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Substrate and cofactor binding to nitrile reductase: a mass spectrometry based study

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Supporting information

Kinetic measurements:

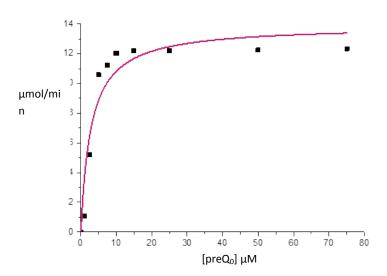


Figure S1. Michaelis-Menten curve

In order to determine the K_m for PreQ₀, the concentration of the cofactor was kept constant at 100 μ M, while changing the substrate concentration from 1-75 μ M. Error margins or standard deviation 5-8%.

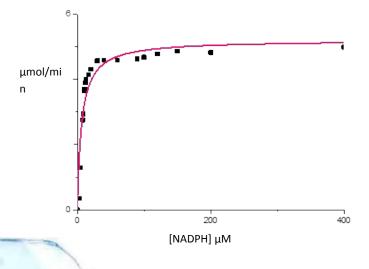


Figure S2. Michaelis-Menten curve

In order to determine the K_m for NADPH, the concentration of PreQ₀ was kept constant at 50 μ M, while changing the cofactor concentration from 1-400 μ M. Error margins Standard deviation 5-8%.

Figure S3. Native gel electrophoresis

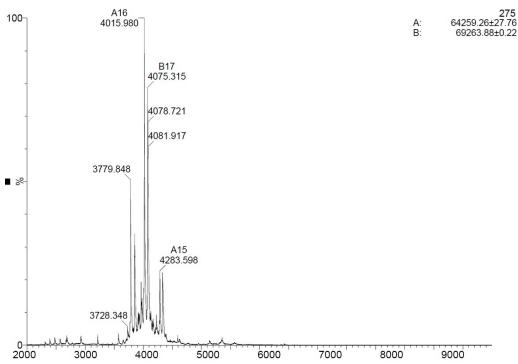


Figure S4. Mass spectra of free EcoNR, A16 multiple charge ion (z=16 $^+$) of dimeric EcoNR- 2 tags, B17 multiple charge ion (z =17 $^+$) of dimeric EcoNR-1 tag.

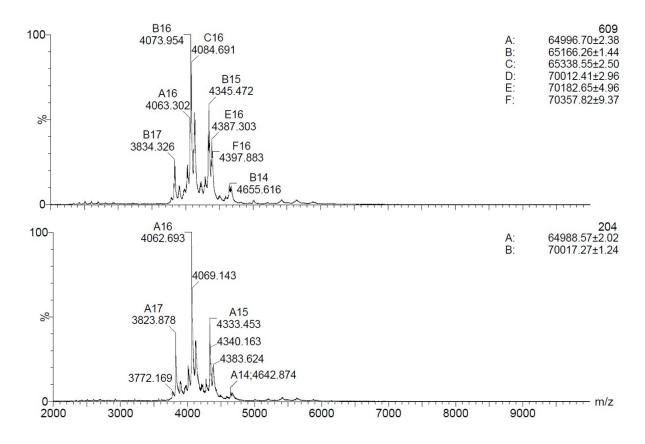


Figure S5. A) *Eco*NR treated with one equivalent of $PreQ_0$ and NADPH each per monomer B) The mass peak m/z=4063 is correlated to the mass of the dimeric *Eco*NR +NADPH when an excess of cofactor was used.

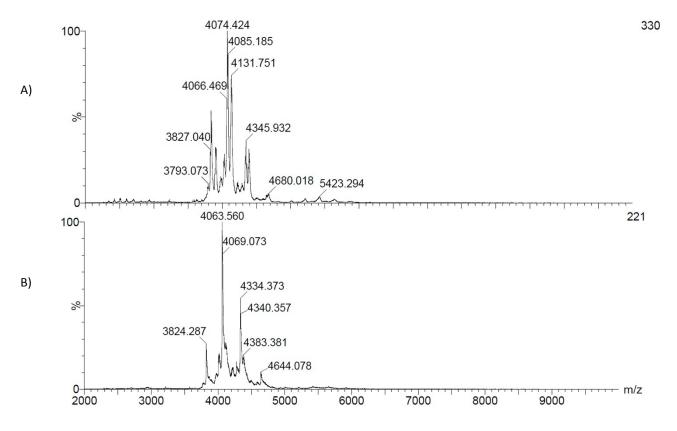


Figure S6. A) *Eco*NR treated with one equivalent of NADPH and $PreQ_0$ each per monomer B) The mass peak m/z=4063 is correlated to the mass of the dimeric *Eco*NR +NADPH when an excess of cofactor was used.

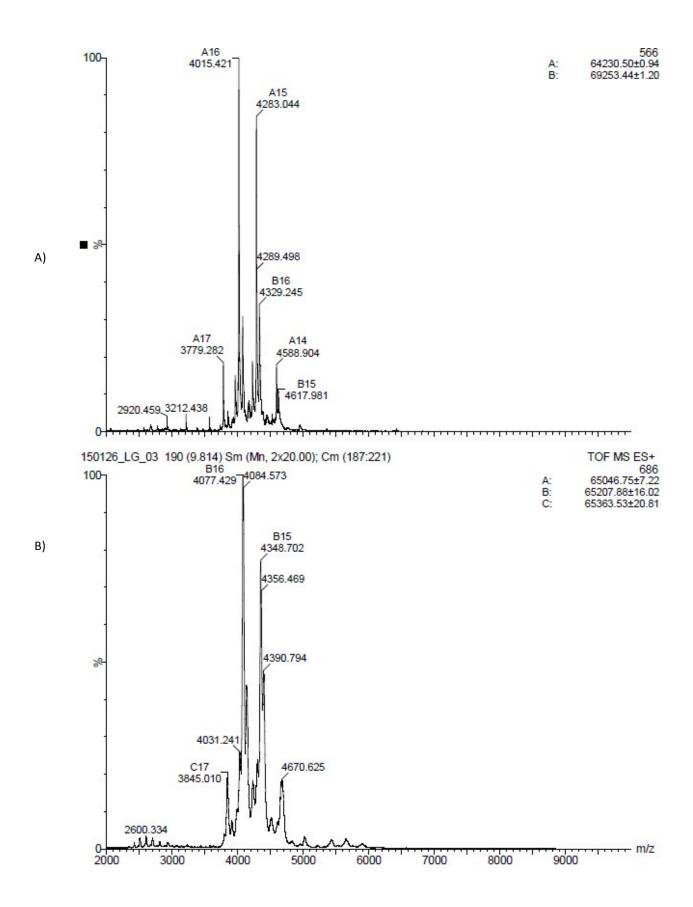


Figure S7. A) EcoNR with AMP and B) followed by the equimolar addition of PreQ₀ and NADPH.

NMR spectrum was recorded on a Bruker Avance spectrometer at 400 MHz (¹H) at room temperature.

$^{\rm 1}H$ NMR in DMSO-d $_{\rm 6}$

