

Electronic Supplementary Information (ESI)

Tricarboxylate-based Gd^{III} coordination polymers exhibiting large magnetocaloric effects

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Table S1. Selected bond lengths (Å) and angles (°) for **1^a**

O2—Gd1 ^{#1}	2.381(2)	Gd1—O3	2.401(2)
O1—Gd1 ^{#2}	2.364(2)	Gd1—O1W	2.411(2)
O4—Gd1 ^{#3}	2.321(2)	Gd1—O7	2.444(2)
Gd1—O6 ^{#4}	2.375(2)	Gd1—O5	2.462(2)
O4 ^{#5} —Gd1—O2 ^{#6}	148.13(7)	O1 ^{#2} —Gd1—O5	141.53(7)
O1 ^{#2} —Gd1—O2 ^{#6}	81.35(7)	O6 ^{#4} —Gd1—O5	73.55(7)
O6 ^{#4} —Gd1—O2 ^{#6}	79.67(7)	O2 ^{#6} —Gd1—O5	125.01(7)
O4 ^{#5} —Gd1—O3	89.08(8)	O3—Gd1—O5	71.10(7)
O1 ^{#2} —Gd1—O3	71.30(7)	O1W—Gd1—O5	137.95(7)
O6 ^{#4} —Gd1—O3	144.55(7)	O7—Gd1—O5	69.36(7)
O2 ^{#6} —Gd1—O3	119.41(7)	O3—Gd1—O1W	137.83(8)
O4 ^{#5} —Gd1—O1W	75.82(8)	O4 ^{#5} —Gd1—O7	140.90(7)
O1 ^{#2} —Gd1—O1W	71.56(8)	O1 ^{#2} —Gd1—O7	100.95(8)
O6 ^{#4} —Gd1—O1W	73.81(8)	O6 ^{#4} —Gd1—O7	100.74(7)
O2 ^{#6} —Gd1—O1W	73.28(8)	O2 ^{#6} —Gd1—O7	69.80(7)

^aSymmetry transformations used to generate equivalent atoms: #1: $x, -y+5/2, z-1/2$; #2: $-x+1, -y+2, -z+2$; #3: $x, -y+3/2, z-1/2$; #4: $-x, -y+2, -z+2$; #5: $x, -y+3/2, z+1/2$; #6: $x, -y+5/2, z+1/2$.

Table S2. Selected bond lengths (Å) and angles (°) for **2^a**

O2—Gd1	2.374(11)	Gd1—O5 ^{#5}	2.366(12)
Gd1—O6 ^{#4}	2.359(12)	Gd1—O3	2.379(12)
Gd1—N1	2.637(14)	Gd1—O1 ^{#6}	2.388(11)
Gd1—O4	2.419(11)	Gd1—O1W	2.418(11)
O6 ^{#4} —Gd1—O5 ^{#5}	75.6(4)	O5 ^{#5} —Gd1—O4	133.4(5)
O6 ^{#4} —Gd1—O2	86.9(4)	O2—Gd1—O4	128.9(4)
O5 ^{#5} —Gd1—O2	82.0(5)	O3—Gd1—O4	72.1(4)
O6 ^{#4} —Gd1—O3	138.8(4)	O1 ^{#6} —Gd1—O4	78.8(4)
O5 ^{#5} —Gd1—O3	145.6(4)	O1W—Gd1—O4	136.7(4)
O2—Gd1—O3	98.2(5)	O6 ^{#4} —Gd1—N1	76.0(4)
O6 ^{#4} —Gd1—O1 ^{#6}	105.4(4)	O5 ^{#5} —Gd1—N1	137.2(4)
O5 ^{#5} —Gd1—O1 ^{#6}	77.3(4)	O2—Gd1—N1	65.1(4)
O2—Gd1—O1 ^{#6}	152.3(4)	O3—Gd1—N1	69.5(4)
O3—Gd1—O1 ^{#6}	88.8(4)	O1 ^{#6} —Gd1—N1	141.5(4)
O6 ^{#4} —Gd1—O1W	148.5(4)	O1W—Gd1—N1	121.6(4)
O5 ^{#5} —Gd1—O1W	74.5(4)	O4—Gd1—N1	64.6(4)
O2—Gd1—O1W	79.4(4)	O3—Gd1—O1W	71.8(4)
O6 ^{#4} —Gd1—O4	73.1(4)	O1 ^{#6} —Gd1—O1W	77.4(4)

^aSymmetry transformations used to generate equivalent atoms: #1: $-x+1/2, y+1/2, -z+1/2$; #4: $-x+1/2, y-1/2, -z+1/2$; #5: $x, y-l, z$; #6: $x-l, y, z$.

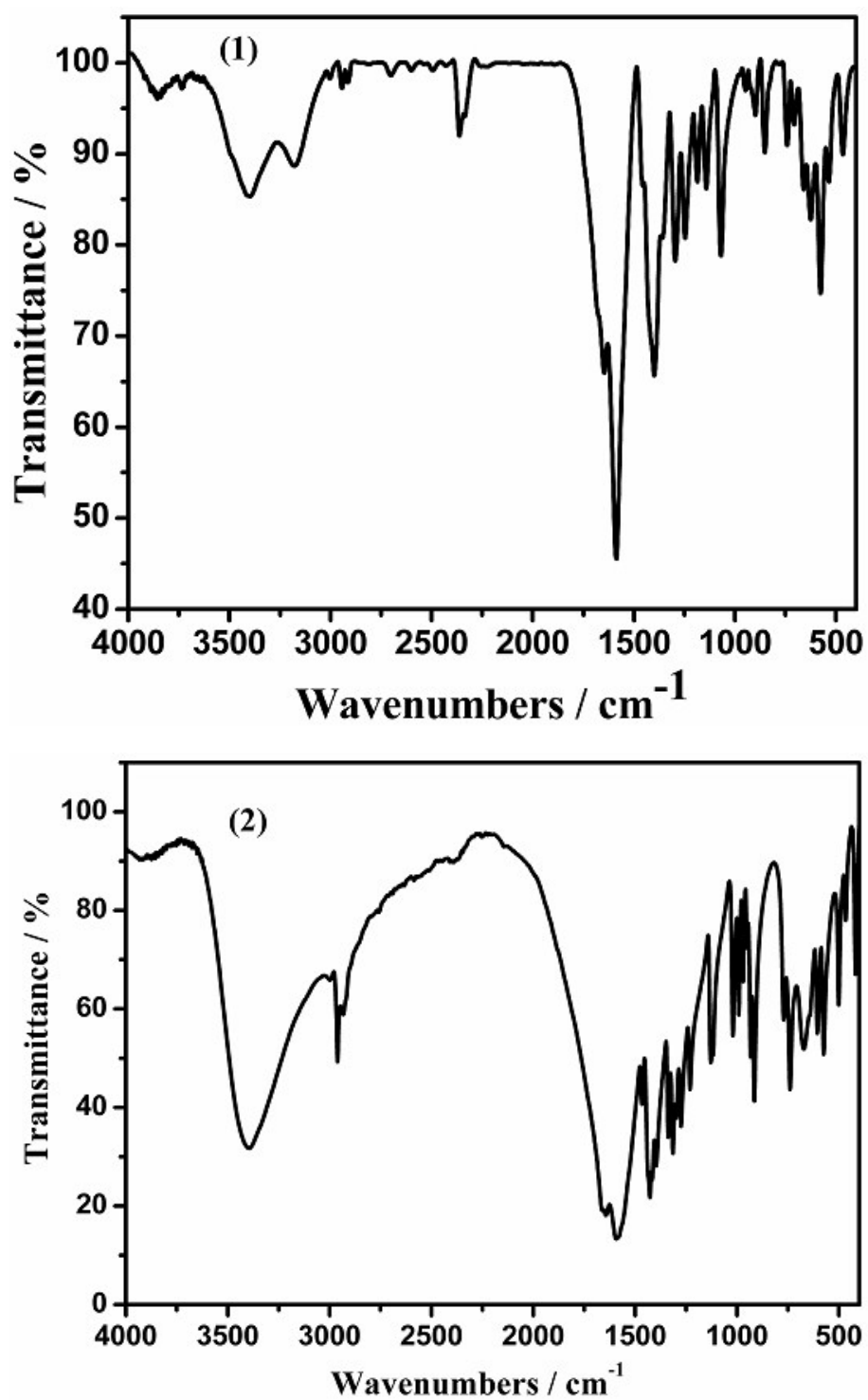
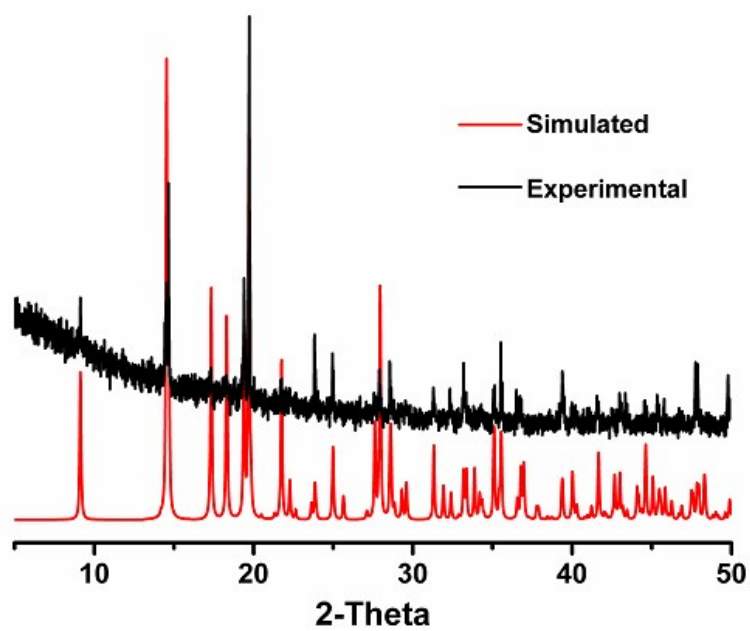
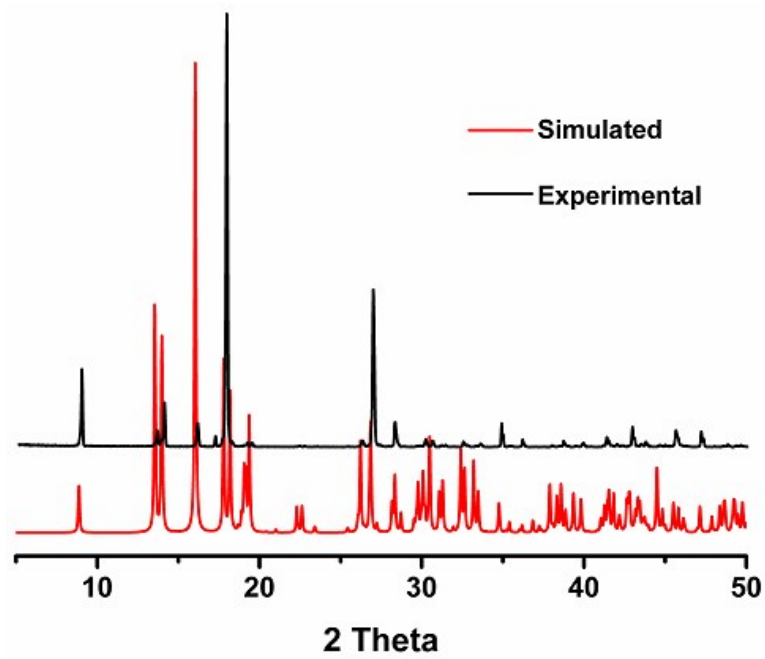


Fig. S1. The IR spectra of complexes 1 and 2.



(a)



(b)

Fig. S2. XRPD patterns of **1** (a) and **2** (b).

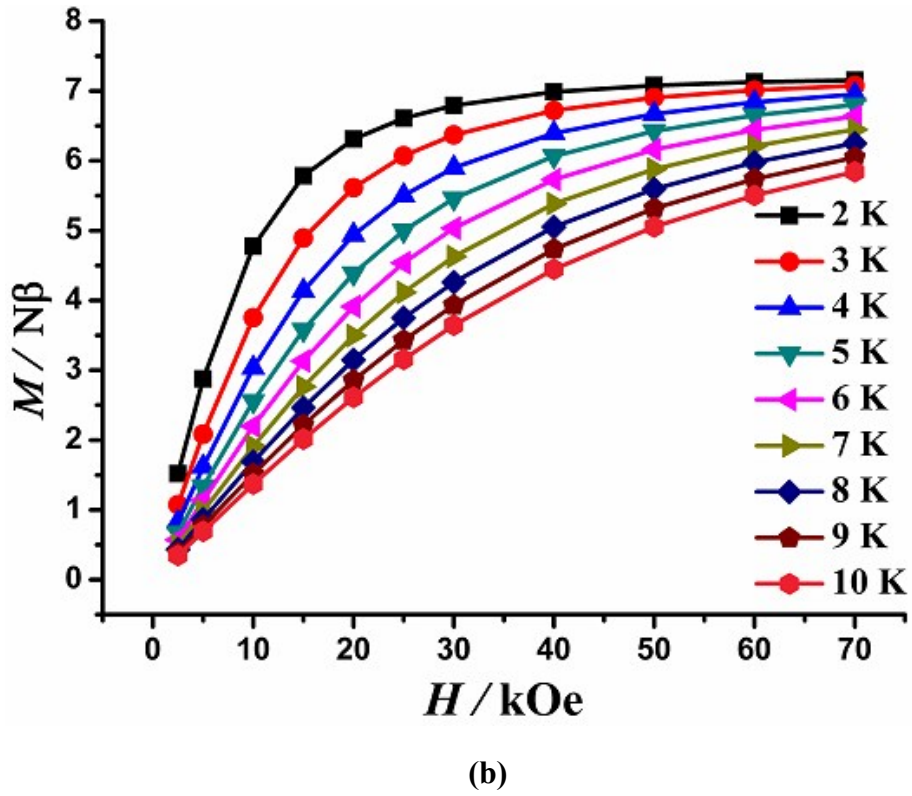
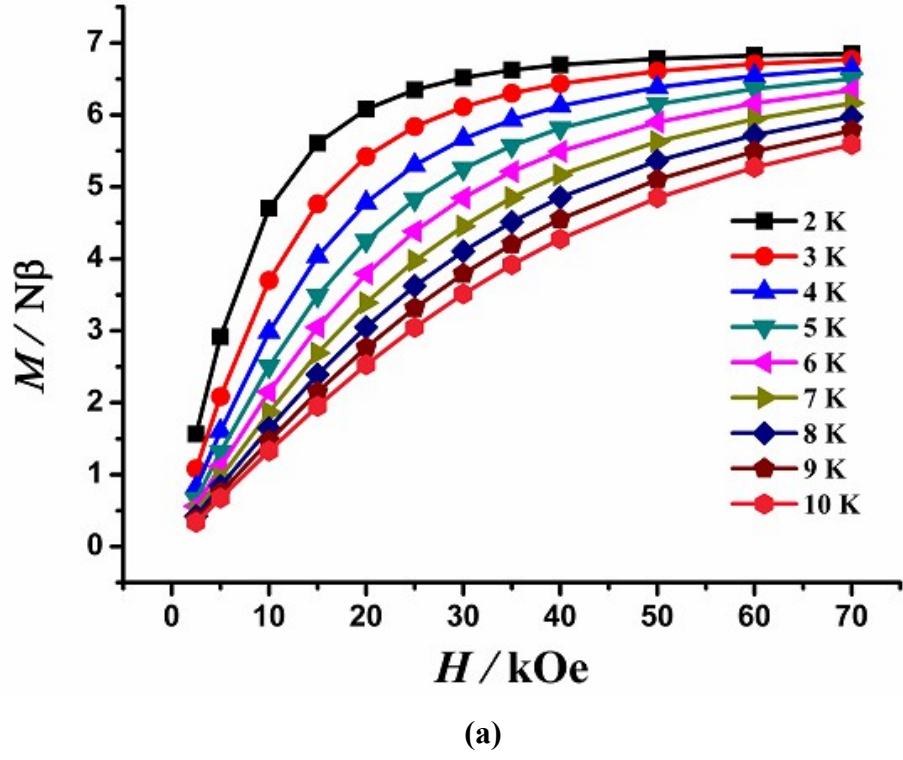


Fig. S3. The M vs. H plots of 1 (a) and 2 (b) at $T = 2\text{--}10$ K and $H = 2.5\text{--}70$ kOe.