

Supporting Information

Catalytic oxidation of benzene over ruthenium-cobalt bimetallic catalysts and the mechanism study

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Contents

1. The characterizations and evaluations data for various samples
2. The XPS data
3. CO₂ selectivity of benzene oxidation over Ru-5Co/TiO₂
4. The TEM images
5. *In-situ FTIR* study

Table S1. The characterizations and evaluations data for various samples

Catalyst	Ru content (wt.%) ^a	S _{BET} (m ² /g)	T ₅₀ (°C)	T ₉₀ (°C)
Ru/TiO ₂	1 (1.01)	52.34	221	229
Ru-1Co/TiO ₂	1 (1.02)	51.91	212	220
Ru-3Co/TiO ₂	1 (1.04)	51.62	219	227
Ru-5Co/TiO ₂	1 (1.01)	51.23	204	215
Ru-10Co/TiO ₂	1 (1.03)	50.01	217	225
5Co-Ru/TiO ₂	1 (1.00)	51.07	219	227
(Ru-5Co)/TiO ₂	1 (0.97)	51.32	218	227

^a The data in parenthesis show the values measured by ICP-OES.

Table S2. The XPS data for the bimetallic Ru-Co catalysts.

Catalyst	Ru _{cus} (At. %)	Ru _{cus} +O _{ot} (At. %)	Ru ⁴⁺ (At. %)	Co (At. %)	Ru/Co	O _{ads} /O _{latt}
Ru-1Co/TiO ₂	0.211	0.103	0.122	0.663	0.778	0.083
Ru-3Co/TiO ₂	0.216	0.099	0.131	0.926	0.482	0.065
Ru-5Co/TiO ₂	0.243	0.141	0.125	0.639	0.796	0.116
Ru-10Co/TiO ₂	0.171	0.095	0.101	1.009	0.364	0.073
5Co-Ru/TiO ₂	0.197	0.068	0.112	1.078	0.349	0.071
(Ru-5Co)/TiO ₂	0.178	0.11	0.096	1.095	0.351	0.071

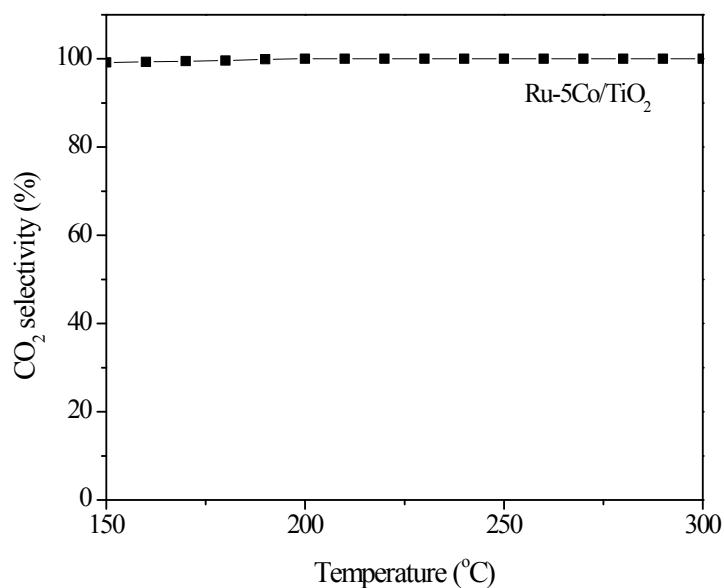


Fig. S1. CO₂ selectivity of benzene oxidation over Ru-5Co/TiO₂.

In the catalytic oxidation of benzene over Ru-5Co/TiO₂, organic byproducts were not observed by GC. Hence, the CO₂ selectivity was calculated using the equation: CO₂ selectivity = [C(CO₂)/(C(CO₂) + C(CO))]. The CO₂ selectivity was above 99% within the temperature range of 150-300 °C due to the high oxidation efficiency of Ru-5Co/TiO₂.

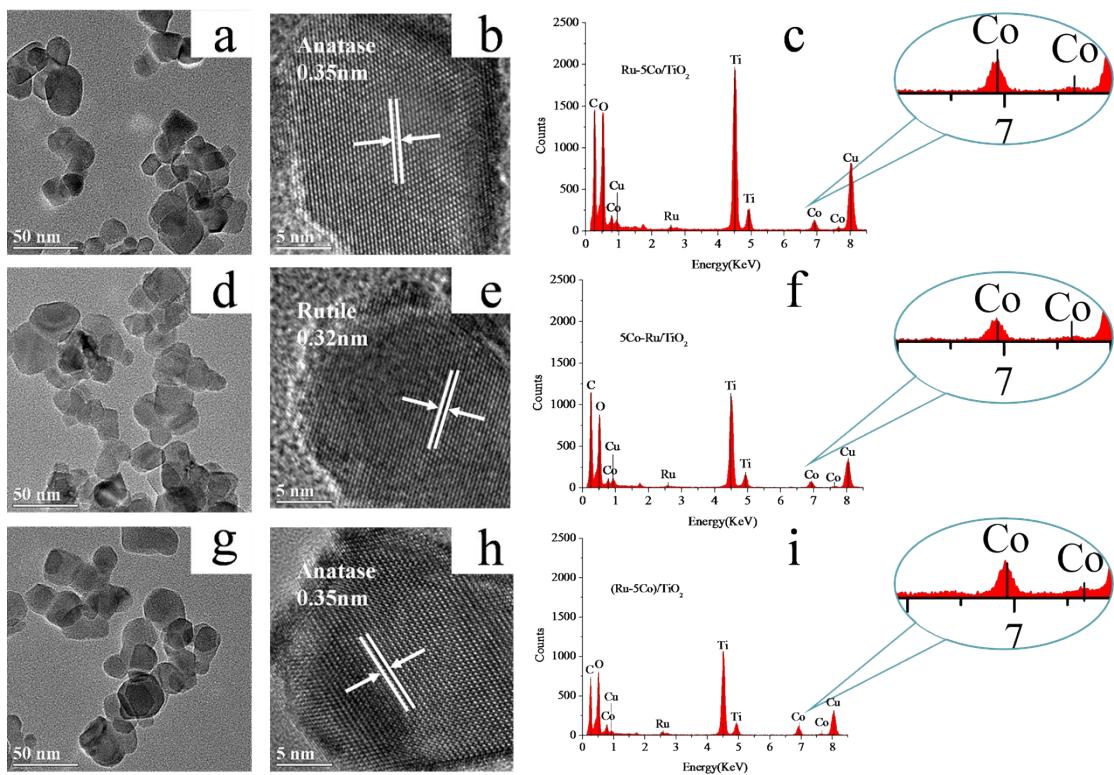


Fig. S2. TEM, HR-TEM, and EDX images for (a-c) Ru-5Co/TiO₂, (d-f) 5Co-Ru/TiO₂, and (Ru-5Co)/TiO₂, respectively

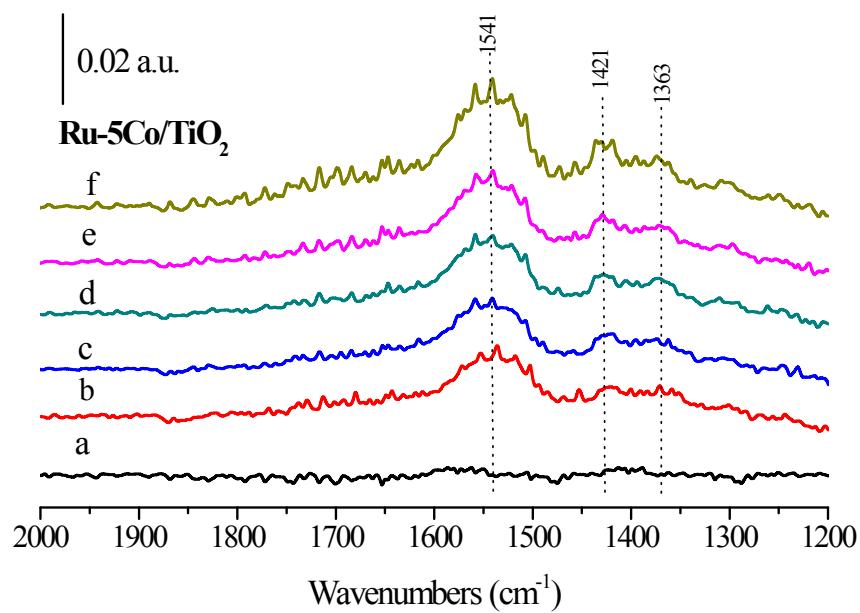


Fig. S3. *In-situ* FTIR spectra for Ru-5Co/TiO₂ collected at 240 °C after (a) 1, (b) 3, (c) 10, (d) 20, (e) 30, (f) 60 min on 500 ppm benzene/Ar stream.