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Electronic Supplementary Information

Effect of protonation on the solvation structure of solute *N*-butylamine in an aprotic ionic liquid

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Solute species	Density / g cm ⁻³		Solute	Cation	Anion
(concentration) ^a	MD ^b	Exp ^c	_		
BuNH ₂ (15 mol%)	1.537	1.475	45	256	256
BuNH ₂ (30 mol%)	1.496	1.414	110	256	256
BuNH ₃ ⁺ (30 mol%)	1.607	1.501	110	256	366

Table S1. Compositions and densities of the systems for the MD simulations.

^a Mol fraction of solute in IL, [C₂mIm][TFSA]

^b Densities obtained from MD simulations.

^c Densities obtained from experiments.



Fig. S1. $S^{\exp}(q)$'s for the (a) 0–30 mol% BuNH₂ and (b) 30 mol% BuNH₃⁺ in [C₂mIm][TFSA] solutions in the whole *q*-range examined in this work.



Figure S2. $r^2[G(r) - 1]$'s for the (a) 0–30 mol% BuNH₂ and (b) 30 mol% BuNH₃⁺ in [C₂mIm][TFSA] solutions in the whole *r*-range examined in this work.



Figure S3. X-ray weighted (a) $G^{\text{MD}}_{\text{intra_total}}(r)$, as a form of $r^2[G(r) - 1]$ obtained from MD simulations for 30 mol% BuNH₂ and BuNH₃⁺ in [C₂mIm][TFSA] solutions, together with the partial $G^{\text{MD}}_{\text{intra}}(r)$'s for (b) solutes (BuNH₂ and BuNH₃⁺): $r^2[G^{\text{MD}}_{\text{intra}}(r) - 1]$, (c) C₂mIm⁺ cation: $r^2[G^{\text{MD}}_{\text{intra}}(r) - 1]$, and (d) TFSA⁻ anion: $r^2[G^{\text{MD}}_{\text{intra}}(r) - 1]$ interactions.



Figure S4. X-ray weighted (a) $G^{\text{MD}}_{\text{inter_total}}(r)$, as a form of $r^2[G(r) - 1]$ obtained from MD simulations for 30 mol% BuNH₂ and BuNH₃⁺ in [C₂mIm⁺][TFSA⁻] solutions, together with the partial $G^{\text{MD}}_{\text{inter}}(r)$'s for (b) solute-solute: $r^2[G^{\text{MD}}_{\text{Bu-Bu}}(r) - 1]$, (c) cation-cation: $r^2[G^{\text{MD}}_{\text{ca-ca}}(r) - 1]$, (d) cation-anion: $r^2[G^{\text{MD}}_{\text{ca-an}}(r) - 1]$, and (e) anion-anion: $r^2[G^{\text{MD}}_{\text{an-an}}(r) - 1]$ interactions.