

**Electronic Supplementary Information for the paper**

**Orbitally induced hierarchy of exchange interactions in zigzag antiferromagnetic state of honeycomb silver delafossite  $\text{Ag}_3\text{Co}_2\text{SbO}_6$**

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Table S1. Known  $\text{Ag}_3\text{M}_2\text{RO}_6$  and  $\text{Ag}_3\text{MRXO}_6$  delafossite-related mixed oxides

Only  $\text{Ag}_2\text{SnO}_3$  [42, 43] has been prepared by direct synthesis and characterized by single-crystal diffraction. All other compounds have been prepared by ion exchange from the sodium or lithium counterparts; they exhibit very similar powder diffraction patterns typical of the rhombohedral  $\text{AgFeO}_2$  type; almost all show one or two weak and diffuse superstructure reflection; however, interpretations of these patterns are different.

Composition	Comments
$\text{Ag}_3(\text{AgSn}_2)\text{O}_6$ [42, 43]	The basic structure is double-layered $\text{P6}_322$ honeycomb type, but there is an incommensurate modulation with $a = 29.22 \text{ \AA}$ .
$\text{Ag}_3\text{LiTi}_2\text{O}_6$ [44]	Powder pattern indexed as a trigonal superstructure. No structure refinement.
$\text{Ag}_3\text{M}_2\text{SbO}_6$ (M = Ni, Zn) [31]	No superlattice reflections found. The structures were refined within disordered $\text{Ag}(\text{M}_{2/3}\text{Sb}_{1/3})\text{O}_2$ $\text{R}\bar{3}m$ model, although the authors assumed high degree of local order.
$\text{Ag}_3\text{LiM}_2\text{O}_6$ (M = Ti, Sn) [45]	No indexing; disordered $\text{Ag}(\text{Li}_{1/3}\text{M}_{2/3})\text{O}_2$ formulas were used although one or two superlattice reflections were visible.
$\text{Ag}_3\text{Co}_2\text{SbO}_6$ [46]	The superstructure was refined within the trigonal $\text{P3}_112$ model. However, accuracy was low, and monoclinic symmetry could not be excluded as discussed in the present paper.
$\text{Ag}_3\text{LiRu}_2\text{O}_6$ [47]	Paradoxically, superstructure was refined omitting the unique superstructure reflection; monoclinic $\text{C2}/m$ model was used although no splitting of reflections from the rhombohedral subcell was detected. It was assumed that each layer was ordered but their stacking was disordered.
$\text{Ag}_3\text{LiRu}_2\text{O}_6$ [48]	No superlattice reflections found. The structure was refined within disordered $\text{Ag}(\text{Li}_{1/3}\text{Ru}_{2/3})\text{O}_2$ $\text{R}\bar{3}m$ model. Lattice parameters differ significantly (by 2.0-2.4%) from the subcell parameters in the preceding work [42].
$\text{Ag}_3\text{LiM}_2\text{O}_6$ (M = Rh, Ir) [49]	Superlattice reflections were found in both X-ray and electron diffraction. Each layer is assumed to be fully ordered. However, the structures were refined within disordered $\text{Ag}(\text{Li}_{1/3}\text{M}_{2/3})\text{O}_2$ $\text{R}\bar{3}m$ model.
$\text{Ag}_3\text{Ni}_2\text{BiO}_6$ [50]	Indexed as a trigonal superstructure ( $\text{P3}_112$ ). No structure refinement.
$\text{Ag}_3\text{NaFeSbO}_6$ [51]	Indexed as a trigonal superstructure ( $\text{P3}_112$ ). Structural model was depicted but not refined.
$\text{Ag}_3\text{LiMTeO}_6$ (M = Co, Ni, Zn) [52], $\text{Ag}_3\text{LiMnSbO}_6$ [53]	Superlattice reflections are visible but not discussed. No indexing, no lattice parameters.
$\text{Ag}_3\text{LiMSbO}_6$ (M = Al, Cr, Fe, Ga) [53]	Indexed as a trigonal superstructure ( $\text{P3}_112$ ). No structure refinement.

Table S2. Experimental and refinement details for  $\text{Ag}_3\text{M}_2\text{SbO}_6$  (M = Co, Zn)

Formula	$\text{Ag}_3\text{Co}_2\text{SbO}_6$	$\text{Ag}_3\text{Zn}_2\text{SbO}_6$
Formula weight	659.2	672.1
Crystal system	monoclinic	monoclinic
Space group	C2/m	C2/m
Z	2	2
a, Å	5.3770(13)	5.3829(2)
b, Å	9.3118(22)	9.3102(3)
c, Å	6.4810(14)	6.5046(2)
$\beta$ , °	106.512(7)	106.306(4)
V, Å <sup>3</sup>	311.12(2)	312.87(2)
Sample preparation	Amorphous admixture (beryllium carbonate or coffee) to reduce texture	
Diffractometer	Rigaku D/max-RC	ARL X'tra
Diffraction geometry	Bragg-Brentano	
Wavelength selection	Secondary-beam monochromator	Solid-state Si(Li) detector
Wavelength (CuK $\alpha$ ), Å	1.5406, 1.5444	
U, kV	55	40
I, mA	180	40
Receiving slit, mm	0.3	0.4
Angular range, °	10–100	12–101.4
Step size, °	0.02	0.02
Count time, s	3	2.4
Number of data points	4500	4470
Number of hkl	179	179
Number of variables	55	57
$R_{\text{wp}}$	0.0769	0.0605
$R_{\text{exp}}$	0.0276	0.0423
$R_{\text{F}}^2$	0.0198	0.0228
$\chi^2$	7.830	2.067

Table S3. Atomic coordinates, site occupancies and thermal displacement parameters for  $\text{Ag}_3\text{M}_2\text{SbO}_6$  (M = Co, Zn)

Atom	$\text{Ag}_3\text{Co}_2\text{SbO}_6$					$\text{Ag}_3\text{Zn}_2\text{SbO}_6$				
	x	y	z	sof	Uiso	x	y	z	sof	Uiso
Sb1	0	0	0	0.458(23)	0.016(4)	0	0	0	0.358(7)	0.0203(9)
M1	0	0	0	0.542(23)	0.016(4)	0	0	0	0.642(7)	0.0203(9)
M2	0	0.6661(10)	0	0.729(11)	0.022(4)	0	0.6688(4)	0	0.679(3)	0.0251(5)
Sb2	0	0.6661(10)	0	0.271(11)	0.022(4)	0	0.6688(4)	0	0.321(3)	0.0251(5)
Ag1	0	1/2	1/2	0.962(12)	0.016(3)	0	1/2	1/2	0.977(3)	0.0131(5)
Ag2	1/2	0.3426(6)	1/2	0.927(7)	0.006(3)	1/2	0.3413(3)	1/2	0.9423(16)	0.0183(3)
O1	0.618(5)	0.364(2)	0.827(4)	1	0.01	0.598(3)	0.3695(6)	0.8258(10)	1	0.01
O2	0.833(7)	1/2	0.173(8)	1	0.01	0.846(3)	1/2	0.1736(18)	1	0.01

Comments. The authors assume that the reported M/Sb mixing is fictitious due to multiple stacking faults whereas each individual layer is essentially ordered with site 1 occupied by Sb only and site 2, by  $\text{M}^{2+}$  only. Displacement parameters for O1 and O2 were fixed to prevent instability and negative values during the structure refinement process.

Table S4. Important interatomic distances (Å) and bond angles (°) in  $\text{Ag}_3\text{M}_2\text{SbO}_6$  (M = Co, Zn)

	$\text{Ag}_3\text{Co}_2\text{SbO}_6$	$\text{Ag}_3\text{Zn}_2\text{SbO}_6$
M1-O2	1.82(3)×2	1.888(13)×2
-O1	1.92(3)×4	1.837(9)×4
average	1.88	1.85
M2-O1	2.06(2)×2	2.171(12)×2
-O2	2.24(4)×2	2.227(11)×2
-O1	2.34(2)×2	2.322(8)×2
Average	2.21	2.24
Ag1-O2	2.05(5)×2	2.051(11)×2
Ag2-O1	2.04(2)×2	2.052(6)×2
O2-Ag1-O2	180.0	180.0
O1-Ag2-O1	168.8(10)	165.3(2)
M1-O1-M2	102.5 (11)	101.7(4)
M1-O1-M2	93.5(7)	95.0(2)
M2-O1-M2	89.8(9)	86.7(3)
M1-O2-M2	98.94(2)×2	98.159(2)×2
M2-O2-M2	87.22(1)	89.791(1)

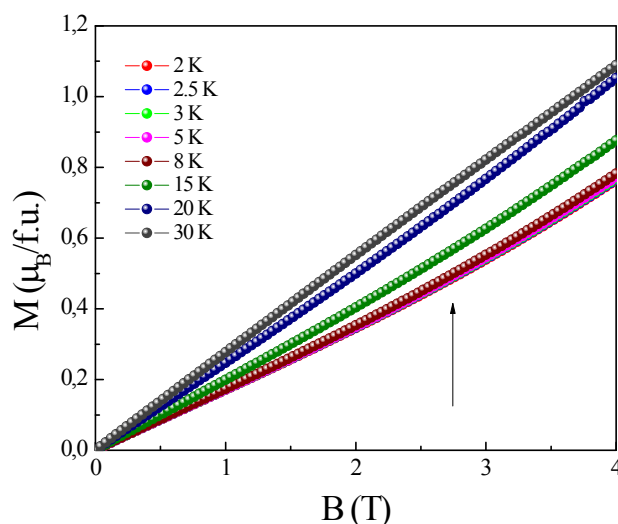


Fig. S1. The magnetization curves for  $\text{Ag}_3\text{Co}_2\text{SbO}_6$  at various temperatures.

#### References

- [31] R. Nagarajan, S. Uma, M.K. Jayaraj, J. Tate, A.W. Sleight. *Solid State Sci.* **4**, 787 (2002).
- [42] C. Linke, M. Jansen. *Z. anorg. allgem. Chem.* **623**, 1441 (1997).
- [43] T. Oku, A. Carlsson, J.-O. Bovin, C. Svensson, L.R. Wallenberg, C. Linke, M. Jansen. *Acta Cryst. B* **56**, 363 (2000).
- [44] V.B. Nalbandyan. *Russ. J. Inorg. Chem.* **45**, 1652 (2000).
- [45] Y. Hosogi, H. Kato, A. Kudo. *J. Mater. Chem.* **18**, 647 (2008).
- [46] V.V. Politaev, V.B. Nalbandyan, A.A. Petrenko, I.L. Shukaev, V.A. Volotchaev, B.S. Medvedev. *J. Solid State Chem.* **183**, 684 (2010).
- [47] S.A.J. Kimber, C.D. Ling, D.J.P. Morris, A. Chemseddine, P.F. Henry and D.N. Argyriou, *J. Mater. Chem.* **20**, 8021 (2010).
- [48] K. Ramesha, A.S. Prakash, M. Sathiya, G. Madras, A.K. Shukla. *Mater. Sci. Eng. B* **176**, 141 (2011).
- [49] V. Todorova, A. Leineweber, L. Kienle, V. Duppel, M. Jansen. *J. Solid State Chem.* **184**, 1112 (2011).
- [50] R. Berthelot, W. Schmidt, S. Muir, J. Eilertsen, L. Etienne, A.W. Sleight, M.A. Subramanian, *Inorg. Chem.* **51**, 5377 (2012).
- [51] V.V. Politaev, V.B. Nalbandyan. *Solid State Sci.* **11**, 144 (2009).
- [52] V. Kumar, N. Bhardwaj, N. Tomar, V. Thakral, S. Uma, *Inorg. Chem.* **51**, 10471 (2012).
- [53] N. Bhardwaj, A. Gupta, S. Uma. *Dalton Trans.* **43**, 12050 (2014).