

## High yield conversion of cellulosic biomass into 5-Hydroxymethylfurfural and a study of reaction kinetics of cellulose-to-HMF in a biphasic system

Luqman Atanda,<sup>a</sup> Muxina Konarova,<sup>a</sup> Qing Ma,<sup>a</sup> Swathi Mukundan,<sup>a</sup> Abhijit Shrotri,<sup>b</sup> and Jorge Beltramini<sup>\*a</sup>

<sup>a</sup> Nanomaterials Centre, Australian Institute for Bioengineering and Nanotechnology and School of Chemical Engineering, The University of Queensland, Brisbane, St. Lucia 4072, Australia

<sup>b</sup> Institute for Catalysis, Hokkaido University, Kita 21 Nishi 10, Kita-Ku, Sapporo 001-0021, Japan

\*To whom correspondence should be addressed, E-mail: [j.beltramini@uq.edu.au](mailto:j.beltramini@uq.edu.au)

Table S1 Furan yield calculation from biomass feedstock

mass of feed stock = 7.5 g

Untreated		Alkali-organocat treated	
<i>Sugarcane bagasse</i>			
	%	g	%
cellulose	41.5	3.11	84.4
xylan	24.1	1.81	6.56
<i>Rice husk</i>			
	%	g	%
cellulose	35.2	2.64	82.0
xylan	18.7	1.40	5.47

$$\text{mole of glucose} = \frac{\text{mass of cellulose}}{162.14}$$

$$\text{mole of xylose} = \frac{\text{mass of xylan}}{132.12}$$

$$\text{total moles} = \text{mole of glucose} + \text{mole of xylose}$$

$$\text{HMF yield} = \frac{\text{moles of HMF produced}}{\text{total moles}}$$

$$\text{furfural yield} = \frac{\text{moles of furfural produced}}{\text{total moles}}$$

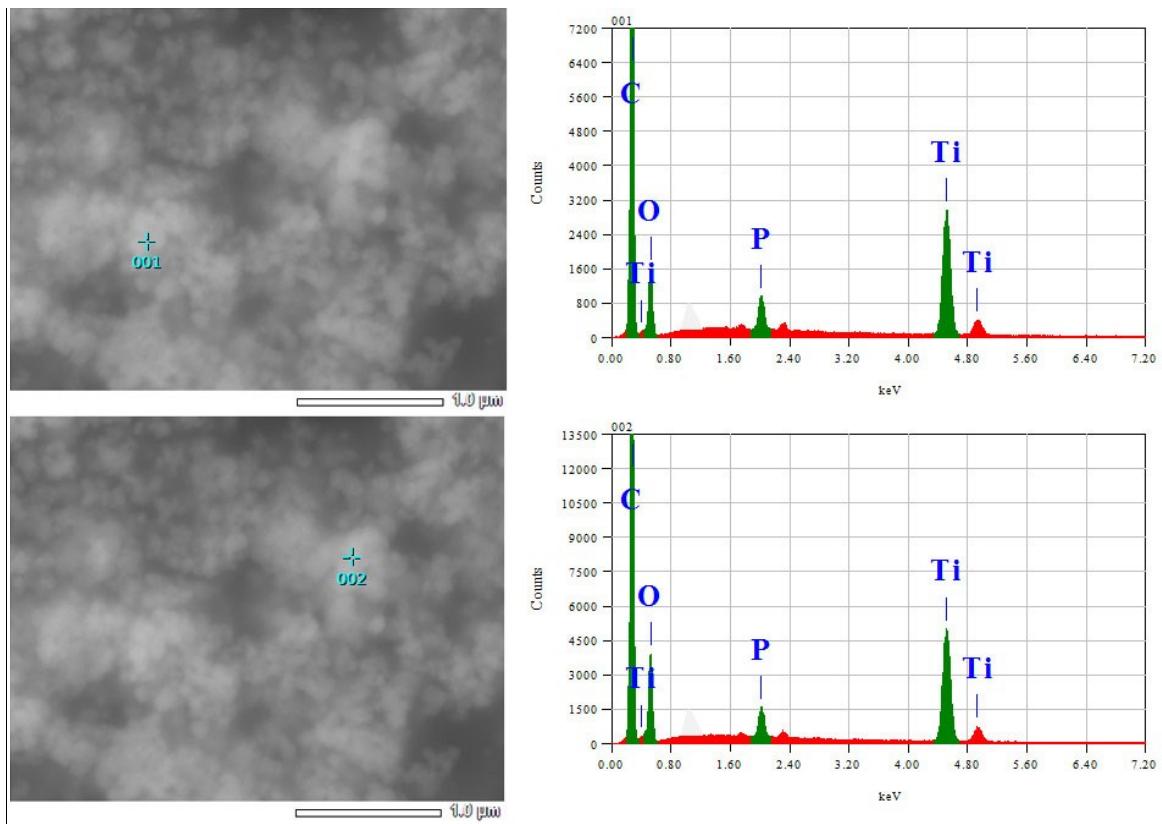


Fig. S1 EDS analysis of phosphated-TiO<sub>2</sub> nanoparticle.

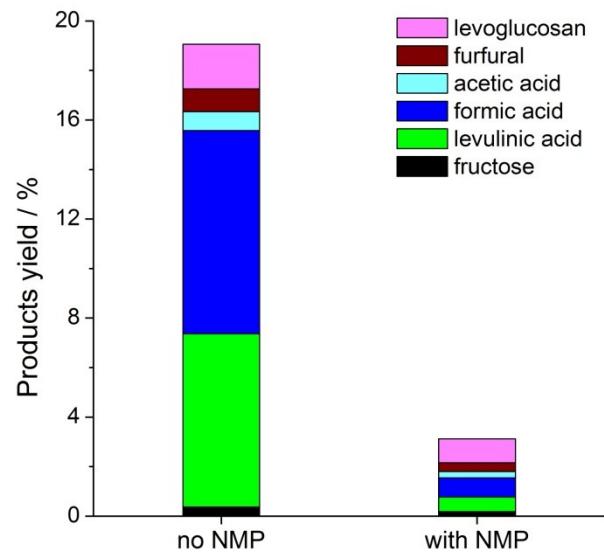


Fig. S2 Product distribution of glucose conversion on phosphated-TiO<sub>2</sub> catalyst. Reaction conditions: 10 g glucose, 1.25 g catalyst, 448 K, 80 min, 100 mL solvent volume (water–organic solvent 3:7 v/v). MeTHF:NMP 6:1 v/v for the modified reaction system.

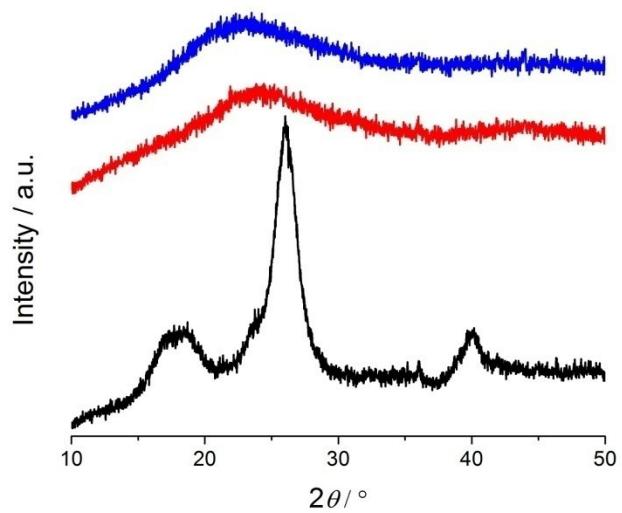
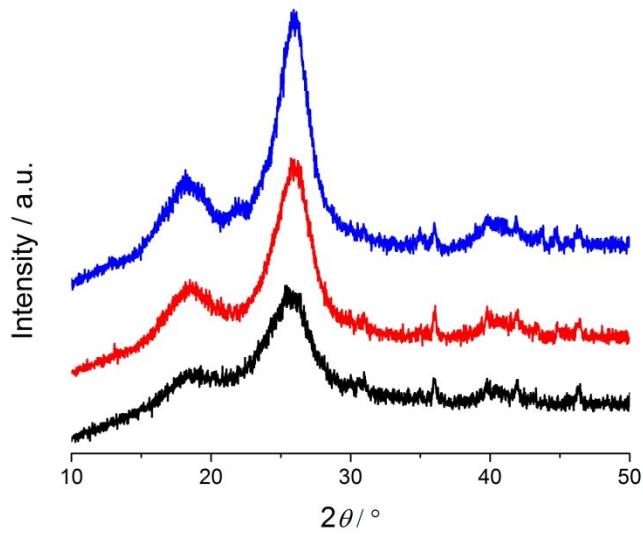
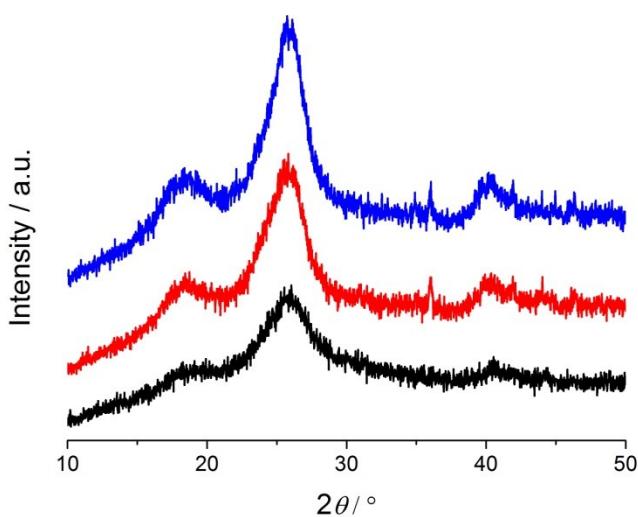


Fig. S3 XRD pattern of microcrystalline cellulose (black), ball-milled cellulose (red), mechanocatalytically depolymerized cellulose (blue)



a) sugarcane bagasse



b) rice husk

Fig. S4 XRD pattern of biomass substrates: a) sugarcane bagasse and b) rice husk. untreated (black), alkali-treated (red), organocat-alkali treated (blue).

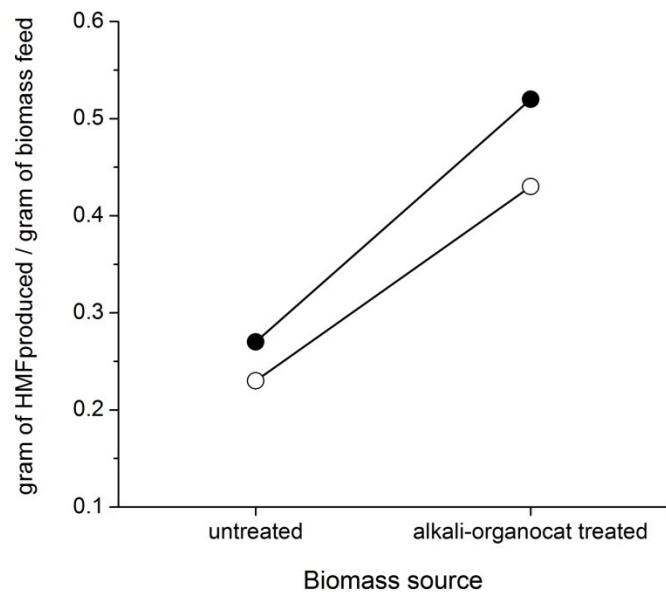


Fig. S5 HMF production per quantity of biomass source.

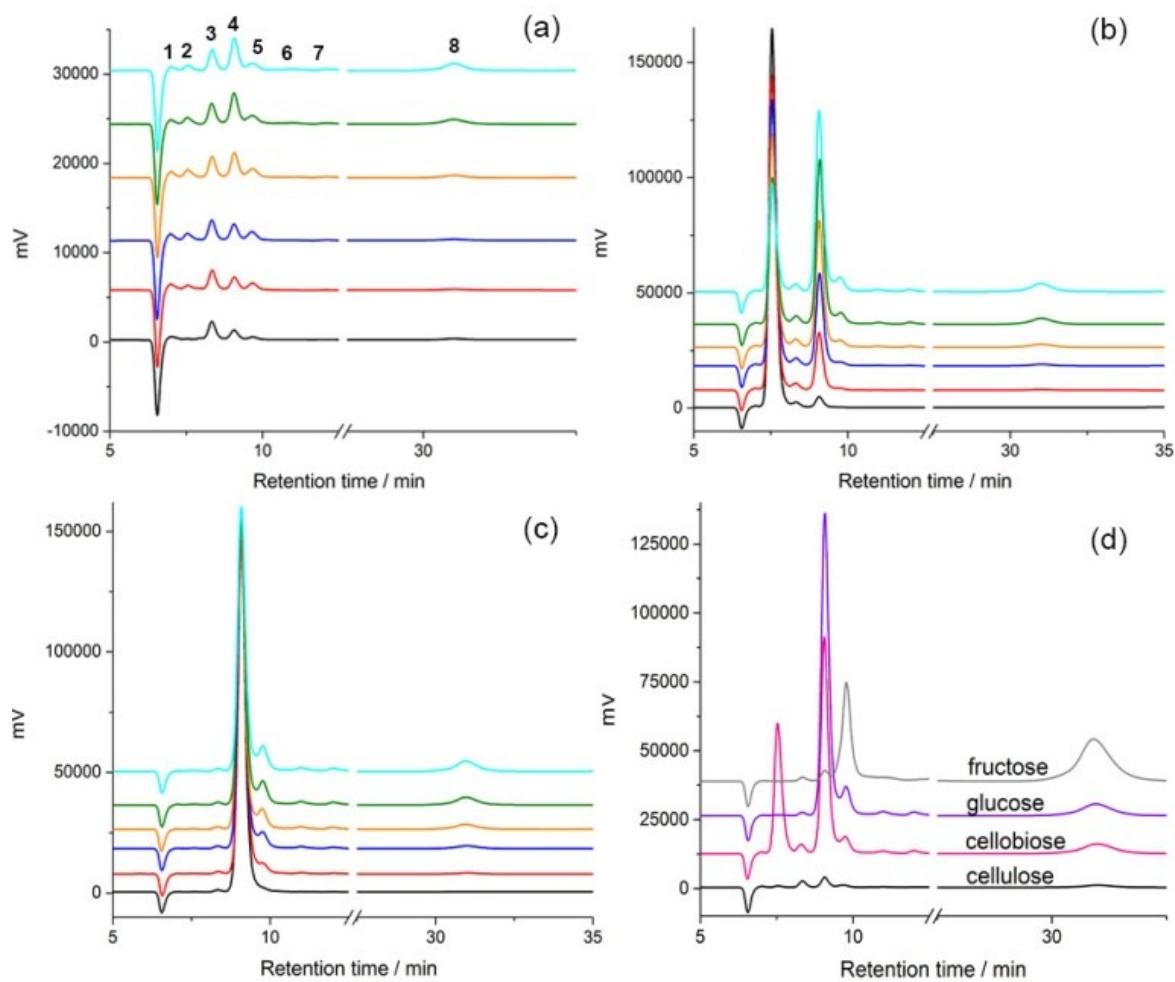


Fig. S6 HPLC analytical profile of carbohydrates conversion measured using RID detector. a) cellulose, b) cellobiose, c) glucose.<sup>a</sup> Reaction conditions: 1 g substrate, 0.4 g cat. wt. (0.2 g)<sup>a</sup>, 453 K (433 K)<sup>a</sup>, 100 mL water. d) comparison of carbohydrates conversion at 60 min reaction time. Reaction time: 0 min (black), 10 min (red), 20 min (blue), 30 min (orange), 45 min (green) and 60 min (cyan). Products: cellobiose – 2, glucose – 4, fructose – 5, HMF – 8, unknowns – 1, 3, 6 and 7.

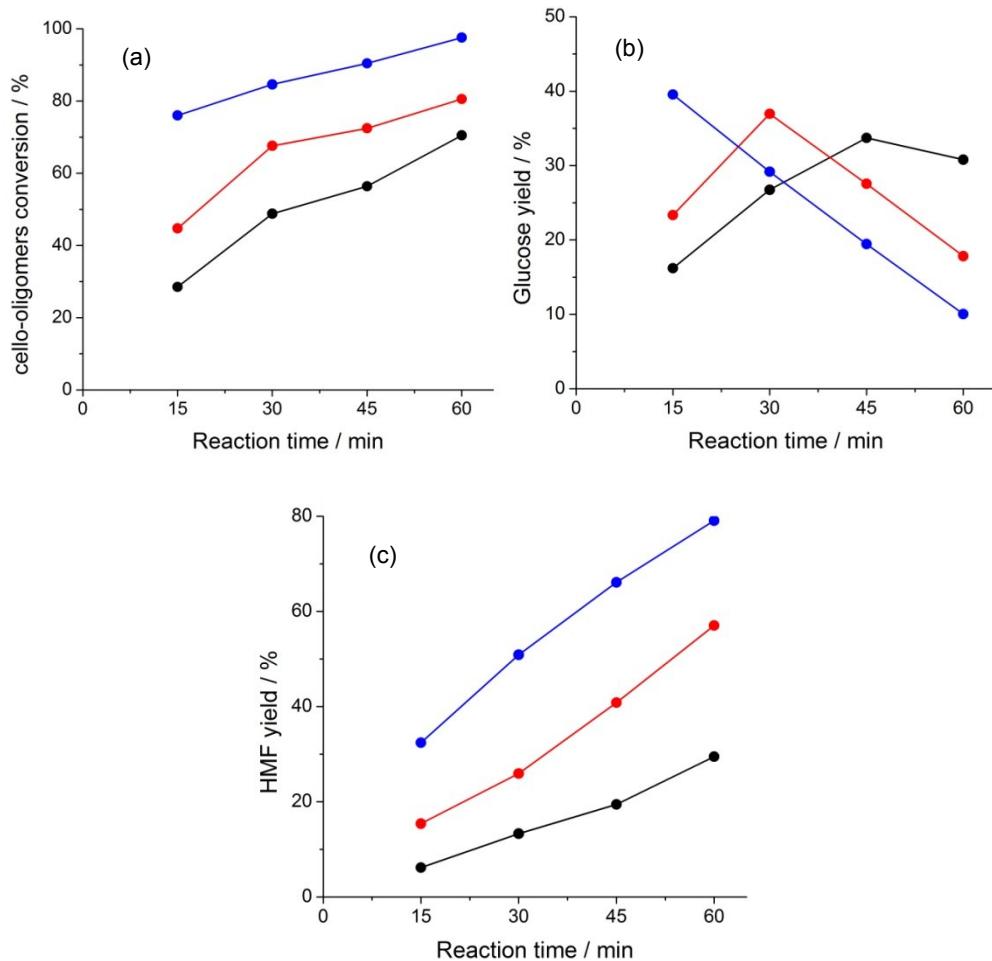


Fig. S7 Experimental rate data as a function of reaction temperature and time for: a) cello-oligomers conversion and yields of b) glucose and c) HMF. Reaction conditions: 10 g substrate, 1.25 g catalyst, 100 mL solvent volume (water–MeTHF/NMP 3:7 v/v). Reaction temperature: 428 K (black), 438 K (red) and 448 K (blue); reaction time: 15–60 min.