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Supplementary Information

Electrocatalytic behavior of freely-diffusing and immobilized synthetic flavins in aqueous media

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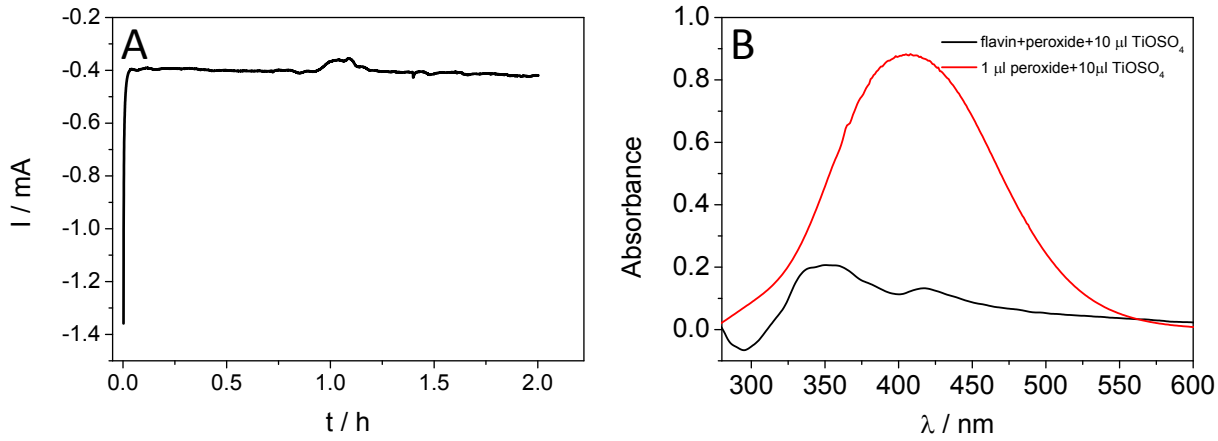

Fig. S1 (A) Chronoamperometric curve recorded during the exhaustive electrolysis of flavin **1**. (E = -0.4 V, t = 2 h, c_{FL1} = 0.5 mM).(B) UV-vis spectra of the electrolyzed sample after adding $TiOSO_4$ (black curve), and UV-vis spectra of H_2TiO_4 complex (red curve).

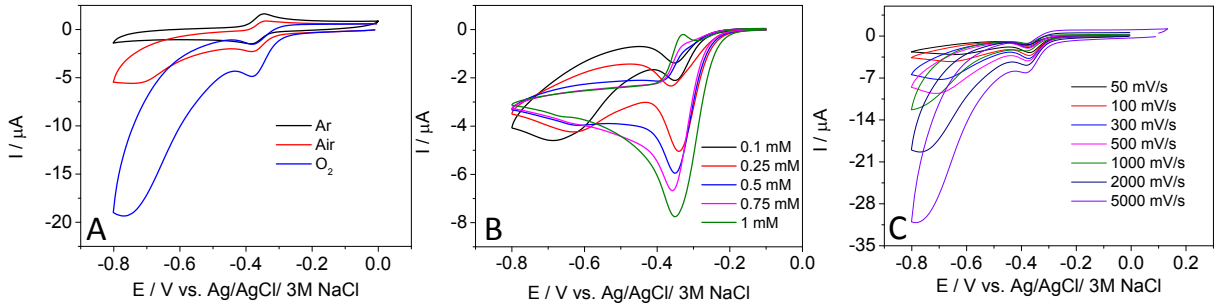

Fig. S2 (A) Comparison of CVs for flavin **2** (GC electrode, pH = 7.0, 0.1 M phosphate buffer solution, varying O_2 concentration (c_{FL1} = 0.1 mM, υ = 2 Vs^{-1})). (B) Comparison of the CV traces for the synthetic flavin **2** at different flavin concentrations (c_{O2} = 2.25 mM, υ=0.1 Vs^{-1}). (C) Scan rate-dependence of the catalytic current during CV measurements (c_{FL2} = 0.1 mM, c_{O2} = 2.25 mM).