

Supporting for “Elucidation of Charge-Transfer SERS Selection Rules by Considering the Excited State
Properties and the Role of Electrode Potential”

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Table S1. Details of excited state energies, oscillator strength and type of transitions.* refers to selected CT state.

BP86	State number	Energy (eV)	Oscillator strength	Type of transition	$\Delta q_{s_0 \rightarrow s_n}^{cluster}$
Py-Ag ₆	1	1.64	0.0007	CT	-0.76
	2	1.80	0.0020	CT*	-0.74
	3	2.13	0.0000	CT	-0.77
	4	2.31	0.0000	CT	-0.80
	5	2.34	0.0119	IC	+0.03
	6	2.47	0.0006	IC	+0.02
	7	2.77	0.0006	IC	+0.00
	8	3.03	0.0104	IC+CT	-0.31
	9	3.10	0.0188	CT	-0.02
	10	3.18	0.0000	IC	
	11	3.19	0.0123	CT	
	12	3.26	0.1174	CT	
	13	3.28	0.0011	CT	
	14	3.32	0.6527	CT+IC	
	15	3.34	0.0000	IC	
	16	3.40	0.0039	CT	
	17	3.42	0.0000	CT	
	18	3.44	0.0000	CT	
	19	3.45	0.8544	IC+CT	
	20	3.50	0.0000	CT	
Py-Ag ₂₀	1	1.65	0.0002	CT	
	2	1.67	0.0035	CT	
	3	1.69	0.0010	CT+IC	
	4	1.71	0.0046	CT+IC	
	5	1.73	0.0001	CT+IC	
	6	1.74	0.0002	CT+IC	
	7	1.75	0.0002	CT+IC	
	8	1.76	0.0004	IC	
	9	1.80	0.0001	CT	
	10	1.82	0.0003	CT	
	11	1.84	0.0053	CT+IC	
	12	1.87	0.0018	CT+IC	
	13	1.90	0.0018	CT+IC	
	14	1.91	0.0001	CT+IC	
	15	1.92	0.0046	IC	
	16	1.92	0.0047	CT+IC	
	17	1.96	0.0115	IC	
	18	1.99	0.0000	IC	
	19	2.03	0.0010	CT+IC	
	20	2.06	0.0002	IC	
BHandHLYP	State number	Energy (eV)	Oscillator strength	Type of transition	
Py-Ag ₆	1	2.67	0.0037	IC	

2	2.68	0.0396	IC
3	3.04	0.0046	IC
4	3.15	0.2211	IC
5	3.17	0.0000	IC
6	3.30	0.0594	CT*
7	3.33	0.0000	IC
8	3.54	1.4287	IC+CT
9	3.72	1.6802	IC+CT
10	3.85	0.0000	CT

Table S2. Magnitude of excited state gradients of $S_0 \rightarrow S_{CT1}$, $S_0 \rightarrow S_{IC}$ and $S_0 \rightarrow S_{IC} + CT$ transitions of Py-Ag₆ and $S_0 \rightarrow S_{CT1}$ transition of Py-Ag₂₀.

Py-Ag ₆	Py-Ag ₂₀
$S_0 \rightarrow S_{CT1}$ 1 Ag : 0.000000010 -0.001038126 0.005372618 2 Ag : -0.000000003 0.008288789 -0.006575878 3 Ag : -0.000000009 0.000000104 0.003019725 4 Ag : 0.000000034 -0.000000054 -0.021139899 5 Ag : 0.000000002 0.001038128 0.005372591 6 Ag : -0.000000018 -0.008288773 -0.006575880 7 C : -0.034546041 0.000000019 -0.033934891 8 C : -0.015919198 0.000000188 0.032500184 9 C : -0.000000041 0.000000011 -0.039160754 10 C : 0.015919208 -0.000000065 0.032500194 11 C : 0.034546028 0.000000098 -0.033934927 12 N : 0.000000018 -0.000000033 0.064175446 13 H : -0.000000038 0.000000002 0.000644164 14 H : -0.001708642 -0.000000001 -0.002240211 15 H : 0.001119383 0.000000027 0.001140038 16 H : -0.001119385 0.000000028 0.001140054 17 H : 0.001708656 0.000000001 -0.002240199	$S_0 \rightarrow S_{CT1}$ 1 Ag : 0.001888819 -0.000504461 -0.000000010 2 Ag : 0.006951272 0.010426942 -0.000000067 3 Ag : 0.005906932 -0.005014766 0.013536162 4 Ag : 0.005908978 -0.005022673 -0.013532376 5 Ag : 0.003238070 -0.005920361 -0.000000388 6 Ag : 0.001948532 0.002177444 0.007186505 7 Ag : 0.001947725 0.002180026 -0.007181835 8 Ag : 0.007580336 0.012689477 0.000001512 9 Ag : 0.005941622 -0.007204620 0.008587170 10 Ag : 0.009937878 -0.007201212 -0.008589827 11 Ag : -0.007154148 0.001078740 0.000000174 12 Ag : -0.007761119 0.004511116 -0.012906218 13 Ag : -0.007760492 0.004511533 0.012906214 14 Ag : -0.009565492 0.007148398 -0.008343259 15 Ag : -0.009567217 0.007152345 0.008334840 16 Ag : -0.007452054 -0.010263383 -0.000888474 17 Ag : -0.007452758 -0.010258389 0.000888385 18 Ag : -0.004384693 0.012156911 0.000001160 19 Ag : -0.005065807 -0.006422134 -0.013055357 20 Ag : -0.005066300 -0.006423162 0.013055603 21 H : 0.027134271 0.000482193 -0.000001162 22 C : -0.014477785 -0.000040009 0.000000784 23 C : 0.012388464 0.000001599 0.005036591 24 C : 0.012389450 0.000000863 -0.005038053 25 C : -0.011472125 -0.000063279 0.011434754 26 C : -0.011472020 -0.000061953 -0.011433241 27 H : -0.000633362 0.000020029 0.000150018 28 H : -0.000633304 0.000019797 -0.000150047 29 H : 0.000744968 -0.000014702 -0.000429934 30 H : 0.000747017 -0.000014819 0.000429951 31 H : 0.001381648 0.000007424 0.000000059
$S_0 \rightarrow S_{CT2}$ Ag : 0.000000013 -0.001022314 0.005277610 Ag : -0.000000001 0.008371974 -0.006422384 Ag : -0.000000001 0.000000092 0.003098701 Ag : 0.000000025 -0.000000058 -0.018997390 Ag : 0.000000005 0.001022303 0.005277551 Ag : -0.000000010 -0.008371967 -0.006422403 C : -0.010930489 0.000000298 0.051331204 C : -0.005837519 -0.000000007 -0.050630729 C : -0.000000207 -0.000000064 0.008851400 C : 0.005837526 -0.000000016 -0.050630715 C : 0.010930478 0.000000231 0.051331149 N : 0.000000047 -0.000000192 0.011337616 H : -0.000000012 -0.000000021 -0.001329347 H : 0.000342615 -0.000000004 -0.000964243 H : -0.000874033 0.000000009 -0.000045279 H : 0.000874023 -0.000000031 -0.000045294 H : -0.000342618 -0.000000009 -0.000964212	
$S_0 \rightarrow S_{IC}$ 1 Ag : 0.000000000 -0.007572917 -0.004272147 2 Ag : 0.000000004 -0.006750293 0.008114323 3 Ag : 0.000000008 0.000001157 -0.003315632 4 Ag : 0.000000009 0.000000370 -0.007825433 5 Ag : -0.000000009 0.007570768 -0.004271677 6 Ag : -0.000000010 0.006751067 0.008113702 7 C : -0.001695980 0.000000093 0.000189622 8 C : 0.000349256 -0.000000144 0.000358498 9 C : -0.000000199 0.000000141 -0.000162448 10 C : -0.000349235 -0.000000099 0.000358579 11 C : 0.001695961 0.000000096 0.000189596 12 N : 0.000000073 -0.000000103 0.002081991 13 H : -0.000000013 -0.000000020 -0.000290067 14 H : -0.000760972 0.000000022 0.000126468 15 H : 0.000247383 -0.000000005 0.000034165 16 H : -0.000247378 -0.000000015 0.000034161 17 H : 0.000760961 0.000000010 0.000126499	
$S_0 \rightarrow S_{IC} + CT$ 1 Ag : 0.000000014 0.001560966 0.004855092 2 Ag : -0.000000014 0.005294852 -0.004508054 3 Ag : 0.000000007 0.000000008 0.009461667 4 Ag : 0.000000007 -0.000000012 -0.002873967 5 Ag : 0.000000030 -0.001560953 0.004855110 6 Ag : -0.000000009 -0.005294841 -0.004508024 7 C : 0.001907863 -0.000000033 -0.002082565 8 C : -0.000300536 0.000000187 0.001232167 9 C : -0.000000063 -0.000000099 -0.001239845 10 C : 0.000300538 -0.000000016 0.001232216 11 C : -0.001907851 0.000000003 -0.002082652 12 N : 0.000000009 -0.000000068 -0.006265414 13 H : -0.000000028 0.000000002 -0.000671750 14 H : -0.001458959 -0.000000002 0.001759391 15 H : 0.000725990 0.000000014 -0.000649053 16 H : -0.000725976 0.000000001 -0.000649056 17 H : 0.001458966 -0.000000007 0.001759380	

Table S3. Calculated dimensionless displacements of totally symmetric modes of Py adsorbed on silver electrode in different potentials.

		External electric field ^a							
		+0.0030	+0.0015	0.0000	-0.0015	-0.0030	-0.0045	-0.0060	-0.0090
BP86	v _{6a}	0.9766	0.9446	0.9168	0.8899	0.8652	0.8409	0.8176	0.7744
	v _{9a}	0.1606	0.8816	0.8753	0.8691	0.8636	0.8580	0.8527	0.8428
	v _{8a}	0.9315	0.9255	0.9210	0.9161	0.9118	0.9070	0.9022	0.8928
		External electric field ^a							
			0.0000	-0.0005	-0.0015	-0.0030	-0.0045	-0.0060	-0.0090
BHandHLYP	v _{6a}		0.9351	0.9311	0.9211	0.8931	0.8639	0.8303	0.7572
	v _{9a}		0.9019	0.9000	0.8930	0.8922	0.8808	0.8645	0.8238
	v _{8a}		0.8980	0.8984	0.8978	0.8909	0.8796	0.8634	0.8228
Mode		Ref. 1			Ref. 2				
		ln -0.5 V	ln -0.0 V		-0.25 V	-0.50 V	-0.75 V	-1.00 V	
v _{6a}		0.0860	0.0000		0.0764	0.1107	0.2569	0.1996	
v _{9a}		0.1405	0.1167		0.0940	0.1550	0.4725	0.3818	
v _{8a}		0.1837	0.1499		0.1754	-0.1777	0.3463	0.1956	

^aDirection and magnitude of applied external electric field in atomic unit.

Table S4. Bond length of Ag-N, and also C1-N and C1-C2 in pyridine before and after adsorption on the clusters and in the external electric field.

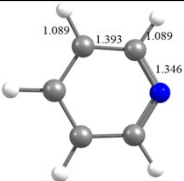
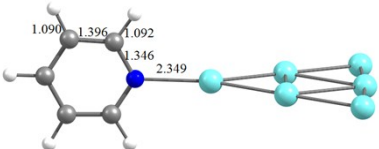
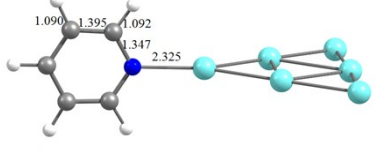
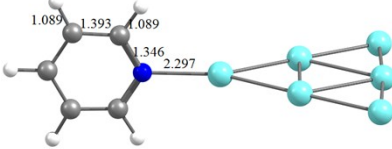
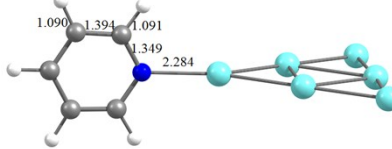
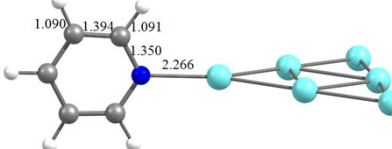
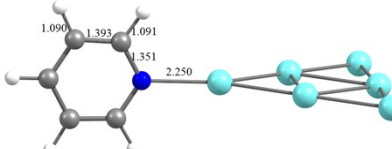
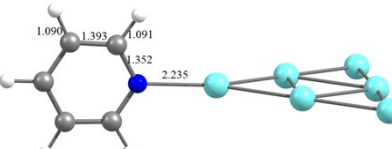
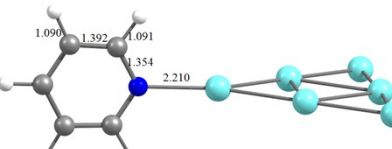
E field		
Py	0.0000	
	+0.0030	
	+0.0015	
Py-Ag ₆	0.0000	
	-0.0015	
	-0.0030	
	-0.0045	
	-0.0060	
	-0.0090	

Table S5. Vibrational energy distribution analysis for ν_{6a} , ν_{8a} and ν_{9a} modes of Py-Ag₆.

Field		VEDA
0.0000	ν_{6a}	11% BEND CCN, 27% BEND CCC, 39% BEND CNC
	ν_{9a}	24 STRE NC, 62% BEND HCC
	ν_{8a}	42% STRE CC, 10% STRE NC
-0.0015	ν_{6a}	11% BEND CCN, 26% BEND CCC, 39% BEND CNC
	ν_{9a}	24% STRE NC, 62% BEND HCC
	ν_{8a}	42% STRE CC, 10% STRE NC
-0.0090	ν_{6a}	11% BEND CCN, 24% BEND CCC, 38% BEND CNC
	ν_{9a}	21% STRE NC , 64% BEND HCC
	ν_{8a}	43% STRE CC, 10% STRE NC

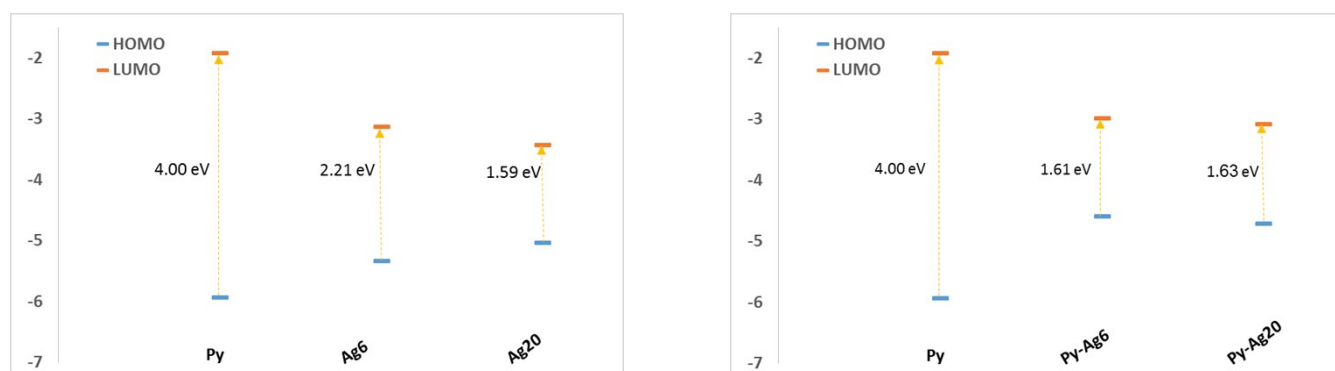


Fig. S1. Left panel: HOMO-LUMO gap of Py and Fermi level of pure clusters. It is apparent that Fermi level of selected cluster is between the HOMO-LUMO gap of Py. Numbers are the energy difference of Fermi level of metal cluster and LUMO energy of Py in eV. Right panel: HOMO-LUMO gap of Py and the combined systems.

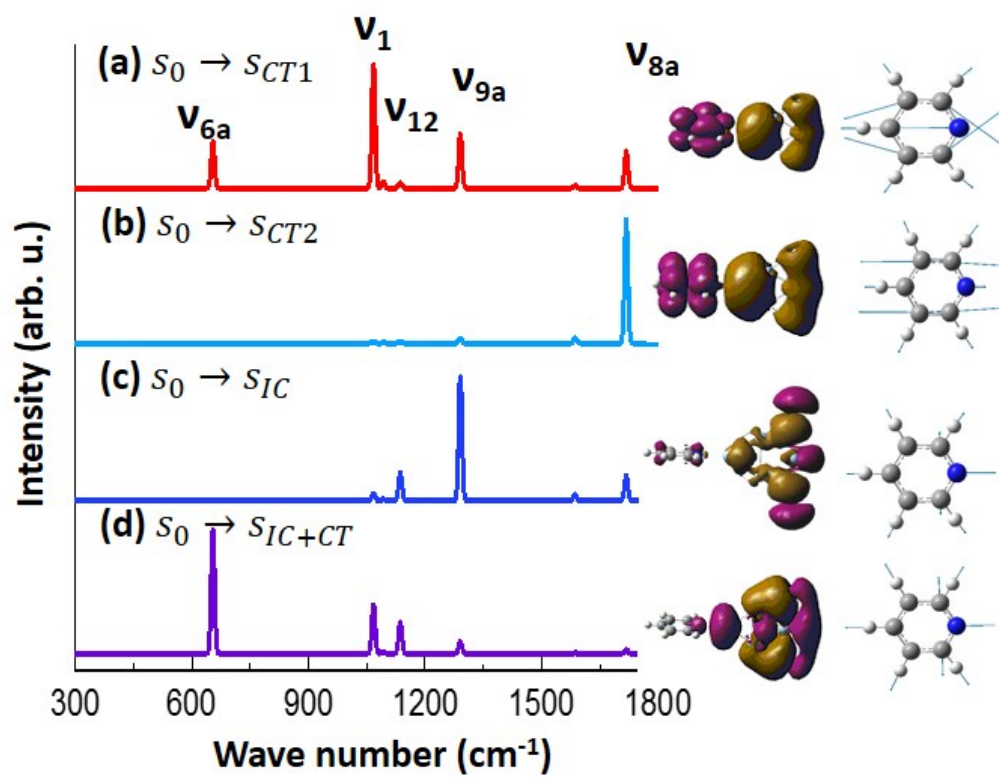


Fig. S2. SERS spectra for a) $S_0 \rightarrow S_{CT1}$, b) $S_0 \rightarrow S_{CT2}$, c) $S_0 \rightarrow S_{IC}$ and d) $S_0 \rightarrow S_{IC+CT}$ transitions of Py-Ag₆, by BHandHLYP level of theory.

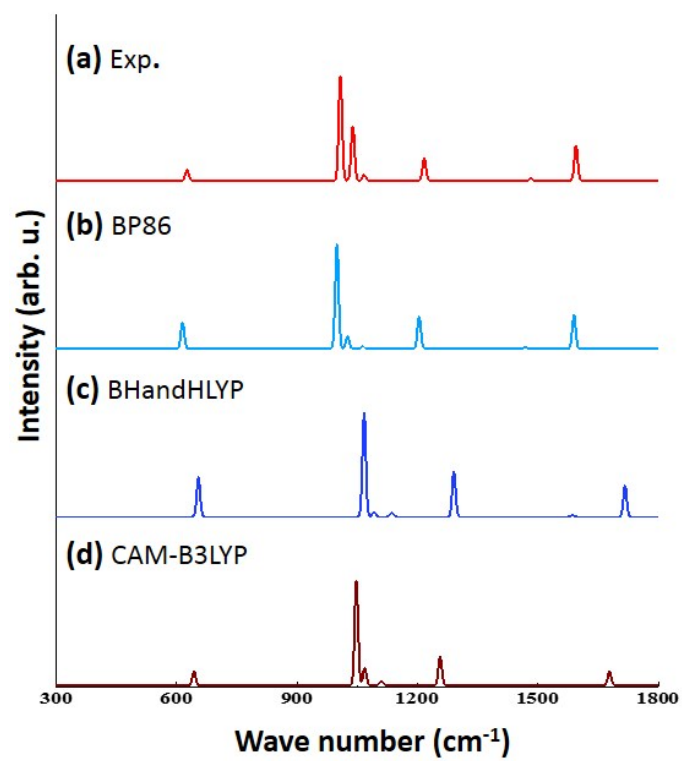


Fig. S3 SERS spectra for $S_0 \rightarrow S_{CT1}$ transition of Py-Ag₆ by BP86, BHandHLYP and CAM-B3LYP functionals in comparison to the experiment.

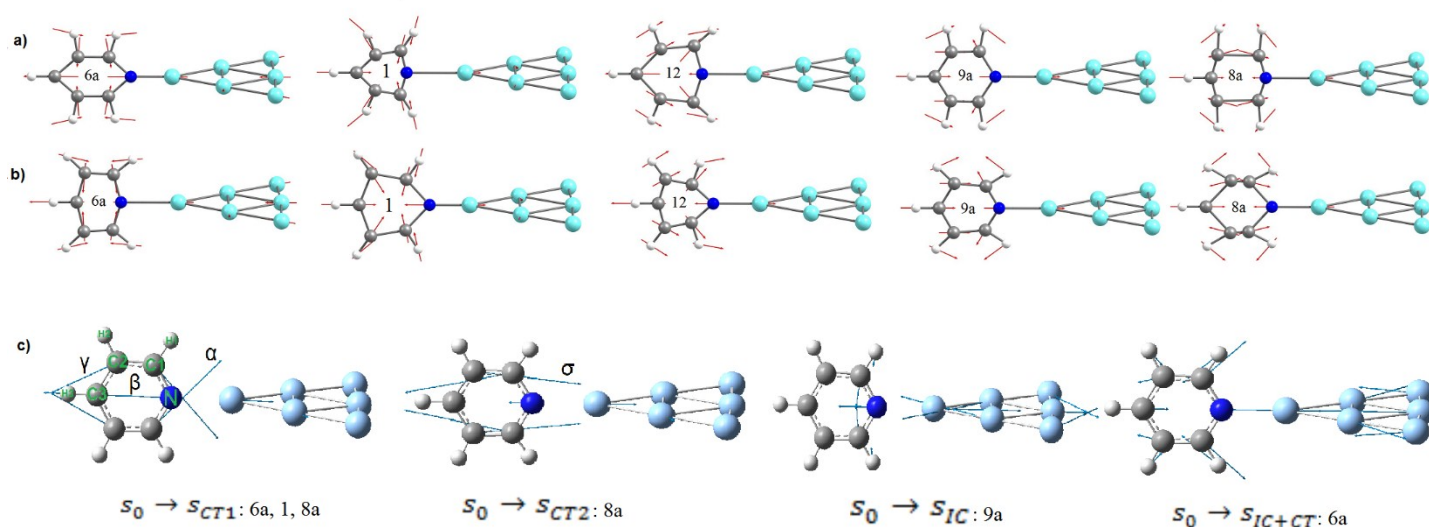


Fig. S4. Deformed structures of Py-Ag₆ upon totally symmetric normal modes in two compressed and extended frames displays in a and b part of the figure. c) Cartesian excited state gradient vectors of selected states of Py-Ag₆. Contribution of each vibrational normal mode in the gradient vectors have been shown under the gradient vectors.

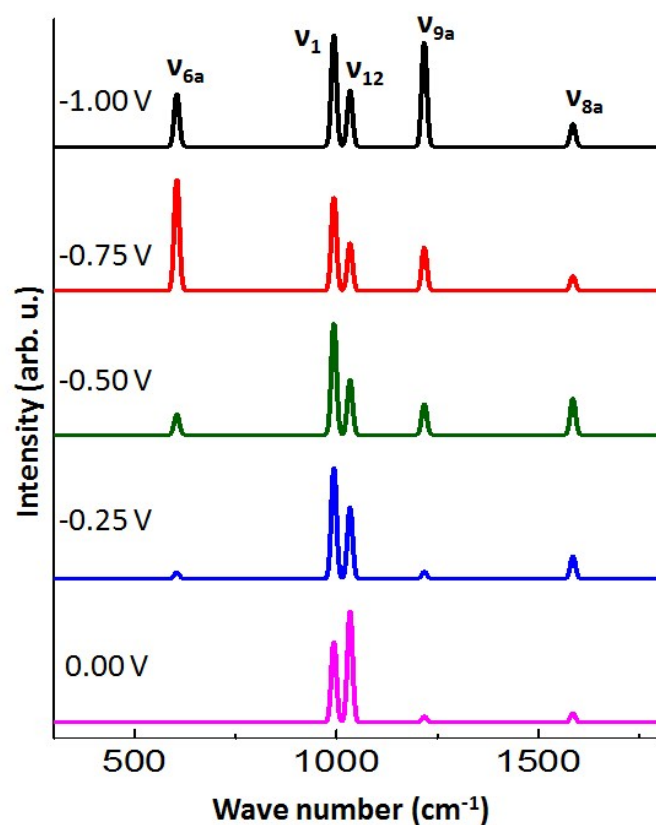


Fig. S5 Fitting of the model parameters of eq. 3 based on experimental data from Ref. 2 was has been performed. In general, the experimental dimensionless displacements are determined along with finding a set of parameters i.e. homogeneous broadening, excitation energy and transition dipole moments by fitting of the experimental relative intensities and peaks position.

References

1. J. R. Lombardi and R. L. Birke, *J. Phys. Chem. C*, 2008, **112**, 5605.
2. J. F. Arenas, I. López Tocón, J. C. Otero and J. I. Marcos, *J. Phys Chem.*, 1996, **100**, 9254.