Application of modified CNTs with Ti(SO₄)₂ in

selective oxidation of dimethyl ether

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FT-IR spectra

FT-IR spectra were measured on a Bruker Tensor 27 instrument with MCT detector.

Thermal gravimetric analysis (TGA)

TGA was conducted on a Setaram Setsys Evolution. Weight loss was measured during a heat treatment up to 800°C with a rate of 5°C /min under air.



Fig. ESI-1 TG profiles of the 40%SO₄²⁻/CNTs calcined at 120°C.



Fig. ESI-2 FT-IR of CNTs.



Fig. ESI-3 H₂-TPR-MS of 40%SO₄²⁻/CNTs calcined at 240°C.



Fig. ESI-4 TEM images of the 40%SO₄²⁻/CNTs calcined at 200 °C (a), 240 °C (b), 300 °C (c), 400 °C (d), 450 °C (e) and 500 °C (f).



Fig. ESI-5 O_2 -TPD-MS of CNTs and 40%SO₄²⁻(Ti)/CNTs calcined at 400°C



Fig. ESI-6 DME-TPSR-MS of 40%SO₄²⁻(Ti)/CNTs calcined at 400°C



Fig. ESI-7 TEM images of the $40\% SO_4(H_2SO_4)/CNTs$ calcined at $240^\circ C.$



Fig. ESI-8 NH₃-TPD profiles of 40%SO₄ 2 -(H₂SO₄)/CNTs and 40%SO₄ 2 -(Ti)/CNTs calcined at 240 $^\circ$ C.

Table ESI-1 Results of NH ₃ -TPD integration		
	Weak acid	
Catalyst	(mmol NH ₃ /g)	
CNTs	0.029	
10% SO4 ²⁻ /CNTs	0.125	
30% SO ₄ ²⁻ /CNTs	0.161	
40% SO4 ²⁻ /CNTs	0.174	
50% SO ₄ ²⁻ /CNTs	0.126	

	Weak acid
Calcination temperature (°C)	(mmol NH ₃ /g)
200	0.537
240	0.354
300	0.201
400	0.174
450	0.153
500	0.143

Table ESI-3 Surface composition results of 40%SO ₄ ²⁻ /CNTs at different calcination temperature from XPS					
Calcination temperature	S^{6+}	Ti ⁴⁺	0		
(°C)	(mol%)	(mol%)	(mol%)		
200	4.15	1.00	28.2		
240	3.55	0.94	25.3		
300	2.53	0.97	19.6		
400	2.06	1.15	15.2		
450	1.81	1.06	13.4		
500	0.60	3.14	16.9		