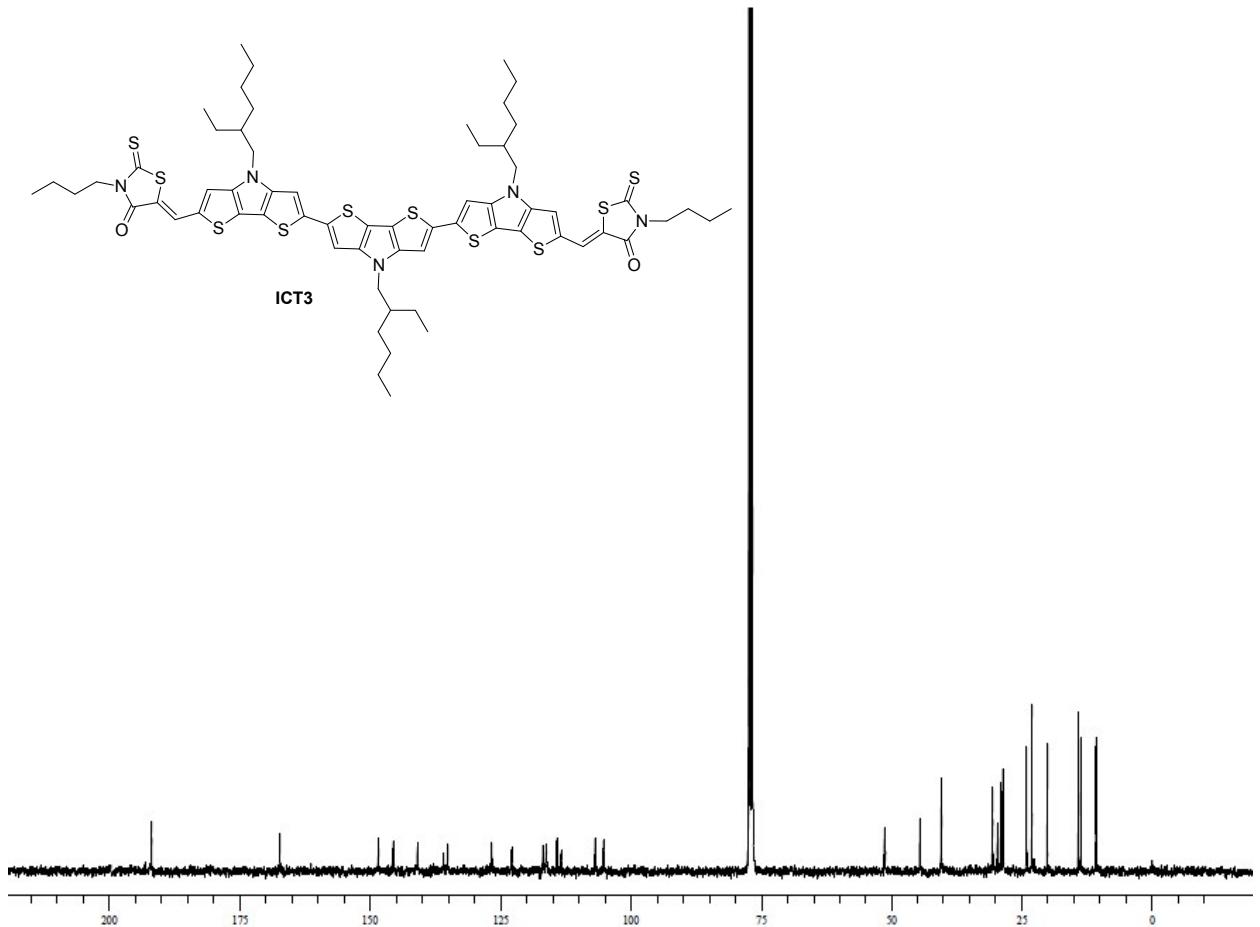


Fig. S2. ^{13}C NMR spectrum of ICT3

M V N RAJU , MNR-NF

Data: MNR0002.1B1[c] 24 Jun 2015 11:40 Cal: tof 8 Jun 2012 18:25
Shimadzu Biotech Axima Performance 2.9.3.20110624: Mode Linear, Power: 69, Blanked, P.Ext. @ 2000 (bin 66)
%Int. 105 mV[sum= 2737 mV] Profiles 1-26 Smooth Gauss 5

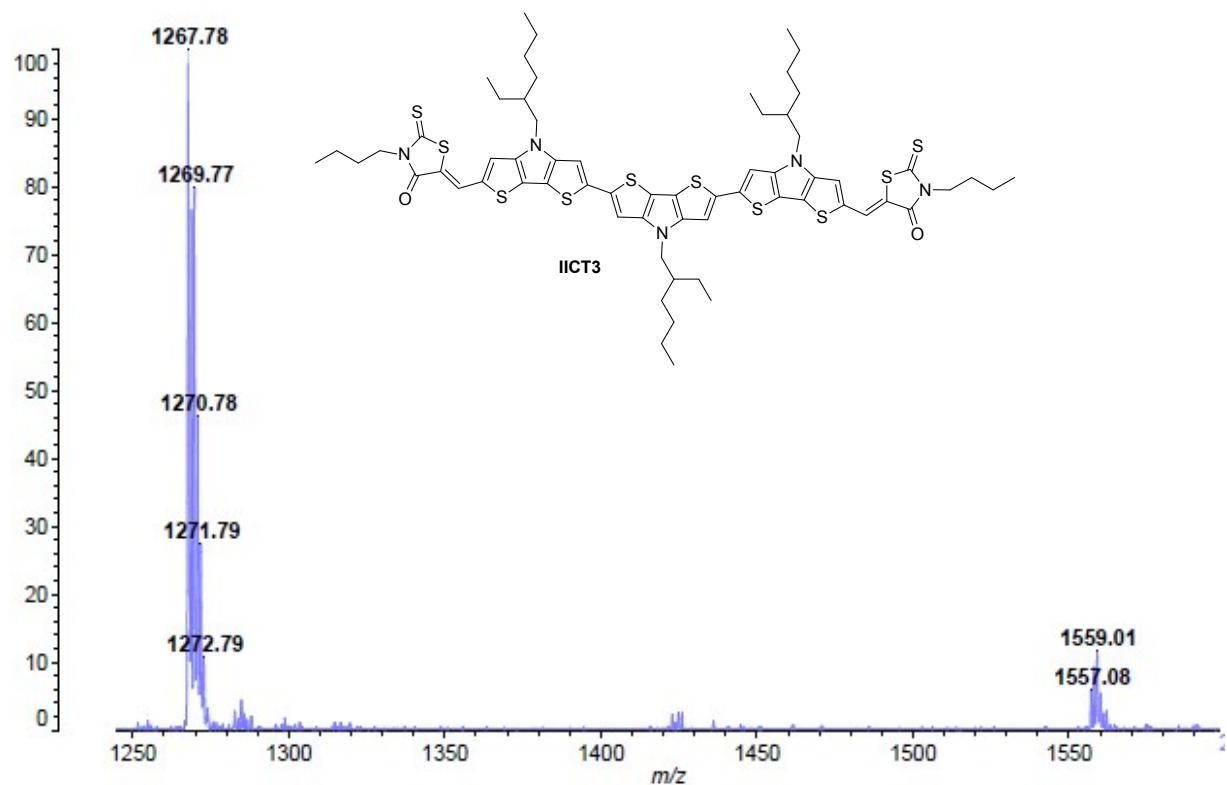


Fig. S3. MALDI-MS spectrum of **ICT3**

Material Code	λ_{max} (nm) ^a	ξ ($\times 10^5 \text{ mol}^{-1} \text{ cm}^{-1}$) ^a	λ_{max} (nm) ^b	$E_{\text{o-o}}$ (eV) ^c	E_g^{opt} (eV) ^d	E_{ox} (V)	E_{HOMO} (eV)	E_{LUMO} (eV) ^e
ICT1	586	1.71	604	1.96	1.68	0.29	-5.39	-3.43
ICT2	594	1.65	608	2.02	1.63	0.38	-5.48	-3.46
<hr/>								
Active layer	J_{sc} (mA/cm ²)	V_{oc} (V)	FF	PCE (%)	R_s (Ωcm^2)	R_{sh} (Ωcm^2)		
ICT1:PC₇₁BM (as cast)	6.84	0.92	0.44	2.77	24.06	510		
ICT2:PC₇₁BM (as cast)	7.26	0.96	0.47	3.27	22.16	564		
ICT1:PC₇₁BM (TSA)	10.15	0.87	0.58	5.12	12.92	686		
ICT2:PC₇₁BM (TSA)	10.68	0.92	0.60	5.90	9.08	746		

^ain dilute chloroform solution (10 μM), ^bin thin film cast from chloroform solution, ^cestimated from intersection of absorption and emission spectra in chloroform solution, ^destimated from $E_g^{\text{opt}}=1240/\lambda_{\text{onset}}$, λ_{onset} is onset absorption wavelength in thin film, ^e $E_{\text{LUMO}}=E_{\text{HOMO}}-E_{\text{o-o}}$.

Fig. S4. Chemical structures of **ICT1** and **ICT2** and their photophysical, electrochemical and photovoltaic data

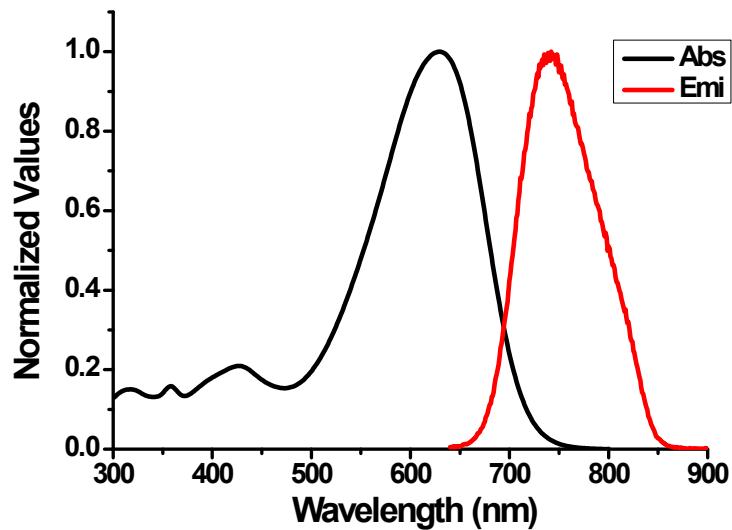


Fig. S5. Absorption and fluorescence spectra of **ICT3** (10 μM) measured in chloroform solution.

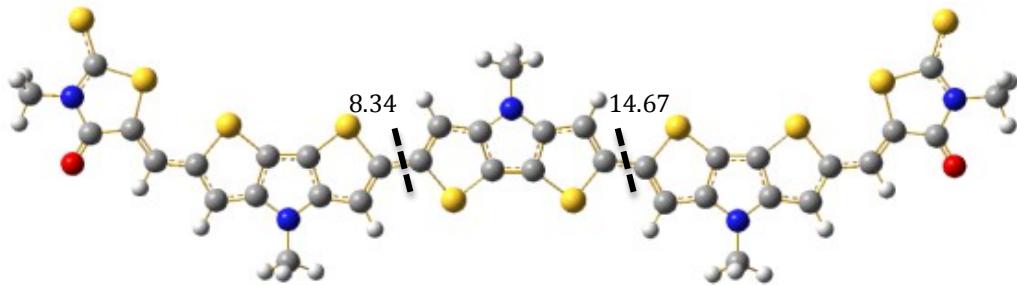


Fig. S6. Optimized structure of **ICT3**.

Table S1 Calculated properties of **ICT3** at M06/6-31G(d,p) level. HOMO and LUMO energies (eV), HOMO–LUMO gap HLG (eV), Optical gap OG (eV), oscillator strength f , the wavelength maxima, the major contribution (%) to the first excited state and the dipole moment μ_g (D).^a

Molecule	HOMO (eV)	LUMO (eV)	HLG (eV)	λ (nm)	OG (eV)	f	Major contribution (%)	μ_g ^a (D)
ICT3	-5.1	-2.69	2.41	628	1.974	3.520	H -1→ L+1(4%) H → L (95%)	9.07
	<i>-5.04</i>	<i>-2.71</i>	<i>2.33</i>	<i>676</i>	<i>1.834</i>	<i>3.707</i>	<i>H -1→ L+1(4%)</i> <i>H → L (93%)</i>	<i>12.27</i>

^a values given in *Italic font* are the values obtained in the solvent phase CF.

Table S2 Calculated properties of **ICT3** at B3LYP/6-31G(d,p) level. HOMO and LUMO energies (eV), HOMO–LUMO gap HLG (eV), Optical gap OG (eV), oscillator strength f , the wavelength maxima, the major contribution (%) to the first excited state and the dipole moment μ_g (D).^a

Molecule	HOMO (eV)	LUMO (eV)	HLG (eV)	λ (nm)	OG	f	Major contribution (%)	μ_g ^b (D)
ICT3	-4.82	-2.76	2.06	664	1.867	3.291	H → L (99%)	9.72
	<i>-4.75</i>	<i>-2.76</i>	<i>1.97</i>	<i>724</i>	<i>1.712</i>	<i>3.484</i>	<i>H → L (98%)</i>	<i>13.15</i>

^a values given in *Italic font* are the values obtained in the solvent phase CF.

Table S3 Calculated properties of **ICT3** at PBE/6-31G(d,p) level. HOMO and LUMO energies (eV), HOMO–LUMO gap HLG (eV), Optical gap OG (eV), oscillator strength f , the wavelength maxima, the major contribution (%) to the first excited state and the dipole moment μ_g (D).^a

Molecule	HOMO (eV)	LUMO (eV)	HLG (eV)	λ (nm)	OG (eV)	f	Major contribution(%)	μ_g (D)
ICT3	-4.82	-2.76	2.06	839	1.4770	1.975	H -1→ L+1 (6%) H→L+2 (3%) H → L (91%)	9.70

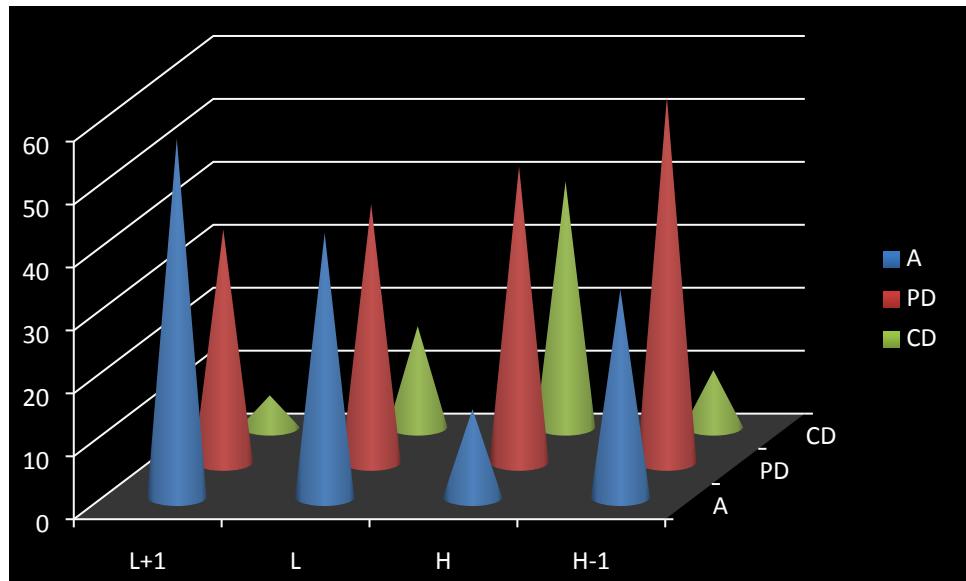


Fig. S7. Graphical representation of percentage contribution of different moieties of **ICT3**.

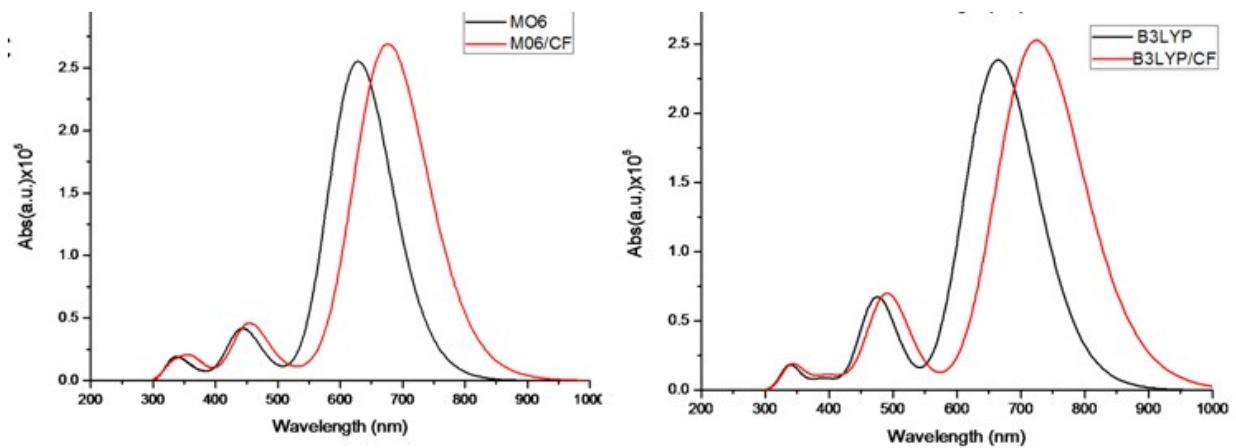


Fig. S8. Simulated UV-Vis Absorption spectra of **ICT3** using M06 and B3LYP in gas phase and solvent chloroform (CF)

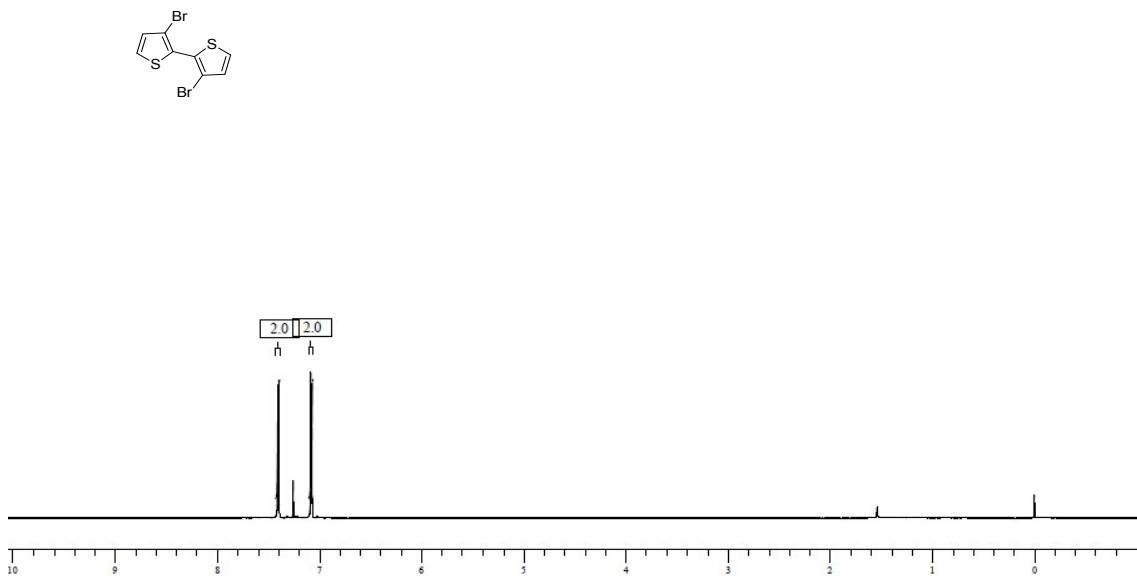


Fig. S9. ¹H NMR spectrum of 1

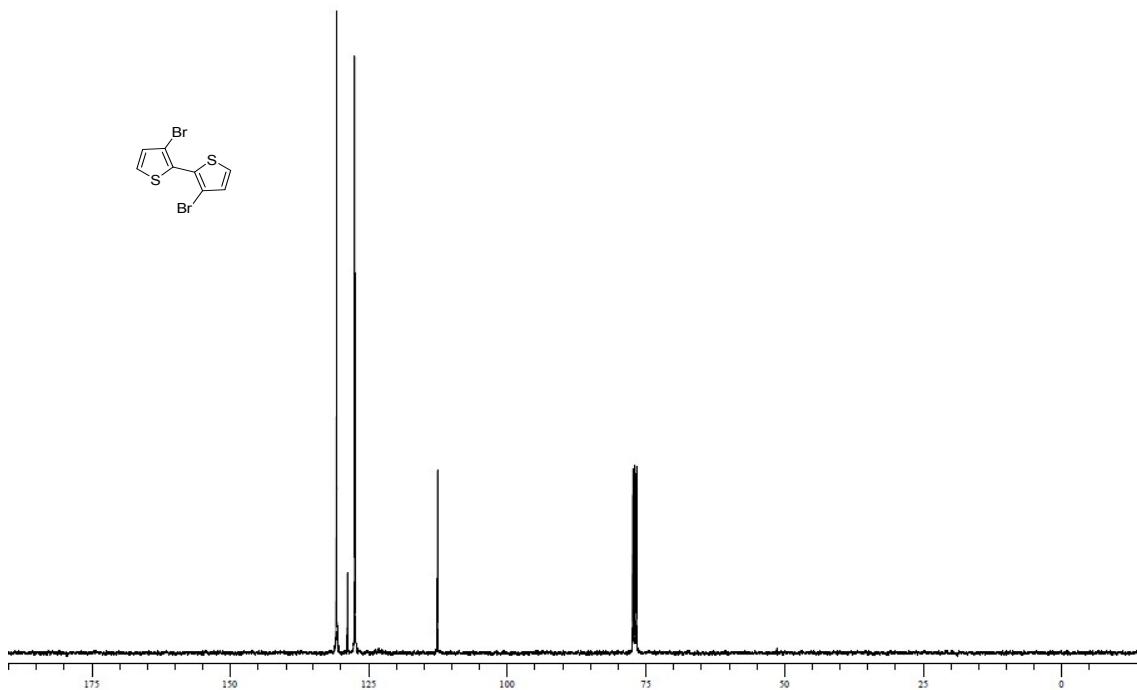


Fig. S10. ¹³C NMR spectrum of 1

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Creation Parameters: Average(1) Time: 0.40..0.43
Comment: JRAO-MR-BRB, EIHRMS ,DIP-300
 $\times 10^6$ Intensity (%)

Experiment Date/Time: 08-09-2016 17:47:56
Ionization Mode: El+

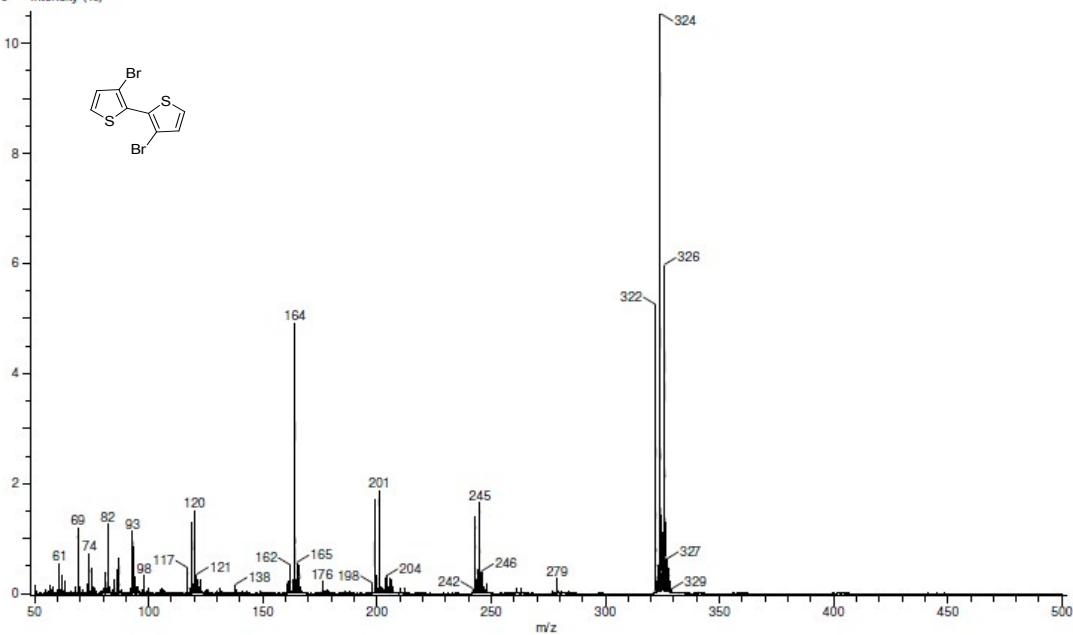


Fig. S11. EI-MS spectrum of 1

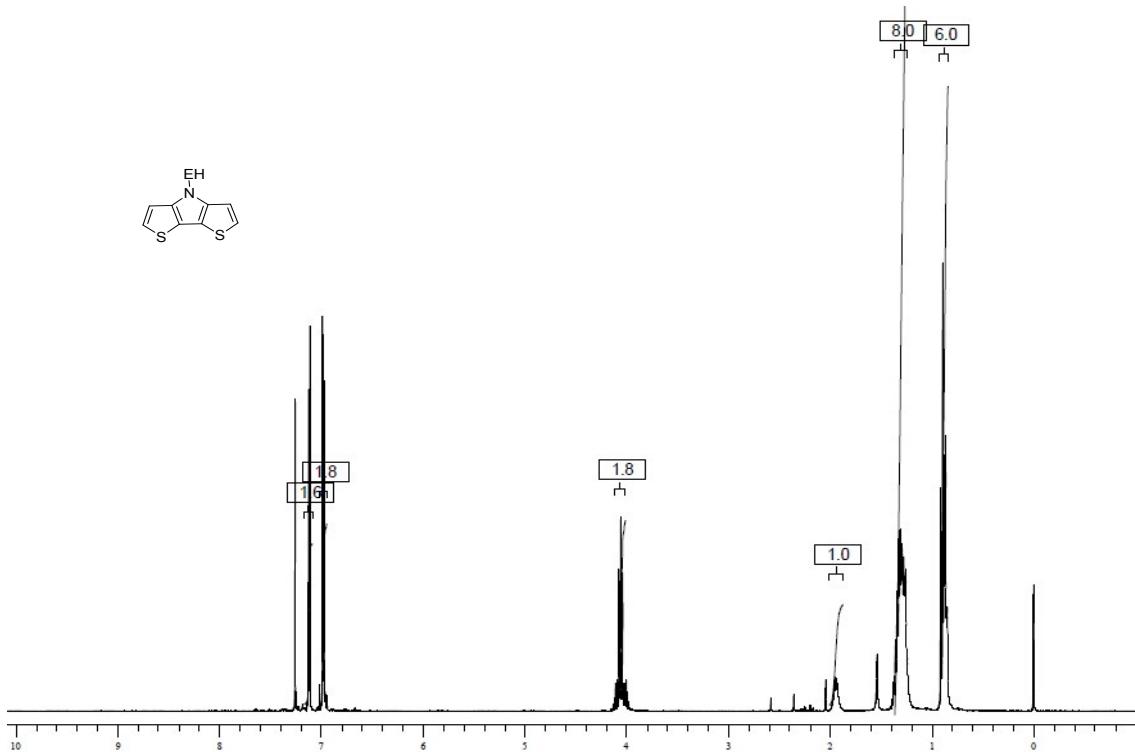


Fig. S12. ^1H NMR spectrum of 2

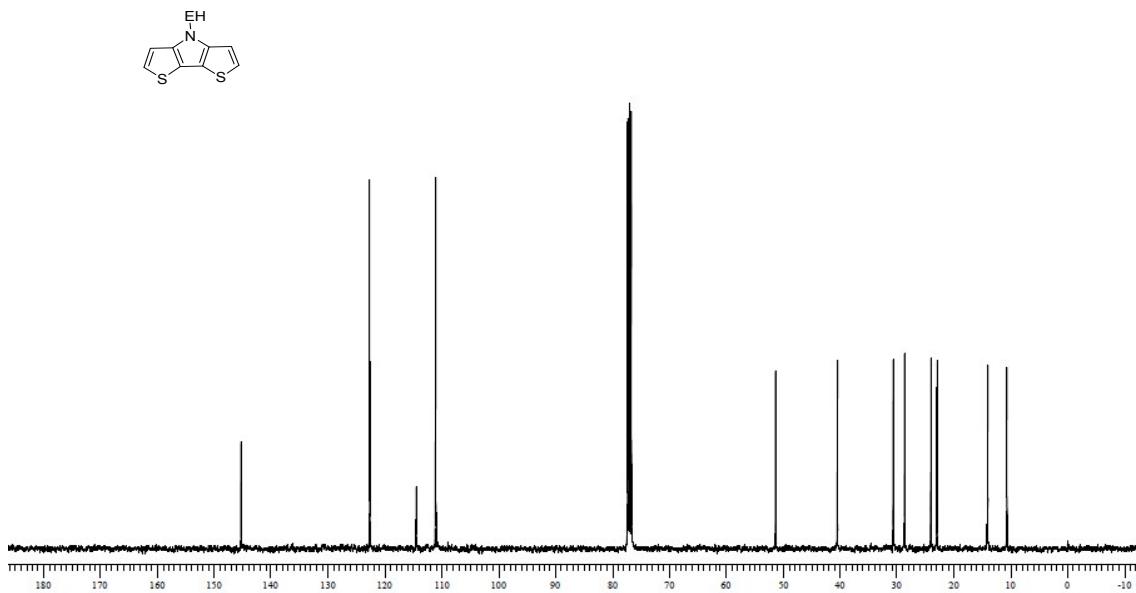


Fig. S13. ^{13}C NMR spectrum of **2**

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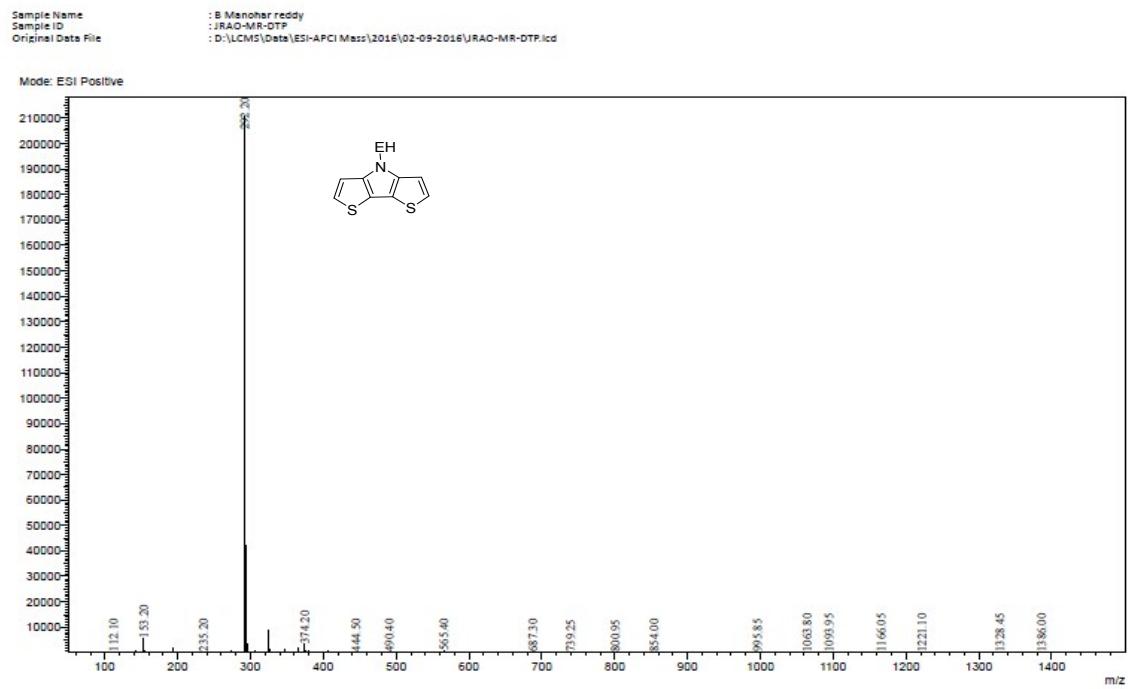


Fig. S14. ESI-MS spectrum of **2**

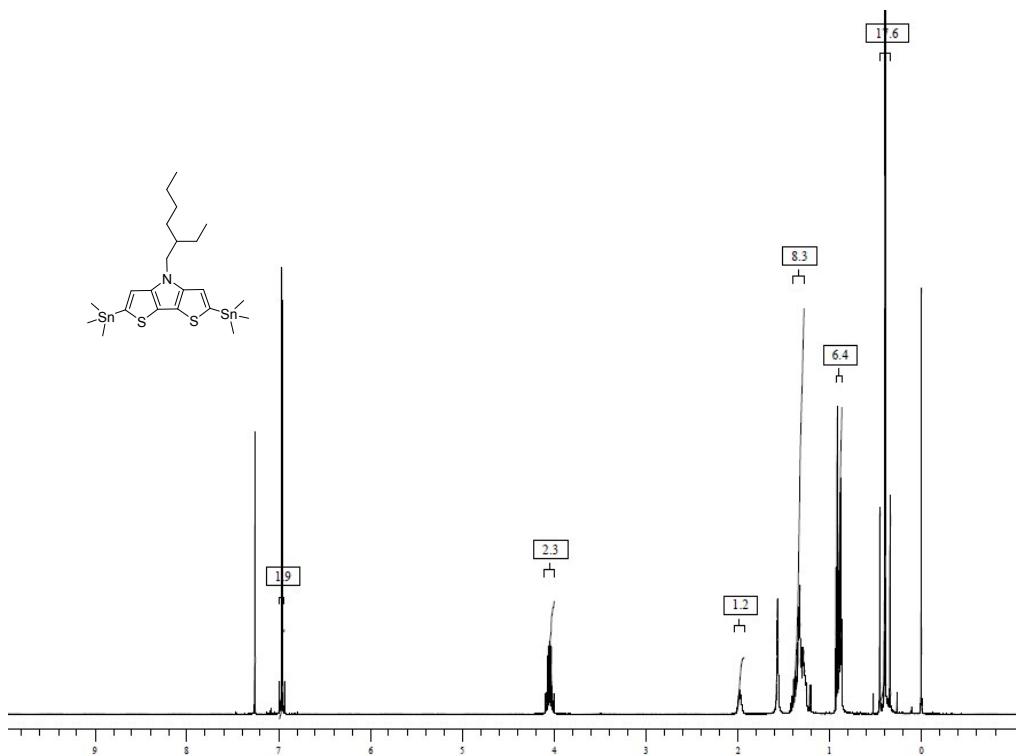


Fig. S15. ^1H NMR spectrum of **3**

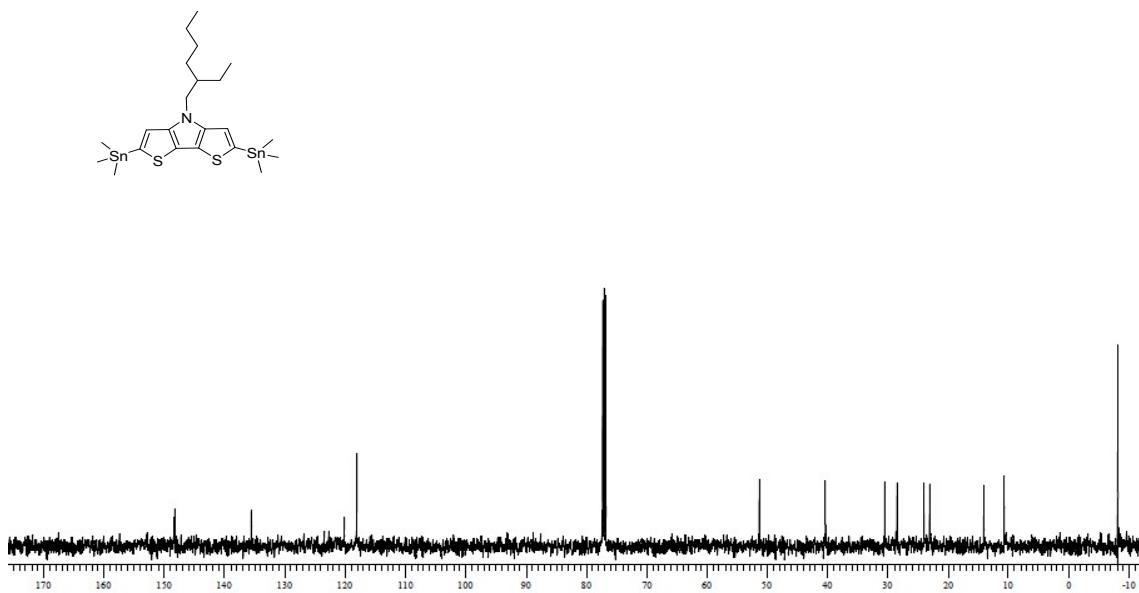


Fig. S16. ^{13}C NMR spectrum of **3**

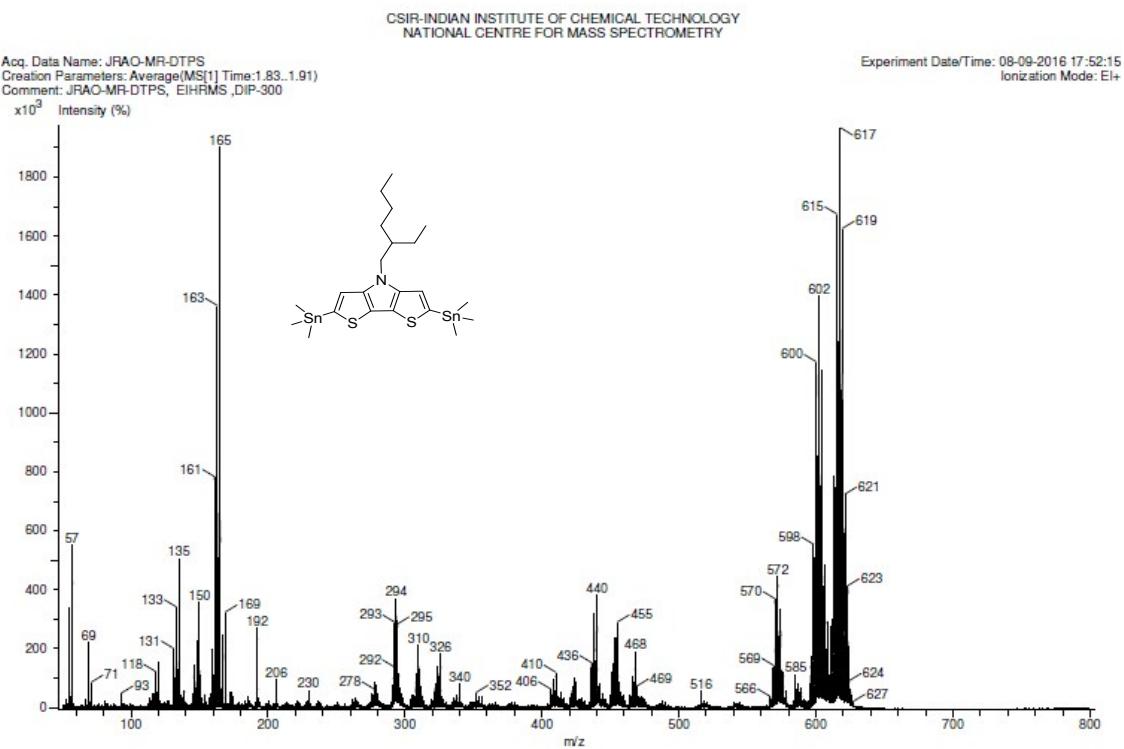


Fig. S17. EI-MS spectrum of 3

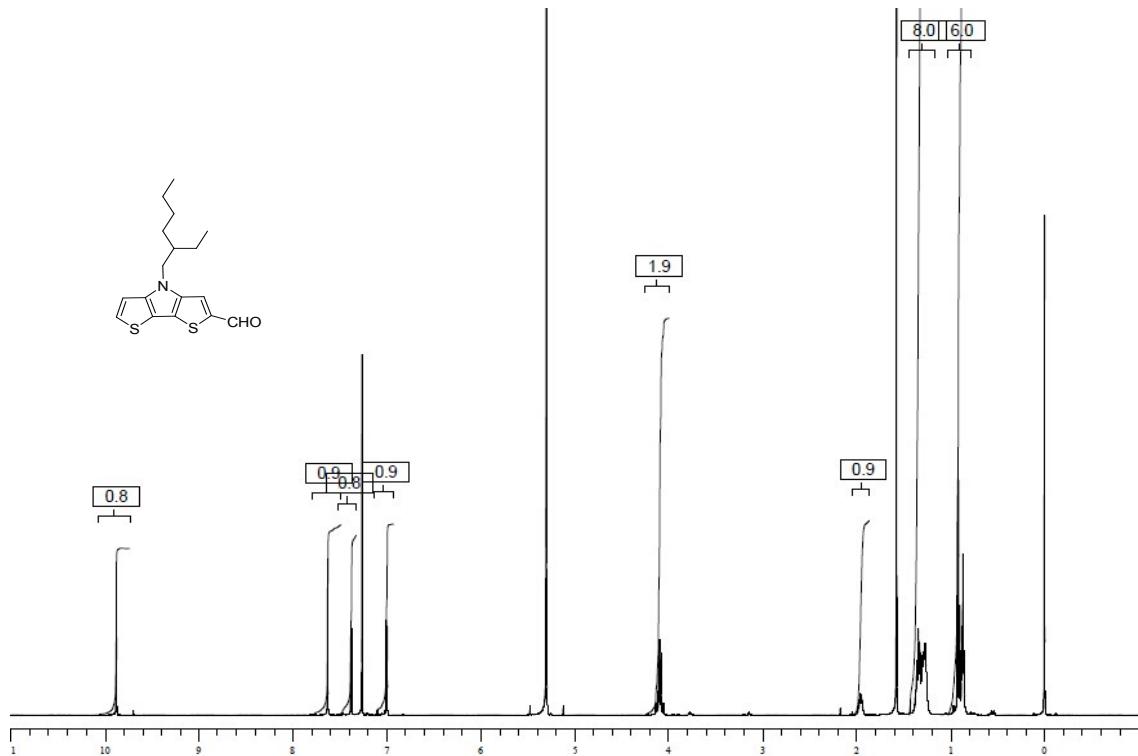


Fig. S18. ^1H NMR spectrum of 4

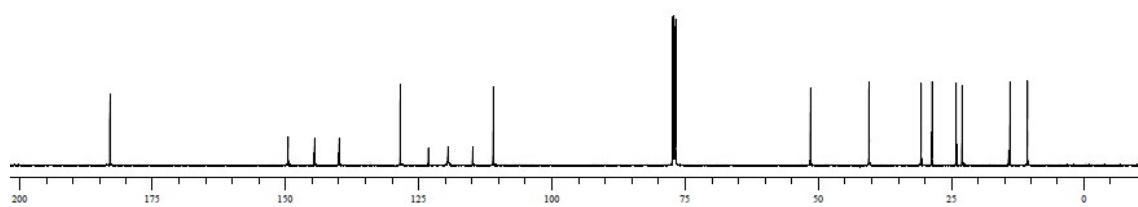
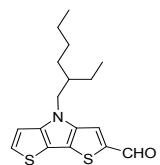


Fig. S19. ¹³C NMR spectrum of 4

==== CPC DIVISION @ CSIR-IICT ===

Sample Name : B Manohar reddy
Sample ID : JRAO-MR-DTPA
Original Data File : D:\LCMS\Datas\ESI-APCI Mass\2016\02-09-2016\JRAO-MR-DTPA.lcd

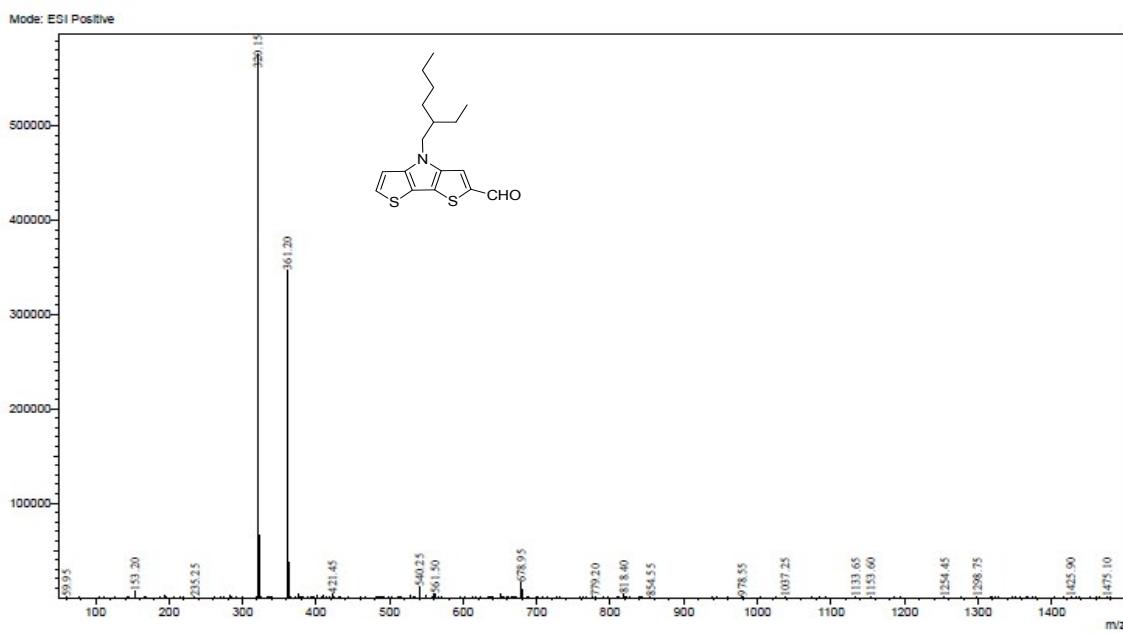


Fig. S20. ESI-MS spectrum of 4

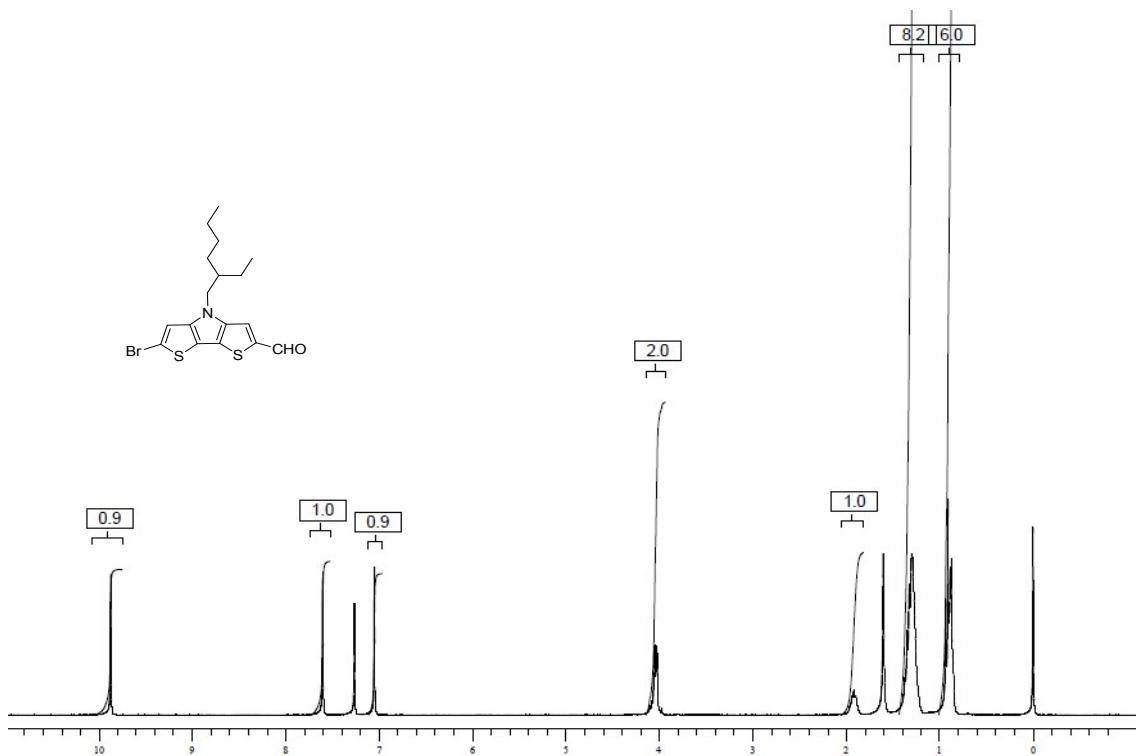


Fig. S21. ^1H NMR spectrum of **5**

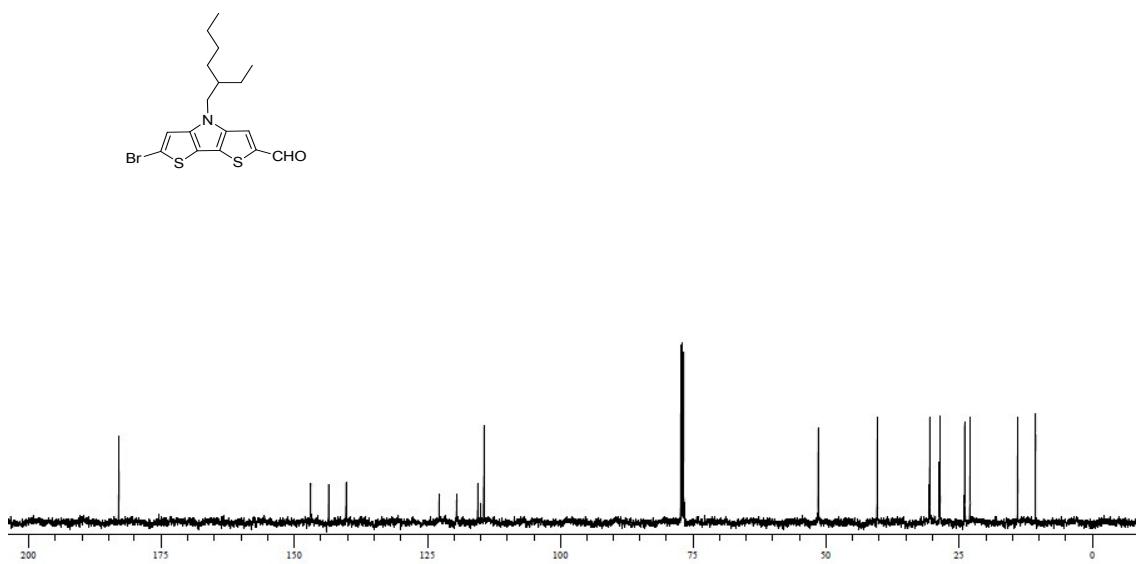


Fig. S22. ^{13}C NMR spectrum of **5**

== CPC DIVISION @ CSIR-IICT ==

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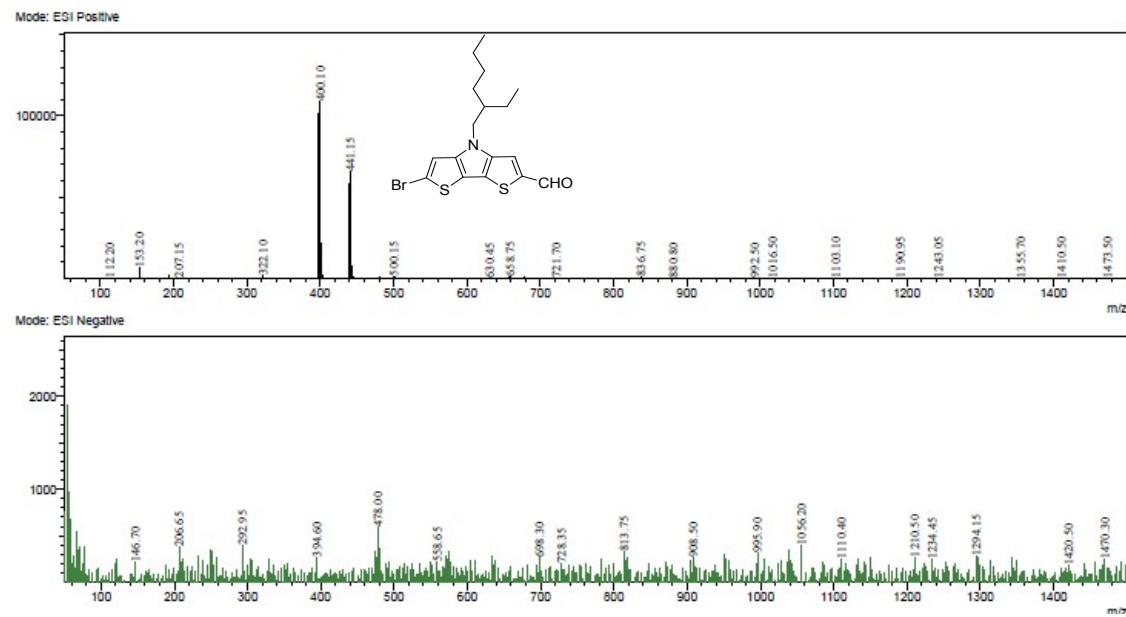


Fig. S23. ESI-MS spectrum of 5

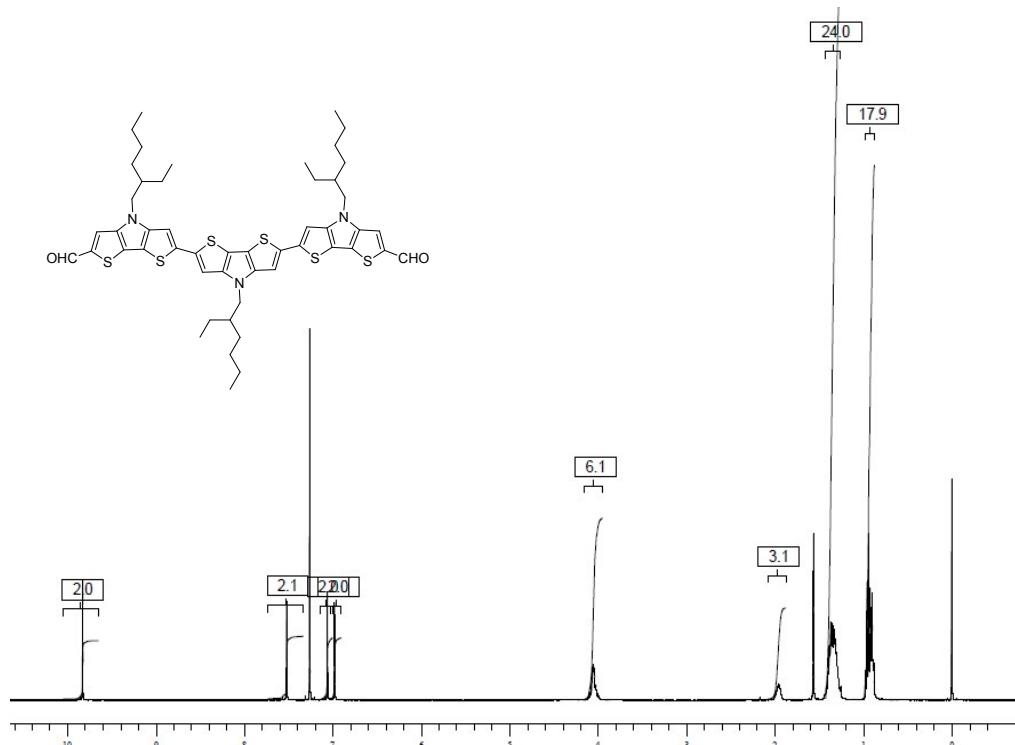


Fig. S24. ^1H NMR spectrum of 6

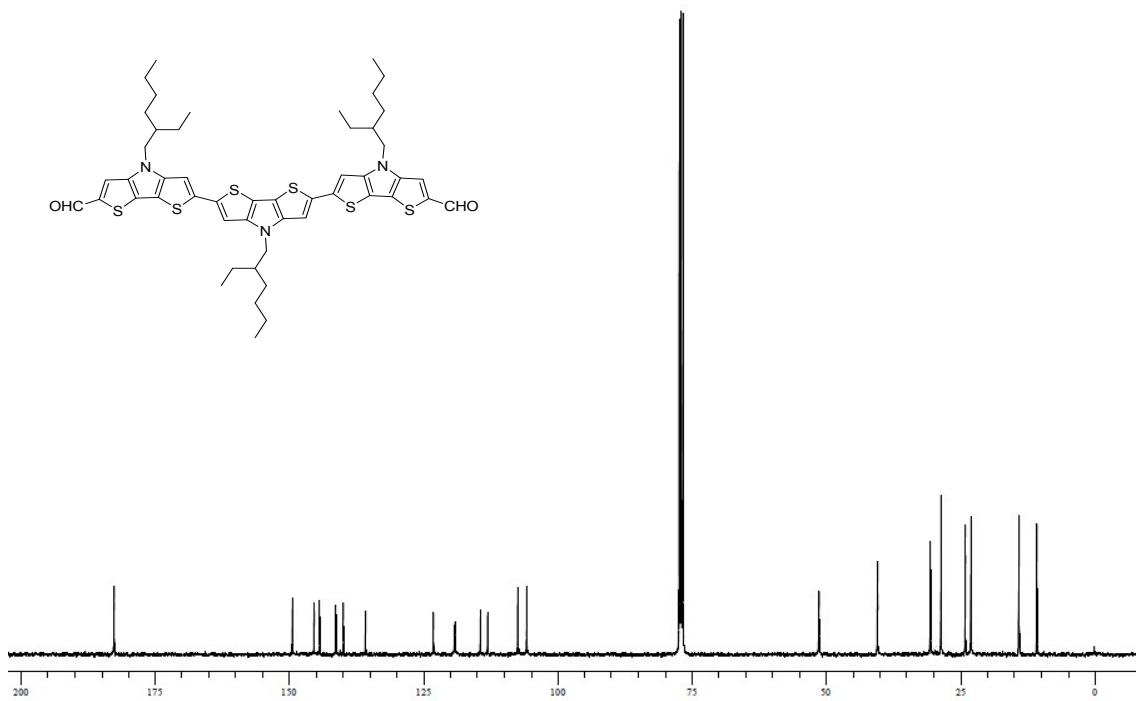


Fig. S25. ^{13}C NMR spectrum of **6**

MANOHAR REDDY , JRAO-MR-NA

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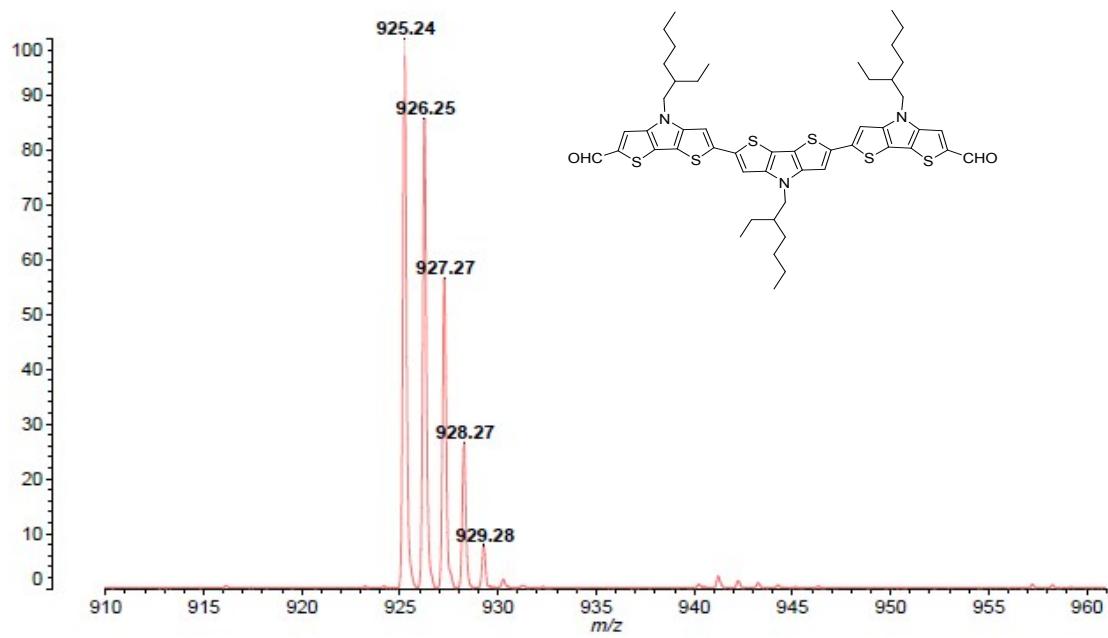


Fig. S26. ESI-MS spectrum of **6**

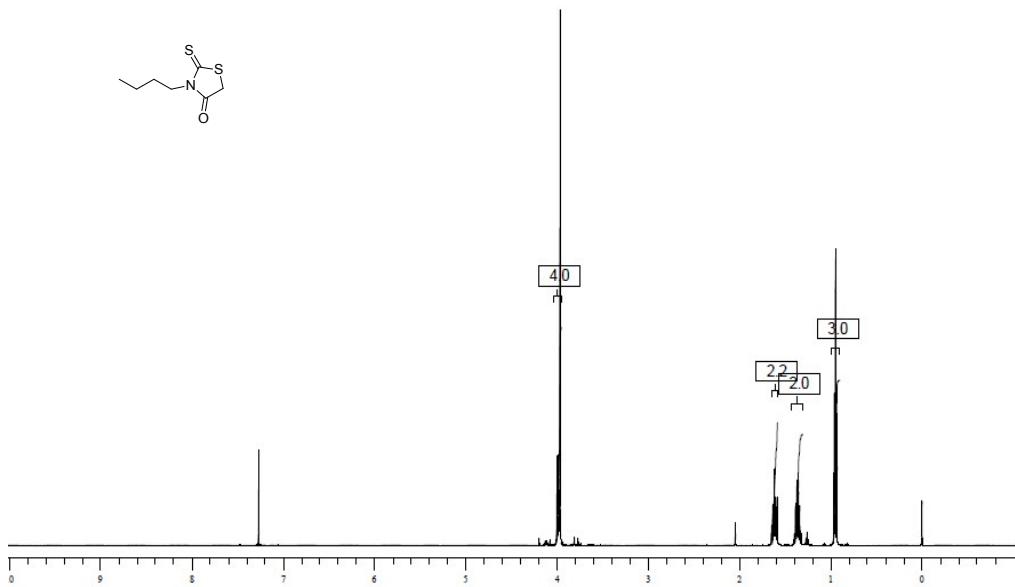


Fig. S27. ¹H NMR spectrum of *n*-butyl rhodanine