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Supporting Information

 μ -Oxo- and Bis(μ -carboxylato)-bridged Diiron(III) Complexes of a 3N Ligand as Catalysts for Alkane Hydroxylation: Steroelectronic Factors of Carboxylate Bridge Determine the Catalytic Efficiency

Mani Balamurugan, a Eringathodi Suresh, b and Mallayan Palanianda vara*

^aSchool of Chemistry, Bharathidasan University, Tiruchirappalli - 620024, Tamil Nadu, India. ^bAnalytical Science Discipline, Central Salt and Marine Chemicals Research Institute, Bhavnagar - 364 002, India.

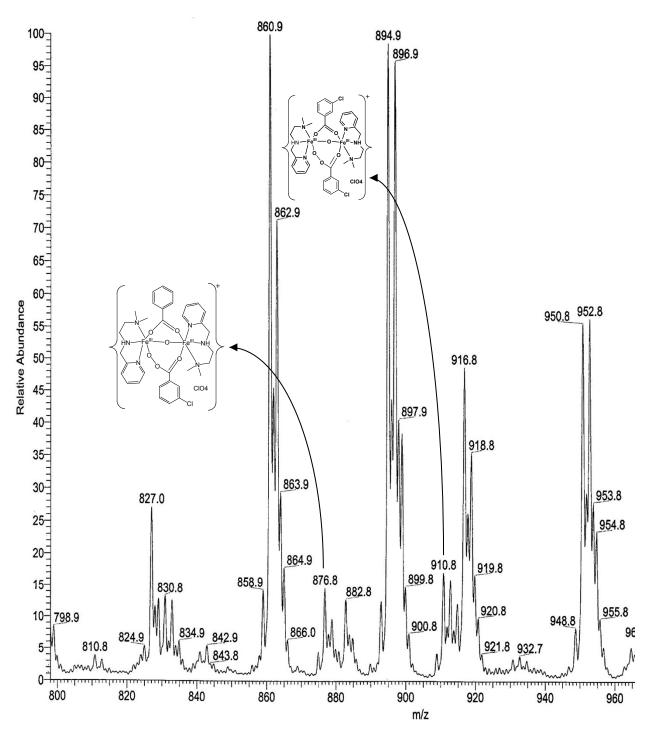


Figure S1. ESI-MS spectrum of the reaction of **4** with 5 eq. of *m*-CPBA and 1 eq. of TEA. Intense peak in the spectrum at m/z = 860.9, 894.8, 916.8 and 950.7 corresponding to $\{[Fe_2^{III}(O)(L)_2(OBz)_2]ClO_4\}^+$, $\{[Fe_2^{III}(O)(L)_2(OBz)_2](OBzCl)\}^+$, $\{[Fe_2^{III}(O)(L)_2(OBzCl)]^+$ and $\{[Fe_2^{III}(O)(L)_2(OBzCl)_2](OBzCl)\}^+$. Less intense peaks in the spectrum are assigned to the *m*-CPBA adducts $\{[Fe_2^{III}(O)(L)_2(OBz)(OOCOC_6H_4Cl)]ClO_4\}^+$ (m/z = 876.8) and $\{[Fe_2^{III}(O)(L)_2(OBzCl)(OOCOC_6H_4Cl)]ClO_4\}^+$.

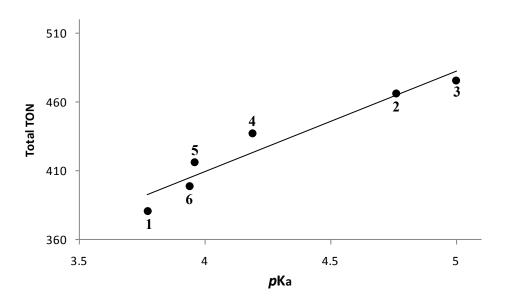


Figure S2.A linear correlation (R^2 , 0.93) between pK_a value of bridging carboxylates and total TON of diiron(III) complexes for adamantane oxidation

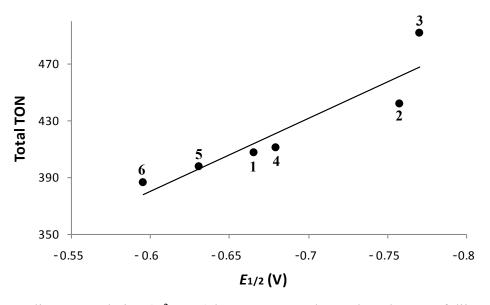


Figure S3. A linear correlation (\mathbb{R}^2 , 0.84) between $E_{1/2}$ value and total TON of diiron(III) complexes for cyclohexane oxidation.

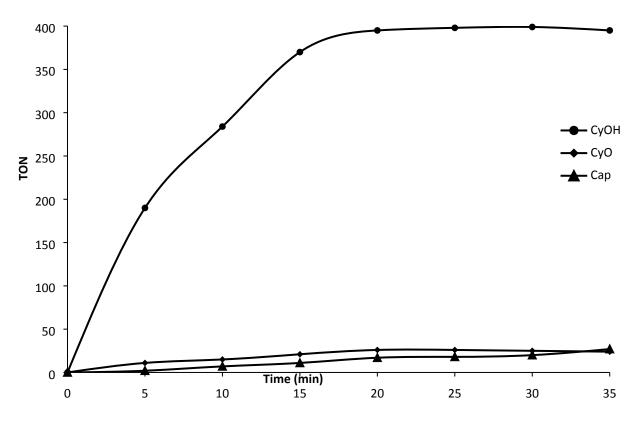


Figure S4. Time dependent oxidation of cyclohexane catalyzed by 2 with m-CPBA