Theoretical study on  $\sigma$ - and  $\pi$ -hole carbon  $\cdots$  carbon bonding interactions: Implications in CFC chemistry.

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# **Electronic Supplementary Information**

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## Cartesian coordinates

### 6.

С	0.000000	0.0000000	-1.0940769
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Η	1.5012712	-0.5094001	0.8921462
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Η	-1.6408257	-0.9290205	-1.2309646
Η	-1.5622816	0.9191775	-1.2311374
С	-1.5982298	-0.0050278	0.6666037
Η	-1.5622816	0.9191775	1.2311374
Н	-1.6408257	-0.9290205	1.2309646

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Η	0.5048252	-1.1263230	-0.8934316
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Cl	1.2870790	-0.7398012	-1.4470331
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F	-0.6507710	0.7042891	1.0778579
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F	1.5756190	0.6262412	-1.0779771
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F	1.2826234	0.3364353	1.0776482

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Η	-0.5403672	5.6526028	-0.8940219
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F	-0.0084658	3.9203596	0.0000000

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Η	3.1331865	1.0444702	0.0000000
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Η	-0.8941953	-5.6087351	-0.4536941
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Η	0.8941953	5.6087351	-0.4536941
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F	0.000000	-3.8552108	0.0064040
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Η	-3.0278018	-0.8915879	0.4878539
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Η	3.0278018	-0.8915879	0.4878539
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F	4.7724054	0.000000	-0.0251268

#### **Additional AIM analyses**

In Fig. S1 the distribution of critical points (CPs) and bond paths for some additional complexes is shown. As it can be observed, for complexes 9 and 16 where a CO molecule acts as electron donor three symmetrically distributed bond CPs (red spheres) connect the halogen and carbon atoms. In addition, a cage CP emerges and further characterizes the interaction. On the other hand, in complexes 11 and 13 it can be noticed three bond CPs that connect the halogen atoms with the carbon atoms of the  $\pi$ -system. These results are in good in agreement with the NBO analysis, which showed orbital contributions involving the halogen atoms of the CFC molecules.



Fig. S1 Distribution of critical points and bond paths in complexes 9, 11, 13 and 15. Bond, ring and cage critical points are represented by red, yellow and green spheres, respectively. The bond paths connecting bond critical points are also represented. The value of the density at the bond critical point ( $\rho$ ·10<sup>2</sup>) is also indicated.