

Supporting Information

The Effects of Varying Solvent for MoS₂ Treatment on its Catalytic Efficiencies for HER and ORR

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Supporting Information

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1. Scanning electron micrograph showing the surface morphologies of the deposited MoS₂ dispersed in various solvents.

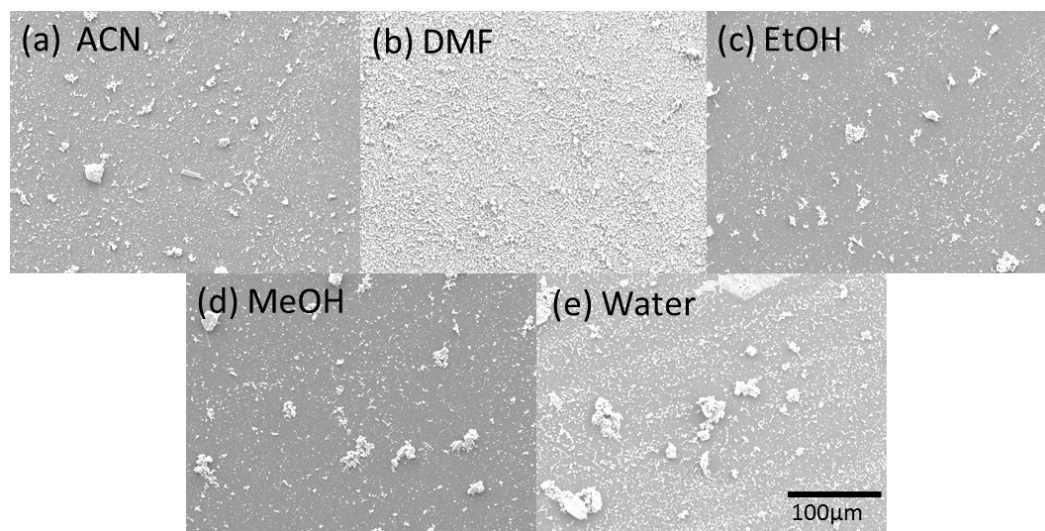


Figure S1: SEM images of MoS₂ drop-casted on the silicon oxide wafer. The aliquots were taken from MoS₂ dispersed in various solvents: (a) ACN (5mg/mL; 1μL), (b) DMF (5mg/mL; 1μL), (c) EtOH (5mg/mL; 1μL), (d) MeOH (5mg/mL; 1μL) and (e) water (1mg/mL; 5μL). Scale bars represent 100 μm. Legend: (*X* mg/mL, *Y* μL), *X* refers to the concentration of the dispersion and *Y* refers to the volume used for drop-casting, with the final MoS₂ loading being constant.

2. Values of the observed heterogeneous electron transfer (HET) rates, k_{obs}^0 , of MoS₂ dispersed in various solvent.

Table S1: Calculated observed heterogeneous electron transfer rate constant k_{obs}^0 for the Fe(CN)₆^{3-/4-} redox probe when MoS₂ is dispersed in various solvents. Bare glassy carbon electrode (GCE) is also shown for reference.

Solvent	k_{obs}^0 (cm s ⁻¹)
ACN	2.99×10^{-3}
DMF	5.90×10^{-4}
EtOH	3.53×10^{-3}
MeOH	2.74×10^{-3}
Water	1.88×10^{-3}
Bare GCE	3.20×10^{-3}

3. Control Experiment using solvent-only modified GCE for HER

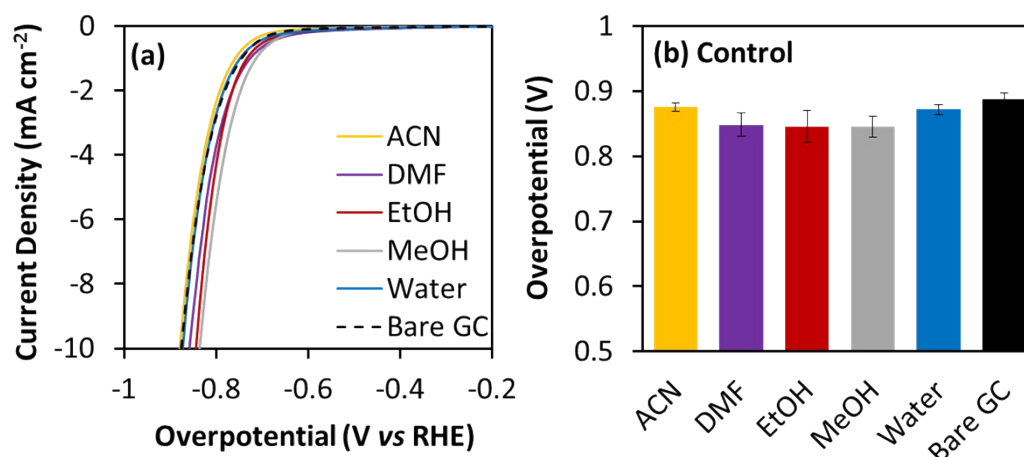


Figure S2: The control experiment for the HER where only solvents (without any material suspended) were drop-casted on bare GCE. (a) Polarization curve for HER and (b) the bar chart shows the overpotential required to reach the specific current density of -10 mA cm^{-2} . Conditions: $0.5 \text{ M H}_2\text{SO}_4$; scan rate: 2 mV s^{-1} . Error bars correspond to standard deviations based on triplicate measurements. Potentials are with respect to RHE.

4. Tafel Plots of the Freshly-Prepared MoS_2 -Dispersion.

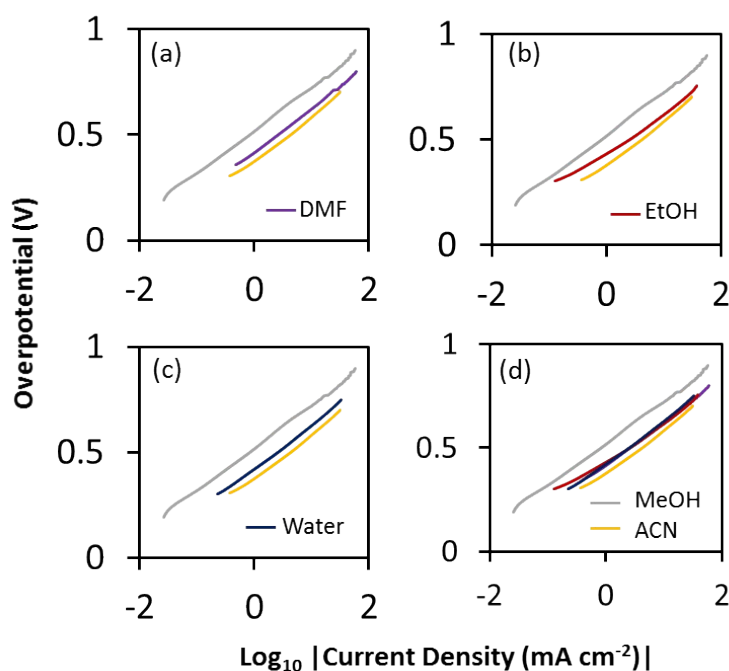


Figure S3. The Tafel plots of the freshly prepared dispersions of MoS_2 in the five solvents. DMF, EtOH and water were separated to ensure clarity.