

ELECTRONIC SUPPLEMENTARY INFORMATION (ESI)

To

Binding of oxime group to uranyl ion: Uranyl complexes with 2-pyridyl ketoximes as ligands

Sokratis T. Tsantis,^a Eirini Zagoraïou,^a Aikaterinh Savvidou,^b Catherine P. Raptopoulou,^b Vassilis Psycharis,^b Lukasz Szyrwił,^c Małgorzata Hołyńska*^d and Spyros P. Perlepes*^{a, e}

This work is dedicated to the loving memory of Malcolm H. Chisholm: A great scientist, a fantastic mentor and a precious friend.

^aDepartment of Chemistry, University, of Patras, 26504 Patras, Greece. E-mail: perlepes@patreas.upatras.gr; Tel: +30 2610 996730

^bInstitute of Nanoscience and Nanotechnology, NCSR "Demokritos", 153 10 Aghia Paraskevi Attikis, Greece

^cDepartment of Chemistry of Drugs, Wrocław Medical University, ul. Borowska 211, 50-556 Wrocław, Poland

^dFachbereich Chemie and Wissenschaftliches Zentrum für Materialwissenschaften, Philips-Universität Marburg, Hans-Meerwein-Strasse, D-35043 Marburg, Germany. E-mail: holynska@staff.uni-marburg.de

^eInstitute of Chemical Engineering Sciences, Foundation for Research and Technology-Hellas (FORTH/ICE-HT), Platani, P.O. Box 1414, 26504 Patras, Greece

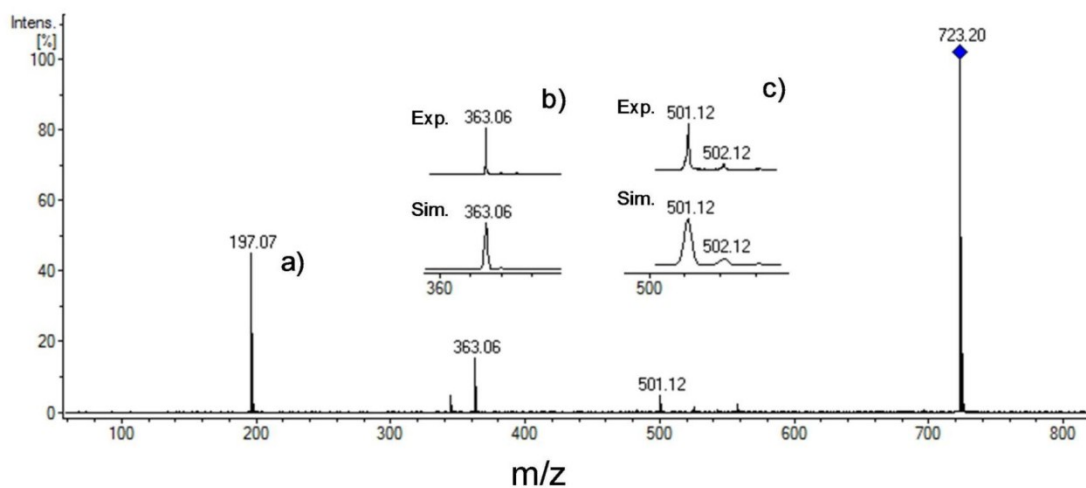


Fig. S1 Collision-induced fragmentation MS/MS spectra of $[\text{UO}_2(\text{O}_2\text{CMe})(\text{phpao})_2]^-$ ($\text{UO}_2\text{C}_{26}\text{H}_{21}\text{N}_4\text{O}_4$, $m/z = 723.20$): a) The phpao^- signal; b) $[\text{UO}_2(\text{OH})_2(\text{O}_2\text{CMe})]^-$ ($\text{UO}_2\text{C}_2\text{H}_5\text{O}_4$); c) $[\text{UO}_2(\text{OH})_2(\text{phpao})]^-$ ($\text{UO}_2\text{C}_{12}\text{H}_{11}\text{N}_2\text{O}_3$).

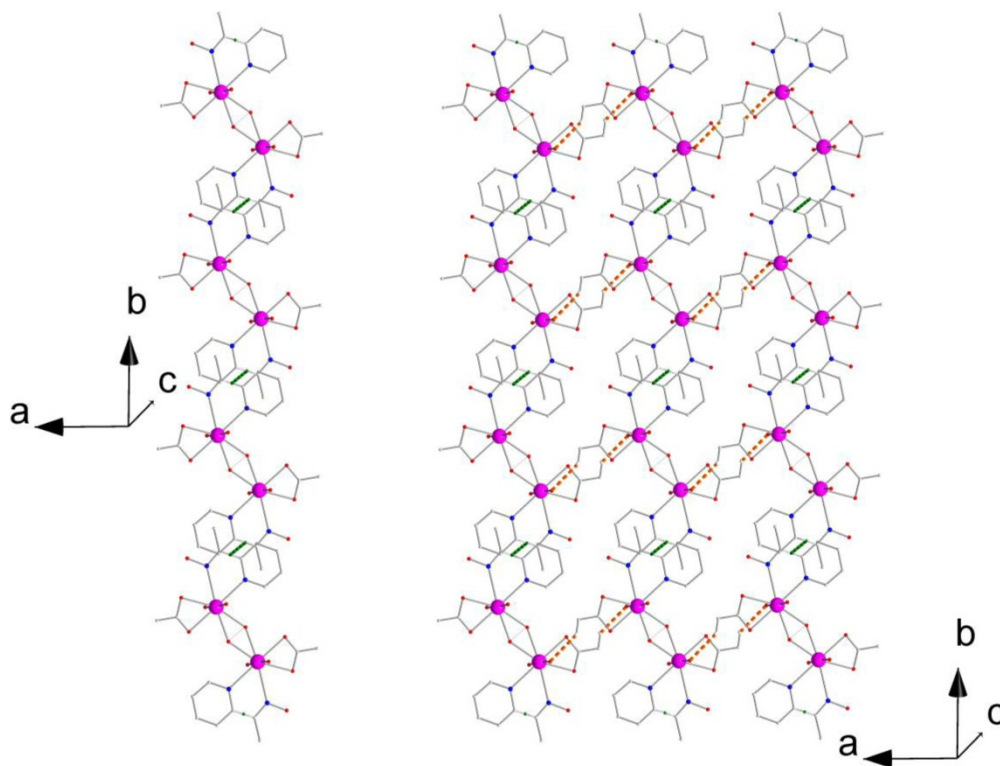


Fig. S2 Chains (left) along the b axis and 2D layers (right) parallel to the (001) plane for complex $[(\text{UO}_2)_2(\text{O}_2)(\text{O}_2\text{CMe})_2(\text{mepaoH})_2]$ (**3**). The dashed dark green and orange lines indicate π - π interactions and $\text{C}_{\text{acetate methyl}}\text{-H}\cdots\text{O}_{\text{uranyl}}$ H bonds.

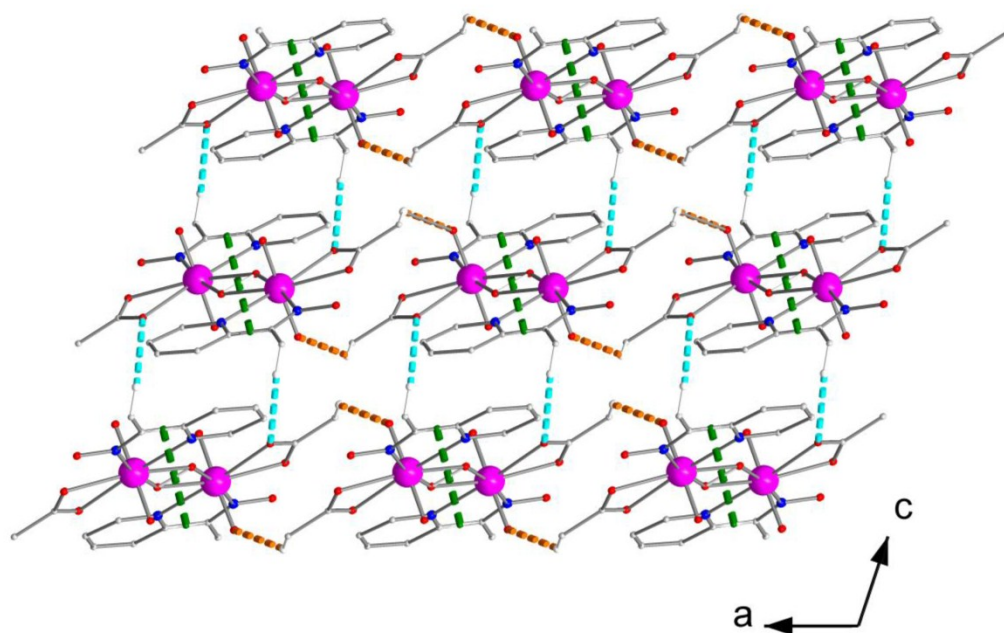


Fig. S3 A part of the 3D architecture in the crystal structure of complex $[(\text{UO}_2)_2(\text{O}_2)(\text{O}_2\text{CMe})_2(\text{mepaoH})_2]$ (**3**). The dashed dark green, turquoise and orange lines indicate π - π interactions, $\text{C}_{\text{oxime methyl}}\text{-H}\cdots\text{O}_{\text{coord.acetate}}$ and $\text{C}_{\text{acetate methyl}}\text{-H}\cdots\text{O}_{\text{uranyl}}$ H bonds, respectively.

Table S1. Intermolecular H bonds in the crystal structures of **1-3**

Interaction D-H \cdots A ^a	D \cdots A (Å)	H \cdots A (Å)	D-H \cdots A(°)	Symmetry operation for A
Complex 1				
O6-H(O6) \cdots O11	2.64	1.77	161.0	x, y, z
Complex 2				
C1 ^b -H1(C1) \cdots C _g 1	3.680(4)	3.17(5)	116(2)	$-x+1, y+1, -z+1$
Complex 3				
C1 ^c -HB(C1) \cdots O5	3.45	2.59	146.3	$x, -y+0.5, z+0.5$
C9 ^d -HC(C9) \cdots O3	3.28	2.55	131.9	$-x+2, -y, -z$

^a D=donor, A=acceptor; ^b C1 is the methyl carbon atom of mepao⁻, not labelled in Fig.4. ^c C1 is the methyl carbon of mepao⁻, not labelled in Fig.6. ^d C9 is the methyl carbon atom of MeCO₂⁻, not labelled in Fig.6.