

Electronic supplementary information (ESI) for the manuscript:

Self-metathesis of 1-butene to propene over SBA-15-supported WO₃

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Table S1. H₂-TPR quantitative analysis results of WO₃-containing catalysts.

Samples	Peak position(°C)	H ₂ uptake (μmol/g)	Theoretical H ₂ uptake (μmol/g) ^a	Reduction degree
WO ₃ /SBA-15	730,773	800.5	2566.9	31.19%
WO ₃ /SBA-15-700	648, 721, 781	1291.4	2506.5	52.52%
WO ₃ -SBA-15-700	457, 609, 732	453.9	2390.7	18.99%

^aTheoretical H₂ uptake was determined as the quantity of H₂ required for the total reduction of WO₃ in the catalysts to W⁰.

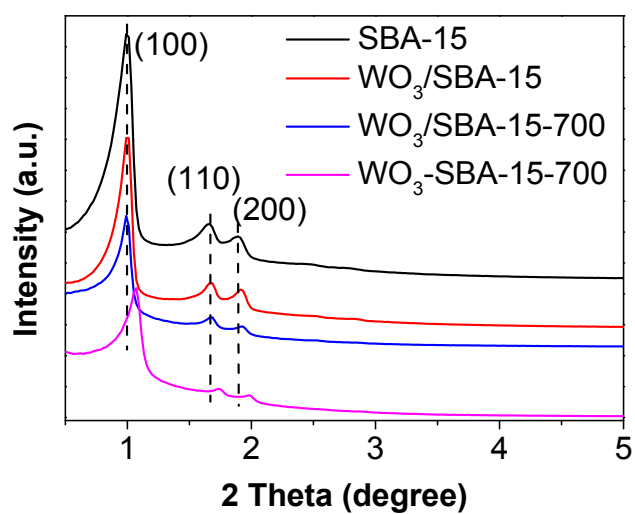


Fig. S1. Small-angle XRD patterns of SBA-15, $\text{WO}_3/\text{SBA-15}$, $\text{WO}_3/\text{SBA-15-700}$ and $\text{WO}_3\text{-SBA-15-700}$.

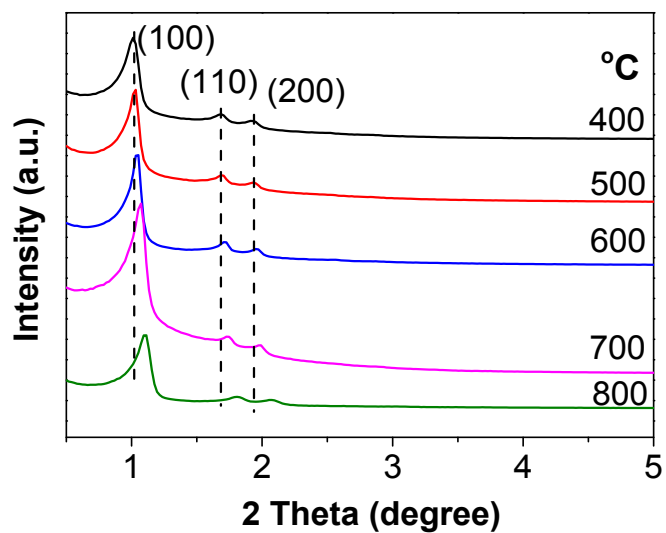


Fig. S2. Small-angle XRD patterns of $\text{WO}_3\text{-SBA-15-x}$ (x represents the calcination temperature, °C).

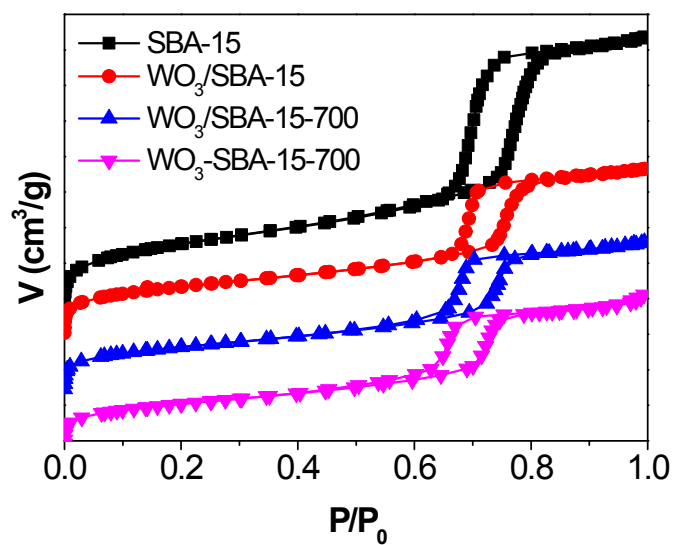


Fig. S3. N_2 adsorption/desorption isotherms of SBA-15, $\text{WO}_3/\text{SBA-15}$, $\text{WO}_3/\text{SBA-15-700}$ and $\text{WO}_3\text{-SBA-15-700}$.

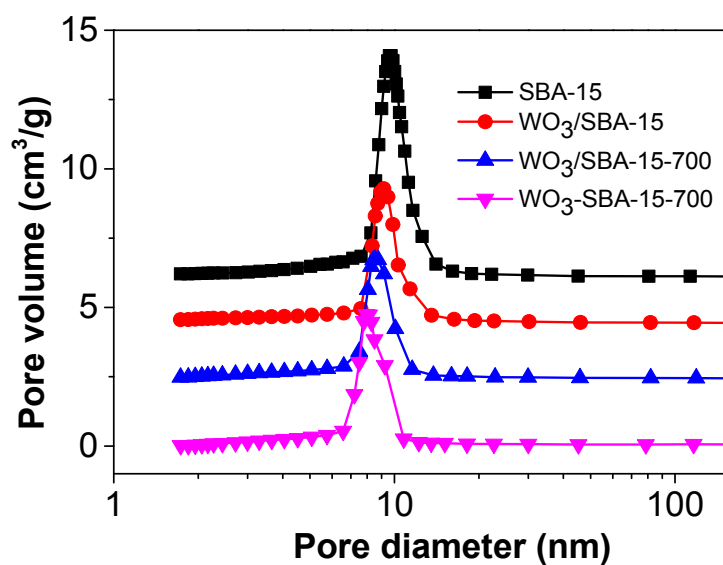


Fig. S4. Pore size distribution of SBA-15, $\text{WO}_3/\text{SBA-15}$, $\text{WO}_3/\text{SBA-15-700}$ and $\text{WO}_3\text{-SBA-15-700}$.

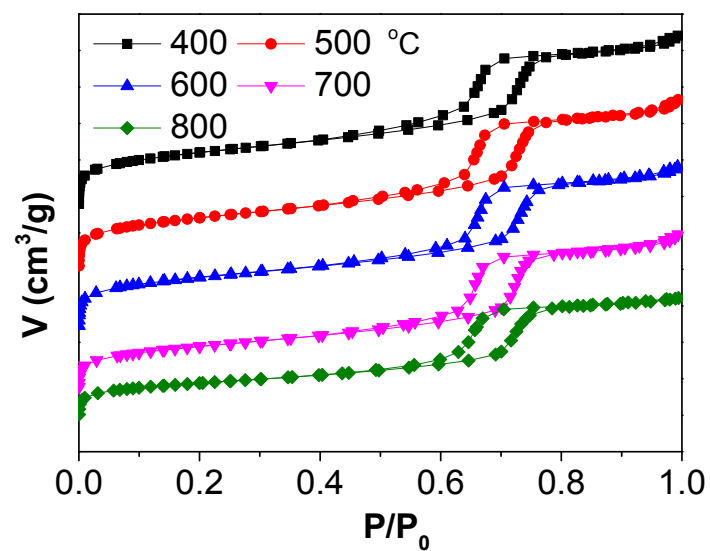


Fig. S5. N₂ adsorption/desorption isotherms of WO₃-SBA-15-x (x represents the calcination temperature, °C).

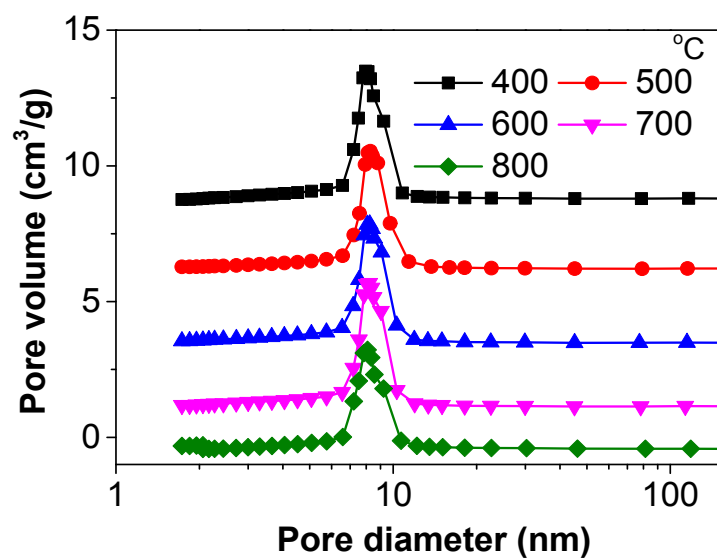


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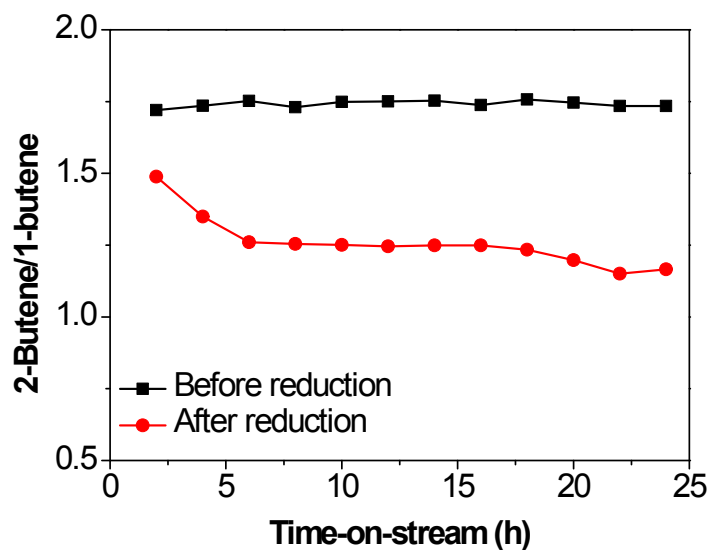


Fig. S7. 2-Butene/1-butene ratios obtained at different reaction time over $\text{WO}_3\text{-SBA-15-700}$ (■) and its reduced counterpart in H_2 at 550 °C for 3 h (●) (reaction conditions: 350 °C, 0.1 MPa, WHSV=1 h^{-1}).

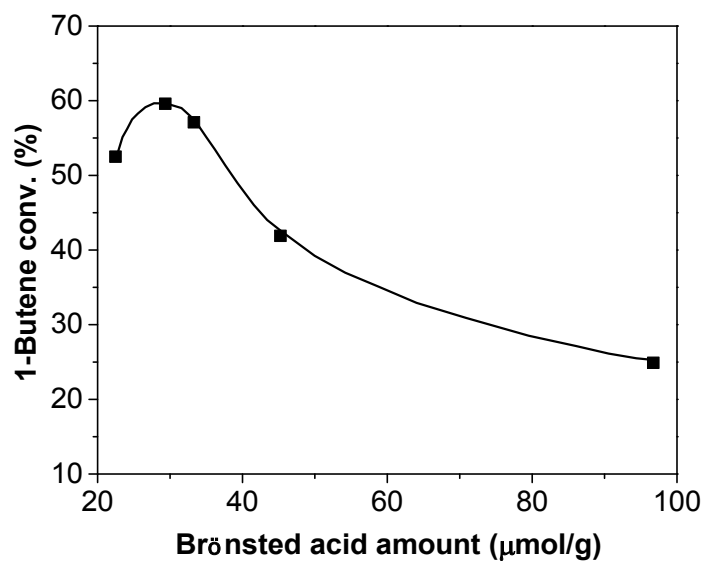
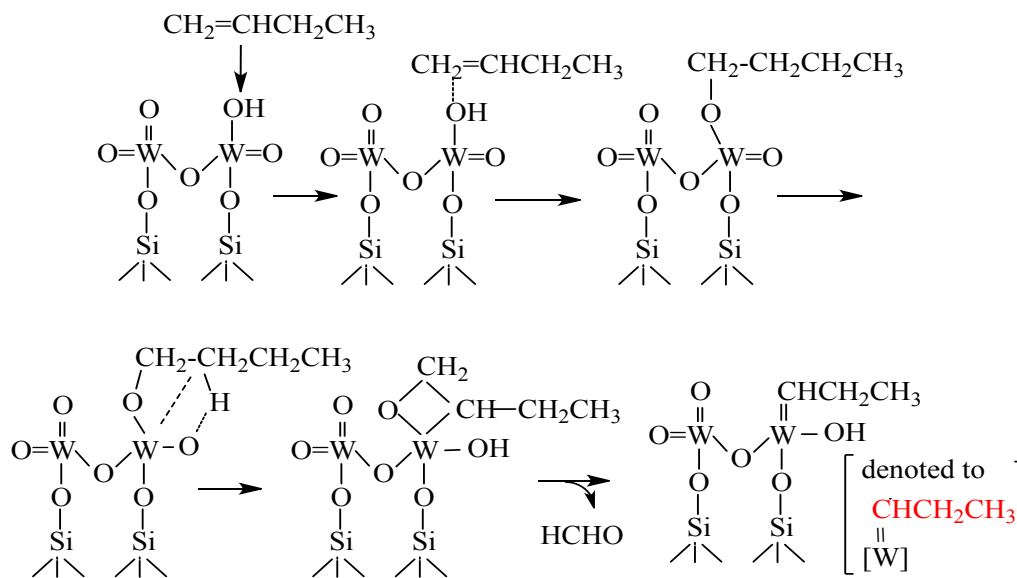
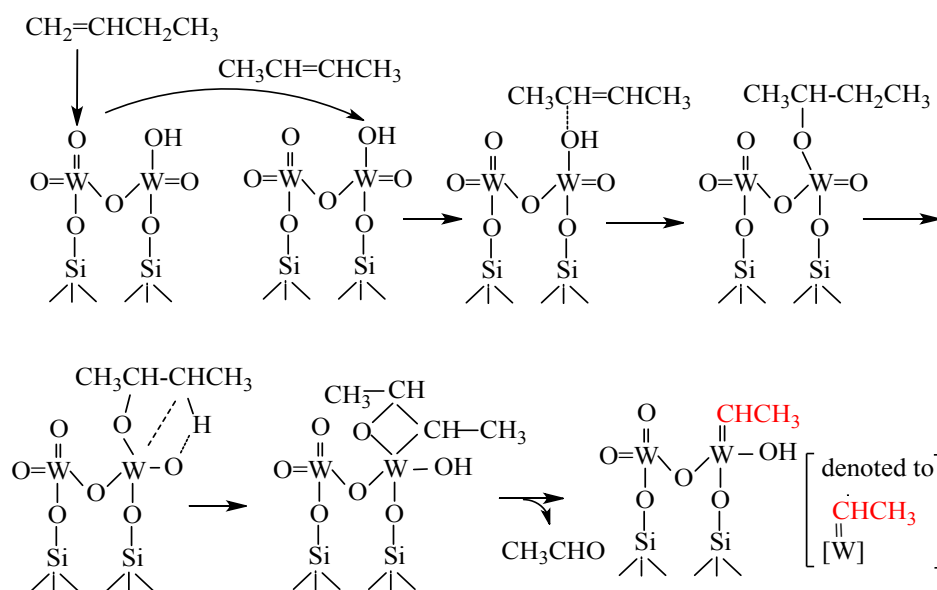


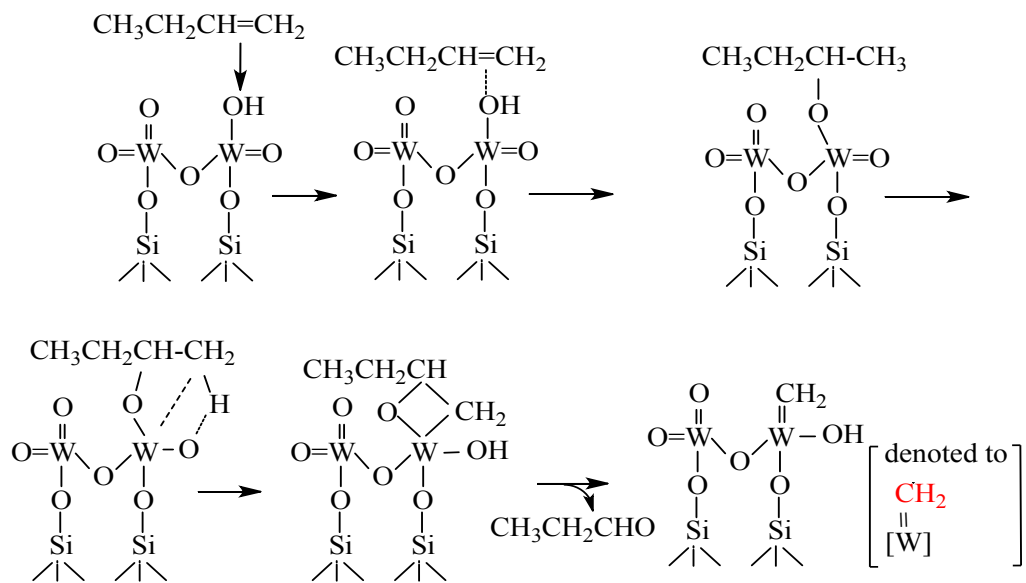
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Scheme S1. Pathway for formation of formaldehyde and $\text{W}=\text{CHCH}_2\text{CH}_3$ species over WO_3 -SBA-15-700.



Scheme S2. Pathway for formation of acetaldehyde and $\text{W}=\text{CHCH}_3$ species over WO_3 -SBA-15-700.



Scheme S3. Pathway for formation of propylaldehyde and $\text{W}=\text{CH}_2$ species over WO_3 -SBA-15-700.