

Supporting Information for:

Probing Bistability in Fe^{II} and Co^{II} Complexes with an Unsymmetrically Substituted Quinonoid Ligand

Margarethe van der Meer,^{a,b} Yvonne Rechkemmer,^{c,b} Frauke D. Breitgoff,^c Sebastian Dechert,^d Raphael Marx,^c María Dörfel,^c Petr Neugebauer,^c Joris van Slageren,^{c,*} and Biprajit Sarkar^{a,*}

^a Institut für Chemie und Biochemie, Freie Universität Berlin, Fabeckstraße 34-36, D-14195 Berlin, Germany. Email: biprajit.sarkar@fu-berlin.de

^c Institut für Physikalische Chemie, Universität Stuttgart, Pfaffenwaldring 55, D-70569 Stuttgart, Germany. Email: ipcjosl@ipc.uni-stuttgart.de

^d Institut für Anorganische Chemie, Georg August Universität Göttingen, Tammannstraße 4, D-37077 Göttingen, Germany.

^b Both authors contributed equally to this work.

X-ray**Table S1.** Crystal data and refinement details for **[1]**BF₄·2CH₂Cl₂ and **[2]**BF₄·2CH₂Cl₂.

compound	[1] BF ₄ ·2 CH ₂ Cl ₂	[2] BF ₄ ·2CH ₂ Cl ₂
empirical formula	C ₃₈ H ₄₀ BCl ₆ F ₄ FeN ₅ O ₃	C ₃₈ H ₄₀ BCl ₆ CoF ₄ N ₅ O ₃
formula weight	970.11	973.19
<i>T</i> [K]	133(2)	140(2)
crystal size [mm ³]	0.50 × 0.33 × 0.17	0.18 × 0.10 × 0.08
crystal system	monoclinic	monoclinic
space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>
<i>a</i> [Å]	15.7289(6)	15.925(3)
<i>b</i> [Å]	15.3978(4)	15.360(3)
<i>c</i> [Å]	17.4980(7)	17.764(3)
α [°]	90	90
β [°]	94.213(3)	92.216(4)
γ [°]	90	90
<i>V</i> [Å ³]	4226.4(3)	4342.0(14)
<i>Z</i>	4	4
ρ [g/cm ³]	1.525	1.489
<i>F</i> (000)	1984	1988
μ [mm ⁻¹]	0.798	0.824
<i>T</i> _{min} / <i>T</i> _{max}	0.7432 / 0.8680	0.6637 / 0.7452
θ-range [°]	1.681 - 25.698	1.685 - 25.112
<i>hkl</i> -range	±19, -18 - 17, ±21	-18 - 17, ±18, ±21
measured refl.	44734	34967
unique refl. [<i>R</i> _{int}]	7959 [0.0541]	7707 [0.0492]
observed refl. (<i>I</i> > 2σ(<i>I</i>))	6992	5201
data / restraints / param.	7959 / 9 / 593	7707 / 8 / 580
goodness-of-fit (<i>F</i> ²)	1.112	1.059
<i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0695, 0.1725	0.0605, 0.1435
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0784, 0.1779	0.0946, 0.1629
resid. el. dens. [e/Å ³]	-0.930 / 0.894	-0.668 / 0.729

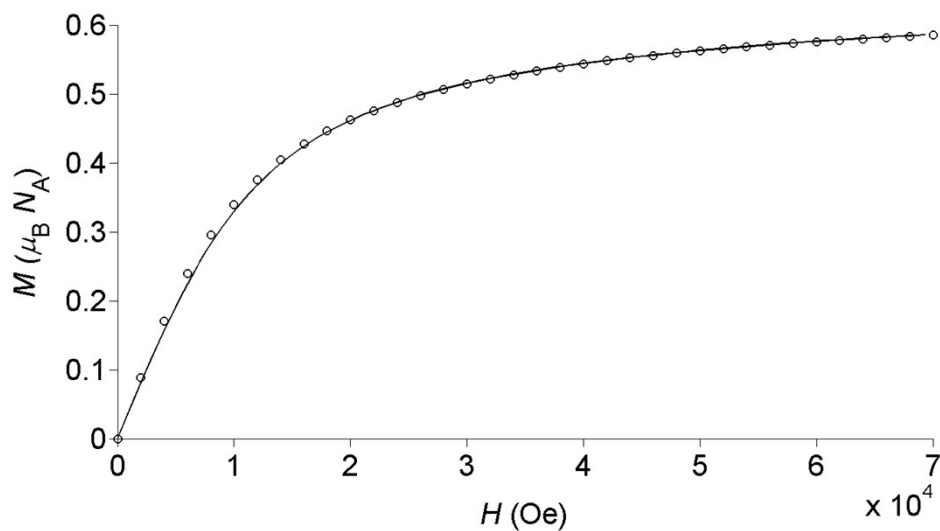


Fig. S1 Magnetization curve for **[1]BF₄** measured at 1.8 K. The best fit with the parameters given in the text is shown as a solid line.

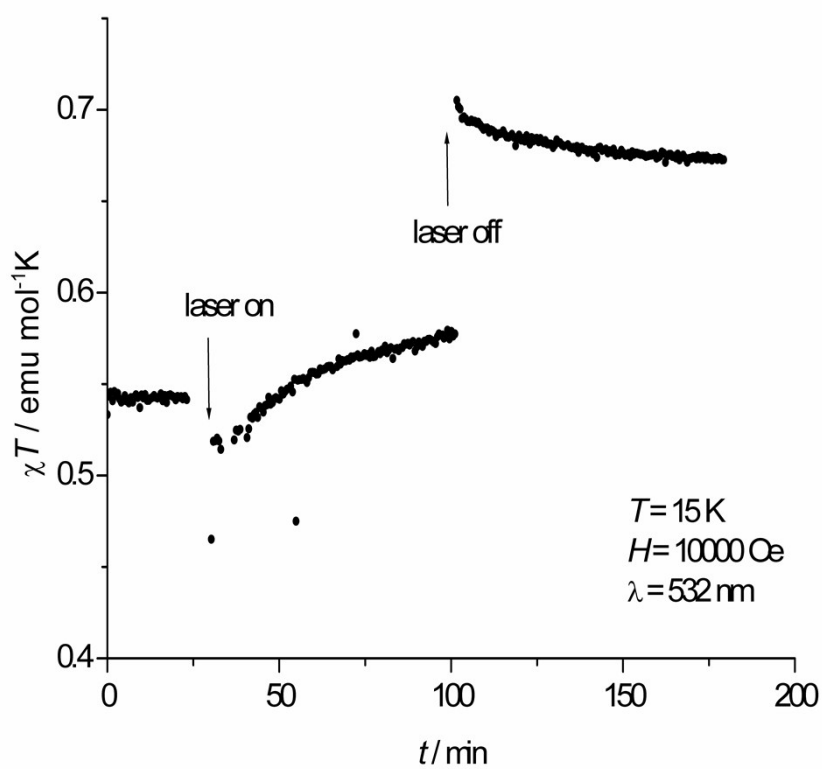


Fig. S2. Isothermal LIESST study of **[1]BF₄** at 15 K. The effect of sample heating due to laser irradiation can clearly be observed by jumps in χT on switching on and off the irradiation.

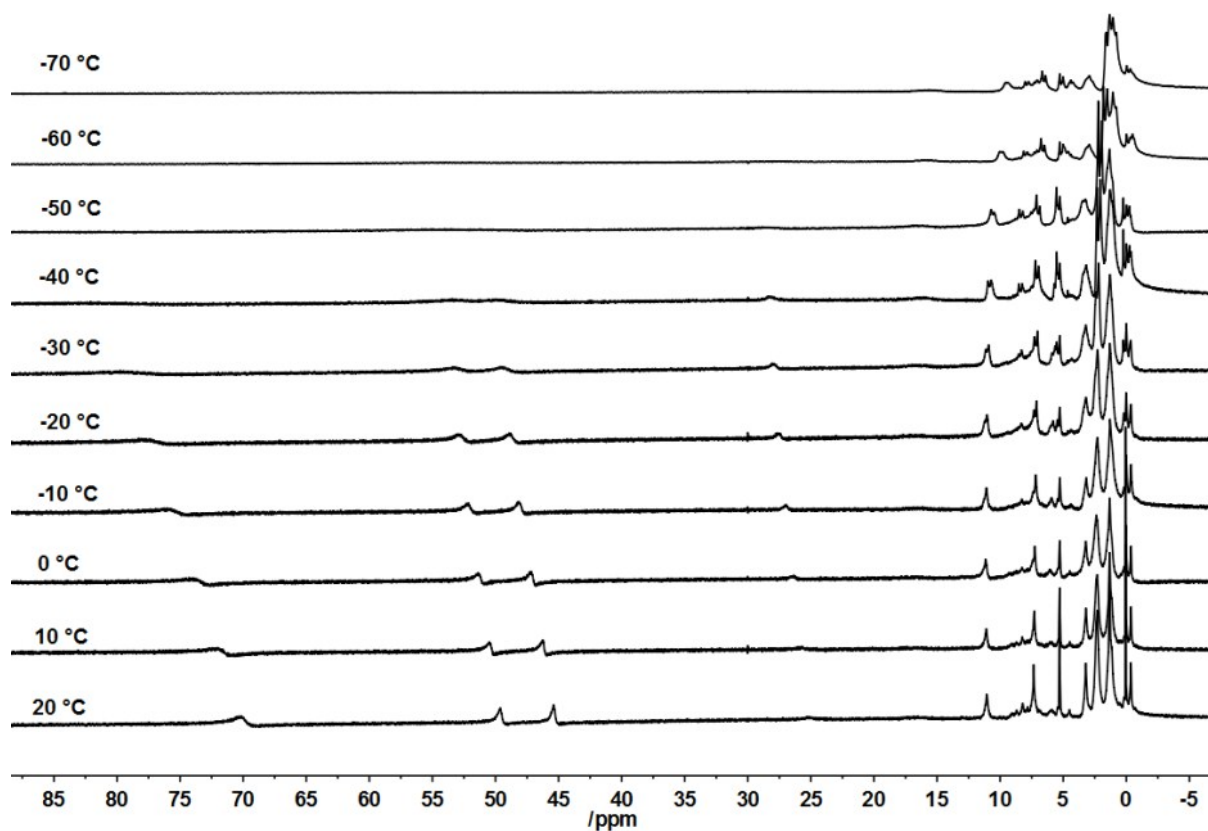


Fig. S3 ^1H -NMR of $[1]^+$ measured at $20\text{ }^\circ\text{C}$ to $-70\text{ }^\circ\text{C}$ in CD_2Cl_2 .

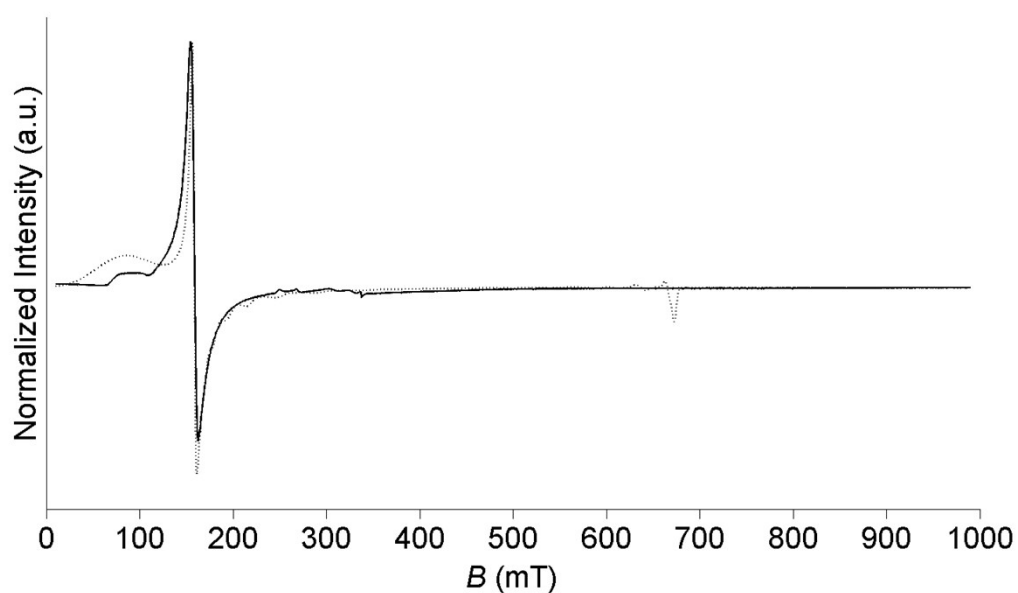


Fig. S4 X-band EPR measurement of $[1]\text{BF}_4$ recorded at 4 K . The best fit with the parameters given in the text is shown as a dotted line.

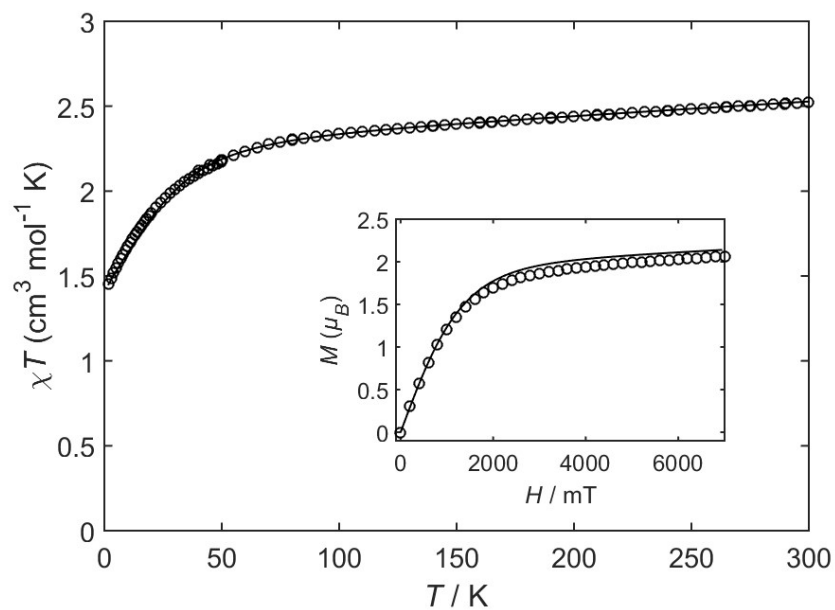


Fig. S5 χT as a function of T for the complex $[2]\text{BF}_4$ measured at 1000 Oe. Inset: Magnetization curve for $[2]\text{BF}_4$ measured at 1.8 K. Simulations with the parameters given in the text are shown as solid lines.

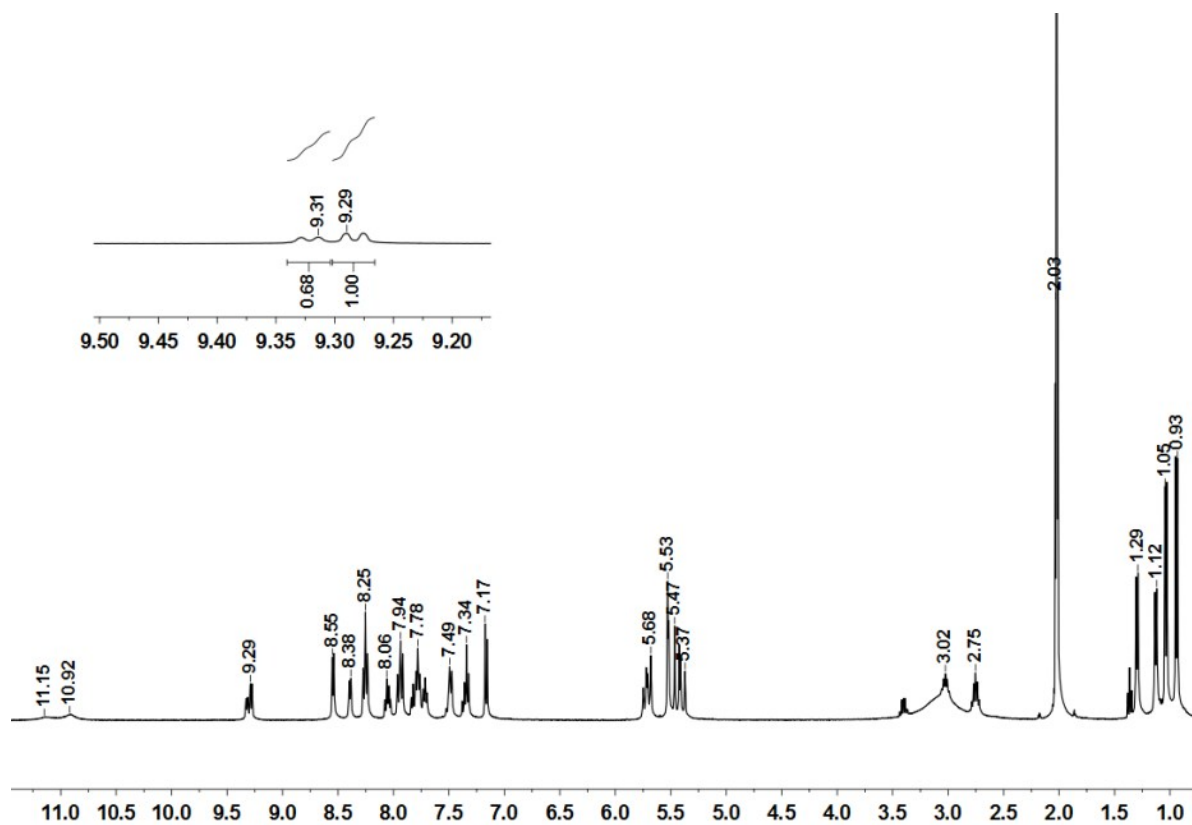


Fig. S6 ^1H NMR spectrum of $[2]^{2+}$ in acetone- d_6 measured after 12 h of dissolving the sample.

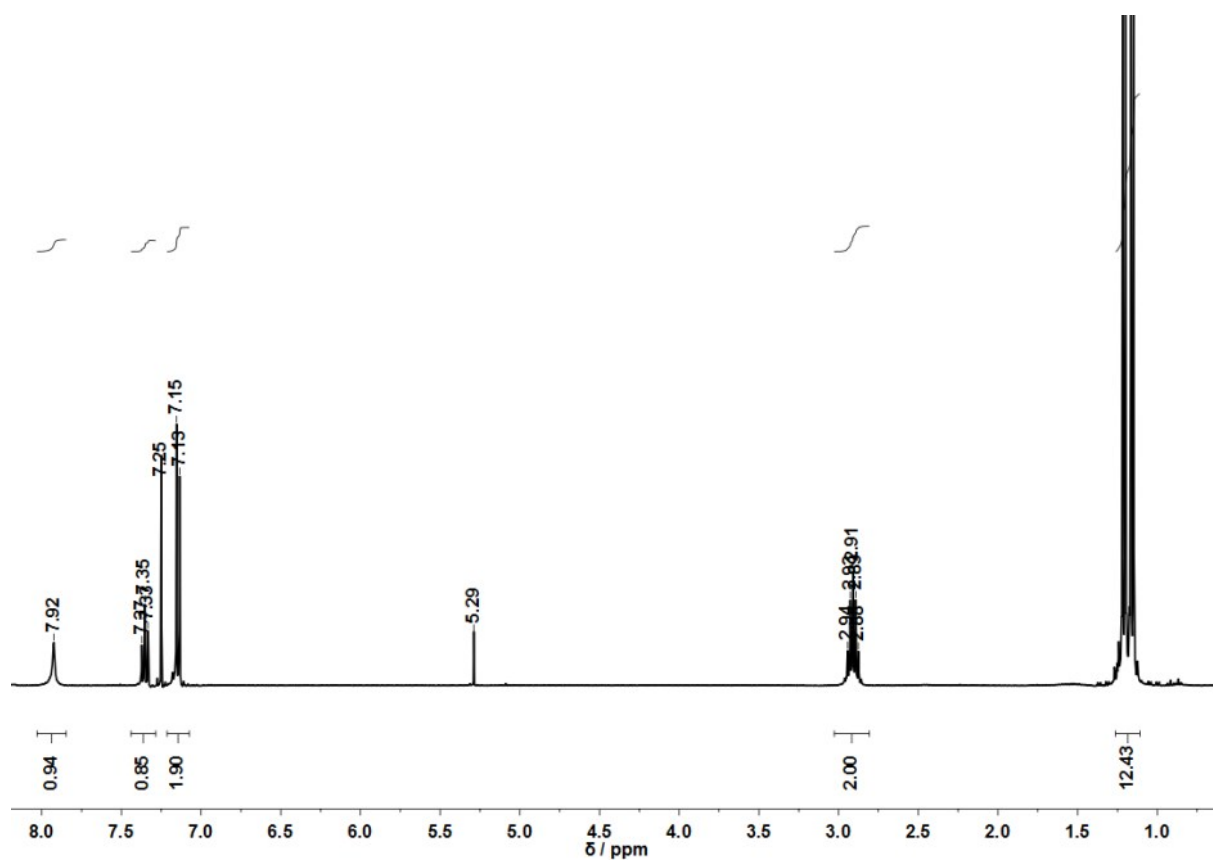


Fig. S7 ¹H NMR spectrum of **H₂L** in CDCl₃. Sample contains a residue of CH₂Cl₂.

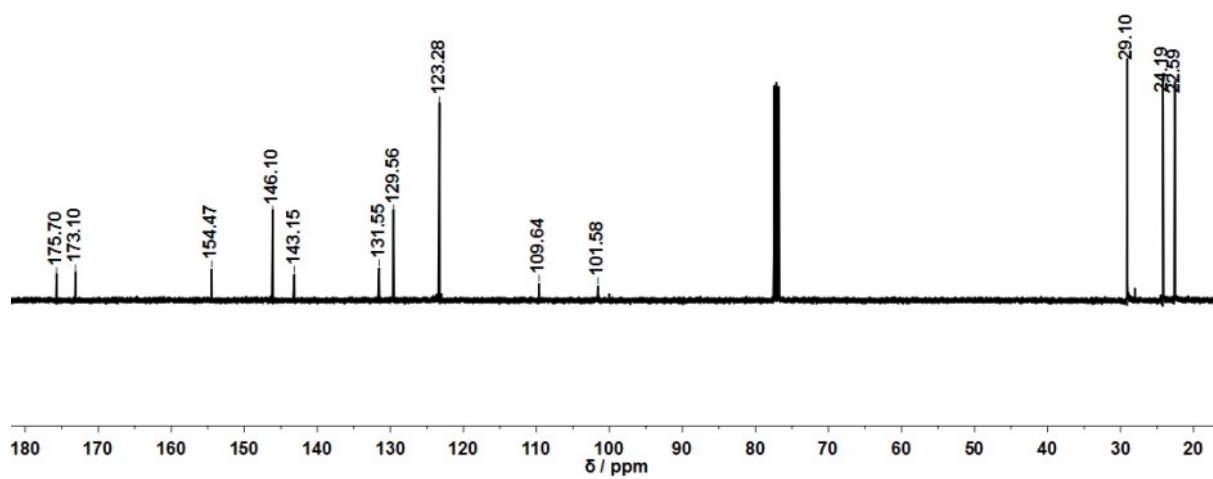


Fig. S8 ¹³C NMR spectrum of **H₂L** in CDCl₃.