

Electronic supplementary information for

**Double-active site synergistic catalysis in Ru-TiO₂ toward benzene
hydrogenation to cyclohexene with largely enhanced selectivity**

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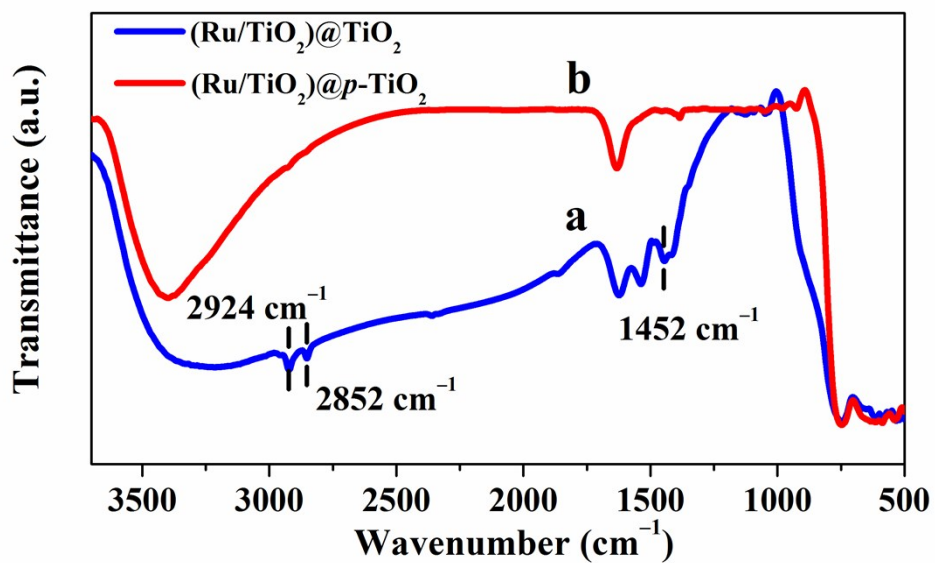


Fig. S1 FTIR spectra of (a) (Ru/TiO₂)@TiO₂ and (b) (Ru/TiO₂)@*p*-TiO₂.

Table S1. Catalytic performance of various catalysts

Catalysts	Additive	Conv. (%)	S_{CHE} (%)	Y_{CHE} (%)	Ref.
Ru/TiO ₂	-	73.1	32.6	23.8	This work
(Ru/TiO ₂)@ <i>p</i> -TiO ₂	-	98.1	76.6	75.1	This work
Ru/ZnO-ZrO _x (OH) _y	-	77.5	72.3	56.0	[1]
Ru _{1.0} Cu _{0.5} /MgAl-LDH	-	71.8	61.3	44.0	[2]
RuCoB/ γ -Al ₂ O ₃	-	62.7	45.7	28.8	[3]
Ru-Zn/HfO ₂	ZnSO ₄	60.0	82.8	49.7	[4]
Ru-La-B/ZrO ₂	ZnSO ₄	80.8	66.5	53.7	[5]
Ru/BEN (IP-HY)	ZnSO ₄	61.5	45.2	27.8	[6]
Ru-Zn/HAP-1	ZnSO ₄	69.8	47.3	33.0	[7]
Ru/B-ZrO ₂	ZnSO ₄	80.0	60.0	48.0	[8]
Ru/P25	ZnSO ₄	88.0	69.0	61.0	[9]

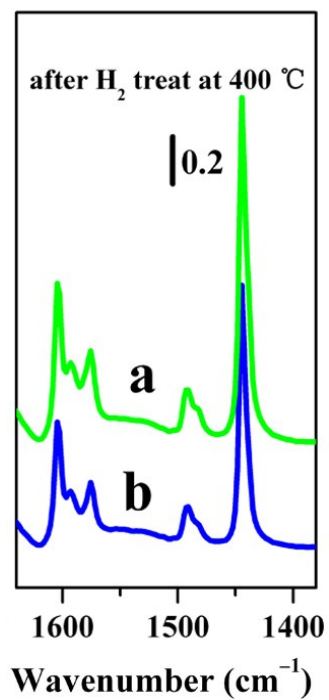


Fig. S2 Py-IR spectra of: (a) Ru/TiO₂ and (b) (Ru/TiO₂)@*p*-TiO₂ after H₂ treatment at 400 °C.

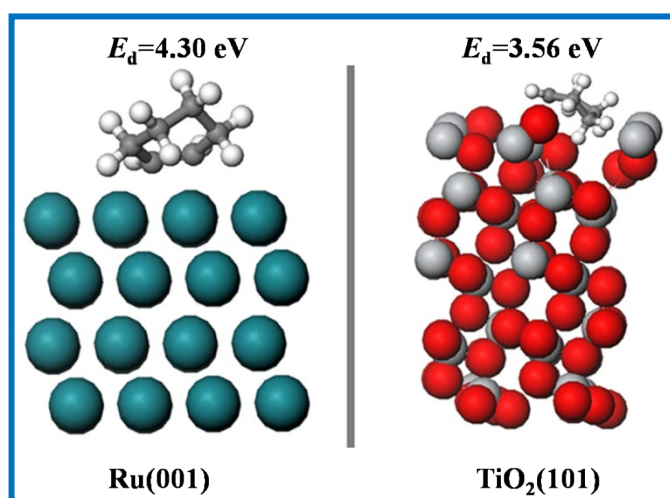


Fig. S3 Calculated adsorption models of cyclohexene on the typical Ru(001) and TiO₂(101) facet, respectively.

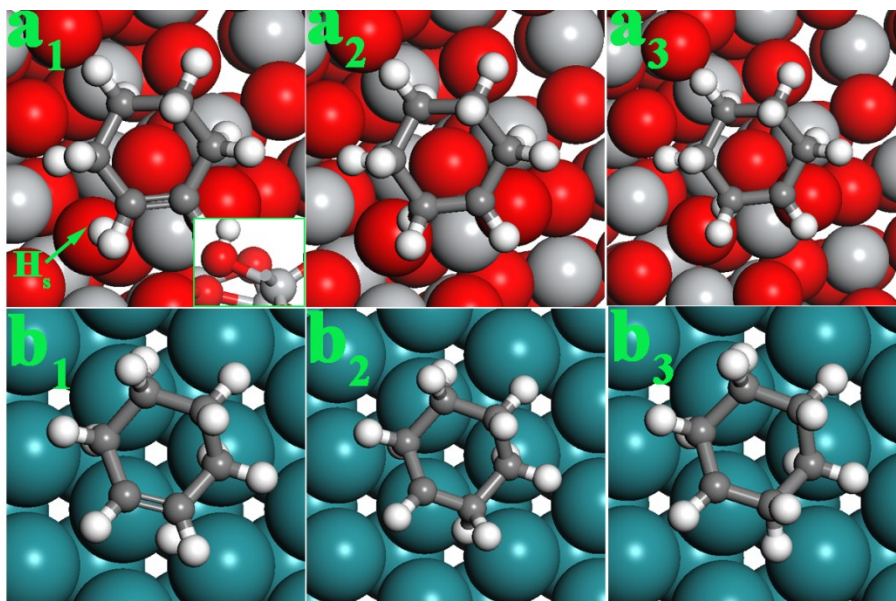


Fig. S4 Calculated hydrogenation models of cyclohexene with one hydrogen atom to C_6H_{11} over (a) $TiO_2(101)$ facet and (b) $Ru(001)$ facet, respectively. (a_1 , b_1) Initial states, (a_2 , b_2) transient states and (a_3 , b_3) final states. Pale gray, red, blue balls are Ti, O and Ru, respectively.

References:

- [1] H. Liu, T. Jiang, B. Han, S. Liang, W. Wang, T. Wu and G. Yang, *Green Chem.*, 2011, **13**, 1106–1109.
- [2] J. Liu, S. Xu, W. Bing, F. Wang, C. Li, M. Wei, D. G. Evans and X. Duan, *ChemCatChem*, 2015, **7**, 846–855.
- [3] G. Y. Fan, W. D. Jiang, J. B. Wang, R. X. Li, H. Chen and X. J. Li, *Catal. Commun.*, **2008**, **10**, 98–102.
- [4] N. Hajime and K. Mitsuo, *U. S. Pat.* 1988, 4734536.
- [5] S. Liu, Z. Liu, Z. Wang, S. Zhao and Y. Wu, *Appl. Catal., A: General*, 2006, **313**, 49–57.
- [6] W. Wang, H. Liu, T. Wu, P. Zhang, G. Ding, S. Liang, T. Jiang and B. Han, *J. Mol. Catal. A: Chem.*, 2012, **355**, 174–179.
- [7] P. Zhang, T. Wu, T. Jiang, W. Wang, H. Liu, H. Fan, Z. Zhang and B. Han, *Green Chem.*, 2013, **15**, 152–159.
- [8] G. Zhou, Y. Pei, Z. Jiang, K. Fan, M. Qiao, B. Sun and B. Zong, *J. Catal.*, 2014, **311**, 393–403.
- [9] G. Zhou, R. Dou, H. Bi, S. Xie, Y. Pei, K. Fan, M. Qiao, B. Sun and B. Zong, *J. Catal.*, 2015, **332**, 119–126.