

Figure S1

Figure S2

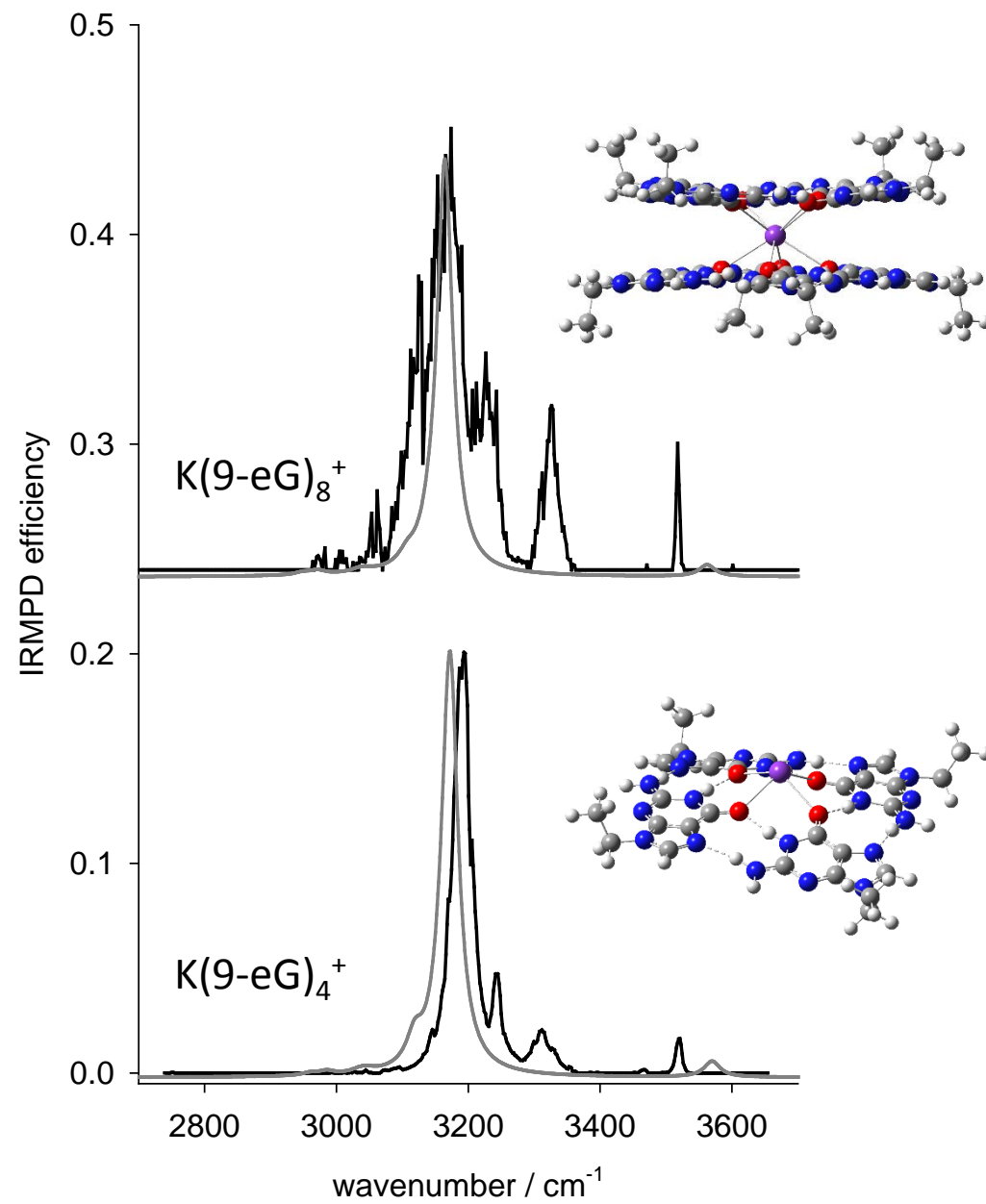
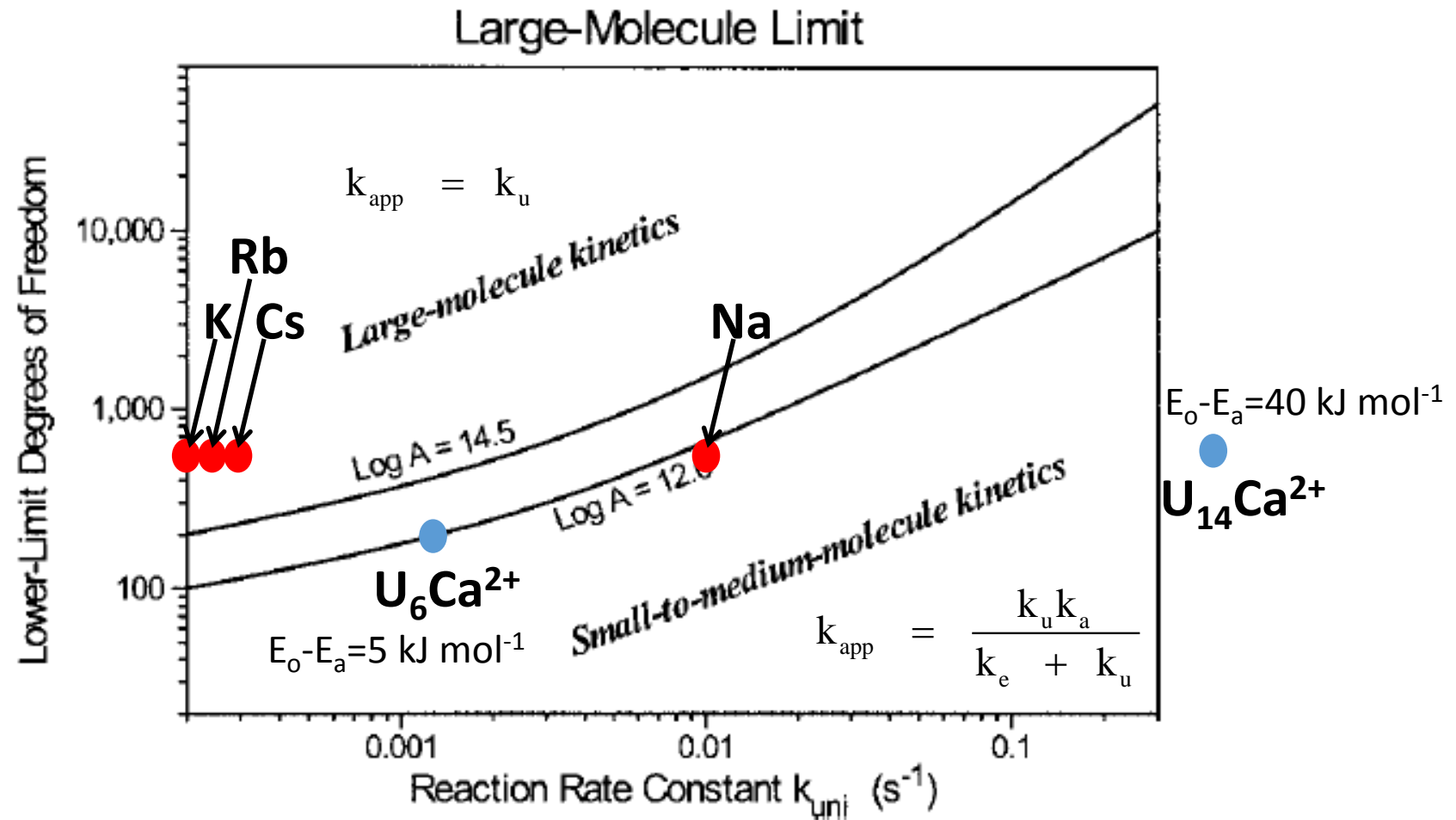


Figure S3



Dunbar, R.C. *Mass Spectrom. Rev.* **2004**, 23, 127.

● E.A.L. Gillis, M. Demireva, K. Nanda, G.J.O. Beran, E. Williams, and T.D. Fridgen *Phys. Chem. Chem. Phys.* **2012**, 14, 3304.

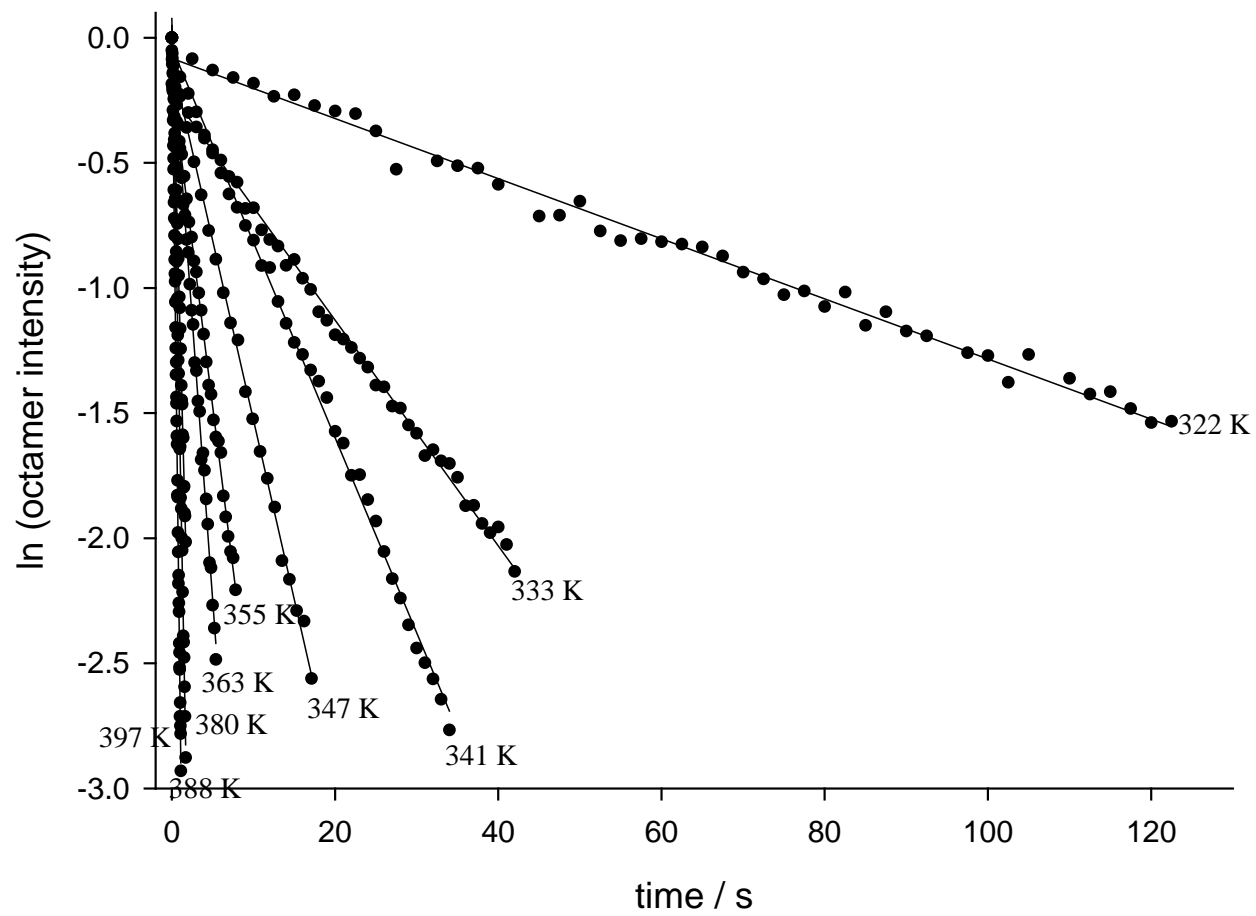
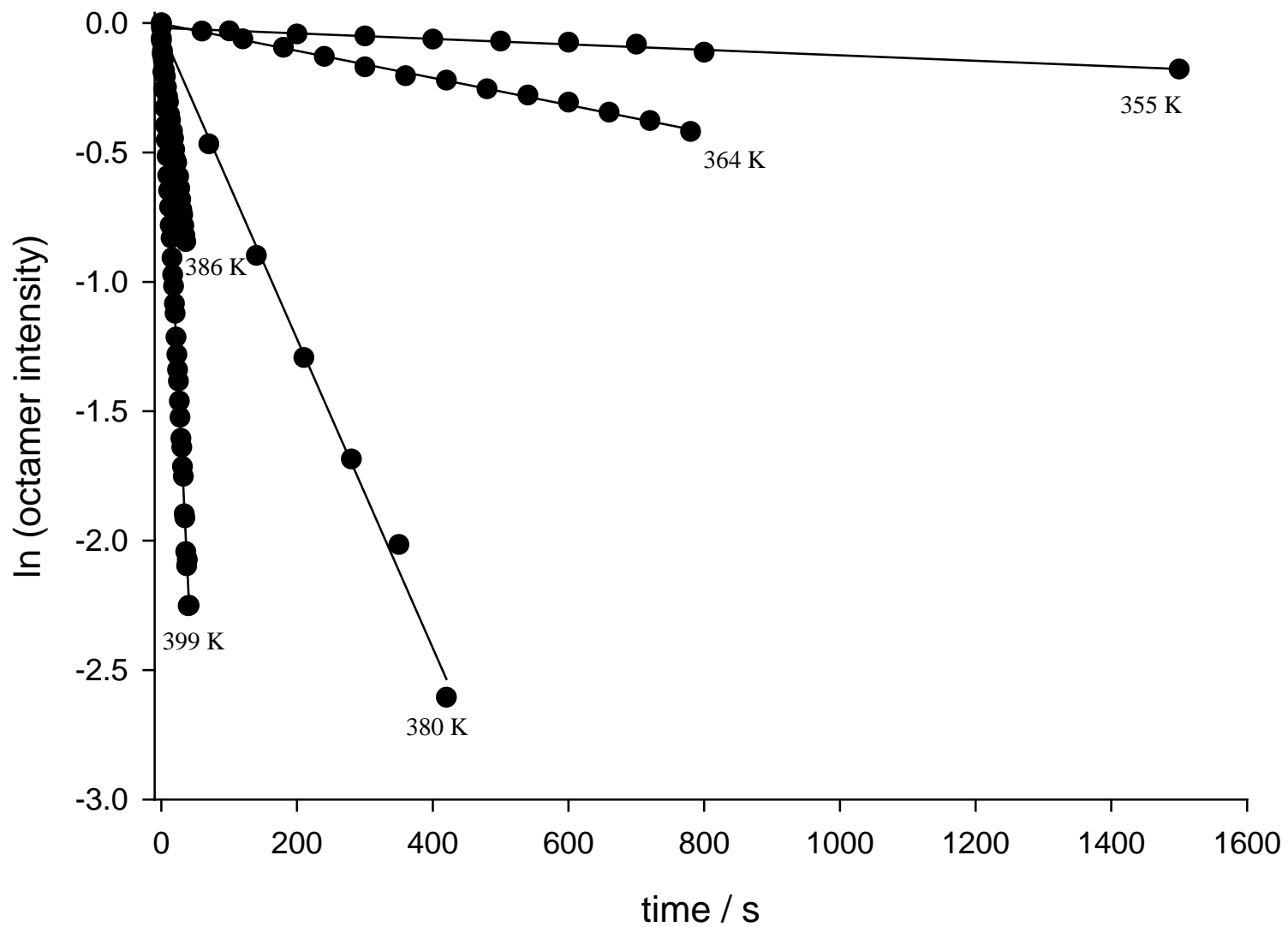
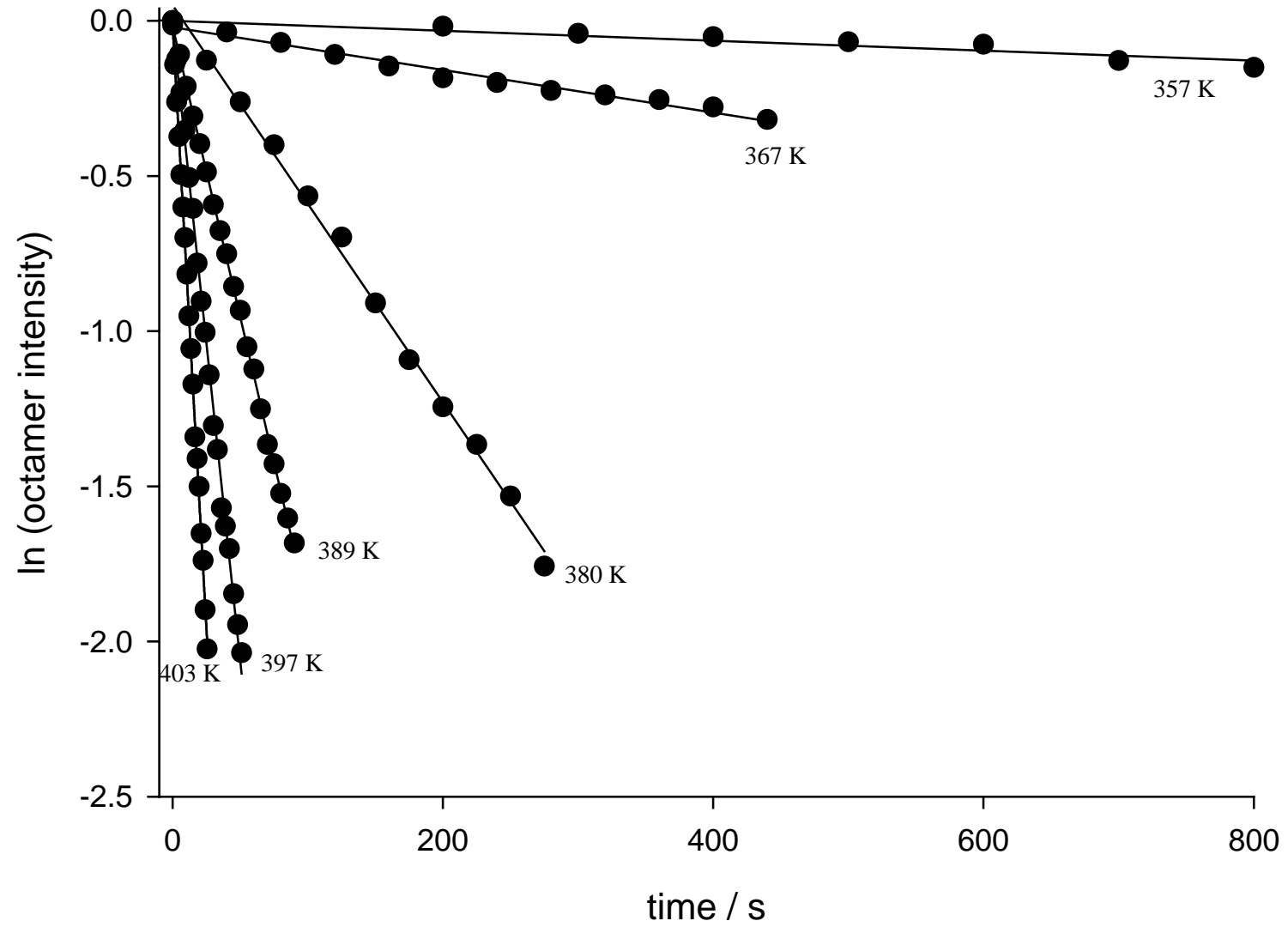
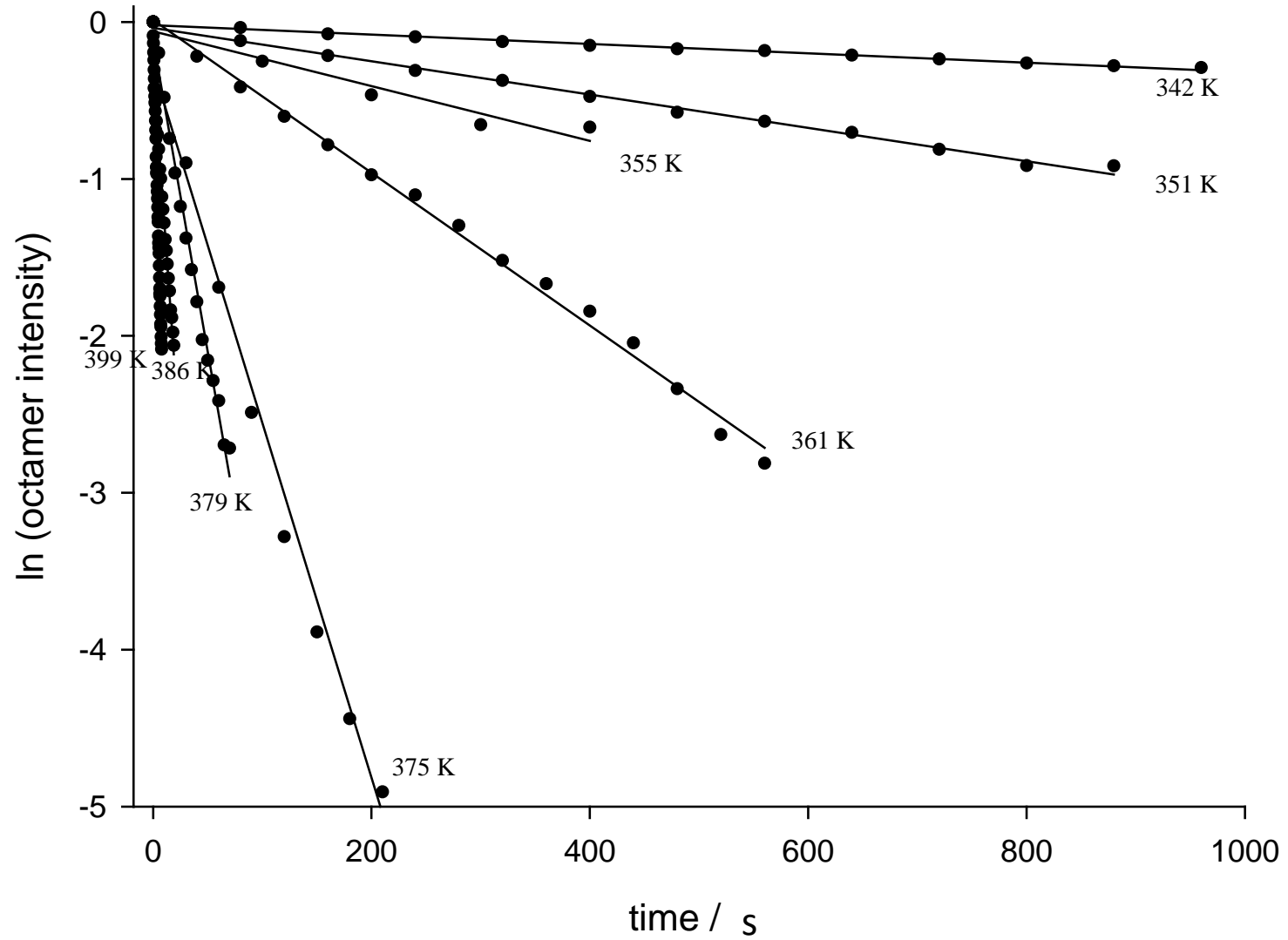


Figure S4.

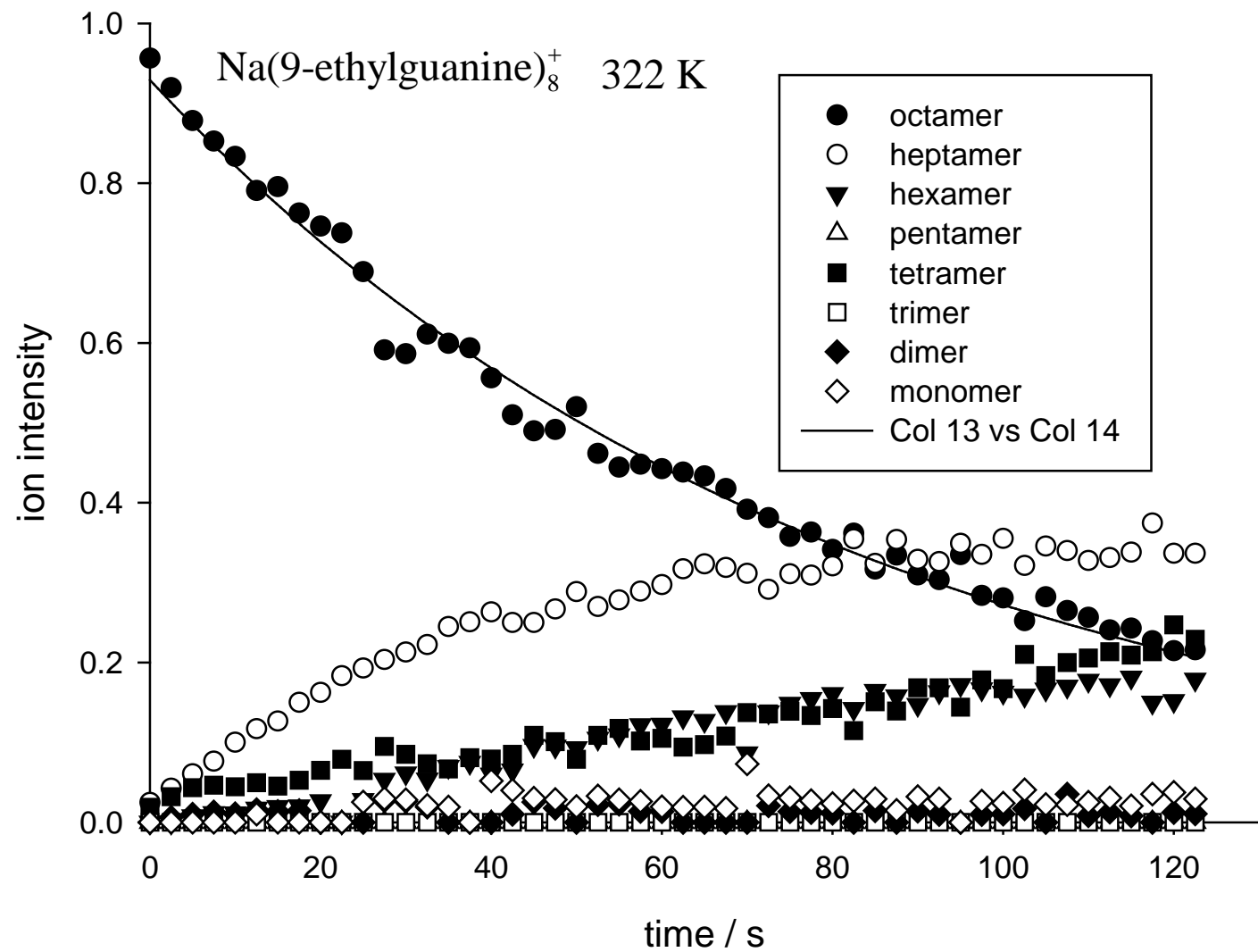


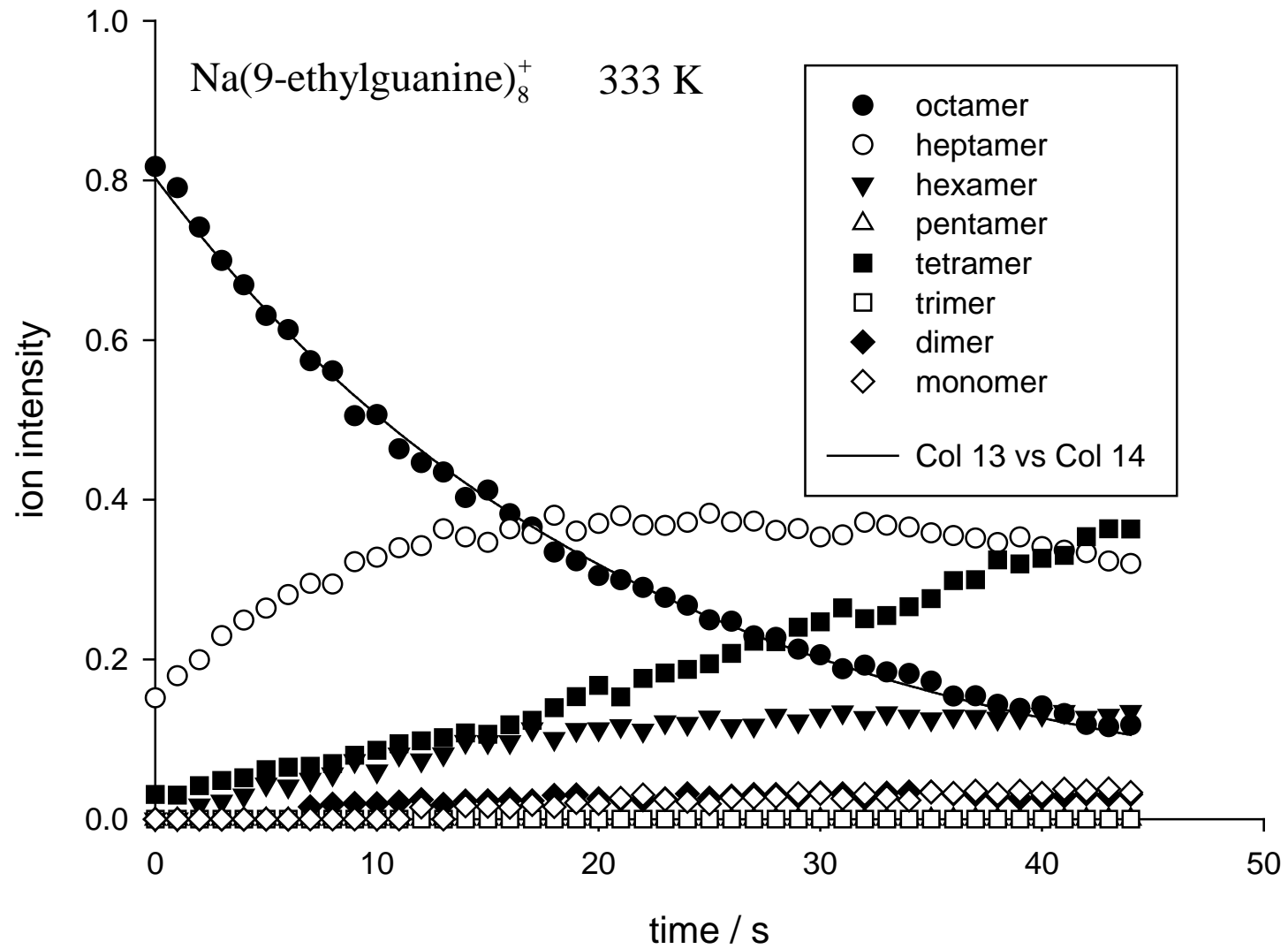


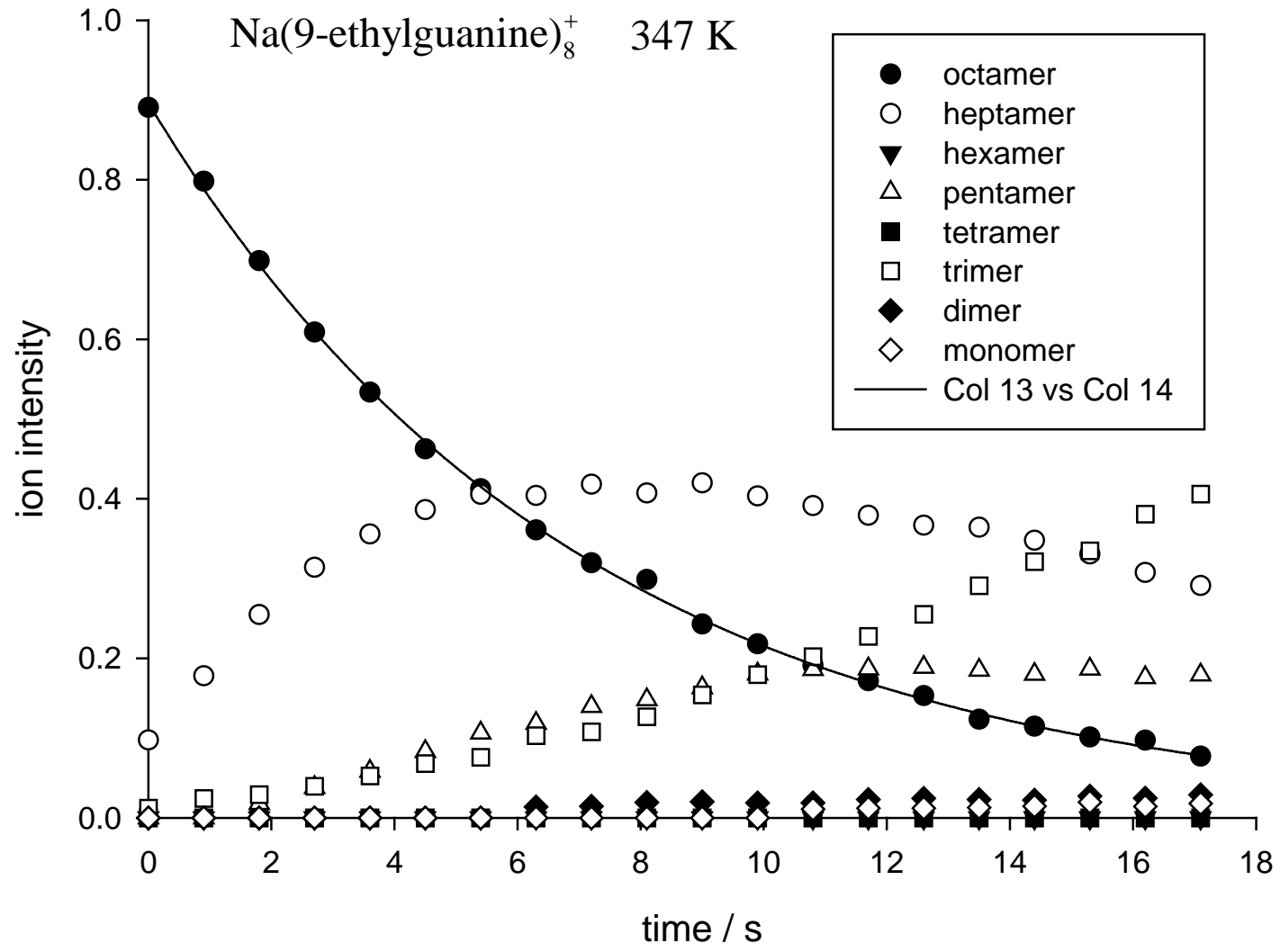


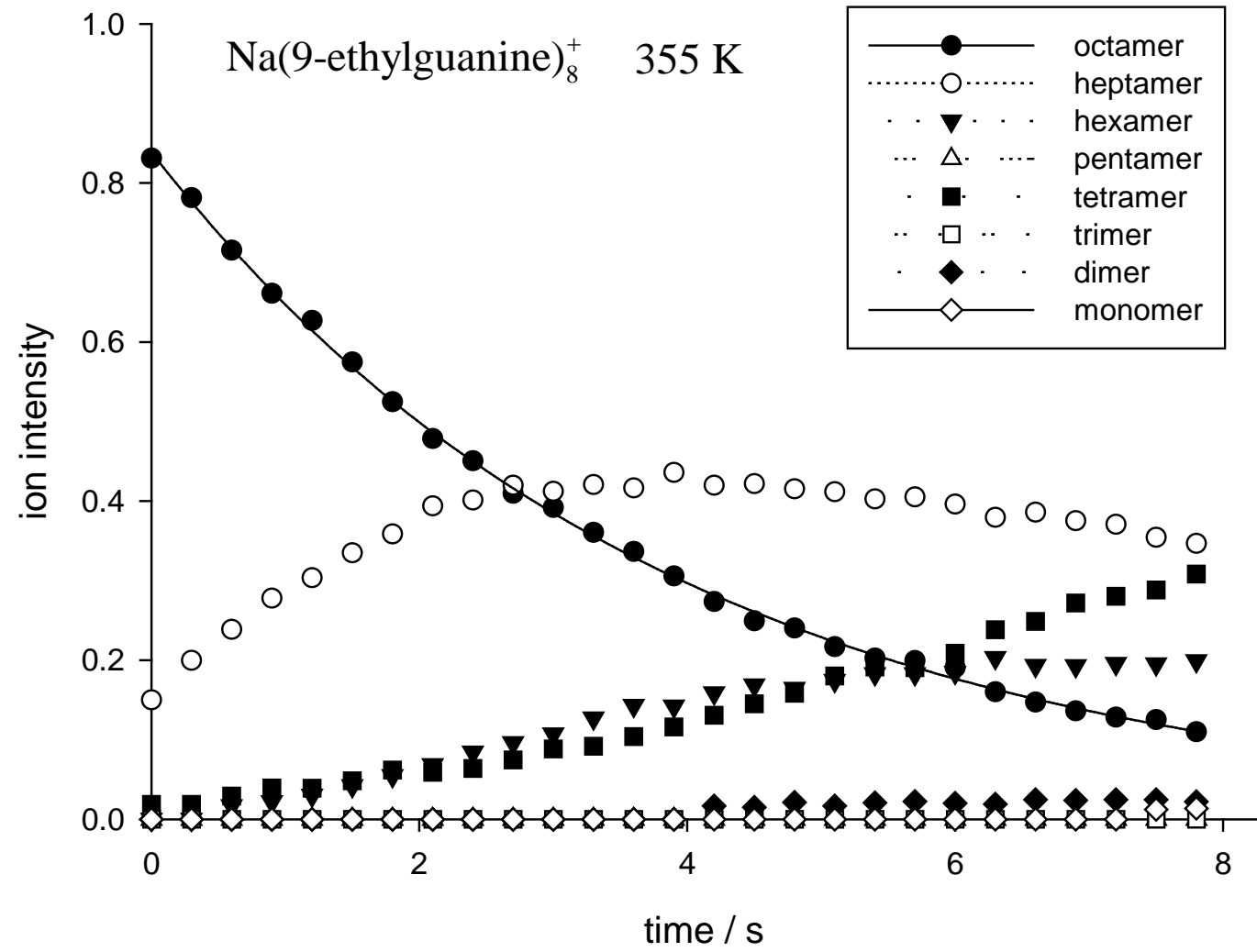
$\text{Na}(9\text{eG})_8^+$	$\text{K}(9\text{eG})_8^+$	$\text{Rb}(9\text{eG})_8^+$	$\text{Cs}(9\text{eG})_8^+$
$k_{322} = 1.20(\pm 0.02) \times 10^{-2} \text{ s}^{-1}$	—	—	—
$k_{333} = 4.45(\pm 0.04) \times 10^{-2} \text{ s}^{-1}$	—	—	—
$k_{341} = 7.48(\pm 0.07) \times 10^{-2} \text{ s}^{-1}$	—	—	$k_{342} = 1.98(\pm 0.09) \times 10^{-4} \text{ s}^{-1}$
$k_{347} = 1.42(\pm 0.02) \times 10^{-1} \text{ s}^{-1}$	—	—	$k_{351} = 1.06 (\pm 0.03) \times 10^{-3} \text{ s}^{-1}$
$k_{355} = 2.60(\pm 0.02) \times 10^{-1} \text{ s}^{-1}$	$k_{355} = 1.06(\pm 0.03) \times 10^{-4} \text{ s}^{-1}$	$k_{357} = 2.13(\pm 0.16) \times 10^{-4} \text{ s}^{-1}$	$k_{355} = 1.63(\pm 0.22) \times 10^{-3} \text{ s}^{-1}$
$k_{363} = 4.47(\pm 0.04) \times 10^{-1} \text{ s}^{-1}$	$k_{364} = 5.24(\pm 0.07) \times 10^{-4} \text{ s}^{-1}$	$k_{367} = 6.88(\pm 0.32) \times 10^{-4} \text{ s}^{-1}$	$k_{361} = 4.87(\pm 0.10) \times 10^{-3} \text{ s}^{-1}$
$k_{380} = 1.21(\pm 0.02) \text{ s}^{-1}$	$k_{380} = 5.70(\pm 0.02) \times 10^{-3} \text{ s}^{-1}$	$k_{380} = 6.40(\pm 0.11) \times 10^{-3} \text{ s}^{-1}$	$k_{379} = 3.97(\pm 0.11) \times 10^{-2} \text{ s}^{-1}$
$k_{388} = 1.73(\pm 0.02) \text{ s}^{-1}$	$k_{386} = 2.10(\pm 0.03) \times 10^{-2} \text{ s}^{-1}$	$k_{389} = 1.88(\pm 0.01) \times 10^{-2} \text{ s}^{-1}$	$k_{386} = 9.19(\pm 0.09) \times 10^{-2} \text{ s}^{-1}$
$k_{397} = 2.68(\pm 0.02) \text{ s}^{-1}$	$k_{398} = 5.36(\pm 0.04) \times 10^{-2} \text{ s}^{-1}$	$k_{403} = 7.77(\pm 0.06) \times 10^{-2} \text{ s}^{-1}$	$k_{399} = 267(\pm 0.02) \times 10^{-1} \text{ s}^{-1}$

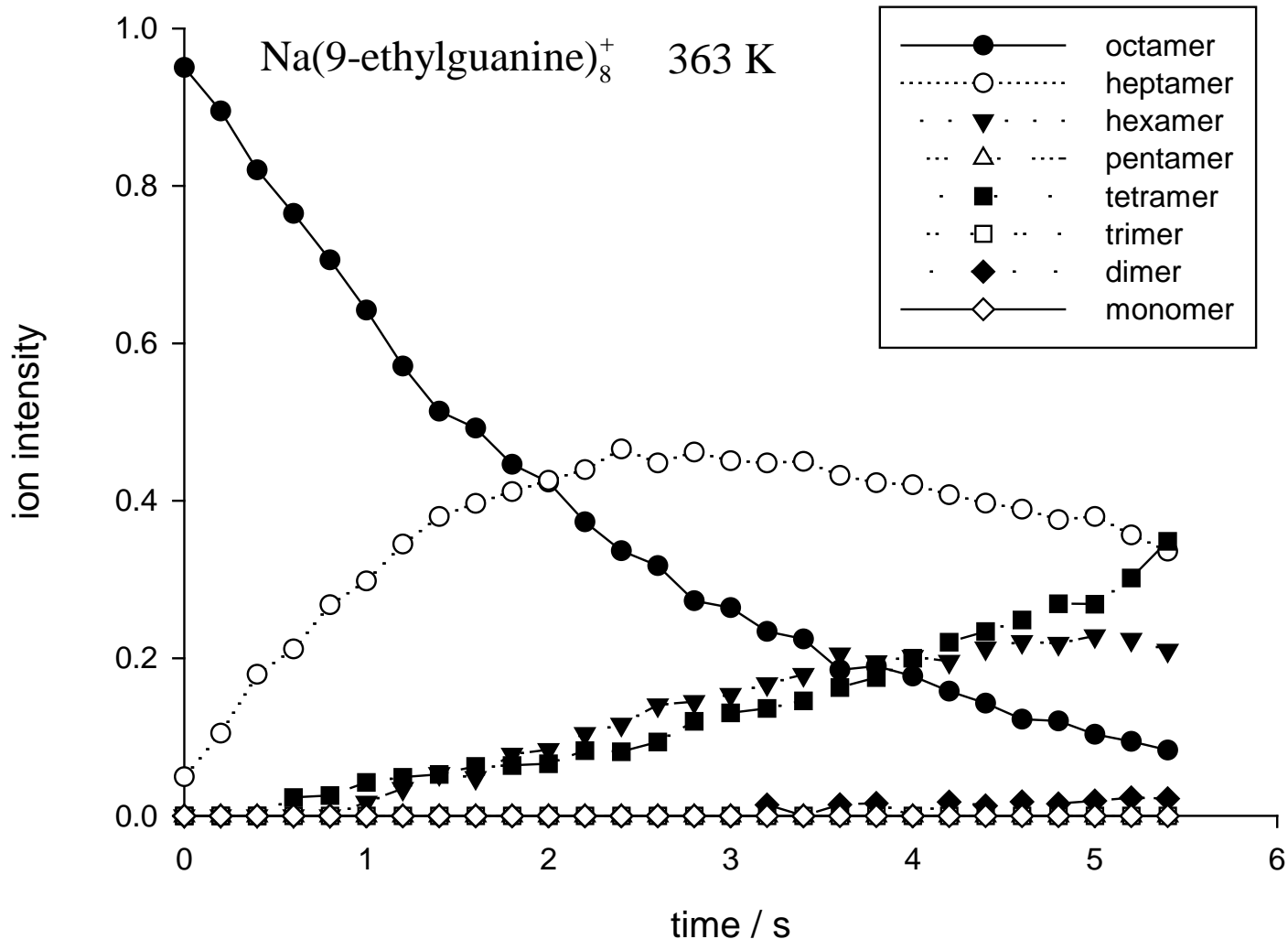
Table 1

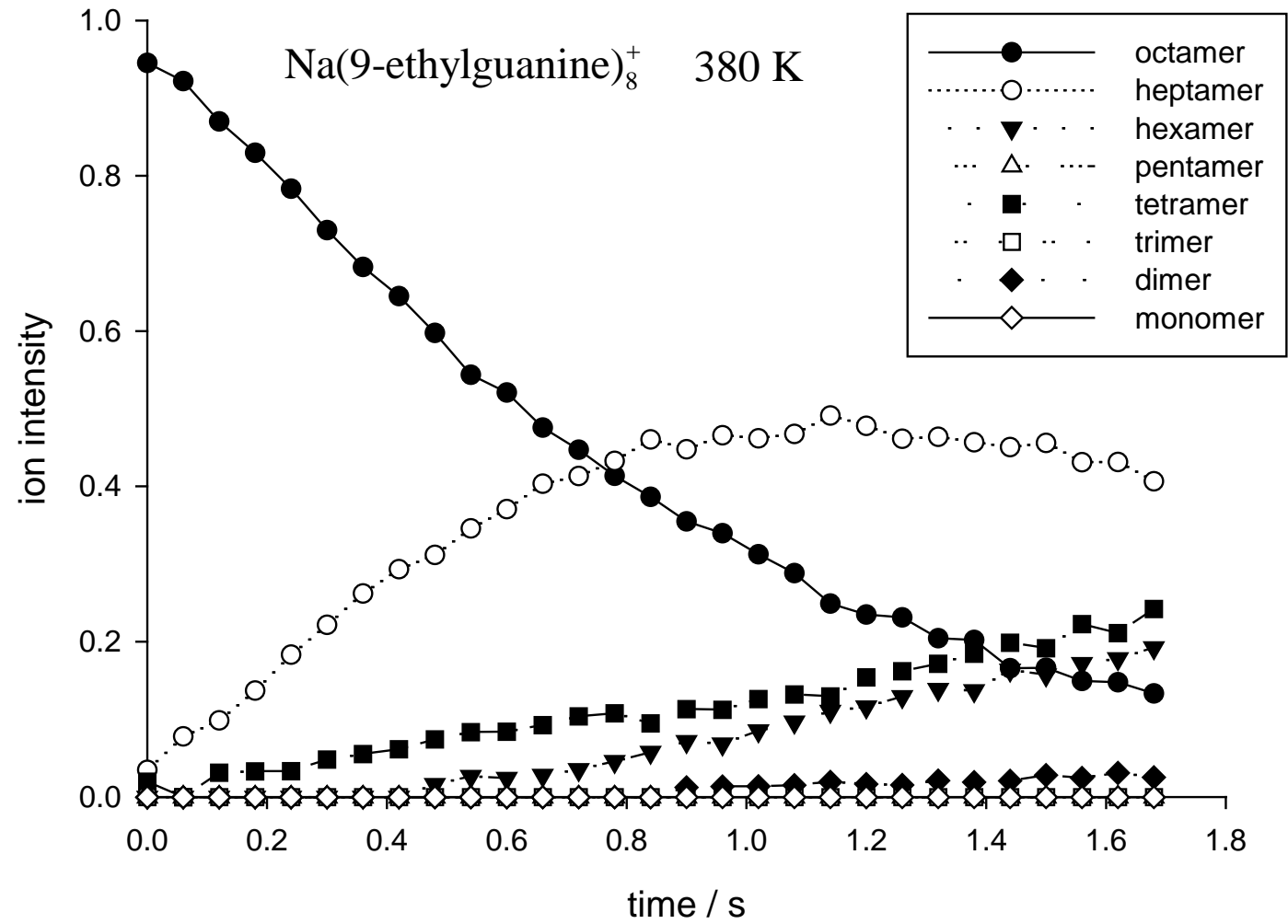


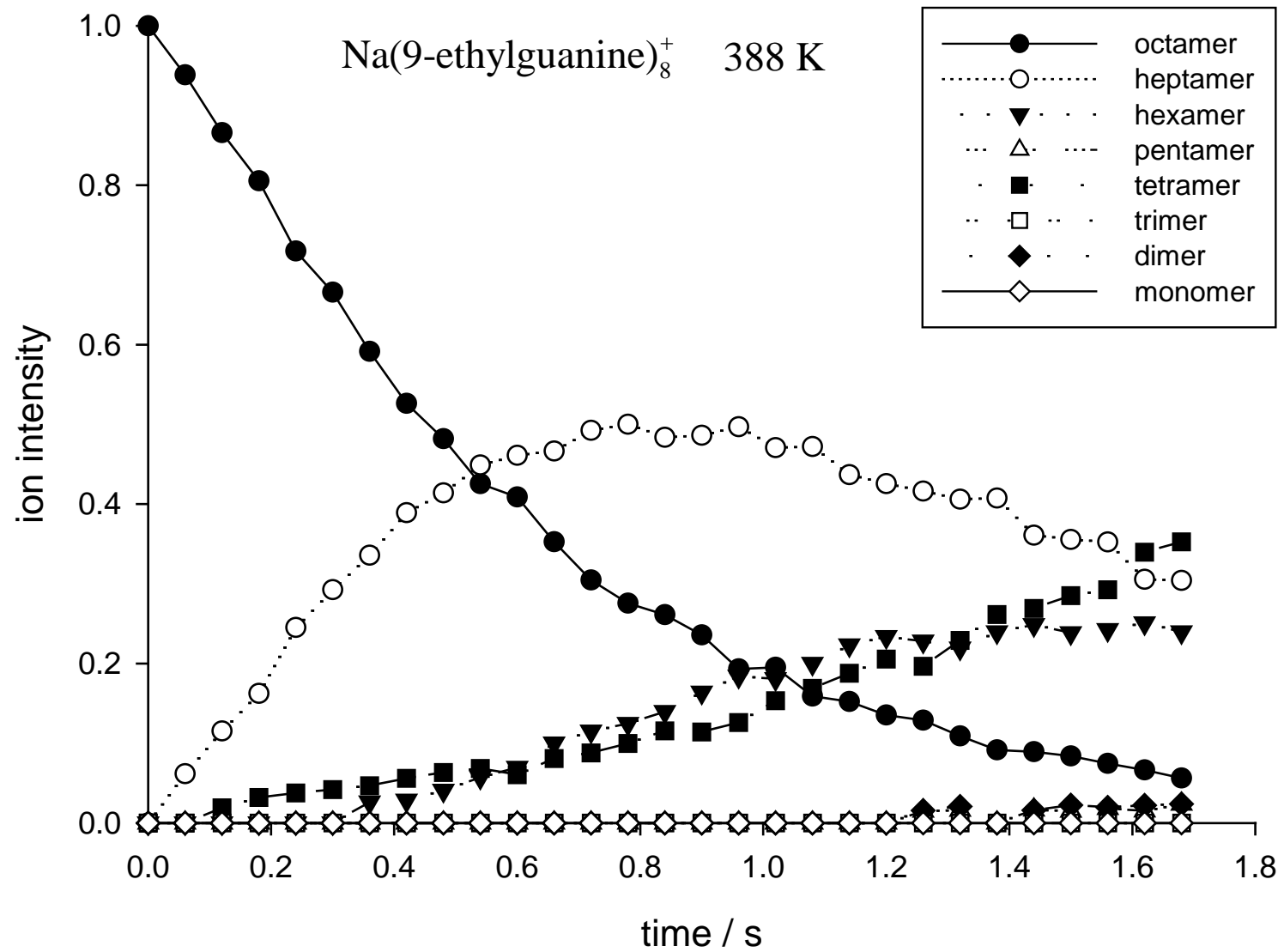


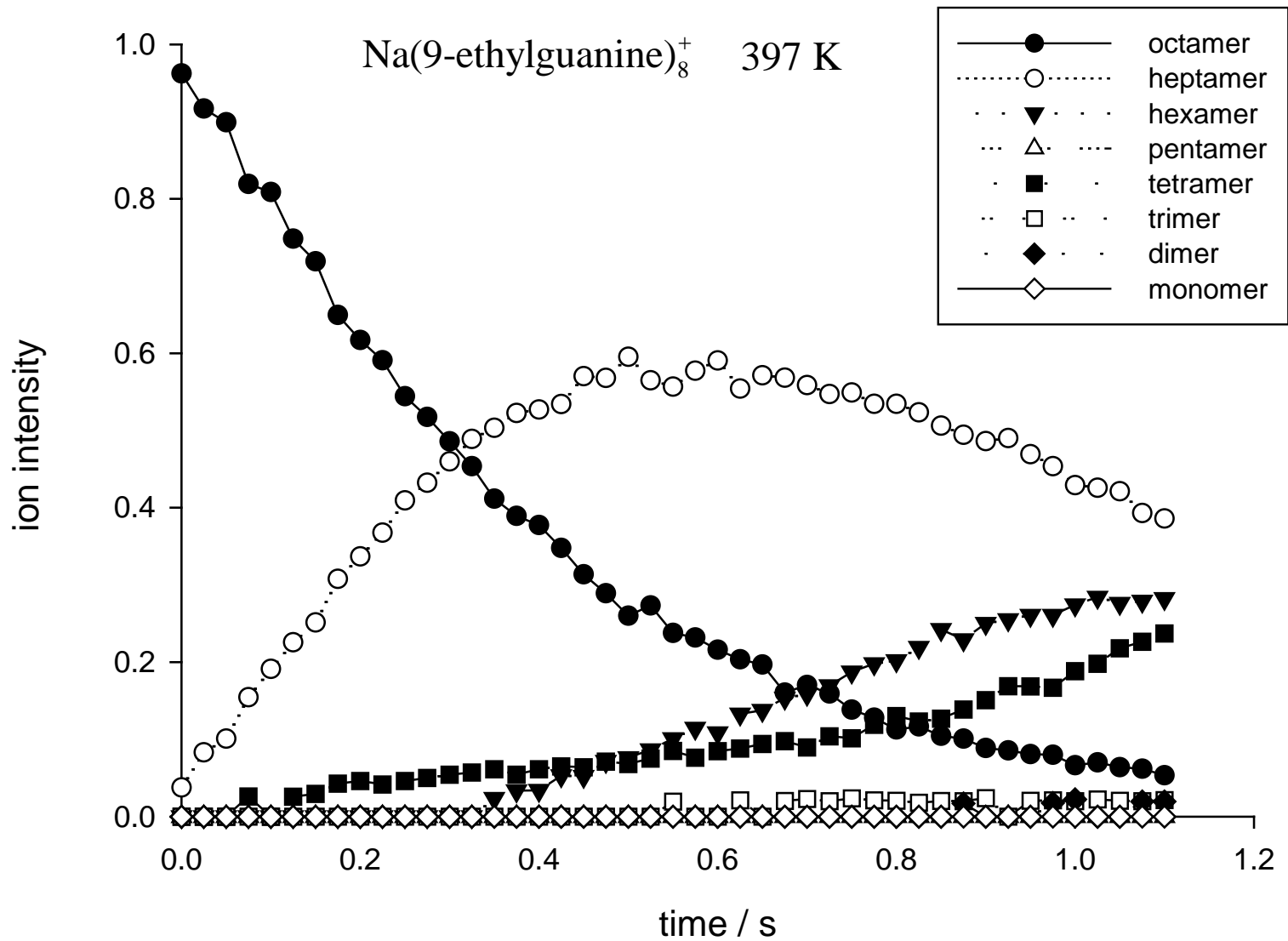


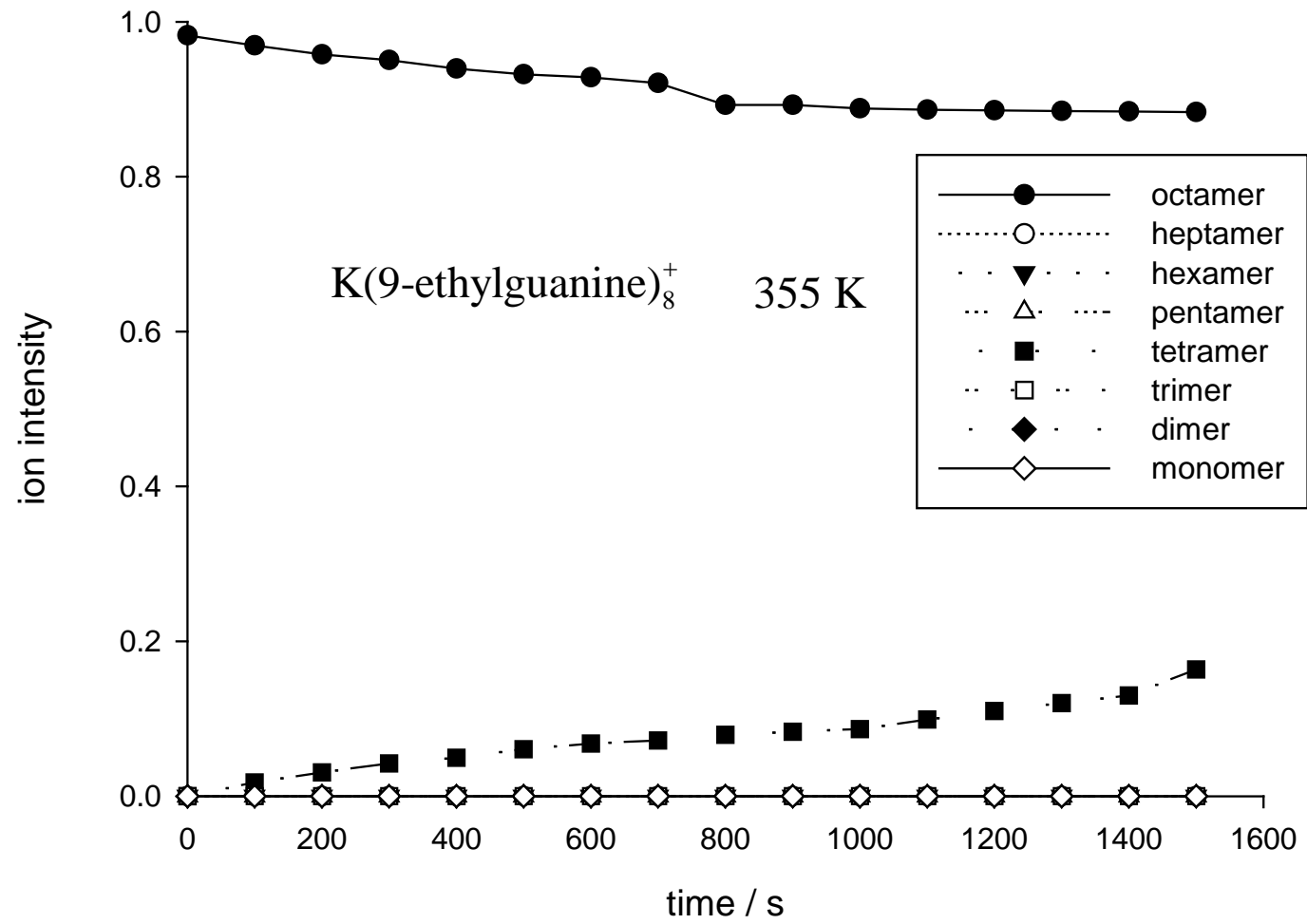


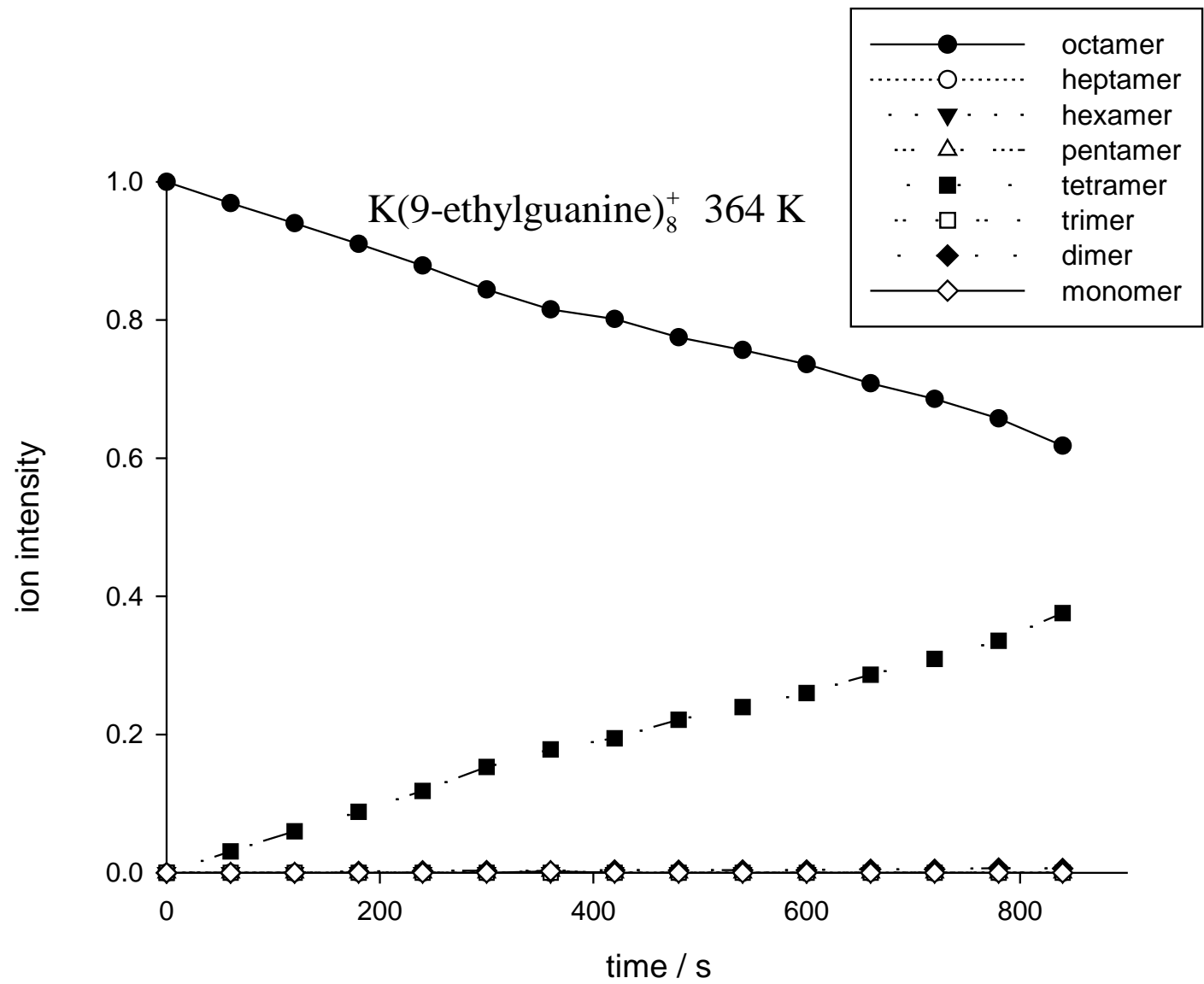


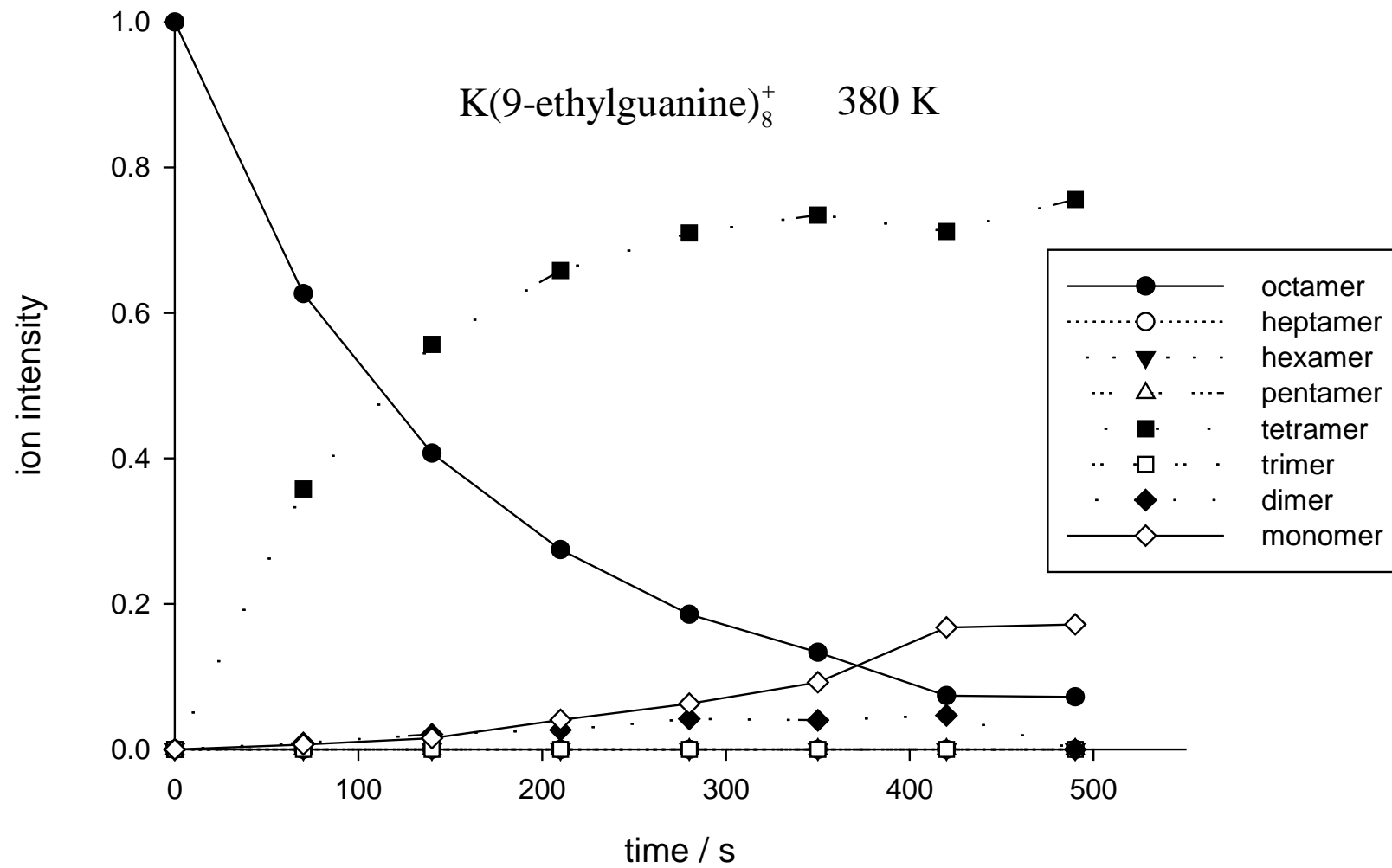


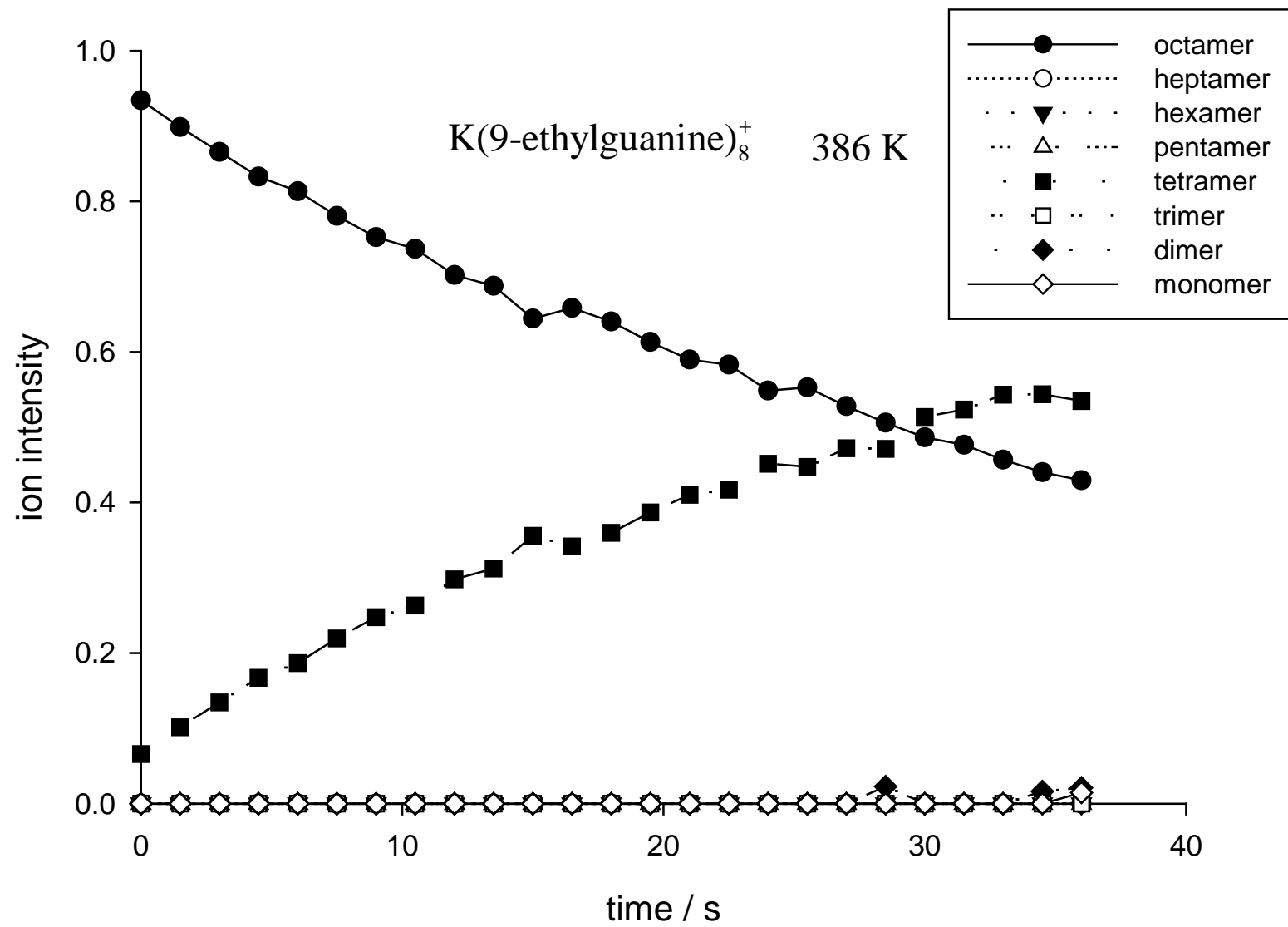


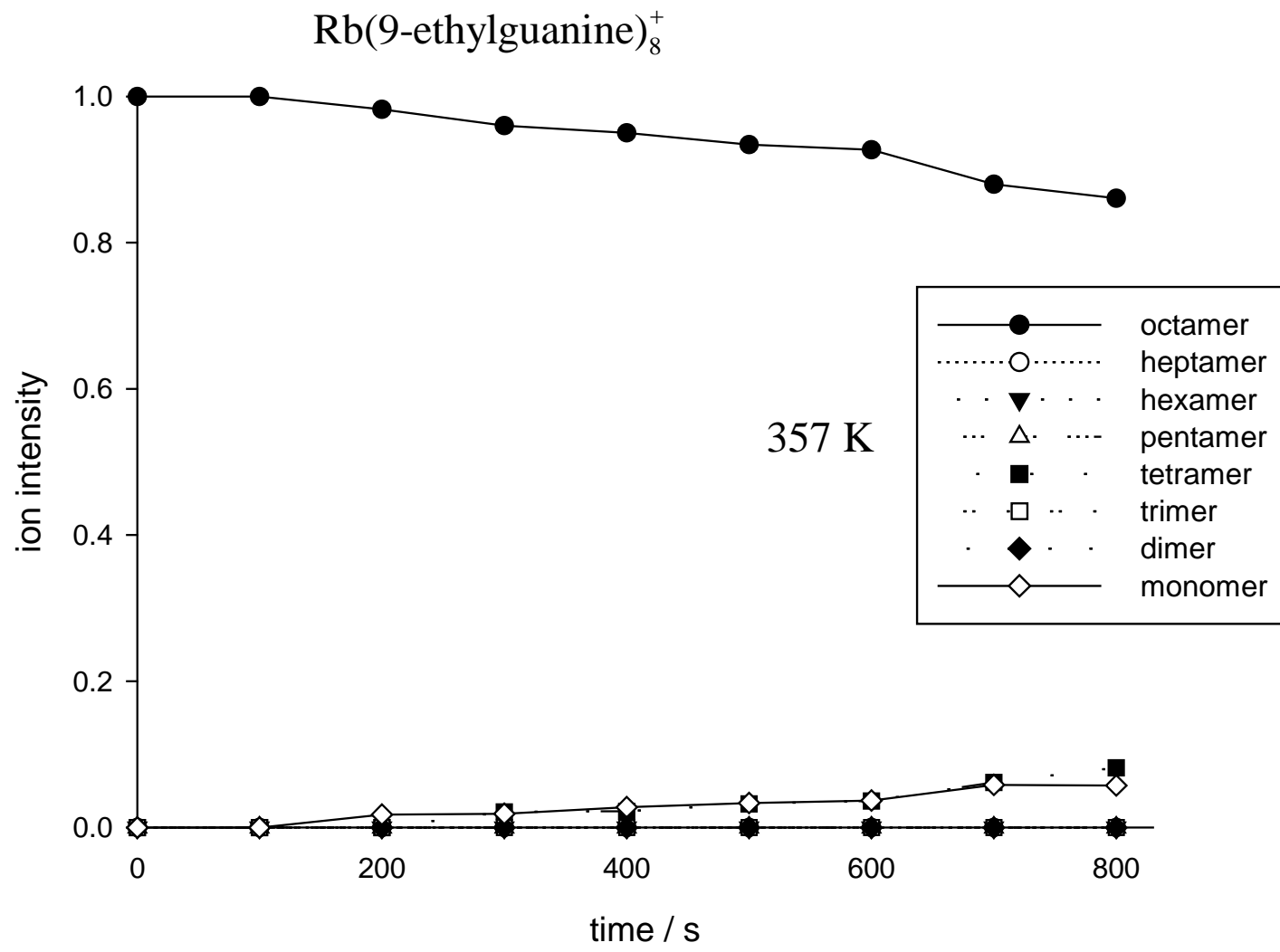


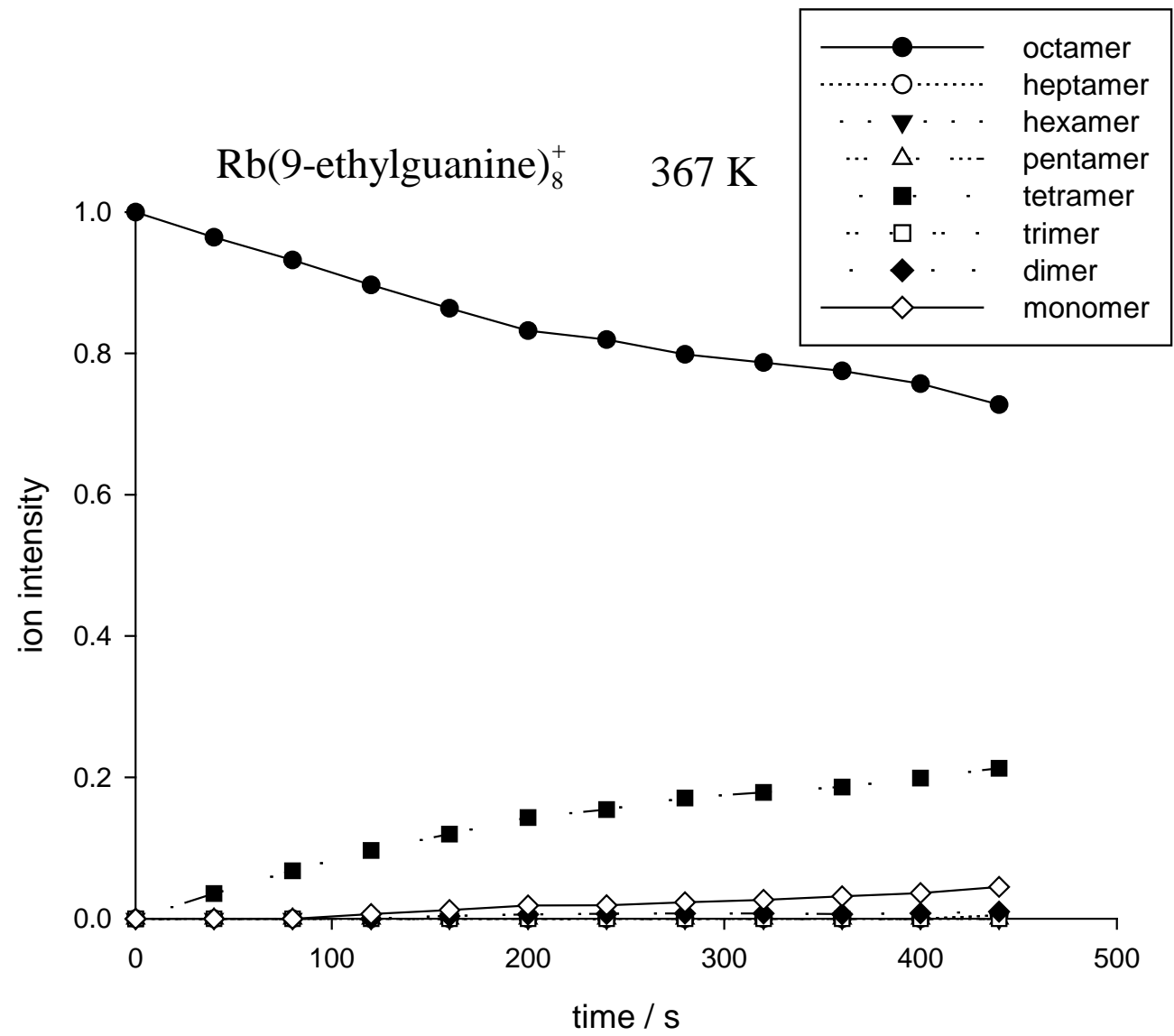


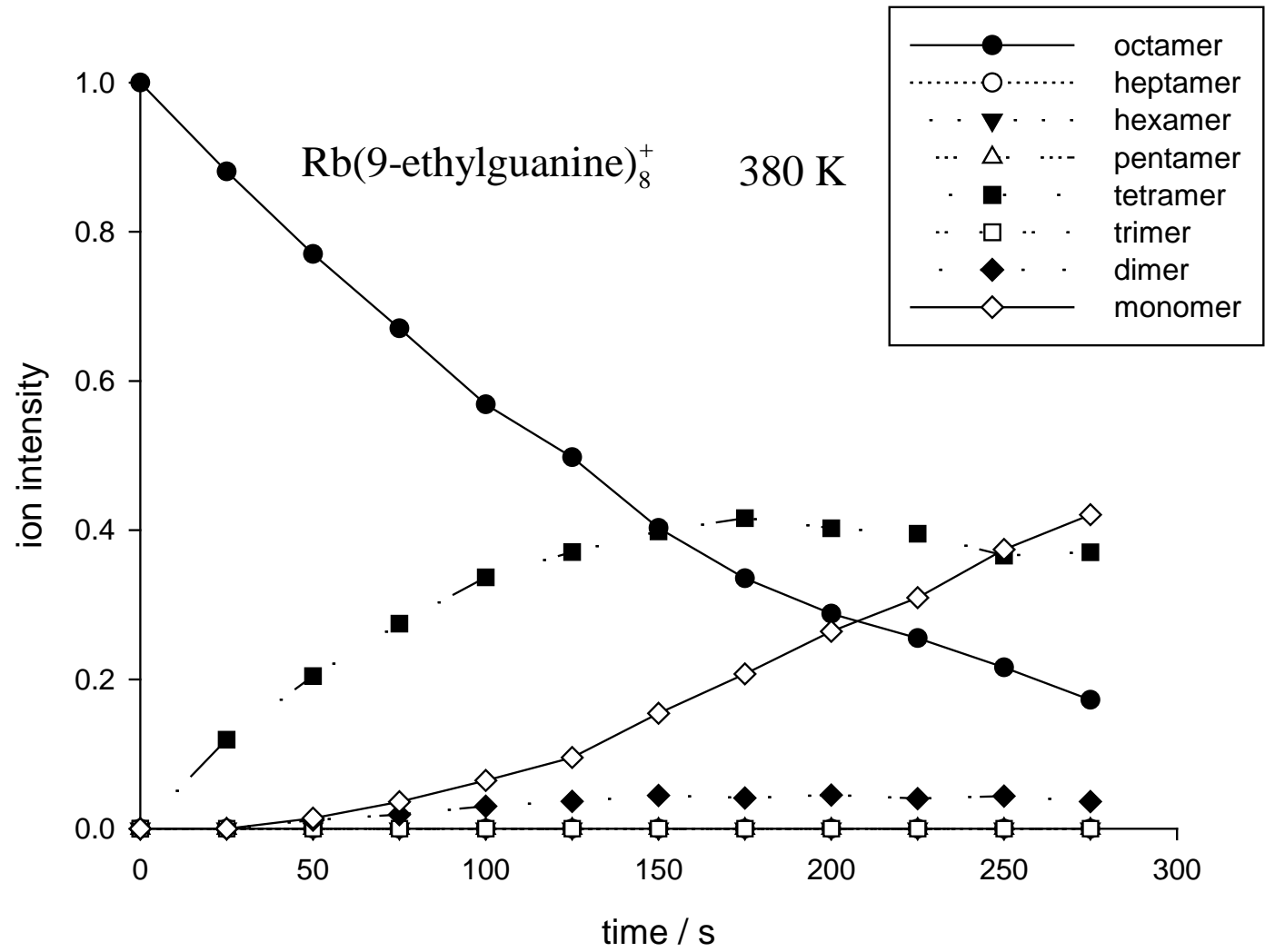


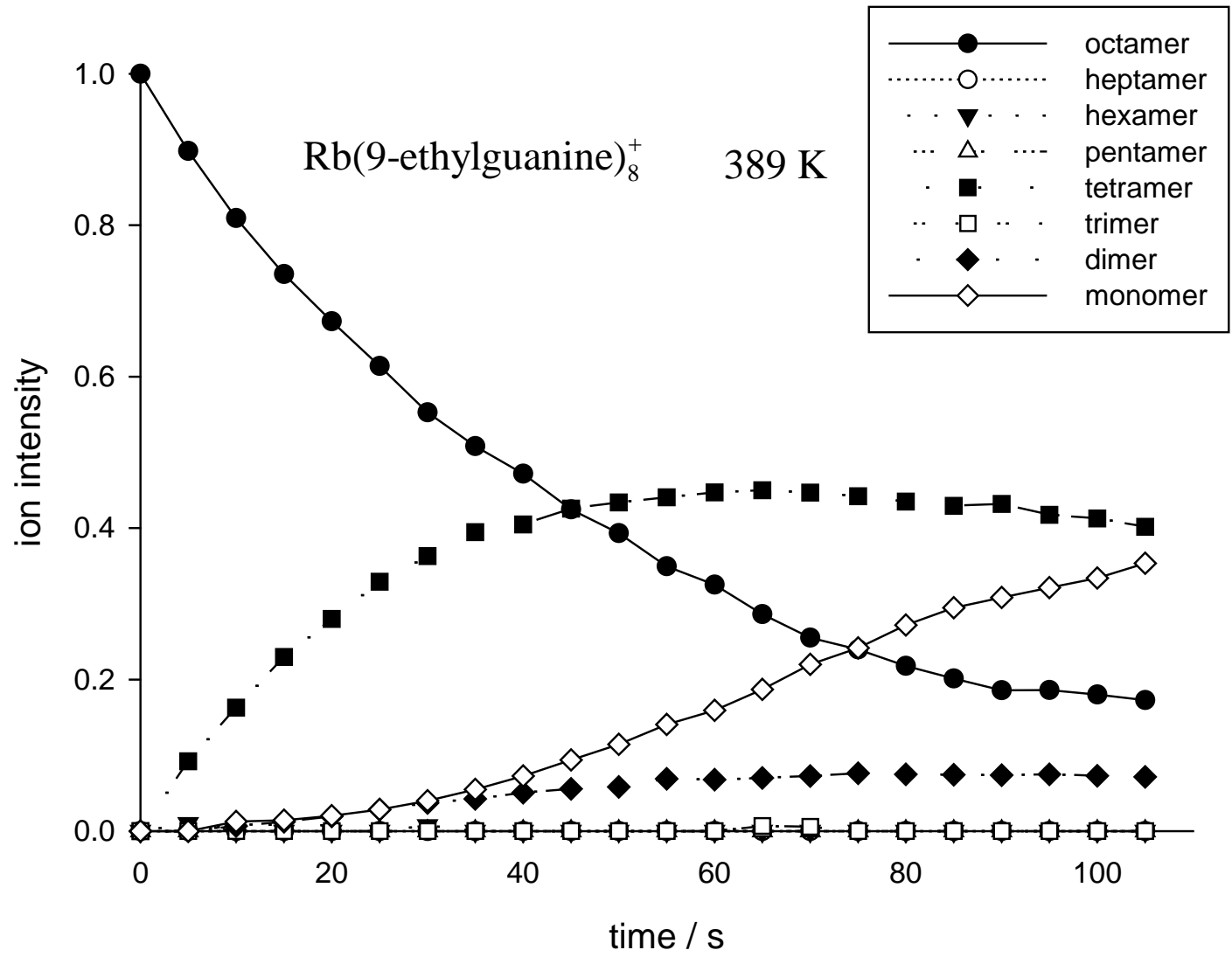


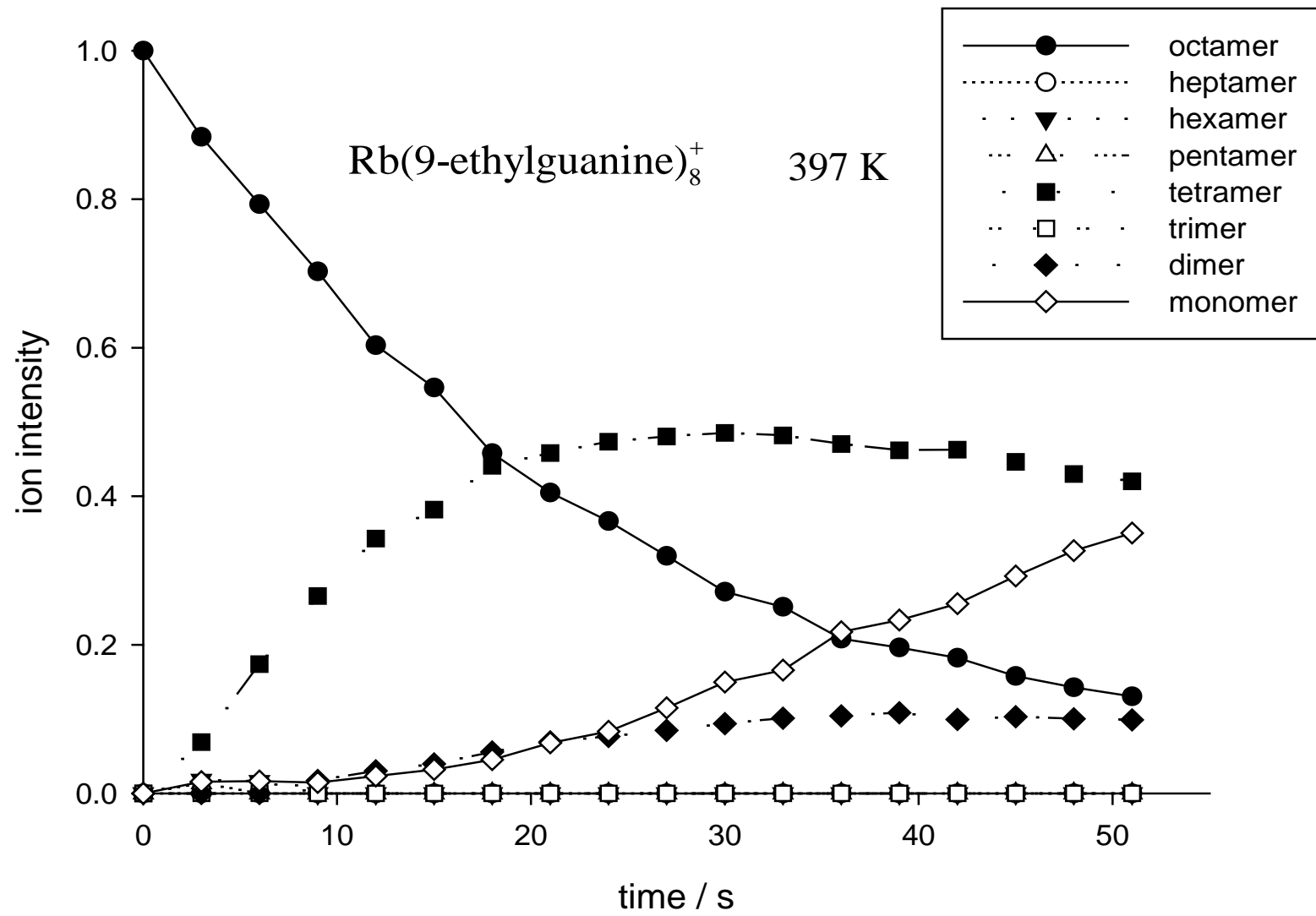


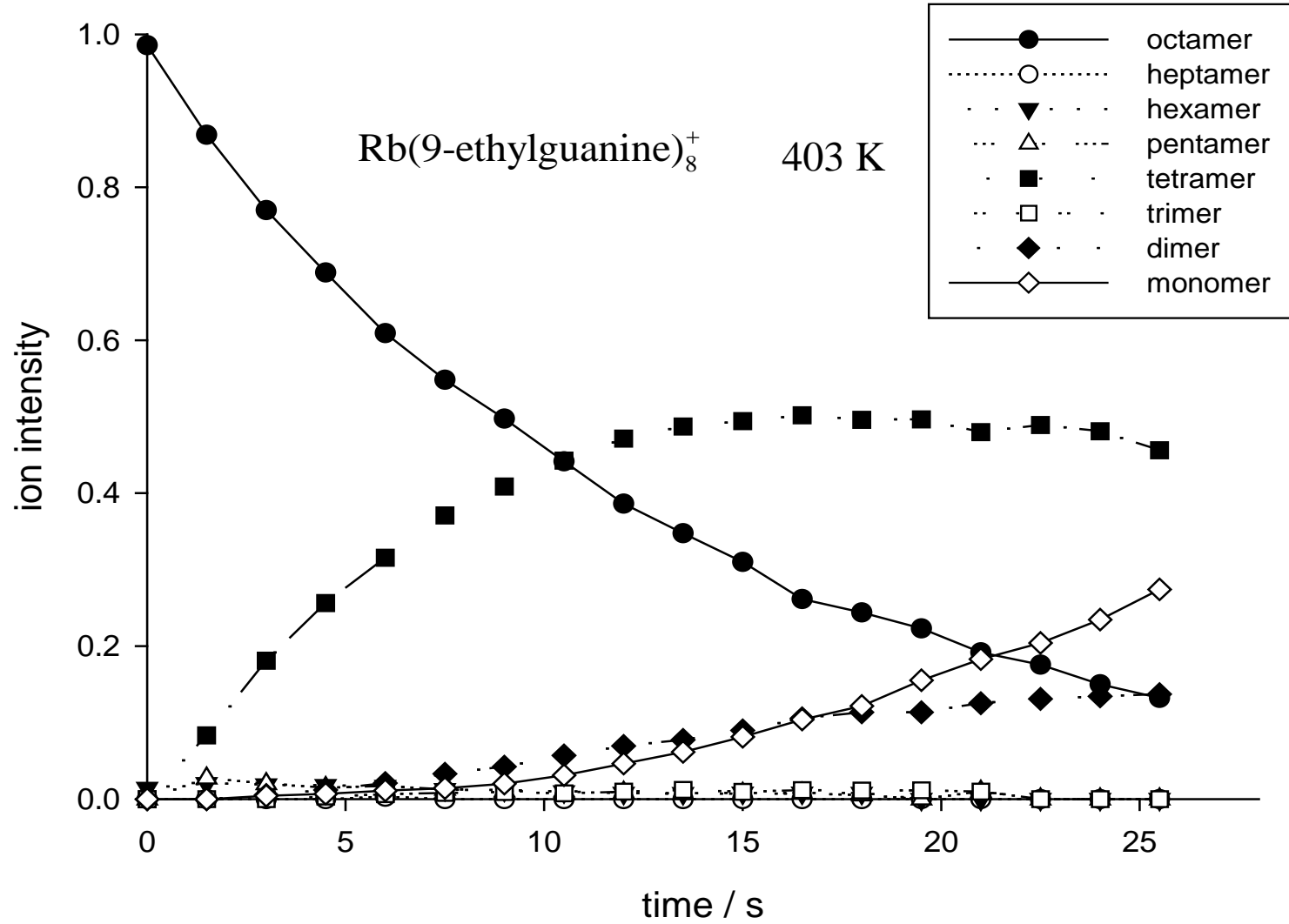


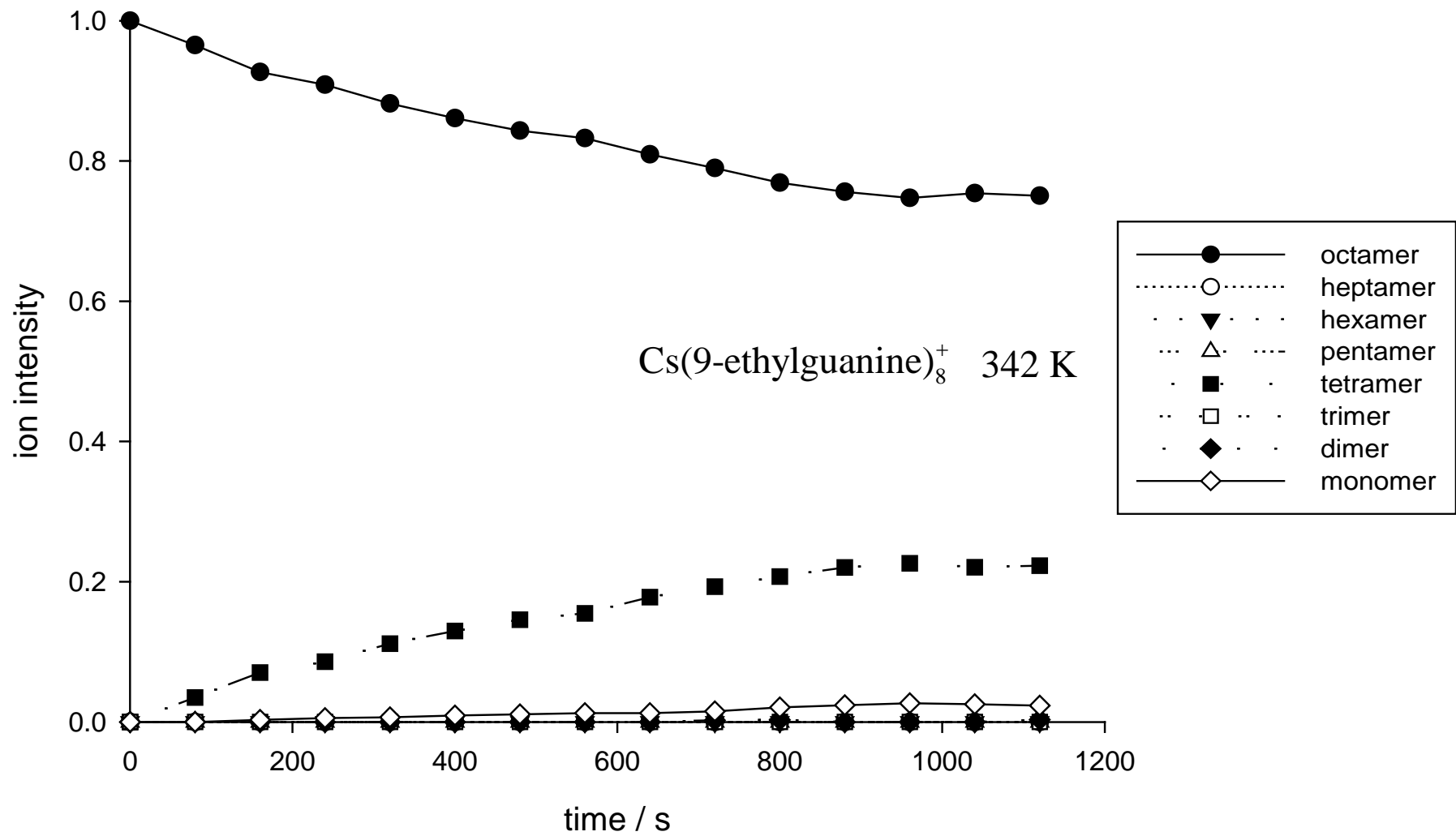


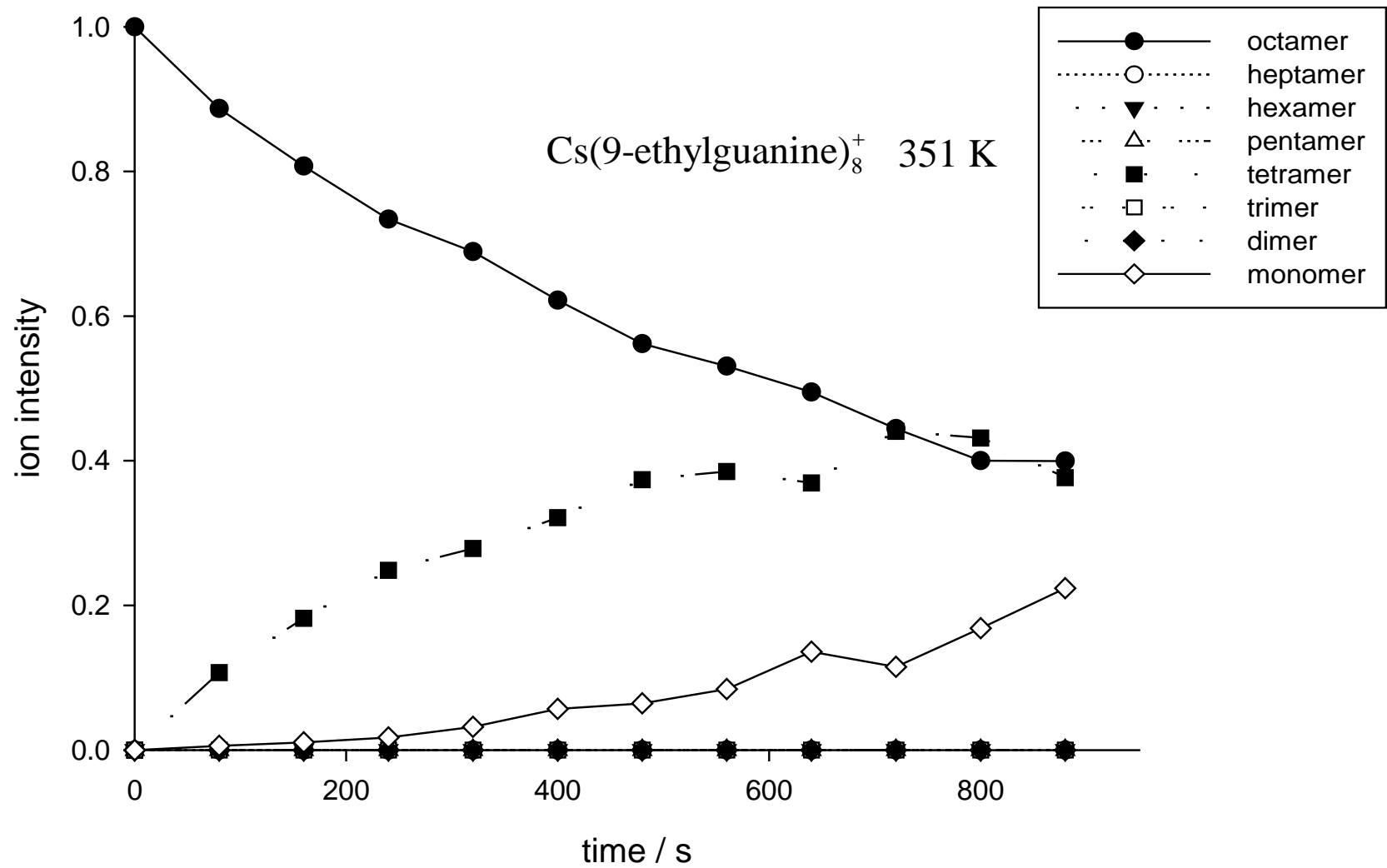


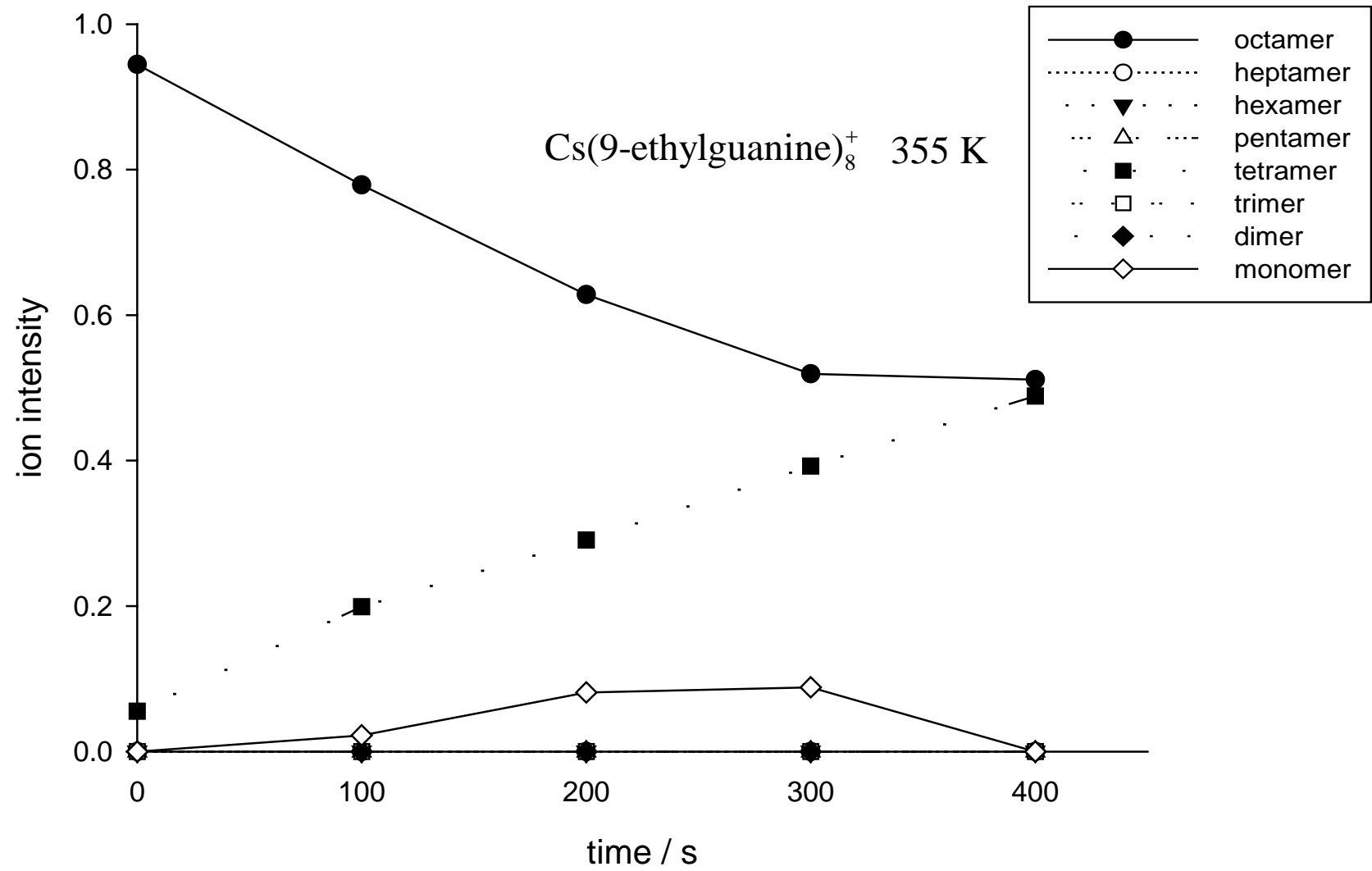


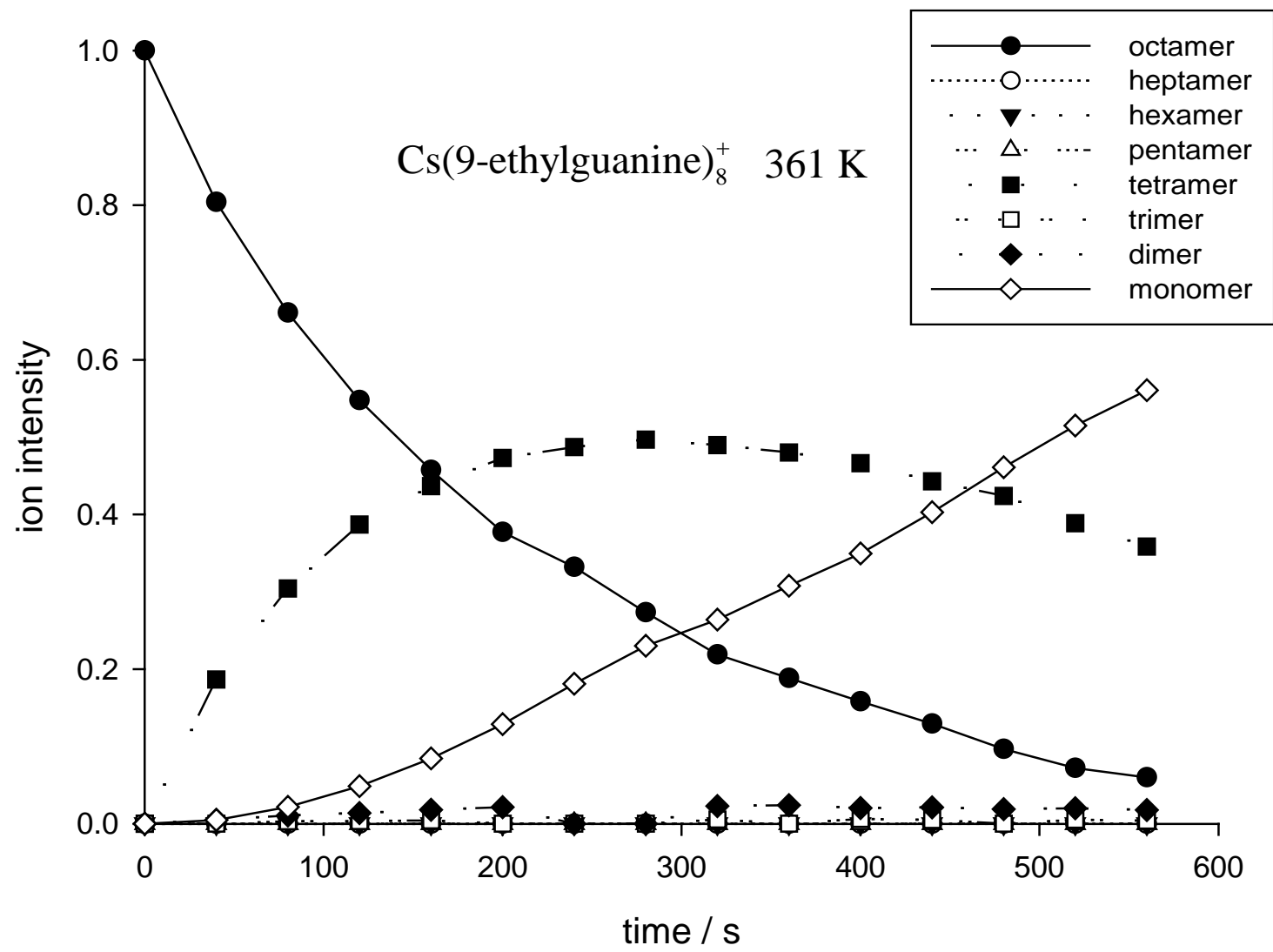


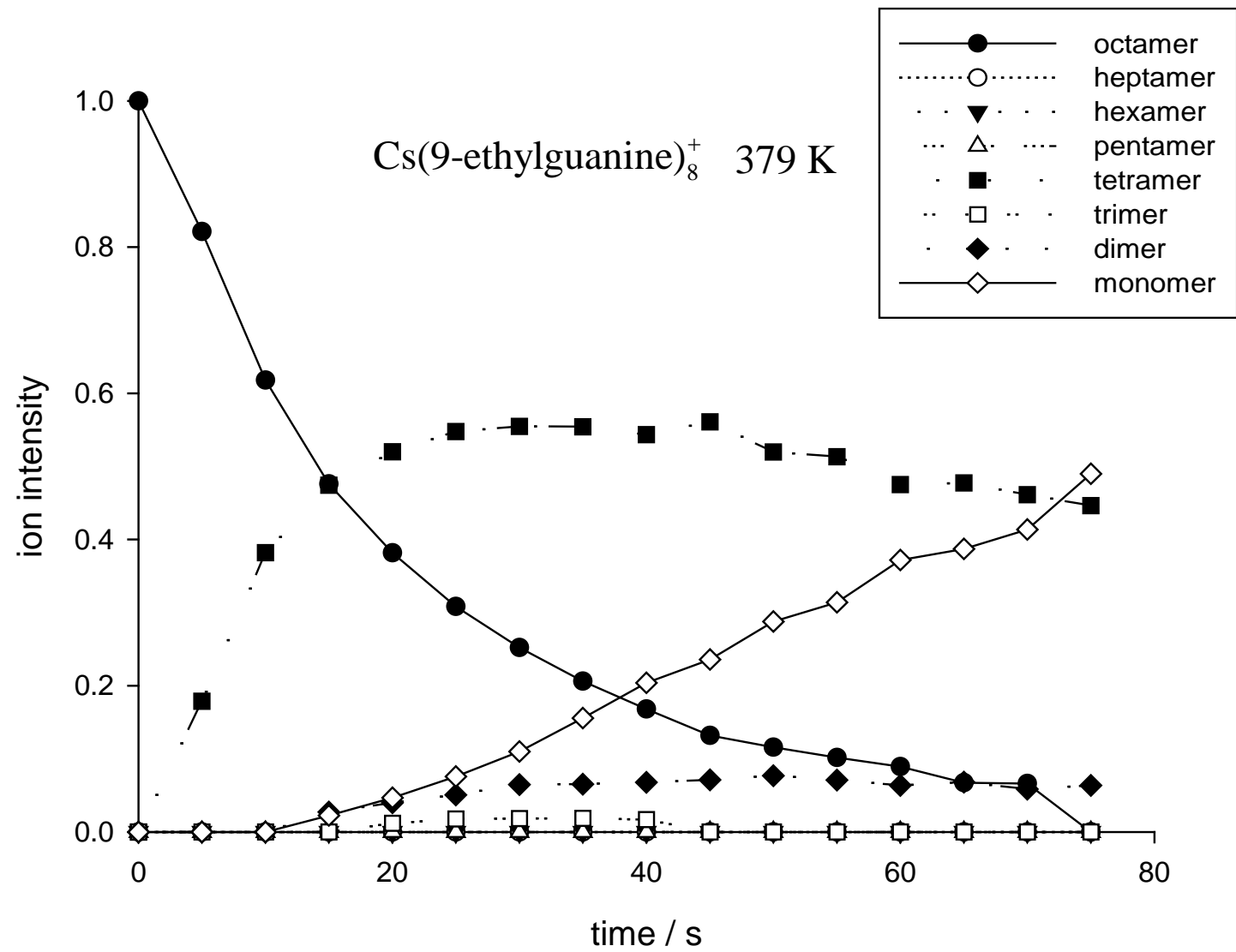


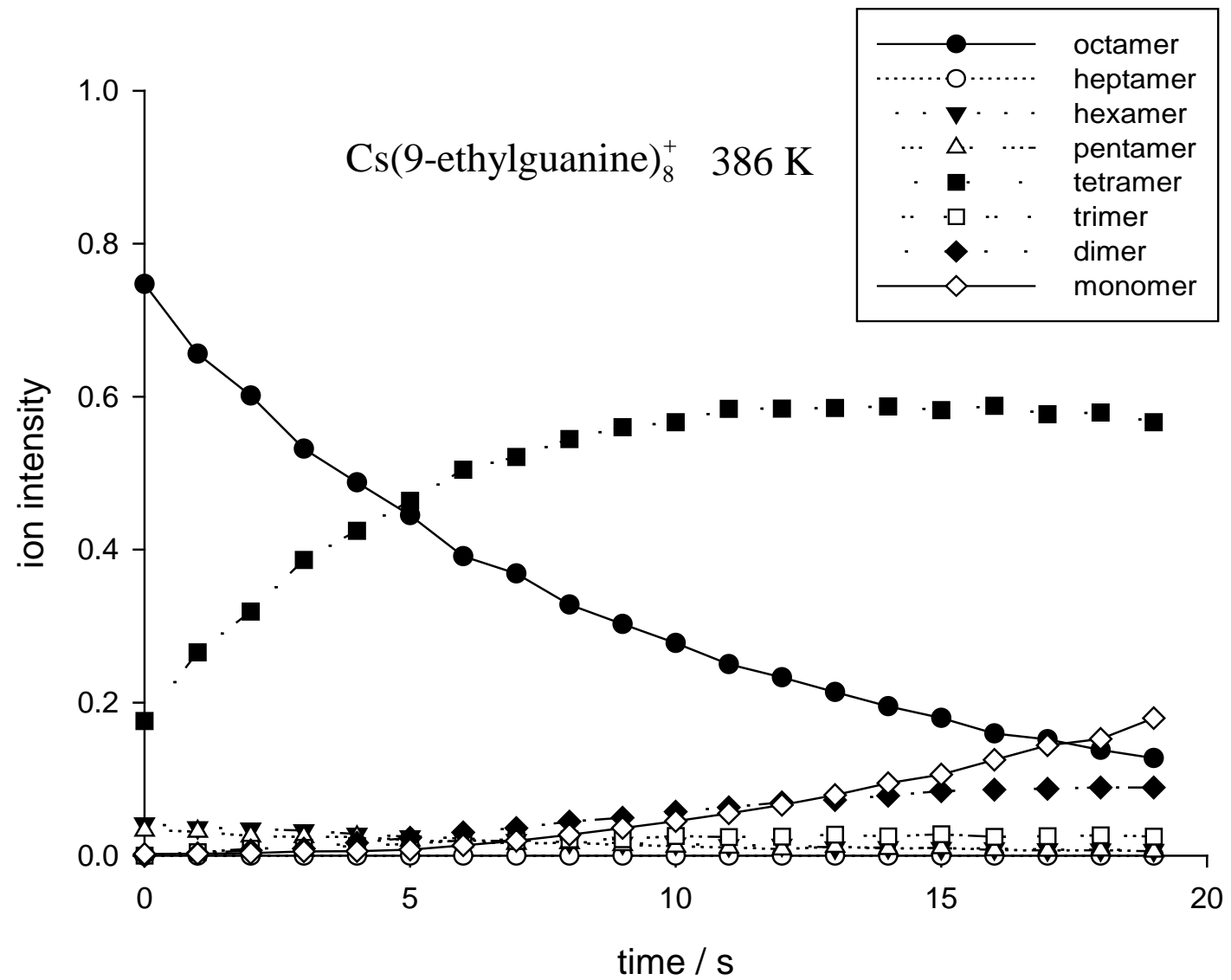


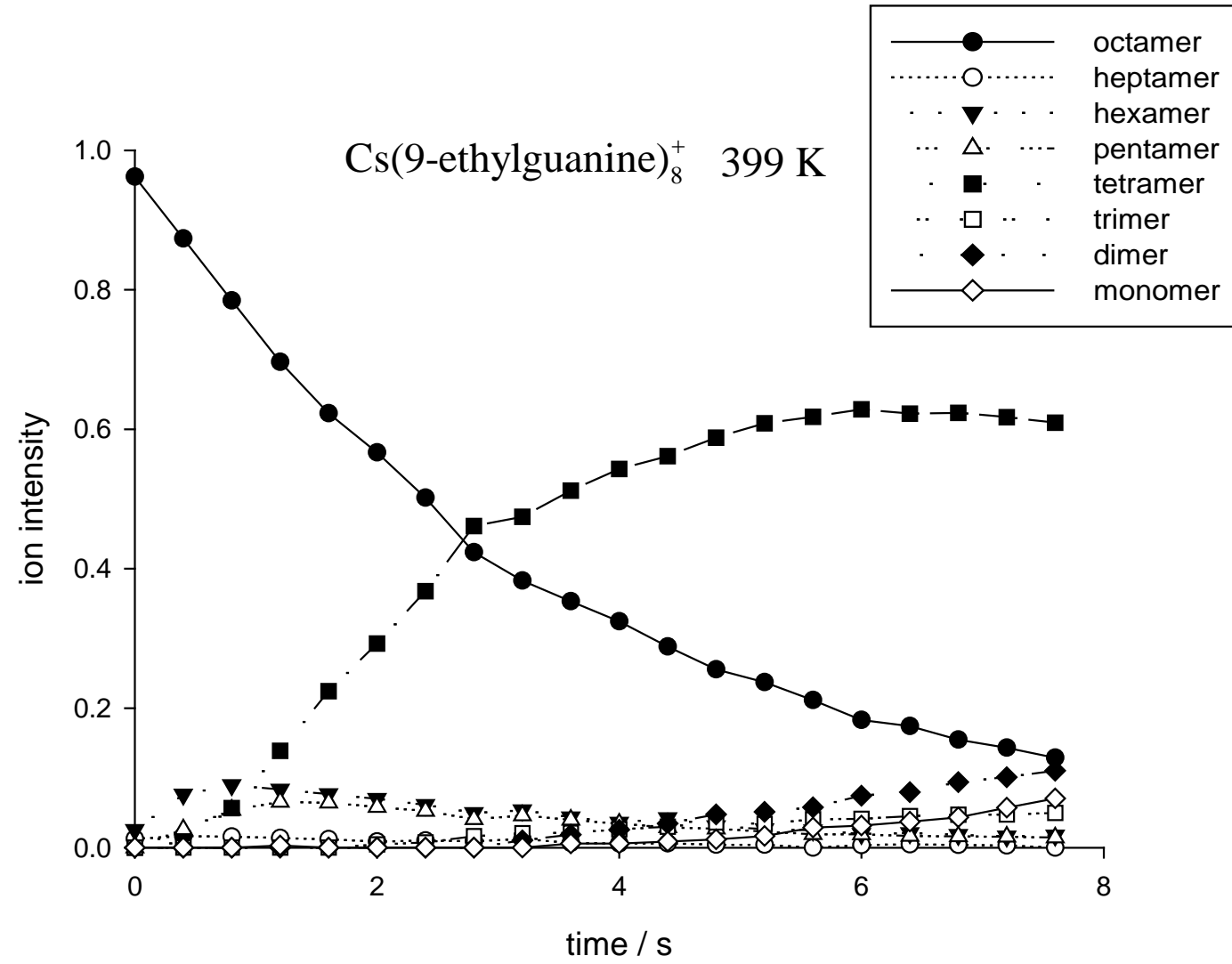




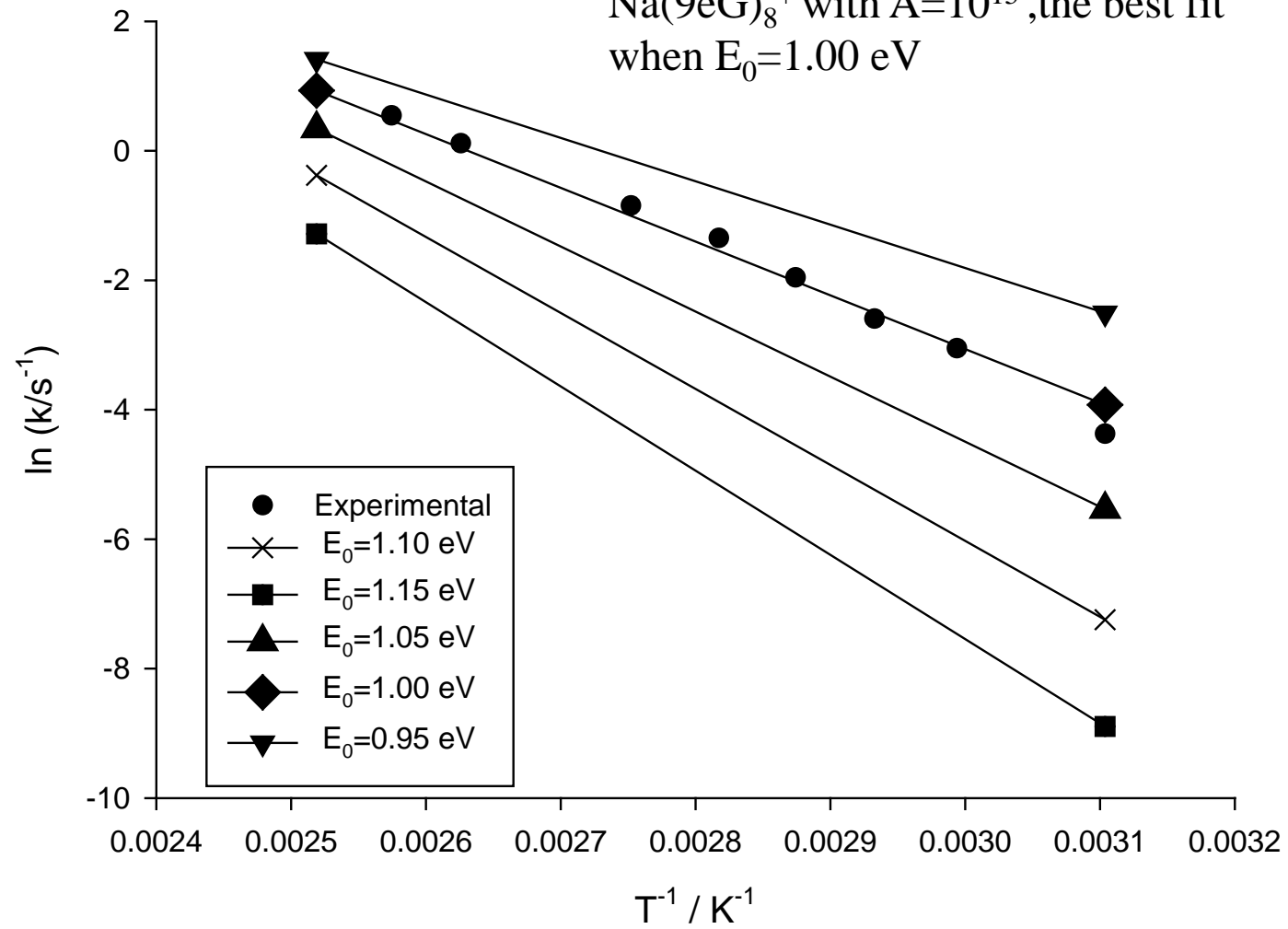




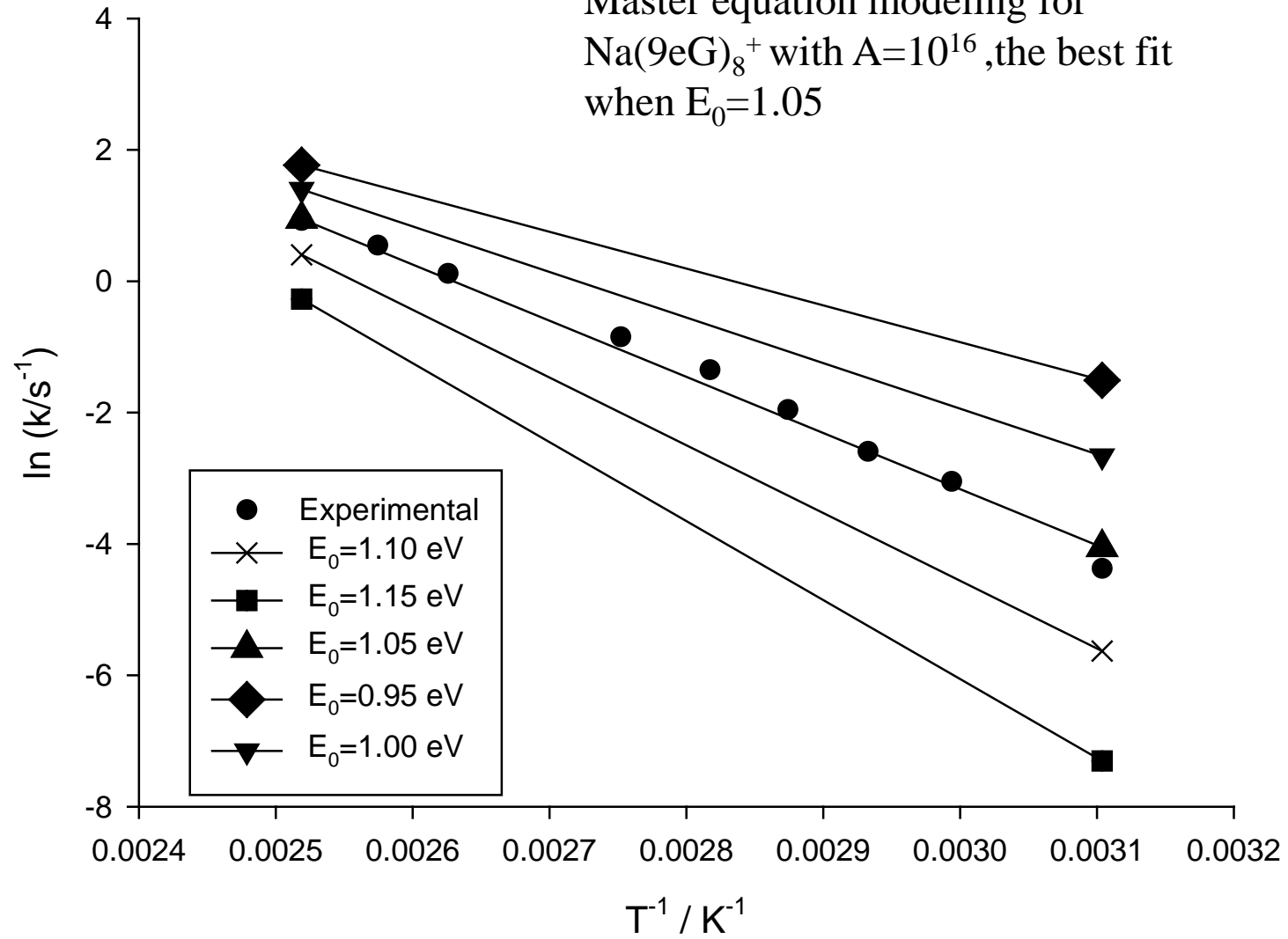




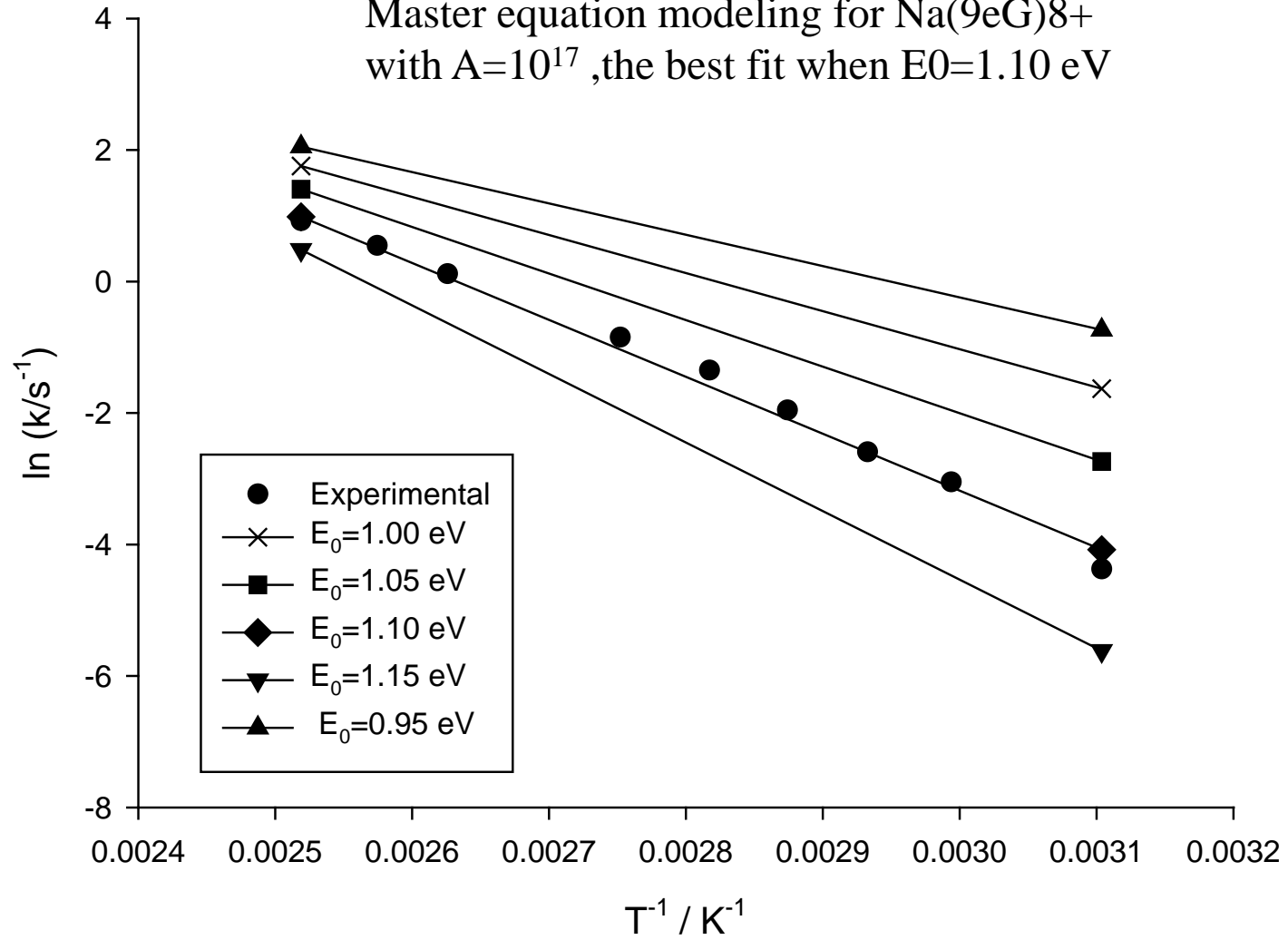
Master equation modeling for
 $\text{Na}(9eG)_8^+$ with $A=10^{15}$, the best fit
when $E_0=1.00$ eV



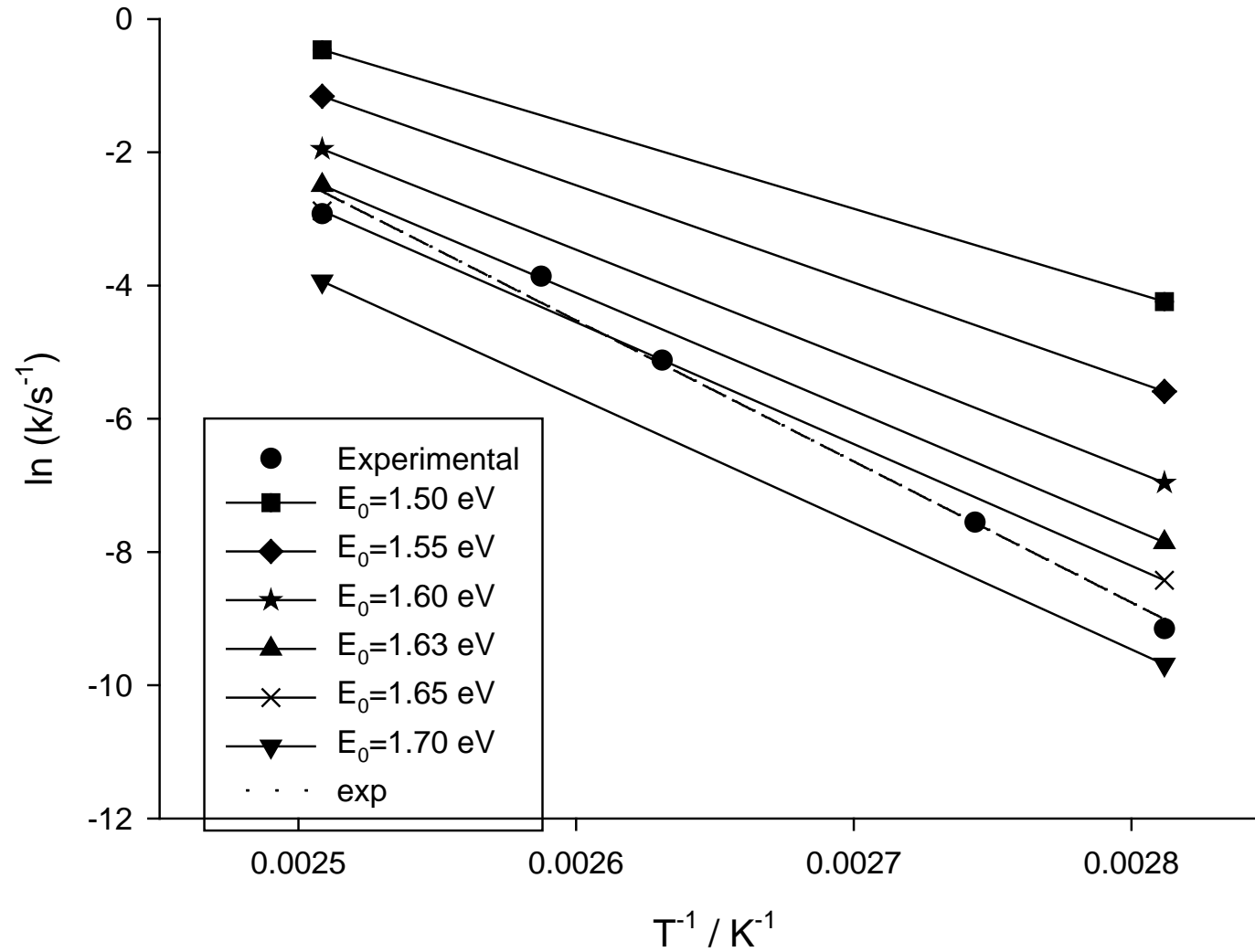
Master equation modeling for
 $\text{Na}(9\text{eG})_8^+$ with $A=10^{16}$, the best fit
when $E_0=1.05$



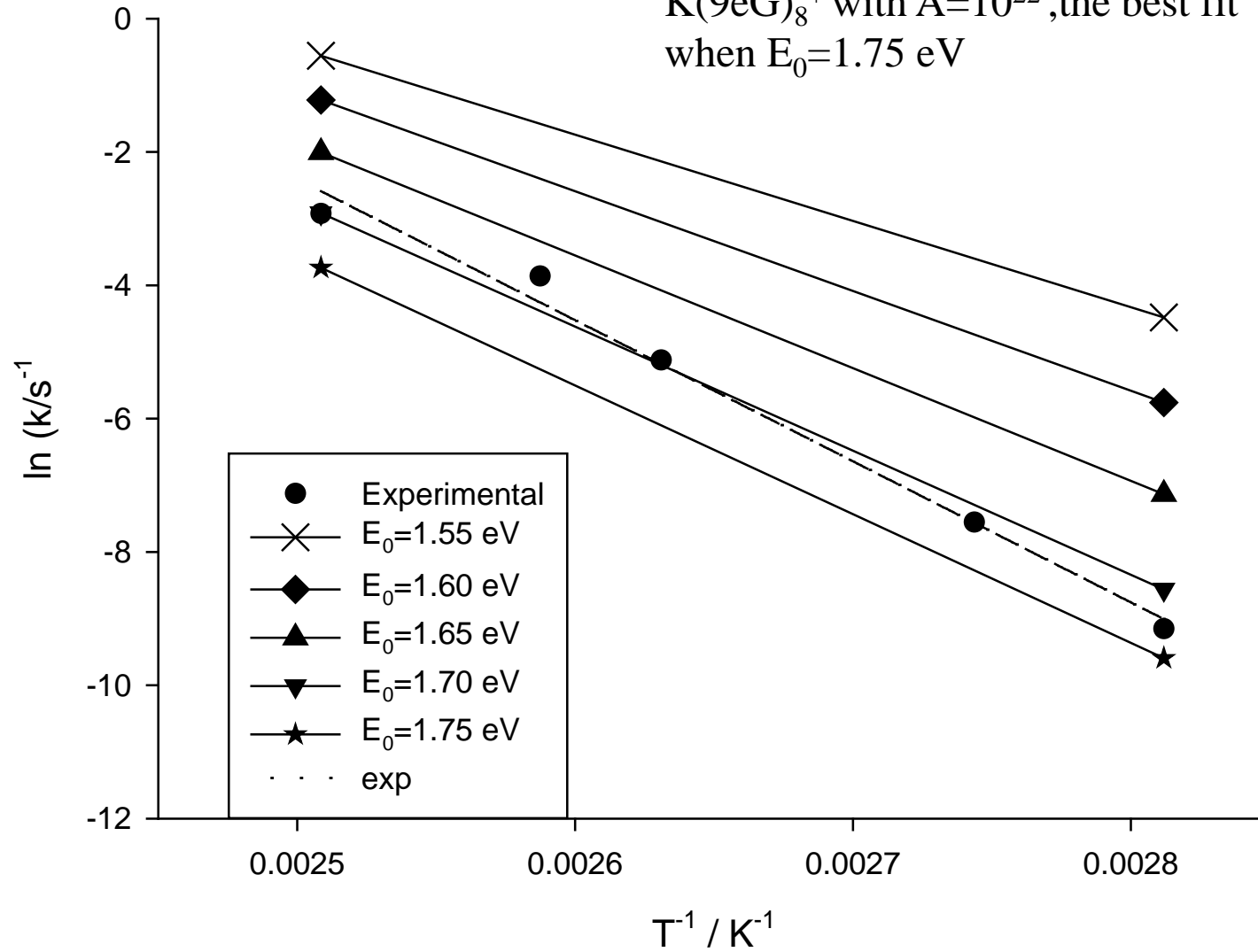
Master equation modeling for Na(9eG)8+
with $A=10^{17}$, the best fit when $E_0=1.10$ eV



Master equation modeling for $\text{K}(9\text{eG})_8^+$ with
 $A=10^{21}$, the best fit when $E_0=1.70$ eV



Master equation modeling for
 $\text{K}(9\text{eG})_8^+$ with $A=10^{22}$, the best fit
when $E_0=1.75$ eV



Master equation modeling for
 $\text{K}(9\text{eG})_8^+$ with $A=10^{23}$, the best fit
when $E_0=1.79$ eV

