

Electronic Supplementary Information

Engineering of ZSM-5 zeolite crystals for enhanced lifetime in the production of light olefins *via* 2-methyl-2-butene cracking

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Table S1. Average product selectivity evidenced over the zeolite catalysts during a single catalytic cycle in the conversion of methanol to olefins. Conditions: $T = 723$ K, $P = 1$ bar, $WHSV = 6 \text{ g}_{\text{MeOH}} \text{ g}_{\text{zeolite}}^{-1} \text{ h}^{-1}$.

Sample	Lifetime ^a (h)	$S_{\text{C2=}}^b$ (Cmol%)	$S_{\text{C3=}}^b$ (Cmol%)	$S_{\text{C4=}}^b$ (Cmol%)	STY^c ($\text{g}_{\text{C2-4}} \text{ g}_{\text{zeolite}}^{-1} \text{ h}^{-1}$)	$S_{\text{C1-3alk}}^b$ (Cmol%)	$S_{\text{C4-7alk}}^b$ (Cmol%)	$S_{\text{C6-8ar}}^b$ (Cmol%)
OH-M	18	10	27	17	3.28	5	32	10
OH-M-H	39	14	25	9	2.64	4	31	9
OH-N	27	8	20	12	2.38	7	36	15
OH-N-H	54	8	20	10	2.25	4	37	10
F-M	19	10	21	12	2.59	6	26	15
F-N	25	9	22	13	2.64	4	27	13
F-N-H	42	7	24	11	2.49	3	33	9

^a Time on stream during which $X_{\text{MeOH}} > 80\%$. ^b Selectivity to ethene ($S_{\text{C2=}}$), propene ($S_{\text{C3=}}$), butenes ($S_{\text{C4=}}$), light alkanes ($S_{\text{C1-3alk}}$), gasoline-range alkanes ($S_{\text{C4-7alk}}$), and to benzene, toluene, and xylene aromatics ($S_{\text{C6-8ar}}$). ^c Space-time yield of C2-4 olefins.

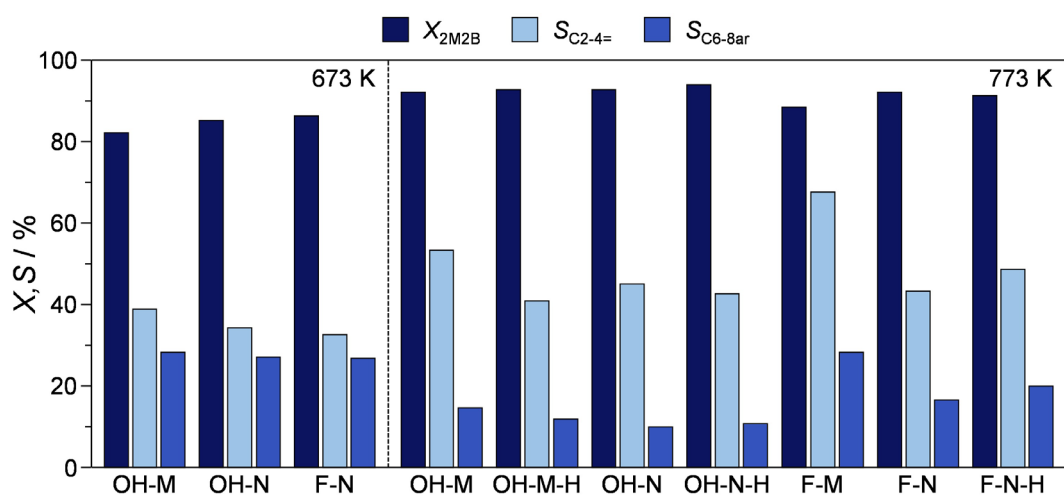


Figure S1. Conversion and selectivity over selected ZSM-5 catalysts in 2M2B cracking at different temperatures. Conditions: $t = 2$ h, $P = 1$ bar, $WHSV = 12 \text{ g}_{2M2B} \text{ h}^{-1} \text{ g}_{\text{catalyst}}^{-1}$.

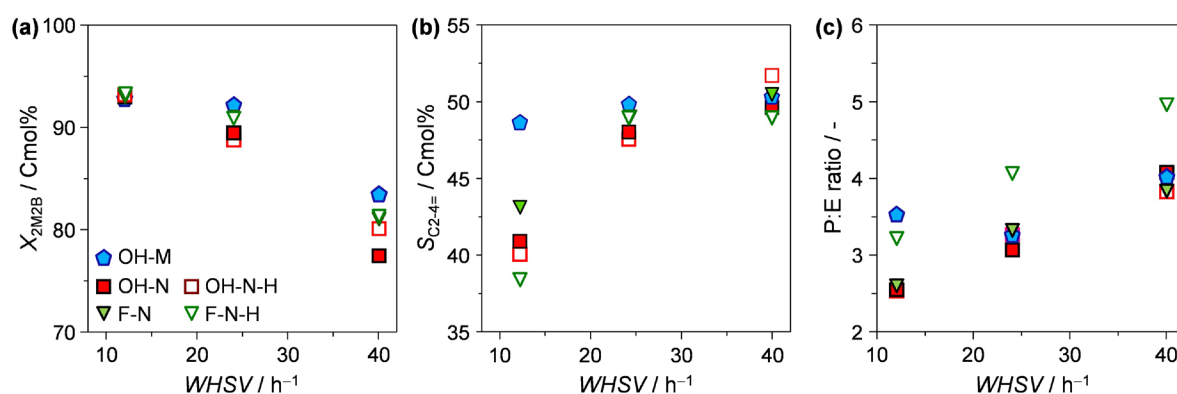


Figure S2. (a) Conversion, (b) olefin selectivity, and (c) the propylene:ethylene ratio in 2M2B cracking over selected ZSM-5 catalysts *versus* the weight hourly space velocity. Conditions: $t = 2$ h, $T = 773$ K, $P = 1$ bar.

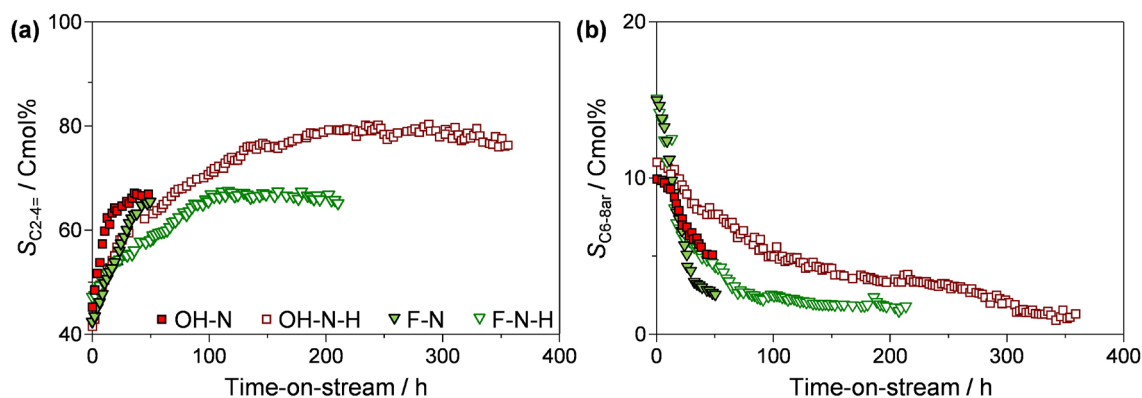


Figure S3. Selectivity to (a) light olefins ($S_{C2-4=}$) or (b) aromatics (S_{C6-8ar}) *versus* time-on-stream in 2M2B cracking over the nanosized zeolites and their desilicated analogues. Conditions: $T = 773$ K, $P = 1$ bar, $WHSV = 12 \text{ g}_{2M2B} \text{ g}_{zeolite}^{-1} \text{ h}^{-1}$.

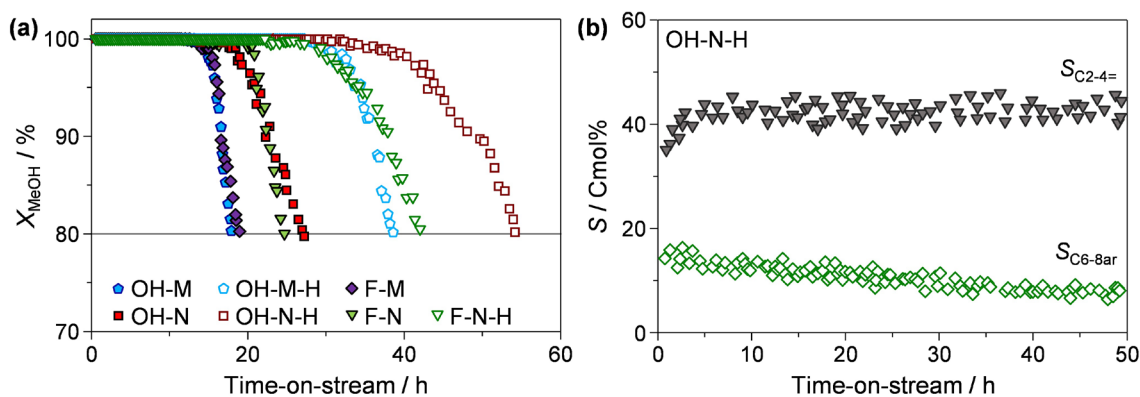


Figure S4. (a) Conversion and (b) selectivity to light olefins ($S_{C2-4=}$) or aromatics (S_{C6-8ar}) *versus* time-on-stream in the conversion of methanol to olefins over the ZSM-5 zeolites studied. The grey line indicates the degree of conversion at which a single catalytic cycle was defined. Conditions: $T = 723$ K, $P = 1$ bar, $WHSV = 6 \text{ g}_{MeOH} \text{ g}_{zeolite}^{-1} \text{ h}^{-1}$.