Electronic Supplementary Material (ESI) for Catalysis Science & Technology. This journal is © The Royal Society of Chemistry 2016

Supporting Information



Fig. S1 The potential energy profile of uncatalyzed coupling reaction of CO_2 and PO calculated at the M06/6-311+G(2d,2p) (PCM)//B3PW91/6-31G(d,p) level along with the structures of intermediate and transition state. (Only the most favorable route is presented)



Fig. S2 The ball-and-stick structures of some key transition states and intermediates in routes 1-4. Bond lengths are in angstroms.







Fig. S3 The ball-and-stick structures of some key transition states and intermediates in routes 5-7. Bond lengths are in angstroms.





Fig. S4 The ball-and-stick structures of some key transition states and intermediates in routes 8-10 along with the ts11-1 and cp11-1. Bond lengths are in angstroms.



Fig. S5 Snapshots of the NEt₃(HE)Br ILs system after 5ns simulation. The N atoms are shown blue, O atoms are shown red, C atoms are shown green, Br atoms are shown pink, and H atoms are shown white.

number	atom	type	description				
1	N1	n4	sp ³ N with four connected atoms				
2	C1	c3	sp ³ C				
3	H1	hx	H bonded to C next to positively charged group				
4	H2	hx	H bonded to C next to positively charged group				
5	C2	c3	sp ³ C				
6	H3	hc	H bonded to aliphatic carbon without electrwd. group				
7	H4	hc	H bonded to aliphatic carbon without electrwd. group				
8	H5	hc	H bonded to aliphatic carbon without electrwd. group				
9	C3	c3	sp ³ C				
10	H6	hx	H bonded to C next to positively charged group				
11	C4	c3	sp ³ C				
12	H7	hx	H bonded to C next to positively charged group				
13	H8	hx	H bonded to C next to positively charged group				
14	C5	c3	sp ³ C				
15	H9	hc	H bonded to aliphatic carbon without electrwd. group				
16	H10	hc	H bonded to aliphatic carbon without electrwd. group				
17	H11	hc	H bonded to aliphatic carbon without electrwd. group				
18	C6	c3	sp ³ C				
19	H12	hx	H bonded to C next to positively charged group				
20	H13	hx	H bonded to C next to positively charged group				
21	C7	c3	sp ³ C				
22	H14	h1	H bonded to aliphatic carbon with 1 electrwd. Group				
23	H15	h1	H bonded to aliphatic carbon with 1 electrwd. Group				
24	01	oh	Oxygen in hydroxyl group				
25	H16	ho	Hydroxyl group				
26	C8	c3	sp ³ C				
27	H17	hc	H bonded to aliphatic carbon without electrwd. group				
28	H18	hc	H bonded to aliphatic carbon without electrwd. group				
29	H19	hc	H bonded to aliphatic carbon without electrwd. group				
30	H20	hx	H bonded to C next to positively charged group				
31	Br1	br	any bromine				

 Table S1 Atom types and their definitions in GAFF.

route	1	2	3	4	5	6	7	8	9	10
cpn-1	-3.36	-3.30	-3.70	-3.23	-21.31	-20.88	-16.88	-14.54	-18.97	-18.60
tsn-1	17.75	19.56	16.06	16.64	0.97	0.44	1.45	1.48	-5.37	0.62
cpn-2	4.65	8.84	5.66	5.24	-19.63	-17.25	-9.02	-5.08	-13.89	-11.44
cpn-3	-11.28	-6.49	-14.84	-16.58	-28.09	-23.35	-19.40	-16.39	-19.75	-24.55
tsn-3	3.25	8.29	1.89	0.26	-8.15	-6.30	-10.12	-9.65	-12.83	-15.55
cpn-4	-22.17	-21.58	-25.67	-13.12	-30.35	-29.47	-34.57	-33.32	-36.50	-39.92

Table S2 All relative energies (kcal/mol) of routes 1-10 calculated at the M06/6-311+G(2d,2p) (PCM)//B3PW91/6-31G(d,p) level. Letter n indicates route number.